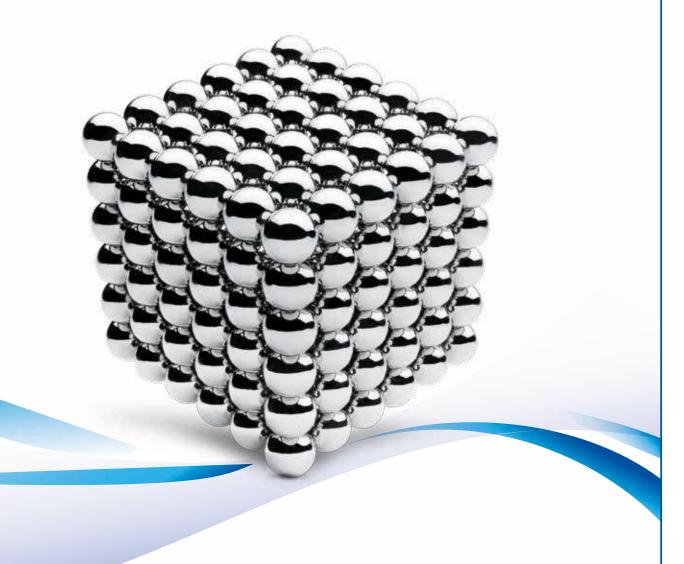


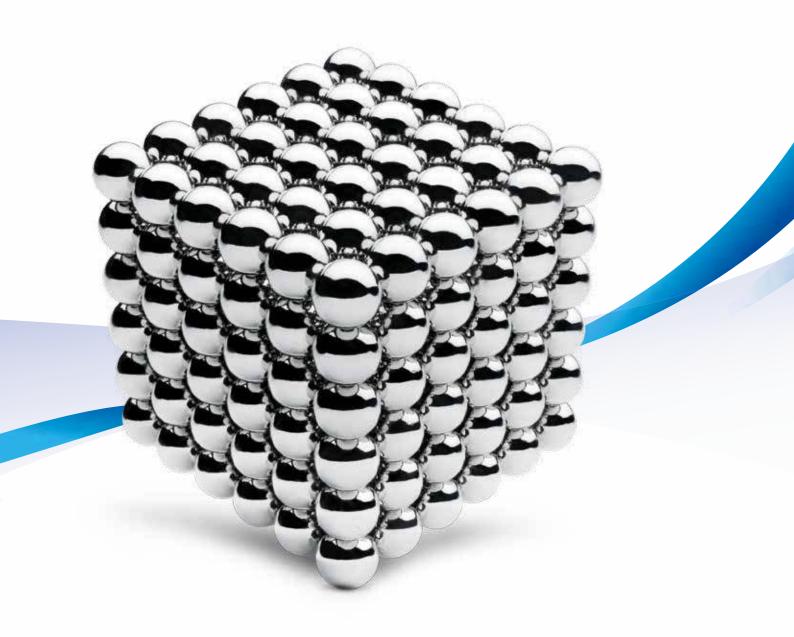
# Perfection in every dimension. Inorganic Reagents catalog





# Inorganic Reagents for laboratory use

Every customer has a different goal. We make reaching yours easier with a clear, helpful guide and straightforward directions. No matter what kind of information you seek, the Inorganic Reagents catalog will help you find exactly what you want – and quickly! Search by special product categories, the A-Z alphabetical listing (index) or the glossary.



### • Experience and innovation

The unique experience of the Merck own history enables us to turn our eyes on to the future and combine tradition with modernity to provide respectable and innovative products. Our extensive expertise and rich heritage enable us to deliver products that are both practical and revolutionary – a huge product range with globally renowned trademarks.

### • Quality, certification and accreditation

Outstanding quality and purity are of utmost importance to us. This ensures that our products are consistently reliable.

### • Confidence and convenience

We believe that entrepreneurial success starts with people. That's why our mission statement defines us as a successful and responsible company. Our goal is to operate a worldwide business that produces meaningful benefits for you – our customers, market partners and community. With Merck Millipore Inorganic Reagents, you can expect a dependable and lasting partnership.

### • Safety and environmental protection

In order to ensure the high quality and safety of our products, Merck Millipore offers a large range of advanced packaging options for safe transport, storage and product protection. This underlines our goal of sustainable protection of people and the environment.

### Service

We work in close cooperation with you, our customers to develop innovative solutions based on your specific requirements and provide you an extensive range of product information.

Special product categories | Explore broad product ranges for numerous applications in your laboratory. These categories allow you to search within your specific application, such as volumetric solutions, reference materials or cleaning applications.

Alphabetical listing | The A to Z product listing spans over 1000 individual products and includes special features to make your search easier.



www.merckmillipore.com/inorganic-reagents

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### www.merckmillipore.com/classical-inorganic-analysis

### www.merckmillipore.com/instrumental-inorganic-analysis



### www.merckmillipore.com/safety-products













Metals and metal oxides



Volumetric solutions



Karl Fischer reagents and standards



Reference materials



Fluxes for XRF



**Caustic alkalis** 

and bases

High purity acids and bases



High purity salts



Absorption and filtration



Absorbents for spilled liquids



Drying agents



Auxiliaries for purification and sample preparation



Cleaning

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Cleaning applications

# History

### MERCK – living innovation.

The company's foundation was laid in 1668 when Friedrich Jacob Merck purchased a pharmacy in Darmstadt, which was later renamed »Engel Apotheke«. In the years that followed, the pharmacy was passed from father to son until in 1816 it came under the leadership of Heinrich Emanuel Merck. Industrial scale production was started in 1827 – and with it the birth of Merck as a company. Since then, the name Merck has stood for high quality pharmaceutical and chemical products that benefit our customers.





From the very beginning, our aim has been to deliver products of the highest standards. This was best expressed by Heinrich Emanuel Merck in his letter to a customer in 1851:

»I herewith guarantee the purity of my preparations and undertake to reimburse you for any damage that may arise through one of my preparations being impure.«

Heinrich Emanuel Merck

This dedication to quality was evident from the start of production with »Merck's Guaranteed Pure Reagents« and the addition of »Pro Analysis« in 1888. These measures assured consequent implementation and careful formulation to the highest purity standards, based on comparative chemical analysis. All products carried the seal »as a warranty sign for the identity and quality of the product«. And this is still valid today. Merck Inorganic Reagents are committed to superior quality, reliability and innovation, focused on our customer needs. As reliable partners, we offer a comprehensive range of premium products and services based on your requirements. We work closely with our customers, listen to your needs and challenges and integrate this knowledge into future products and services. Thanks to our cooperation today, we can develop the products you need tomorrow.

In 2010 Merck took over Millipore Corporation, one of the most leading life science companies. With the newly formed division "Merck Millipore" we strengthened our leading position in the life science market.

We always go the extra mile to ensure a lasting and dependable partnership.

# Quality management

Always high quality - just choose your grade

At Merck Millipore, we believe our tradition of world-class quality in reagents and chemicals is a reputation worth upholding. You can see how we act on this commitment on three fronts: validation, regulations and accreditation.

#### Quality

We maintain our own state-of-the-art laboratories to test and certify every single product in-house. Highly qualified specialist staff is key to this procedure. As a result, you can expect batch-to-batch consistency in any chemical or reagent you choose.

During our tests – even for complex instrumental and wet chemical procedures – we always integrate the latest developments in technology and methods. Adherence to international standards and legal requirements also occupies a central position in our work.

But ultimately, quality to us encompasses more than »product quality«. As our customer, you deserve more than just product purity or batch-to-batch consistency. You deserve highest quality service, from rapid delivery to sound advice and packaging that's right for your application. In short: service that exceeds your expectations. And that's what we offer you every day.

#### Regulations

Whether you manufacture products nationally or internationally, you're faced with a host of laws and guidelines. It can be difficult to maintain an overview of which requirements you need to meet where – especially when those requirements change. This is where a capable partner can help point the way.

You can also expect proven safety thanks to an unrivaled range of comprehensive specifications with up to 60 parameters, all combined with the declaration of international standards, such as ISO, ACS and Reag. Ph Eur. This means Merck Millipore analytical reagents comply with the official quality guidelines for pharmacopoeia analysis worldwide. The benefit to you? Customized, unique quality joined with the highest possible degree of reliability – a relationship that makes a real difference in your everyday lab tasks.

#### Accreditation

As an integral part of the way we comply with regulations, accreditation forms another cornerstone of our quality assurance policy. It acts as a stamp of approval from qualified external organizations, confirming that our quality is as good as we say it is. Our calibration labs, for example, are accredited, and our standards are measured against international standards of the National Institute of Standards and Technology (NIST), a U.S.-based body.

One way we ensure that we continue to provide distinguished chemicals and reagents is our continuous effort to expand the implementation of quality norms like DIN EN ISO / IEC 17025.

Take advantage of first-rate chemicals and reagents that are just right for your application! Our different quality grades are designed to accommodate your needs, whether you have to meet international standards, follow safety regulations, require both bulk and small quantities, or have an application unlike any other. You're sure to find a perfect fit in our product range.

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Deutsche Akkreditierungs German Accreditation Bod	
Entrusted according to Section 8 subsection subsection 3 AAASostarGitv Separatory to the Multilateral Agreements of EX, ILAC and UV for Mutual Recognition	n] AbhStelleC in connection with Section 1
Accreditation	
The Deutsche Alkindtierungsstelle Grittet	Correst Accreditation Body) attests that
Merck KGaA Frankfurter Str.250, 64293 Darmstadt	
with the calibration laboratory.	
Merck KGaA, Kalibrierlaboratorium N Frankfurter Str.250, 64293 Danmstadt	
is composed under the terms of DN EN ISS following field	DARC 19935-3008 to carry and califerations in the
Chemical analysis, reference matertals:	
<ul> <li>pří valut</li> <li>Electrolytic conductivity</li> <li>Mass fraction of elements in stars</li> <li>Mass fraction of thrimutils stands</li> </ul>	
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#### More information

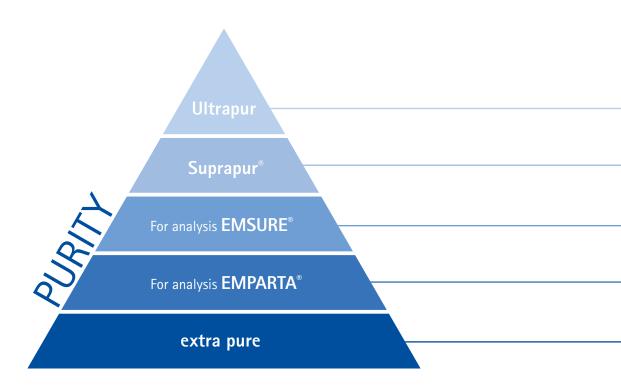
Turn the page to explore our quality grades

## Exceptional quality is our standard

Merck Millipore places the greatest importance on the high quality of each individual product. From development to delivery, we take the utmost care to achieve outstanding quality, particularly in the sensitive field of analytical reagents.

#### Always high quality - just choose your grade

Take advantage of first-rate chemicals and reagents that are perfectly suited to your individual application! Our various quality grades are designed to accommodate your every wish. Whether you need to meet international standards, follow safety regulations, require large or small quantities, or have an unique application, you are sure to find the ideal solution in our product range.



### Specification

	Standards	of the	American	Chemical	Society	
--	-----------	--------	----------	----------	---------	--

- ISO Standards of the International Organization for Standardization
- Reag. Ph Eur Reagents specified by the European Pharmacopoeia

### Quality grades for your individual needs

#### Ultrapur

High purity acids, hydrogen peroxide | Highly sophisticated instrumental trace analysis, e.g. ICP | Specified parameters in ppt range

#### Suprapur®

High purity acids, bases and salts | Instrumental trace analysis, e.g. AAS | Specified parameters in ppb range

#### **EMSURE**®

Acids, caustics and salts | For analysis in highly demanding and regulated markets, e.g. Pharma QC | Reag. Ph Eur, ACS, ISO et al | Up to 60 specified parameters

#### **EMPARTA®**

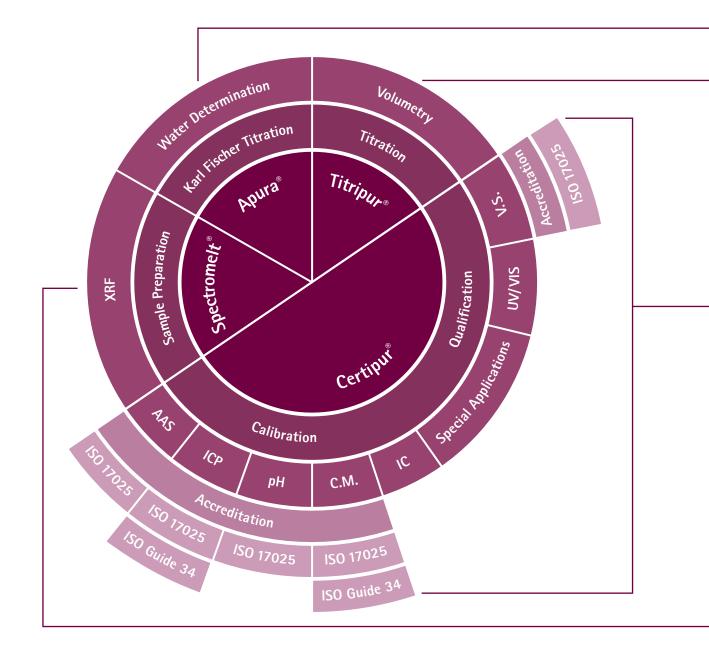
Acids | For analysis in routine applications and classical analysis | ACS | Up to 15 specified parameters

#### extra pure

Acids, caustics and salts | For preparative lab applications and cleaning purposes

# Our quality standards in instrumental analysis

The more sensitive the instrumental detection becomes, the more important it is to use highly pure reagents. System error caused by impurities in the auxiliary reagents are difficult to quantify. Merck Millipore has a specially tailored range of standards and reagents to satisfy all purity requirements.



### Specification / Traceability

- ISO International Organization for Standardization
- NIST National Institute of Standards and Technology, USA
- PTB Physical Technical Institute, Germany
- ag. Ph Eur Reagents specified by the European Pharmacopoeia
  - USP United States Pharmacopoeia requirements for reagents

### Quality grades for your individual needs

#### Apura®

Karl Fischer reagents | Titration according to Karl Fischer method for water determination

#### Titripur®

Volumetric solutions | Sophisticated titration | Reag. Ph Eur, Reag. USP | Traceable to NIST standards

#### Certipur®

Reference materials | Calibration and monitoring of instruments for instrumental analysis | ISO Guide 34 | ISO / IEC 17025 | Traceable to NIST and PTB standards

C.M. = Conductivity measurement | V.S. = Volumetric standards

#### Spectromelt<sup>®</sup>

Reagents for X-ray fluorescence sample preparation | Fluxes and tabletting aid

# Analytical reagents specified for pharmacopoeia analysis

Merck Millipore offers several hundred inorganic chemical reagents for pharmacopoeia analysis – in particular acids, salts, caustics, bases, indicators and special reagents. Our products bear the designation »for analysis EMSURE® ACS, Reag. Ph Eur«. This extensive range provides you with everything you need – from one trusted source.

In fact, no other manufacturer supplies such a wide variety of products. Furthermore, you have the confidence of working with the appropriate reagents for your particular needs that meet all quality requirements. Our analytical reagents for pharmacopoeia analysis not only comply with the ACS standard, but also with Reag. Ph Eur. Thus, our products meet all reagent specifications described by the American as well as the European Pharmacopoeia.

Analytical reagents from Merck Millipore fulfill the following quality guidelines:

- ACS, and referred by Reag. USP
- Reag. Ph Eur

Through compliance with these comprehensive standards, we are creating a new level of quality for analytical reagents, thus providing you with the highest possible degree of reliability – worldwide. Take advantage of our premium analytical reagents to fulfill the fundamental prerequisites of your scientific work and successfully pass all audits.

#### More information

### Safety and environment

The prevention of accidents, employee safety and the security of your working environment are of utmost importance to you. As they are for Merck Millipore.

#### Safety is our top priority

You can rely on Merck Millipore products, safety instructions, supporting materials and global experts to help you achieve the highest possible degree of safety.

We are committed to acting in an environmentally responsible way and safeguarding the health of every individual working with Merck Millipore products. Our Safety and Environmental Management is based on statutory legislation and the principles of Responsible Care\*.

All applicable laws and directives are strictly adhered to. Merck Millipore's Safety System follows international standards such as ISO and REACH. We take our responsibility seriously to assure your occupational health and wellbeing. High safety standards in production, processing and packaging are reflected in every Merck Millipore product you use in your laboratory. These high standards are achieved through:

- State-of-the-art Safety Data Sheets with detailed product information
- Specially designed packaging suited to the properties of each product (safe break bottles, safe caps etc.)
- Compilation of relevant environmental data on product properties, storage, use and disposal (SafeDat, ChemDat, Store Card)
- Extensive product tests prior to market launch
- Reliable storage and supply of products
- Exemplary and fast implementation of new GHS labelling

The basis of our joint success: innovative and reliable products that benefit humanity and safeguard the environment.

### Environmental protection, occupational safety and health are at the core of all our innovations!

#### Safety and environment – characteristics

> For easy detection, safety and/or sustainable characteristics of our products are highlighted with this symbol.

www.merckmillipore.com/protection > www.responsibility.merck.de > www.responsiblecare.org

### Packaging

In order to ensure high product quality and safety, sophisticated and innovative packaging is a pre-condition. Besides offering premium quality chemicals and reagents, Merck Millipore has invested decades into developing the most advanced packaging concepts in the field of chemistry.

#### **Essential protection**

Reliable packaging is essential to protect users and the environment, as well as to safeguard our high-quality reagents from impurities and contamination. This is why Merck Millipore offers a wide range of exceptional packaging options for safe transport, storage and userfriendly handling. All our packing materials have been tested for quality and permeability, thereby ensuring the purity of the product. Thus, not only the container itself, but also the closing system and the outer transportation box are optimized as a packaging system.

#### Packaging department

Merck Millipore's in-house packaging department is exclusively responsible for testing, developing and approving packaging materials.

In fact, our package testing facility for hazardous chemicals is authorized by the German Institute for Materials Research and Testing (Bundesanstalt für Materialforschung und -prüfung – BAM).



#### Measures to ensure high-quality, optimized packaging:

Packaging	• Design of innovative packaging concepts to make the use of reagents safe,
development	easy and more convenient

- Developing sophisticated packaging for new products
- Continuous improvements and inspections of our existing packaging
- Creating ecological and economical recycling concepts

#### Packaging testing • Conformity with all safety requirements

- Maintenance of quality even after longer periods of storage
- No interferences between packaging material and contents
- No contamination of the high purity reagents
- User-friendly packaging
- Ecological and economical package recycling

In order to maintain the high quality and reliability of all our packaging, we perform extensive tests before we use it for our chemicals. For each batch of packaging material, we prepare statistical test samples for the most critical packaging characteristics, for example:

- Testing of primary packaging with respect to purity
- Cooling stress tests of glass bottles
- Compression resistance test with corrugated board boxes
- Drop test with corrugated board boxes for hazardous goods
- Auditing of suppliers at regular intervals
- Hygiene management: exclusion of contamination by production procedure
- Consistent quality of packaging through process stability



# Variety of packaging types

Merck Millipore offers a unique range of packaging types in various sizes for reagents and chemicals. Here you will find more detailed information about the most important packaging types for Inorganic Reagents.

inliner

large quantities



### Your advantages

- Packaging is always compatible with the product
- Safe and convenient handling, storage and transportation
- Optimal protection of chemicals and reagents from contamination
- Application-oriented packaging
- Wide choice of packaging materials and sizes

#### Glass bottle



Merck Millipore's amber glass bottle, made from premium blank glass, is one of our standard forms of packaging for basic liquid reagents. It is characterized by ideal handling, storage and transportation properties.

- Premium amber blank glass of hydrolytic class 3, which remains inert even to aggressive chemicals
- The glass used minimizes the leaching of impurities by the contents
- Impermeable to air and water vapor, resulting in a high level of stability and allowing long storage of the content
- Amber glass protects the content against light
- The glass bottles are produced in a clean environment thereby safeguarding reagents against contamination
- The special shape of the opening allows an optimum pouring process
- The secure base and low center of gravity make it difficult for the bottle to tip over
- Special S28 and S40 closure system developed by Merck Millipore

Safebreak bottle Our exceptionally safe glass bottle for especially dangerous acids

- Glass bottle with eco-friendly PE jacket
- If the glass breaks, the splinters and acid are caught safely by the PE jacket
- Offers all the benefits of conventional glass bottles
- S40 closure system
- The Safebreak bottle can be recycled together with conventional glass bottles



#### More information

- ▶ See paragraph »Closure systems for liquids and solids« on page 23
- ▶ See chapter »Acids for analysis« on page 94

# Packaging overview

#### HDPE narrow necked bottle



The HDPE narrow necked bottle is another standard form of packaging for liquid reagents. The bottle's low tare weight makes it easy to handle and gives it ideal transportation properties.

- Made from high-density polyethylene (HDPE)
- The high level of purity of the HDPE used means that it has little influence on the contents
- Integrated handles (for 2.5 | and 5 |) ensure comfortable handling and dosage
- The special shape of the opening allows an optimum pouring process
- The narrow base enables efficient use of laboratory storage space
- Low tare weight makes it easy to handle and keeps transport costs down
- 2.5 l bottle is extremely stable thanks to its special base geometry (even with pressure producing chemicals)
- Depending on the chemicals, the bottles are colored to protect against unwanted UV radiation
- Special S28 and S40 closure systems

For certain products such as high purity hydrogen peroxide, we offer specially developed bottles which are adapted to the product's individual properties.

#### HDPE wide necked bottle



Merck Millipore mainly uses HDPE wide necked bottles for solids. With their square base and large opening, the bottles allow solids to be handled easily and safely.

- Made from high-density polyethylene (HDPE)
- The high level of purity of the HDPE used means that it has little influence on the contents
- Wide opening for easy withdrawal
- Square base allows optimum use of storage space in the laboratory and during transportation
- S38 to S85 closure systems

#### More information

▶ See paragraph »Closure systems for liquids and solids« on the next page

#### Closure systems



Merck Millipore's exclusive closure systems are the ideal complement to our glass, HDPE and Safebreak bottles. The screw caps and special openings of the bottles were developed especially for use with chemicals and make the bottles extremely leakproof.

- Leakproof thanks to special shape
  - 3-zone closure
  - No sealing ring necessary
- Tamper-proof seal
- Made from polypropylene or PTFE, depending on the product

Closure systems for liquids and solids in glass,

HDPE and Safebreak bottles (S40, S28, S38, S60 and S85)

#### SafetyCap Our special closure for chemicals which build up pressure in the bottle.



- Allows gas to be released in case of internal pressure
- Absolutely leakproof against liquids
  - No liquid can leak, even if the bottle is tipped
- No additional PE bag needed
- The semi-permeable PTFE valve prevents gas or liquid from entering the bottle
- No protruding parts that can break off

#### More information

▶ See chapter »Acids for analysis« on page 94

# Packaging overview

Corrugated board box with PE inliner



Our standard packaging for large quantities of solids. The exceptionally high-quality PE inliner provides the chemicals with ideal protection from contamination while also preventing substances from escaping.

- The PE inliner used to protect the product is produced in clean room conditions
- Corrugated board boxes are glued in a water-resistant manner in accordance with DIN 53133. This means that they remain stable even in damp conditions and provide optimum protection for the product
- Stable construction allows stacking
- The size of the corrugated board boxes is designed to fit the size of the CP-5 chemicals pallet, guaranteeing optimum use of space and safety when loading

Packaging for larger quantities of liquid chemicals

For larger quantities of liquid reagents, we offer packaging that ensures safe and convenient handling.

#### Canisters

- HDPE canisters are available in 5 l to 25 l
- For light-sensitive chemicals, the canisters are colored blue
- All canisters have a KS 60 x 6 male screw as standard
- Specially shaped canisters are used for liquids which build pressure

#### Drums

- HDPE drums up to 200 l
- CombiDrum: steel drum with PE inliner and special finishing

#### Intermediate Bulk Container (IBC)

• Made from HDPE, for liquids up to 950 l



#### Titripac®

Titripac<sup>®</sup> is an innovative and safe packaging for ready-to-use volumetric solutions, buffer solutions and other aqueous solutions.



- Available in 4 l and 10 l
- Outer packaging is made from corrugated board, the inner bag from composite foil
- Thanks to its absolutely leakproof, sealed packaging system, Titripac<sup>®</sup> ensures that the quality of the solution remains constant from the very first to the very last drop
- No additional checks on the solution are necessary
- Contamination from air, carbon dioxide or microorganisms is excluded
- Easy to use thanks to the integrated tap, which can be connected directly to a titrator using an adaptor
- Reduced packaging waste since the box and inner bag are disposed of separately

Bags for portion-sized quantities of ready-to-use buffer and conductivity solutions.

#### Sachets



- The bags are made from composite film, completely leakproof and guarantee the stability of the solution
- Portion-sized quantities for single use, so that a fresh solution is ready for any application
- Packed in boxes of 30 sachets

#### More information

- ▶ See chapter »Volumetric solutions« on page 132
- See chapter »Standards, reference materials and buffers« on page 172

# Packaging overview

Bottle made from borosilicate glass



Our standard packaging for high purity liquid reagents of Suprapur® quality.

- Clear glass bottle of hydrolytic class 1 (DURAN\*)
- Especially high purity level and very low leach-out behavior by acids
- No contamination of chemicals, so that reagents retain a constant purity level

Bottle made from perfluoralkoxy copolymer (PFA)



Our standard packaging for high purity reagents of Ultrapur quality.

- Perfluoralkoxy copolymer (PFA) is an extremely chemically-resistant plastic
- · High purity reagents, with maximum impurities within the ppt or ppq range, are protected from contamination and interaction with the packaging
- The quality of the chemical is retained

Plastic ampoules (Titrisol® ampoules)

Titrisol® ampoules are used for concentrates in the production of buffer solutions, element standard solutions and volumetric solutions.



- Each ampoule contains a precisely defined amount of substance, which is normally filled up to 1 liter final volume. Any concentration required, however, can be created by dilution
- Titrisol<sup>®</sup> ampoules are made from polyethylene
- Easy to open
- Space-saving storage
- Long shelf life

Glass ampoule

We use especially high-quality glass ampoules for standards and reference materials.



- Clear or brown glass of hydrolytic class 1
- · Hermetically sealed so that the reagent remains stable
- Predetermined breaking point for ease of opening

# Packaging for shipping The same high standards that we set for the primary packaging of our reagents, also apply to our transport packaging. This ensures that our premium products reach you safely and completely intact.

- The packaging for shipping is also subject to our strict quality assurance
- Corrugated board boxes are mainly used for shipping
- Corrugated board boxes are glued in a water-resistant manner in accordance with DIN 53133. This means that they remain stable even in damp conditions and provide optimum protection for the product
- Merck Millipore delivers on CP-5 chemicals pallets (1140 x 760 cm)
- The use of disposable pallets means that they are always clean and flawless



### Packaging overview

#### Labels

Labeling products safely is extremely important. This is why Merck Millipore uses varnished paper labels which are resistant to most chemicals. PE labels are also used in exceptional cases.

Our labels are resistant to abrasion and forgery proof. They are applied with a glue specially developed for use in the chemicals sector and with the respective packaging.

Needless to say, all substances carry GHS labels. By June 1, 2015, all mixtures will also feature GHS labels. Merck Millipore's labels contain a wide range of useful information about our products (see next page).

#### GHS - the first-ever globally uniform basis

GHS stands for the Globally Harmonized System of Classification and Labeling of Chemicals. In December 2002, the United Nations published the GHS in the so-called »Purple Book« with a description of harmonized classification and labeling criteria.

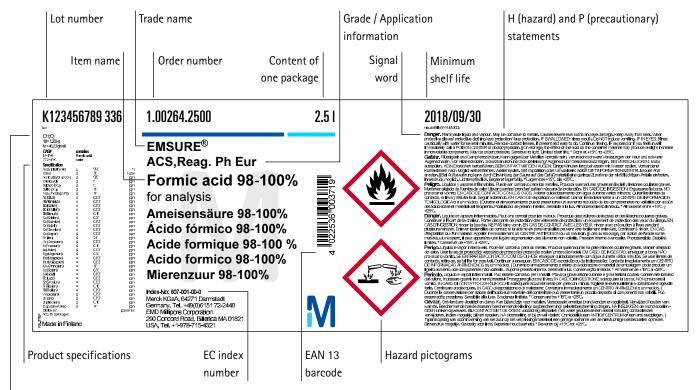
The goal of GHS is to harmonize the various existing classification and labeling systems all over the world. Due to inconsistent evaluation criteria, it has long been the case that the same substance can be classified as poisonous, harmful to health, or even not harmful. This leads to different levels of protection in terms of occupational health and safety, as well as for consumers and the environment. GHS offers the first ever globally uniform basis for the evaluation of substance properties. In doing so, it establishes the requirement of a globally high protection level for human health and the environment.

This harmonized system defines criteria for classification and labeling, including hazardous substance labeling, and requirements for the creation of Safety Data Sheets.

More information

www.merckmillipore.com/ghs > www.merckmillipore.com/safety

#### **Product** label



Empirical formulas | Density | Molar mass

# Index A-B

#### Inorganic Reagents A

Product	Cat. No.	Pag
Absorption tube for CO <sub>2</sub> (sodium hydroxide on support)	101562	24
Absorption tube for $H_2O$ (molecular sieve 0.3 nm with indicator)	106107	24
Acetic acid (glacial) 100 % anhydrous for analysis EMSURE® ACS, ISO, Reag. Ph Eur	100063	10
	100063	11
Acetic acid (glacial) 100 % for analysis EMPARTA® ACS	101830	11
Acetic acid (glacial) 100 % Suprapur®	100066	22
Acetic acid 1 mol/l prepared from raw materials acc. Ph Eur Titripur®	199061	14
Acetic acid 30 % for analysis Reag. Ph Eur	159166	11
Acetic acid 96 % for analysis EMSURE®	100062	11
Acetic acid for 1000 ml, $c(CH_3COOH) = 0.1 \text{ mol/l} (0.1 \text{ N}) \text{ Titrisol}^{\textcircled{B}}$	109944	14
Acetic acid for 500 ml, c(CH <sub>3</sub> COOH) = 1 mol/l (1 N) Titrisol®	109951	14
Acetic anhydride for analysis EMSURE® ACS, ISO, Reag. Ph Eur	100042	11
Acridine orange zinc chloride double salt [C.I. 46005] for microscopy Certistain®	115931	30
Alizarin red S mono sodium salt [C.I. 58005] for analysis and indicator	106278	29
Alizarin-3-methylamine-N,N-diacetic acid dihydrate for the spectrophotometric determination of fluoride	101010	29
Alkali blue [C.I. 42765] indicator	109196	29
Alkali blue solution indicator	109198	29
Alloy acc. to Wood melting point about 73°C bar diameter about 1 cm	106001	12
Aluminium (foil) for analysis 0.3 mm thickness, 30 mm width EMSURE®	101057	12
Aluminium ammonium sulfate dodecahydrate for analysis EMSURE® ACS	101031	7
Aluminium fine powder, stabilized about 2 % fat	101056	12
Aluminium ICP standard traceable to SRM from NIST Al(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 $\%$ 1000 mg/l Al Certipur®	170301	18
Aluminium ICP standard traceable to SRM from NIST Al(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 $\%$ 10000 mg/l Al Certipur <sup>®</sup>	170371	18
Aluminium nitrate nonahydrate for analysis EMSURE®	101063	7
Aluminium oxide fibers for gooch crucibles	115754	24
Aluminium potassium sulfate dodecahydrate for analysis EMSURE® ACS, Reag. Ph Eur	101047	7
Aluminium standard 1000 mg Al, (AlCl <sub>3</sub> in H <sub>2</sub> O) Titrisol <sup>®</sup>	109967	18
Aluminium standard solution traceable to SRM from NIST Al(NO $_3$ ) $_3$ in HNO $_3$ 0.5 mol/l 1000 mg/l Al Certipur®	119770	18
Amido black 10 B [C.I. 20470] for electrophoresis	101167	29
Amidosulfuric acid extra pure	100219	11
Amidosulfuric acid for analysis EMSURE®	100103	11
Ammonia solution 25 % for analysis EMSURE®	105432	12
Ammonia solution 25 % Suprapur®	105428	22
Ammonia solution 28 – 30 % for analysis EMSURE® ACS, Reag. Ph Eur	105423	12
Ammonia solution 32 % extra pure	105426	12
Ammonium acetate for analysis EMSURE® ACS, Reag. Ph Eur	101116	7
Ammonium amidosulfonate for analysis (for detection of sulfonamide in blood) EMSURE® ACS, Reag. Ph Eur	101220	7
Ammonium bromide for analysis EMSURE® ACS	101125	7
Ammonium carbamate for analysis EMSURE®	101134	7
Ammonium carbonate for analysis EMSURE® ACS, Reag. Ph Eur	159504	7
Ammonium cerium(IV) nitrate for analysis EMSURE® ACS, Reag. Ph Eur	102276	7

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Product	Cat. No.	Page
Ammonium cerium(IV) nitrate solution $c(NH_a)_2$ Ce(NO <sub>3</sub> ) <sub>6</sub> = 0.1 mol/I (0.1 N) Titripur <sup>®</sup>	102277	142
Ammonium cerium(IV) sulfate dihydrate for analysis EMSURE® ACS	102273	70
Ammonium chloride 99.995 Suprapur®	101143	232
Ammonium chloride for analysis EMSURE® ACS, ISO, Reag. Ph Eur	101145	70
Ammonium dihydrogen phosphate 99.99 Suprapur®	101440	232
Ammonium dihydrogen phosphate for analysis EMSURE® ACS, Reag. Ph Eur	101126	70
Ammonium fluoride for analysis EMSURE® ACS	101164	70
Ammonium heptamolybdate tetrahydrate for analysis EMSURE® ACS, Reag. Ph Eur	101182	70
Ammonium iron(II) sulfate hexahydrate for analysis EMSURE®, ISO	103792	71
Ammonium iron(II) sulfate solution for 250 ml, $c[(NH_4)_7Fe(SO_4)_2] = 0.1 mol/l (0.1 N) Titrisol®$	109864	142
Ammonium iron(III) sulfate dodecahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	103776	70
Ammonium nitrate for analysis EMSURE® ACS	101188	71
Ammonium peroxodisulfate for analysis EMSURE® ACS, Reag. Ph Eur	101201	71
Ammonium standard solution traceable to SRM from NIST NH <sub>4</sub> Cl in H <sub>7</sub> O 1000 mg/l NH <sub>4</sub> $^{+}$ Certipur <sup>®</sup>	119812	203
Ammonium sulfate 99.9999 Suprapur®	101209	232
Ammonium sulfate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	101217	71
Ammonium thiocyanate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	101213	71
Ammonium thiocyanate solution $c(NH_4SCN) = 0.1 \text{ mol/I} (0.1 \text{ N}) \text{ Titripur}^{\otimes}$	109079	142
Ammonium thiocyanate solution for 1000 ml, $c(NH_{4}SCN) = 0.1 \text{ mol/l} (0.1 \text{ N}) \text{ Titrisol}^{\otimes}$	109900	142
Ammonium thiosulfate 98 %+	101208	127
Anion multi-element standard I 1000 mg/l: F⁻, PO₄³⁻, Br⁻ in H₂O Certipur®	111437	203
Anion multi-element standard II 1000 mg/l: Cl <sup>-</sup> , NO <sup>3-</sup> , SO <sub>4</sub> <sup>2-</sup> in H <sub>2</sub> O Certipur <sup>®</sup>	111448	203
Antimony ICP standard traceable to SRM from NIST ${ m Sb_2O_3}$ in HCl 7 $\%$ 1000 mg/l Sb Certipur®	170302	180
Antimony powder for analysis particle size <150 µm EMSURE®	107832	127
Antimony standard solution traceable to SRM from NIST Sb $_2O_3$ in HCl 2 mol/l 1000 mg/l Sb Certipur®	170204	186
Antimony(III) chloride for analysis EMSURE®	107838	127
Antimony(III) oxide extra pure	107835	127
Antimony(III) oxide for analysis EMSURE®	107836	127
Arsenazo III metal indicator	110107	298
Arsenic ICP standard traceable to SRM from NIST $H_3AsO_4$ in $HNO_3$ 2 – 3 % 1000 mg/l As Certipur®	170303	180
Arsenic lumps for analysis (protective gas: nitrogen) EMSURE®	100115	127
Arsenic standard 1000 mg As, (As <sub>2</sub> O <sub>5</sub> in H <sub>2</sub> O) Titrisol <sup>®</sup>	109939	185
Arsenic standard solution traceable to SRM from NIST $H_3AsO_4$ in $HNO_3$ 0.5 mol/l 1000 mg/l As Certipur®	119773	186
L(+)-Ascorbic Acid for analysis EMSURE® ACS, ISO, Reag. Ph Eur	100468	110
Aurin tricarboxylic acid ammonium salt for analysis (reagent for aluminium) ACS	100128	298
Barbituric acid for analysis EMSURE®	100132	110
Barium acetate for analysis EMSURE® ACS	101704	71
Barium carbonate for analysis EMSURE® ACS, Reag. Ph Eur	101714	71
Barium chloride 99.995 Suprapur®	101716	232
Barium chloride dihydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	101719	71

# Index B

#### Inorganic Reagents B

Product	Cat. No.	Pag
Barium chloride solution for 1   measure solution, $c(BaCl_2) = 0.05 \text{ mol/l} (0.1 \text{ N}) \text{ Titrisol}^{\textcircled{m}}$	109962	14
Barium fluoride 99.99 Suprapur®	101722	23
Barium hydroxide octahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	101737	7
Barium ICP standard traceable to SRM from NIST Ba(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 $\%$ 1000 mg/l Ba Certipur®	170304	18
Barium nitrate for analysis EMSURE® ACS	101729	7
Barium perchlorate anhydrous for analysis EMSURE®	101738	7
Barium perchlorate solution in 2-propanol / water (80 : 20) $c(Ba(ClO_4)_2) = 0.005 \text{ mol/l Titripur}^{\textcircled{B}}$	109086	14
Barium standard 1000 mg Ba, (BaCl $_2$ in 7 % HCl) Titrisol $^{\odot}$	109968	18
Barium standard solution traceable to SRM from NIST Ba(NO $_3$ ) $_2$ in HNO $_3$ 0.5 mol/l 1000 mg/l Ba Certipur®	119774	18
Barium sulfate for white standard DIN 5033	101748	21
Benzoic acid volumetric standard, secondary reference material for alkalimetry, traceable to NIST Standard Reference Material (SMR) Certipur®	102401	20
Benzoic acid for analysis EMSURE® Reag. Ph Eur	100136	11
Beryllium ICP standard traceable to SRM from NIST Be <sub>4</sub> O( $C_2H_3O_2$ ) <sub>6</sub> in HNO <sub>3</sub> 2 – 3 % 1000 mg/l Be Certipur <sup>®</sup>	170305	18
Beryllium standard solution $Be_4O(C_2H_3O_2)_6$ in HNO <sub>3</sub> 0.5 mol/l 1000 mg/l Be Certipur <sup>®</sup>	170207	18
2,2'-Bipyridine for analysis (reagent for iron (II) and molybdenum) ACS	103098	29
Bismuth ICP standard traceable to SRM from NIST Bi(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 % 1000 mg/l Bi Certipur <sup>®</sup>	170306	18
Bismuth standard solution traceable to SRM from NIST Bi(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 0.5 mol/l 1000 mg/l Bi Certipur®	119804	18
Bismuth(III) nitrate alkaline for analysis EMSURE® Reag. Ph Eur	101878	7
Bismuth(III) oxide extra pure	101862	12
Boiling chips granules $\sim 1 - 2$ mm	107912	28
Boiling chips granules $\sim 2 - 8$ mm	107913	28
Boric acid 99.9999 Suprapur®	100765	21
Boric acid for analysis EMSURE® ACS, ISO, Reag. Ph Eur	100165	11
Boron ICP standard traceable to SRM from NIST, H <sub>3</sub> BO <sub>3</sub> in H <sub>2</sub> O 1000 mg/l B Certipur®	170307	18
Boron standard solution traceable to SRM from NIST H <sub>3</sub> BO <sub>3</sub> in H <sub>2</sub> O 1000 mg/l B Certipur®	119500	18
Brilliant cresyl blue zinc chloride double salt for microscopy Certistain®	101368	29
Brilliant green (hydrogen sulfate) [C.I 42040] for microbiology	101310	29
Bromide bromate solution for 1000 ml, $c(Br_2) = 0.05 \text{ mol/l} (0.1 \text{ N}) \text{ Titrisol}^{\odot}$	109905	14
Bromide standard solution traceable to SRM from NIST NaBr in H20 1000 mg/I Br <sup>-</sup> Certipur®	119896	20
Bromine 99.9999 Suprapur®	101947	23
Bromine extra pure	101945	12
Bromine for analysis EMSURE® ACS, ISO, Reag. Ph Eur	101948	12
Bromocresol green indicator ACS, Reag. Ph Eur	108121	29
Bromocresol green sodium salt indicator water-soluble ACS	101541	29
Bromocresol purple indicator ACS, Reag. Ph Eur	103025	29
Bromophenol blue indicator ACS, Reag. Ph Eur	108122	29
Bromophenol red indicator	103023	29
Bromothymol blue indicator ACS, Reag. Ph Eur	103026	29
Bromothymol blue sodium salt indicator water-soluble ACS	101895	29
BTS catalyst (about 5 x 3 mm) for gas purification	104182	28

#### Inorganic Reagents B

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Product	Cat. No.	Pag
Buffer concentrate (borate / hydrochloric acid), traceable to SRM of NIST and PTB for 500 ml buffer solution, pH 8.00 +- 0.02 (20°C) Titrisol®	109888	19
Buffer concentrate (boric acid / potassium chloride / sodium hydroxide), traceable to SRM of NIST and PTB for 500 ml buffer solution, pH 10.00 +- 0.05 (20°C) Titrisol®	109890	19
Buffer concentrate (boric acid / potassium chloride / sodium hydroxide), traceable to SRM of NIST and PTB for 500 ml buffer solution, pH 11.00 +- 0.05 (20°C) Titrisol®	109880	19
Buffer concentrate (boric acid / potassium chloride / sodium hydroxide), traceable to SRM of NIST and PTB for 500 ml buffer solution, pH 9.00 +- 0.02 (20°C) Titrisol®	109889	19
Buffer concentrate (citrate / hydrochloric acid), traceable to SRM of NIST and PTB for 500 ml buffer solution, pH 2.00 +- 0.02 (20°C) Titrisol®	109882	19
Buffer concentrate (citrate / hydrochloric acid), traceable to SRM of NIST and PTB for 500 ml buffer solution, pH 3.00 +- 0.02 (20°C) Titrisol®	109883	19
Buffer concentrate (citrate / hydrochloric acid), traceable to SRM of NIST and PTB for 500 ml buffer solution, pH 4.00 +- 0.02 (20°C) Titrisol®	109884	19
Buffer concentrate (citrate / sodium hydroxide), traceable to SRM of NIST and PTB for 500 ml buffer solution, pH 5.00 +- 0.02 (20°C) Titrisol®	109885	19
Buffer concentrate (citrate / sodium hydroxide), traceable to SRM of NIST and PTB for 500 ml buffer solution, pH 6.00 +- 0.02 (20°C) Titrisol®	109886	19
Buffer concentrate (glycine / hydrochloric acid), traceable to SRM of NIST and PTB for 500 ml buffer solution, pH 1.00 +- 0.02 (20°C) Titrisol®	109881	19
Buffer concentrate (phosphate / sodium hydroxide), traceable to SRM of NIST and PTB for 500 ml buffer solution, pH 12.00 +- 0.05 (20°C) Titrisol®	109892	19
Buffer concentrate (phosphate), traceable to SRM of NIST and PTB for 500 ml buffer solution, pH 7.00 +- 0.02 (20°C) Titrisol®	109887	19
Buffer concentrate (potassium chloride / sodium hydroxide), traceable to SRM of NIST and PTB for 500 ml buffer solution, pH 13.00 +- 0.05 (20°C) Titrisol®	109893	19
Buffer concentrate for buffer solution acc. to WEISE for 1000 ml of solution (phosphate), pH 7.20 Titrisol®	109879	19
Buffer solution (acetic acid / sodium acetate) traceable to SRM from NIST and PTB, pH 4.66 (20°C) Certipur®	107827	19
Buffer solution (boric acid / potassium chloride / sodium hydroxide solution), traceable to SRM from NIST and PTB, pH 9.00 (25°C) Certipur®	199003	19
Buffer solution (boric acid / potassium chloride / sodium hydroxide) color coded: blue, traceable to NIST and PTB, pH 10.00 (25°C) Certipur®	199050	19
Buffer solution (boric acid / potassium chloride / sodium hydroxide) color coded: yellow, traceable to NIST and PTB, pH 10.00 (20°C) Certipur®	109400	19
Buffer solution (boric acid / potassium chloride / sodium hydroxide), colored: blue, traceable to SRM from NIST and PTB, pH 9.00 (20°C) Certipur®	109476	19
Buffer solution (boric acid / potassium chloride / sodium hydroxide), traceable to SRM from NIST and PTB, pH 10.00 (20°C) Certipur®	109438	19
Buffer solution (boric acid / potassium chloride / sodium hydroxide), traceable to SRM from NIST and PTB, pH 10.00 (25°C) Certipur®	109409	19
Buffer solution (boric acid / potassium chloride / sodium hydroxide) traceable to SRM from NIST and PTB, pH 10.00 (25°C) Certipur®	199004	19

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#### Inorganic Reagents B

Product	Cat. No.	Pag
Buffer solution (boric acid / potassium chloride / sodium hydroxide) traceable to SRM from NIST and PTB, pH 11.00 (20°C) Certipur®	109462	193
Buffer solution (boric acid / potassium chloride / sodium hydroxide) traceable to SRM from NIST and PTB, pH 9.00 (20°C) Certipur®	109461	19
Buffer solution (boric acid / potassium chloride / sodium hydroxide) traceable to SRM from NIST and PTB, pH 9.00 (25°C) Certipur®	109408	19
Buffer solution (boric acid / sodium hydroxide / hydrogen chloride) traceable to SRM from NIST and PTB, pH 8.00 (20°C) Certipur®	109460	19
Buffer solution (boric acid / sodium hydrogen / hydrogen chloride) traceable to SRM from NIST and PTB, pH 8.00 (25°C) Certipur®	199038	19
Buffer solution (boric acid / sodium hydroxide / potassium chloride) traceable to SRM from NIST and PTB, pH 11.00 (25°C) Certipur®	199021	19
Buffer solution (boric acid / sodium hydrogen / potassium chloride) traceable to SRM from NIST and PTB, pH 11.00 (25°C) Certipur®	199041	19
Buffer solution (citric acid / sodium hydroxide / hydrogen chloride) color: red, traceable to SRM from NIST and PTB, pH 4.00 (20°C) Certipur®	109475	19
Buffer solution (citric acid / sodium hydroxide / hydrogen chloride) color: red, traceable to SRM from NIST and PTB, pH 4.00 (25°C) Certipur®	199054	19
Buffer solution (citric acid / sodium hydroxide / hydrogen chloride) traceable to SRM from NIST and PTB, pH 2.00 (20°C) Certipur®	109433	19
Buffer solution (citric acid / sodium hydroxide / hydrogen chloride) traceable to SRM from NIST and PTB, pH 2.00 (25°C) Certipur®	109442	19
Buffer solution (citric acid / sodium hydroxide / hydrogen chloride) traceable to SRM from NIST and PTB, pH 2.00 (25°C) Certipur®	199012	19
Buffer solution (citric acid / sodium hydroxide / hydrogen chloride) traceable to SRM from NIST and PTB, pH 3.00 (20°C) Certipur®	109434	19
Buffer solution (citric acid / sodium hydroxide / hydrogen chloride) traceable to SRM from NIST and PTB, pH 3.00 (25°C) Certipur®	109444	19
Buffer solution (citric acid / sodium hydroxide / hydrogen chloride) traceable to SRM from NIST and PTB, pH 4.00 (20°C) Certipur®	109435	19
Buffer solution (citric acid / sodium hydroxide / hydrogen chloride) traceable to SRM from NIST and PTB, pH 4.00 (25°C) Certipur®	109445	19
Buffer solution (citric acid / sodium hydroxide / hydrogen chloride) traceable to SRM from NIST and PTB, pH 4.00 (25°C) Certipur®	199064	19
Buffer solution (citric acid / sodium hydroxide) traceable to SRM from NIST and PTB, pH 5.00 (20°C) Certipur®	109436	19
Buffer solution (citric acid / sodium hydroxide) traceable to SRM from NIST and PTB, pH 5.00 (25°C) Certipur®	109446	19
Buffer solution (citric acid / sodium hydroxide) traceable to SRM from NIST and PTB, pH 6.00 (20°C) Certipur®	109437	19
Buffer solution (citric acid / sodium hydroxide) traceable to SRM from NIST and PTB, pH 6.00 (25°C) Certipur®	199036	19
Buffer solution (citric acid / sodium hydroxide) traceable to SRM from NIST and PTB, pH 6.00 (25°C) Certipur®	199016	19
Buffer solution (di-sodium hydrogen phosphate / potassium dihydrogen phosphate) color: green, traceable to SRM from NIST and PTB, pH 7.00 (20°C) Certipur®	109477	19
Buffer solution (di-sodium hydrogen phosphate / potassium dihydrogen phosphate) traceable to SRM from NIST and PTB, pH 7.00 (20°C) Certipur®	109439	19
Buffer solution (di-sodium hydrogen phosphate / potassium hydrogen phosphate) traceable to SRM from NIST and PTB, pH 6.88 (20°C) Certipur®	107294	19
Buffer solution (di-sodium hydrogen phosphate / potassium dihydrogen phosphate), color: yellow, traceable to SRM from NIST and PTB, pH 7.00 (25°C) Certipur®	199057	19
Buffer solution (di-sodium hydrogen phosphate / sodium hydroxide) traceable to SRM from NIST und PTB,	199022	19
pH 12.00 (25°C) Certipur®	199022	19
Buffer solution (di-sodium tetraborate) traceable to SRM from NIST and PTB, pH 9.18 (25°C) Certipur®	199019	19
Buffer solution (di-sodium tetraborate) traceable to SRM from NIST and PTB, pH 9.22 (20°C) Certipur®	101645	19
Buffer solution (glycine / sodium chloride / hydrogen chloride) traceable to SRM from NIST and PTB, pH 1.00 (20°C) Certipur®	109432	19
Buffer solution (glycine / sodium chloride / hydrogen chloride) traceable to SRM from NIST and PTB, pH 1.00 (25°C) Certipur®	109441	19
Buffer solution (potassium dihydrogen phosphate / di-sodium hydrogen phosphate) traceable to SRM from NIST and PTB, pH 7.00 (25°C) Certipur®	109407	19
Buffer solution (potassium dihydrogen phosphate / di-sodium hydrogen phosphate) traceable to SRM from NIST and PTB, pH 7.00 (25°C) Certipur®	199002	19

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	Product	Cat. No.	Page
В	Buffer solution (potassium hydrogen phthalate) traceable to SRM from NIST and PTB, pH 4.01 (25°C) Certipur®	109406	194
	Buffer solution (potassium hydrogen phthalate) traceable to SRM from NIST and PTB pH 4.01 (25°C) Certipur®	199001	197
	Buffer solution for strong acids for Karl Fischer titration Apura®	188035	163
	Buffer solution for strong bases for Karl Fischer titration Apura®	188036	163
	Buffer solutions traceable to SRM from NIST and PTB 10 x pH 4.01 (phthalate), 10 x pH 7.00 (phosphate), 10 x pH 9.00 (borate), pH 4.01 / pH 7.00 / pH 9.00 (25°C) Certipur®	199005	197
	Buffer solutions traceable to SRM from NIST and PTB 10 x pH 4.01 (phthalate)m, 10 x pH 7.00 (phosphate) 10 x pH 10.00 (borate), pH 4.01 / pH 7.00 / pH 10.00 (25°C) Certipur®	199006	197
С	Cadmium acetate dihydrate for analysis EMSURE®	102003	71
	Cadmium coarse powder, for analysis and for filling reductors particle size about 0.3 – 1.6 mm EMSURE®	102001	127
	Cadmium granular, for analysis particle size about 3 – 6 mm EMSURE®	102004	127
	Cadmium ICP standard traceable to SRM from NIST Cd(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 % 1000 mg/I Cd Certipur®	170309	180
	Cadmium standard 1000 mg Cd, (CdCl <sub>2</sub> in H <sub>2</sub> O) Titrisol®	109960	185
	Cadmium standard solution traceable to SRM from NIST Cd(NO <sub>3</sub> ), in HNO <sub>3</sub> 0.5 mol/l 1000 mg/l Cd Certipur®	119777	186
	Cadmium sulfate hydrate for analysis EMSURE® ACS	102027	71
	Calcein indicator for metal determination	102315	298
	Calcium carbonate volumetric standard, secondary reference material for complexometry,	102410	207
	traceable to NIST Standard Reference Material (SRM) Certipur®		
	Calcium carbonate 99.95 Suprapur®	102059	232
	Calcium carbonate precipitated for analysis EMSURE® Reag. Ph Eur	102066	71
	Calcium carbonate precipitated for analysis of silicates EMSURE®	102067	71
	Calcium chloride anhydrous powder Reag. Ph Eur	102378	264
	Calcium chloride anhydrous, granular $\sim$ 1 – 2 mm	102379	264
	Calcium chloride anhydrous, granular $\sim$ 2 – 6 mm	102391	264
	Calcium chloride anhydrous, granular $\sim$ 6 – 14 mm	102392	264
	Calcium chloride dihydrate for analysis EMSURE® ACS, Reag. Ph Eur	102382	72
	Calcium chloride fused, granular about 0.5 – 2.0 mm for elementary analysis	102083	92
	Calcium chloride tetrahydrate 99.995 Suprapur®	102384	232
	Calcium granular particle size about 2 – 6 mm	102053	264
	Calcium hydroxide for analysis EMSURE® ACS, Reag. Ph Eur	102047	72
	Calcium ICP Standard traceable to SRM from NIST Ca(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 % 1000 mg/l Ca Certipur <sup>®</sup>	170308	180
	Calcium ICP Standard traceable to SRM from NIST Ca(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 % 10000 mg/l Ca Certipur <sup>®</sup>	170373	180
	Calcium nitrate tetrahydrate 99.95 Suprapur®	102123	232
	Calcium nitrate tetrahydrate for analysis EMSURE® ACS	102121	72
	Calcium oxide from marble small lumps $\sim$ 3 – 20 mm	102109	240
	Calcium standard 1000 mg Ca, (CaCl $_{ m 2}$ in 6.5 % HCI) Titrisol $^{ m B}$	109943	185
	Calcium standard dissolved in oil c(Ca in standard oil) = 1 g/kg Certipur®	115053	187
	Calcium standard solution traceable to SRM from NIST Ca(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 0.5 mol/l 1000 mg/l Ca Certipur®	119778	186
	Calcium sulfate dihydrate precipitated for analysis EMSURE®	102161	72
	Calcon [C.I. 15705] metal indicator	104594	298
	Calconcarboxylic acid metal indicator	104595	298

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#### Inorganic Reagents C

Product	Cat. No.	Pag
Cerium ICP standard traceable to SRM from NIST Ce(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 $\%$ 1000 mg/I Ce Certipur <sup>®</sup>	170311	18
Cerium(IV) sulfate solution c(Ce(SO <sub>4</sub> ) <sub>2</sub> · 4 H <sub>2</sub> O) = 0.1 mol/l (0.1 N) Titripur <sup>®</sup>	109092	14
Cerium(IV) sulfate tetrahydrate for analysis EMSURE®	102274	7
Certified sec. standard reference buffer solution di-sodium tetraborate decahydrate, directly traceable to primary SRM from NIST / PTB $pH(S) = 9.184 (25^{\circ}C)$ Certipur <sup>®</sup>	107203	19
Certified sec. standard reference buffer solution potassium dihydrogen phosphate / di-sodium hydrogen phosphate directly traceable to primary SRM from NIST / PTB pH (S) = 6.863 (25°C) Certipur®	e, <b>107202</b>	19
Certified sec. standard reference buffer solution potassium hydrogen phosphate / di-sodium hydrogen phosphate, directly traceable to primary SRM from NIST / PTB $pH(S) = 7.416 (25^{\circ}C)$ Certipur®	107205	19
Certified sec. standard reference buffer solution potassium hydrogen phthalate, directly traceable to primary SRM from NIST / PTB pH(S) = 4.005 (25°C) Certipur®	107200	19
Certified sec. standard reference buffer solution potassium tetraoxalate dihydrate, directly traceable to primary SRM from NIST / PTB pH(S) = 1.681 (25°C) Certipur®	107204	19
Cesium chloride 99.995 Suprapur®	102039	23
Cesium chloride extra pure	102041	12
Cesium chloride for analysis EMSURE®	102038	12
Cesium ICP Standard traceable to SRM from NIST CsNO3 in HNO32 – 3 % 1000 mg/l Cs Certipur®	170310	18
Cesium nitrate 99+	102856	12
Cesium standard solution traceable to SRM from NIST CsNO3 in HNO3 0.5 mol/l 1000 mg/l Cs Certipur®	170212	18
Charcoal activated for analysis	102186	24
Charcoal activated granular about 1.5 mm extra pure food grade	102514	24
Charcoal activated powder extra pure food grade	102184	24
Charcoal activated pure	102183	24
Charcoal wood powder	102204	24
Chemizorb® granules absorbent for spilled liquids	101568	24
Chemizorb® H* absorbent and neutralizer for spilled acids, with indicator	101595	24
Chemizorb® HF absorbent and neutralizer for spilled hydrofluoric acid, with indicator	101591	24
Chemizorb® Hg reagents and accessories for absorbent for mercury	112576	24
Chemizorb® Hg reagents refill pack for Ord. No. 1.12576.0001	101569	24
Chemizorb® OH <sup>-</sup> absorbent and neutralizer for spilled alkalis, with indicator	101596	24
Chemizorb® powder absorbent for spilled liquids	102051	24
Chloride standard for 1000 mg Cl <sup>-</sup> , (HCl in $H_20$ ) Titrisol <sup>®</sup>	109871	18
Chloride standard solution traceable to SRM from NIST NaCl in $\rm H_2O~1000~mg/l~Cl^-$ Certipur®	119897	20
Chlorophenol red indicator	103024	29
Chromate standard solution traceable to SRM of NIST $K_2CrO_4$ in $H_2O$ 1000 mg/l $CrO_4^{2-}$ Certipur®	119780	20
Chromium ICP standard traceable to SRM from NIST Cr(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 $\%$ 1000 mg/l Cr Certipur®	170312	18
Chromium ICP standard traceable to SRM from NIST Cr(NO3)3 in HNO3 2 - 3 % 10000 mg/l Cr Certipur®	170374	18
Chromium standard 1000 mg Cr, (CrCl $_3$ in 4.2 % HCl) Titrisol®	109948	18
Chromium standard solution traceable to SRM from NIST Cr(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 0.5 mol/l 1000 mg/l Cr Certipur®	119779	18
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Cobalt ICP standard traceable to SRM from NIST Co(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 % 10000 mg/l Co Certipur®	170375	18
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CombiTitrant 5 one component reagent for volumetric Karl Fischer titration 1 ml / approx. 5 mg $H_2O$ Apura®	188005	15
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Potassium hydroxide solution c(KOH) = 0.5 mol/l (0.5 N) Titripur®	111586	14
Potassium hydroxide solution c(KOH) = 1 mol/l (1 N) Titripur®	109108	14
Potassium hydroxide solution for 1000 ml, c(KOH) = 0.1 mol/l (0.1 N) Titrisol®	109921	14
Potassium hydroxide solution for 1000 ml, c(KOH) = 0.5 mol/l (0.5 N) Titrisol®	109919	14
Potassium hydroxide solution for 1000 ml, c(KOH) = 1 mol/l (1 N) Titrisol®	109918	14
Potassium hydroxide solution in ethanol c(KOH) = 0.1 mol/l (0.1 N) Titripur $^{\textcircled{m}}$	109115	14
Potassium hydroxide solution in ethanol c(KOH) = 0.5 mol/l (0.5 N) Titripur $^{\odot}$	109114	14
Potassium hydroxide solution in isopropanol acc. to DIN 51558 part 1 c(KOH) = 0.1 mol/l (0,1 N) Titripur®	105544	14
Potassium hydroxide solution in methanol c(KOH) = 0.1 mol/l (0.1 N) Titripur®	111587	14
Potassium hydroxide solution in methanol c(KOH) = 0.5 mol/l (0.5 N) Titripur®	109351	14
Potassium hydroxide solution in methanol c(KOH) = 2.0 mol/l (2 N) Titripur®	111787	14
Potassium ICP standard traceable to SRM from NIST KNO $_3$ in HNO $_3$ 2 – 3 $\%$ 1000 mg/l K Certipur®	170342	18
Potassium ICP standard traceable to SRM from NIST KNO $_3$ in HNO $_3$ 2 – 3 $\%$ 10000 mg/l K Certipur®	170377	18
Potassium iodate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	105051	7
Potassium iodate solution for 1000 ml, $c(KIO_3) = 1/60 \text{ mol/I} (0.1 \text{ N}) \text{ Titrisol}^{\textcircled{m}}$	109917	14
Potassium iodate volumetric standard, secondary reference material for iodometry, traceable to NIST SRM Certipur®	102404	20
Potassium iodide 99.995 Suprapur®	105044	23
Potassium iodide for analysis EMSURE® ISO, Reag. Ph Eur	105043	7
Potassium nitrate 99.995 Suprapur®	105065	23
Potassium nitrate for analysis EMSURE® ISO, Reag. Ph Eur	105063	7
Potassium nitrite cryst. for analysis EMSURE® ACS	105067	7
Potassium perchlorate for analysis EMSURE® ACS	105076	7
Potassium permanganate for analysis (max. 0.000005 % Hg) EMSURE® ACS	105084	7
	105084	9
Potassium permanganate for analysis EMSURE® ACS, Reag. Ph Eur	105082	7
Potassium permanganate solution c(KMnO₄) = 0.05 mol/l (0.25 N) Titripur®	480160	14
Potassium permanganate solution for 1000 ml, c(KMnO₄) = 0,002 mol/l (0.01 N) Titrisol®	109930	14
Potassium permanganate solution for 1000 ml, c(KMnO <sub>4</sub> ) = 0.02 mol/l (0.1 N) Titrisol <sup>®</sup>	109935	14
Potassium permanganate solution standardised against oxalate c(KMnO <sub>4</sub> ) = 0.02 mol/l (0.1 N) Titripur <sup>®</sup>	109122	14
Potassium permanganate solution standardised against sodium thiosulfate c(KMnO <sub>4</sub> ) = 0.02 mol/l (0.1 N) Titripur®	109121	14
Potassium peroxodisulfate for analysis (≤0.001 % N) EMSURE <sup>®</sup> ACS, Reag. Ph Eur	105092	7
	105092	9

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Potassium peroxodisulfate for analysis EMSURE®	105091	77
Potassium sodium tartrate tetrahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	108087	77
Potassium standard 1000 mg K, (KCl in H <sub>2</sub> 0) Titrisol®	109924	185
Potassium standard dissolved in oil c(K in standard oil) = 1 g/kg Certipur®	115054	187
Potassium standard solution traceable to SRM from NIST KNO3 in HNO3 0.5 mol/l 1000 mg/l K Certipur®	170230	186
Potassium sulfate 99.999 Suprapur®	105152	219
Potassium sulfate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	105153	77
Potassium sulfide small lumps for analysis EMSURE®	105134	77
Potassium tetrachloroplatinate(II) 99+	119235	126
Potassium tetraoxalate dihydrate certified secondary standard reference material for pH measurement; directly traceable to primary reference material from NIST / PTB, pH(S) = 1.681 (25°C) (DIN 19266) Certipur®	101961	192
Potassium thiocyanate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	105125	77
Praseodymium ICP standard traceable to SRM from NIST $Pr_2O_3$ in HNO $_3$ 2 – 3 % 1000 mg/l Pr Certipur®	170343	181
1-(2-Pyridylazo)-2-naphthol (PAN) metal indicator Reag. Ph Eur	107531	299
4-(2-Pyridylazo)resorcinol monosodium salt monohydrate metal indicator Reag. Ph Eur	107533	299
3,5-Pyrocatecholdisulfonic acid di-sodium salt monohydrate (Tiron) metal indicator	101922	299
Pyrogallol red metal indicator	107534	299
Quartz fine granular, washed and calcined for analysis	107536	284
Quinaldine red indicator Reag. Ph Eur	102282	292
Refractive index standard kit 1 2,2,4-Trimethyl-pentane / water, traceable to SRM from NIST and PTB n (20°C) = 1.3915 Certipur®	108962	209
Refractive index standard kit 2 Toluene / water, traceable to SRM from NIST and PTB n (20°C) = 1.4969 Certipur®	108961	209
Refractive index standard kit 3 1-Methyl-naphthalene / water, traceable to SRM from NIST and PTB n (20°C) = 1.6160 Certipur®	108963	209
Refractive index standard traceable to SRM from NIST and PTB 0° Brix Certipur®	500400	209
Refractive index standard traceable to SRM from NIST and PTB 5° Brix Certipur®	500405	209
Refractive index standard traceable to SRM from NIST and PTB 10° Brix Certipur®	500410	209
Refractive index standard traceable to SRM from NIST and PTB 20° Brix Certipur®	500420	209
Refractive index standard traceable to SRM from NIST and PTB 30° Brix Certipur®	500430	209
Refractive index standard traceable to SRM from NIST and PTB 40° Brix Certipur®	500440	209
Refractive index standard traceable to SRM from NIST and PTB 50° Brix Certipur®	500450	209
Refractive index standard traceable to SRM from NIST and PTB 60° Brix Certipur®	500460	209
Rhenium ICP standard traceable to SRM from NIST NH <sub>4</sub> ReO <sub>4</sub> in H <sub>2</sub> O 1000 mg/l Re Certipur <sup>®</sup>	170344	181
Rhodium ICP standard traceable to SRM from NIST Rh(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 % 1000 mg/l Rh Certipur <sup>®</sup>	170345	181
Rhodium standard 10 mg/l (Rhodium(III) nitrate in nitric acid 0.5 mol/l) internal standard for ICP-MS	108525	181
Rhodizonic acid di-sodium salt indicator for sulfate titration	106595	299
Rubidium chloride for analysis EMSURE®	107615	128
Rubidium ICP standard traceable to SRM from NIST RbNO3 in HNO3 2 – 3 % 1000 mg/l Rb Certipur®	170346	181
Ruthenium ICP standard RuCl <sub>3</sub> in HCl 7 % 1000 mg/l Ru Certipur®	170347	181
Ruthenium(III) chloride hydrate	119247	126

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Safranine O [C.I. 50240] for microscopy Certistain®	115948	29
Samarium ICP standard traceable to SRM from NIST $\rm Sm_2O_3$ in $\rm HNO_3$ 2 – 3 % 1000 mg/l Sm Certipur®	170348	18
Scandium ICP standard traceable to SRM from NIST $Sc_2O_3$ in $HNO_3$ 7 % 1000 mg/l Sc Certipur®	170349	18
Scandium standard solution traceable to SRM from NIST $Sc_2O_3$ in HNO <sub>3</sub> 1 mol/l 1000 mg/l Sc Certipur®	119513	18
Sea sand extra pure	107711	24
Sea sand purified by acid and calcined for analysis	107712	24
Selenium black 99+	107714	12
Selenium ICP standard traceable to SRM from NIST SeO <sub>2</sub> in HNO <sub>3</sub> 2 – 3 $\%$ 1000 mg/l Se Certipur®	170350	18
Selenium standard 1000 mg Se, (SeO $_2$ in HNO $_3$ ) Titrisol®	109915	18
Selenium standard solution traceable to SRM from NIST SeO <sub>2</sub> in HNO <sub>3</sub> 0.5 mol/l 1000 mg/l Se Certipur®	119796	18
SICAPENT® with indicator (phosphorus pentoxide drying agent for desiccators) on inert carrier material	100543	27
Silica gel beads, desiccant $\sim$ 2 – 5 mm	107735	9
	107735	27
Silica gel granules, desiccant ~ 0.2 - 1 mm	101905	27
Silica gel granules, desiccant $\sim$ 2 – 5 mm	101907	27
Silica gel with indicator (orange gel), granulate $\sim$ 1 – 3 mm	101969	9
	101969	27
Silica gel with moisture indicator (brown gel) desiccant $\sim$ 1 – 4 mm	101972	27
Silicic acid precipitated extra pure heavy	100656	28
Silicic acid precipitated extra pure light DAB	100657	28
Silicon anti-foaming agent	107743	9
	107743	28
Silicon ICP standard traceable to SRM from NIST (SiO $_2$ in NaOH 4 %) 10000 mg/l Si Certipur®	170386	18
Silicon ICP standard traceable to SRM from NIST SiO $_2$ in NaOH 2 $\%$ 1000 mg/l Si Certipur®	170365	18
Silicon standard 1000 mg Si, (SiCl $_4$ in 14 % NaOH) Titrisol $^{\textcircled{m}}$	109947	18
Silicon standard solution traceable to SRM from NIST acidic, $(NH_4)_2SiF_6$ in $H_2O$ 1000 mg/l Si Certipur®	112310	18
Silicon standard solution traceable to SRM from NIST SiO $_2$ in NaOH 0.5 mol/l 1000 mg/l Si Certipur®	170236	18
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Silicone high vacuum grease medium	107922	28
Silicone oil for oil baths up to 250°C	107742	28
Silver chloride 99+	119203	12
Silver diethyldithiocarbamate for analysis (reagent for arsenic and antimony) EMSURE® ACS, Reag. Ph Eur	101515	12
Silver ICP standard traceable to SRM from NIST AgNO $_3$ in HNO $_3$ 2 – 3 % 1000 mg/l Ag Certipur®	170352	18
Silver nitrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur,	101512	7
Silver nitrate solution c(AgNO <sub>3</sub> ) = 0.05 mol/l (0.05 N) Titripur®	111718	14
Silver nitrate solution c(AgNO <sub>3</sub> ) = 0.1 mol/l (0.1 N) Titripur <sup>®</sup>	109081	14
Silver nitrate solution c(AgNO <sub>3</sub> ) = 1 mol/l (1 N) Titripur®	109080	14
Silver nitrate solution for 1000 ml, $c(AgNO_3) = 0.1 \text{ mol/l} (0.1 \text{ N}) \text{ Titrisol}^{\otimes}$	109990	14
Silver oxide 99+	119208	12
Silver standard 1000 mg Ag, (AgNO $_3$ in HNO $_3$ 5 %) Titrisol®	109906	18
Silver standard solution traceable to SRM from NIST AgNO3 in HNO3 0.5 mol/l 1000 mg/l Ag Certipur®	119797	18

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Product	Cat. No.	Page
Silver wool for elemental analysis	101506	126
Sodalime pellets with indicator for analysis	106839	242
Sodalime with indicator, granules ~ 1 – 2.5 mm	106733	242
Sodium acetate anhydrous 99.99 Suprapur®	106264	233
Sodium acetate trihydrate for analysis indifferent to potassium permanganate EMSURE® ACS, ISO, Reag. Ph Eur	106267	78
Sodium ammonium hydrogen phosphate tetrahydrate for analysis EMSURE®	106682	78
Sodium arsenite solution $c(NaAsO_2) = 0.05 \text{ mol/l} (0,1 \text{ N}) \text{ Titripur}^{\textcircled{B}}$	106277	143
Sodium borohydride for analysis	106371	93
Sodium bromide 99.995 Suprapur®	106363	233
Sodium carbonate anhydrous 99.999 Suprapur®	106395	233
Sodium carbonate anhydrous for analysis EMSURE® ISO	106392	78
Sodium carbonate anhydrous, for analysis EMSURE® ACS, ISO, Reag. Ph Eur	106393	78
Sodium carbonate decahydrate for analysis EMSURE® ISO, Reag. Ph Eur	106391	78
Sodium carbonate solution for 1000 ml, $c(Na_2CO_3) = 0.05 \text{ mol/l} (0.1 \text{ N}) \text{ Titrisol}^{(0)}$	109940	143
Sodium carbonate volumetric standard, secondary reference material for acidimetry, traceable to NIST SRM Certipur®	102405	207
Sodium chloride 99.99 Suprapur®	106406	233
Sodium chloride for analysis EMSURE® ACS, ISO, Reag. Ph Eur	106404	78
Sodium chloride solution for 1000 ml, c(NaCl) = 0.1 mol/l (0.1 N) Titrisol®	109945	143
Sodium chloride volumetric standard, secondary reference material for argentometry, traceable to NIST SRM Certipur®	102406	207
Sodium dichromate dihydrate for analysis EMSURE® ACS	106336	78
Sodium dihydrogen phosphate anhydrous 99.99 Suprapur®	106370	233
Sodium dihydrogen phosphate dihydrate for analysis EMSURE® Reag. Ph Eur	106342	78
Sodium dihydrogen phosphate monohydrate for analysis EMSURE® ACS, Reag. Ph Eur	106346	78
Sodium disulfite (sodium metabisulfite) for analysis EMSURE® ACS, Reag. Ph Eur	106528	79
Sodium dithionite for analysis	106507	79
Sodium fluoride 99.99 Suprapur®	106450	233
Sodium fluoride for analysis EMSURE® ACS, ISO, Reag. Ph Eur	106449	79
Sodium formate for analysis EMSURE® ACS, Reag. Ph Eur	106443	79
Sodium hexanitrocobaltate(III) [sodium cobalt(III)nitrite] for analysis EMSURE® ACS, Reag. Ph Eur	102521	79
Sodium hydrogen carbonate / sodium carbonate certified secondary standard reference material for pH	101962	192
measurement; directly traceable to primary SRM from NIST / PTB, pH(S) = 10.014 (25°C) (DIN 19266) Certipur®		
Sodium hydrogen carbonate for analysis EMSURE® ACS	106329	79
Sodium hydrogen sulfate monohydrate for analysis EMSURE®	106352	79
Sodium hydroxide 1.000 l	109913	144
for the preparation of dilute volumetric solutions or for direct use c(NaOH) = 5 mol/l (5 N) Combi-Titrisol®		
Sodium hydroxide monohydrate 99.99 Suprapur®	106466	233
Sodium hydroxide pellets for analysis (max. 0.0002 % K) EMSURE® ACS, Reag. Ph Eur	106495	120
Sodium hydroxide pellets for analysis (max. 0.02 % K) EMSURE® ACS, Reag. Ph. Eur	106469	120
Sodium hydroxide pellets for analysis EMSURE® ISO	106498	120
Sodium hydroxide solution 1 mol/l prepared from raw materials acc. Ph Eur Titripur®	199060	144
Sodium hydroxide solution 1 more prepared from have matchais acc. The full httput	105589	226
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Product Sodium hydroxide solution about 32 % (for the determination of nitrogen) for analysis EMSURE®	Cat. No. 105590	Pag
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Sodium hydroxide solution about 32 % extra pure	105587	32
Sodium hydroxide solution $c(NaOH) = 0.02 \text{ mol/l} (0.02 \text{ N}) \text{ Titripur®}$	109142	14
Sodium hydroxide solution $c(NaOH) = 0.1 \text{ mol/l} (0.1 \text{ N}) \text{ Titripur®}$	109141	14
Sodium hydroxide solution c(NaOH) = 0.2 mol/l (0.2 N) Titripur®	109140	14
Sodium hydroxide solution c(NaOH) = 0.25 mol/l (0.25 N) Titripur®	109139	14
Sodium hydroxide solution $c(NaOH) = 0.33 \text{ mol/l} (1/3 \text{ N}) \text{ Titripur}^{\textcircled{B}}$	105595	14
Sodium hydroxide solution c(NaOH) = 0.5 mol/l (0.5 N) Titripur®	109138	14
Sodium hydroxide solution c(NaOH) = 1 mol/l (1 N) Titripur®	109137	14
Sodium hydroxide solution c(NaOH) = 2 mol/l (2 N) Titripur®	109136	14
Sodium hydroxide solution c(NaOH) = 4 mol/l (4 N) Titripur®	111584	14
Sodium hydroxide solution for 1000 ml, c(NaOH) = 0.01 mol/l (0.01 N) Titrisol®	109961	14
Sodium hydroxide solution for 1000 ml, c(NaOH) = 0.1 mol/l (0.1 N) Titrisol®	109959	14
Sodium hydroxide solution for 1000 ml, c(NaOH) = 0.25 mol/l (0.25 N) Titrisol®	109958	14
Sodium hydroxide solution for 1000 ml, $c(NaOH) = 0.5 \text{ mol/l} (0.5 \text{ N}) \text{ Titrisol}^{\textcircled{B}}$	109957	14
Sodium hydroxide solution for 1000 ml, c(NaOH) = 1 mol/l (1 N) Titrisol®	109956	14
Sodium hydroxide solution min. 10 % (1.11) for analysis EMSURE®	105588	12
Sodium hydroxide solution min. 27 % (1.30) for analysis (for the determination of nitrogen) EMSURE®	105591	12
Sodium hydroxide solution min. 45 % for analysis EMSURE®	111360	12
Sodium hydroxide solution prepared from raw materials acc. Ph Eur c(NaOH) = 6 mol/l (6 N) Titripur®	199062	14
Sodium hypochlorite solution (6 – 14 % active chlorine)	105614	10
Sodium ICP standard traceable to SRM from NIST NaNO3 in HNO3 2 – 3 % 1000 mg/l Na Certipur®	170353	18
Sodium ICP standard traceable to SRM from NIST NaNO $_3$ in HNO $_3$ 2 – 3 $\%$ 10000 mg/l Na Certipur®	170381	18
Sodium iodate for analysis EMSURE®	106525	7
Sodium iodide 99.99 Suprapur®	106519	23
Sodium iodide for analysis EMSURE® ACS, Reag. Ph Eur	106523	7
Sodium metaperiodate for analysis EMSURE® ACS, Reag. Ph Eur	106597	8
Sodium molybdate dihydrate for analysis EMSURE®	106521	8
Sodium nitrate 99.99 Suprapur®	106546	23
Sodium nitrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	106537	8
Sodium nitrite for analysis EMSURE® ACS, Reag. Ph Eur	106549	8
Sodium perchlorate monohydrate for analysis EMSURE®	106564	8
Sodium peroxide granular for analysis ACS, ISO	106563	21
Sodium peroxidisulfate for analysis EMSURE®	106609	8
Sodium rod diameter 2.5 cm (protective liquid: paraffin oil)	106260	12
Sodium salicylate for analysis EMSURE®	106601	8
Sodium standard 1000 mg Na, (NaCl in $H_2O$ ) Titrisol <sup>®</sup>	109927	18
Sodium standard dissolved in oil c(Na in standard oil) = 1 q/kg Certipur®	115058	18
Sodium standard dissolved in on citra in standard on $J = 1$ g/kg certipul Sodium standard solution traceable to SRM from NIST NaNO <sub>3</sub> in H <sub>2</sub> O 1000 mg/l Na Certipur <sup>®</sup>	119507	20
Sodium standard solution traceable to SRM from NIST NaNO <sub>3</sub> in HNO <sub>3</sub> 0.5 mol/l 1000 mg/l Na Certipur <sup>®</sup>	170238	18
Sodium sulfate anhydrous 99.99 Suprapur®	106647	23
Sodium sulfate anhydrous for analysis EMSURE® ACS, ISO, Reag. Ph Eur	106649	8 27

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Sodium sulfate anhydrous granulated for organic trace analysis EMSURE®	106639	8 27
Sodium sulfate anhydrous, coarse granules for analysis 0.63 – 2.0 mm EMSURE® ACS	106637	
	106637	27
Sodium sulfate decahydrate for analysis EMSURE® ACS, Reag. Ph Eur	106648	8
Sodium sulfite anhydrous for analysis EMSURE® Reag. Ph Eur	106657	8
Sodium tartrate dihydrate, water standard for volumetric Karl Fischer titration (volumetric standard) Apura®	106664	17
Sodium thiosulfate pentahydrate 99.999 Suprapur®	106509	23
Sodium thiosulfate pentahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	106516	8
Sodium thiosulfate solution $c(Na_2S_2O_3 5 H_2O) = 0.1 \text{ mol/l} (0.1 \text{ N}) \text{ Titripur}^{\otimes}$	109147	14
Sodium thiosulfate solution for 1000 ml, $c(Na_2S_2O_3) = 0,01 \text{ mol/l} (0.01 \text{ N}) \text{ Titrisol}^{\textcircled{s}}$	109909	14
Sodium thiosulfate solution for 1000 ml, $c(Na_2S_2O_3) = 0.1 \text{ mol/l} (0.1 \text{ N}) \text{ Titrisol}^{\otimes}$	109950	14
Sodium tungstate dihydrate for analysis EMSURE®	106673	14
Solvent Oils & Fats, solvent for volumetric Karl Fischer titration with two component reagents for oils and fats Apura®	188016	16
Solvent, solvent for volumetric Karl Fischer titration with two component reagents Apura®	188015	1!
Spectromelt® A 10 (di-lithium tetraborate)	110783	2
Spectromelt® A 100 di-lithium tetraborate	112630	2
Spectromelt® A 1000 Li₂B₄O7 + 0.07 % LiBr	113175	2
Spectromelt® A 12 66 % di-lithium tetraborate / 34 % lithium metaborate	111802	2
Spectromelt® A 14 51 % di-lithium tetraborate / 27 % lanthane(III) oxide / 12 % lithium metaborate / 10 % lithium fluoride	111724	2
Spectromelt® A 20 lithium metaborate	112996	2
Spectromelt® A 22 lithium metaborate / di-lithium tetraborate 22:12	124001	2
Spectromelt® A 50 50 % lithium tetraborate / 50 % lithium metaborate	117057	2
Spectromelt® B 10 di-Sodium tetraborate	106304	2
Spectromelt® C 10 grinding and tabletting aid for X-ray fluorescence spectroscopy	113990	2
Spectromelt® C 15 grinding and tabletting aid for X-ray fluorescence spectroscopy	117062	2
Spectromelt® C 15 MP grinding and tabletting aid micropowder for X-ray fluorescence spectroscopy	117074	2
Spectromelt® C 20 grinding and tabletting aid for X-ray fluorescence spectroscopy	113934	2
Standard oil for atomic absorption spectroscopy	113898	18
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Strontium chloride hexahydrate for analysis EMSURE® ACS	107865	1
Strontium ICP standard traceable to SRM from NIST Sr(NO $_3$ ) $_2$ in HNO $_3$ 2 – 3 % 1000 mg/l Sr Certipur®	170354	18
Strontium nitrate anhydrous 99.99 Suprapur®	107871	23
Strontium nitrate for analysis EMSURE®	107872	8
Strontium standard 1000 mg Sr, (SrCl <sub>2</sub> in 7 % HCl) Titrisol <sup>®</sup>	109993	18
Strontium standard solution traceable to SRM from NIST Sr(NO $_3$ ) $_2$ in HNO $_3$ 0.5 mol/l 1000 mg/l Sr Certipur®	119799	18
Succinic acid for analysis EMSURE®	100682	1
Sudan III, [C.I. 26100]	111747	29
Sulfate standard for 1000 mg SO <sub>4</sub> <sup>2-</sup> , (H <sub>2</sub> SO <sub>4</sub> in H <sub>2</sub> O) Titrisol <sup>®</sup>	109872	18
Sulfate standard solution traceable to SRM from NIST Na <sub>2</sub> SO <sub>4</sub> in H <sub>2</sub> O 1000 mg/l SO <sub>4</sub> <sup>2-</sup> Certipur <sup>®</sup>	119813	20
Sulfur for external use Ph Eur, BP	107983	12

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Sulfur ICP standard traceable to SRM from NIST $H_2SO_4$ in $H_2O$ 1000 mg/l S Certipur®	170355	18
Sulfur ICP standard traceable to SRM from NIST $H_2SO_4$ in $H_2O$ 10000 mg/l S Certipur®	170385	18
Sulfuric acid 1.000 l for the preparation of dilute volumetric solutions or for direct use $c(H_2SO_4) = 2.5 \text{ mol/l}$ (5 N) Combi-Titrisol <sup>®</sup>	109912	14
Sulfuric acid 100 % for conductivity measurements	112223	11
Sulfuric acid 25 % for analysis EMSURE®	100716	11
Sulfuric acid 40 % for determination of gas metabolism acc. to Knipping	109286	11
Sulfuric acid 62 % for analysis, for the determination of fat in cheese (d 1.52)	480531	11
Sulfuric acid 90 – 91 % for Gerber fat determination and determination of nitrates in milk	100729	11
Sulfuric acid 95 – 97 % for analysis (max. 0.005 ppm Hg) EMSURE® ACS, ISO, Reag. Ph Eur	100732	9
	100732	10
	100732	11
Sulfuric acid 95 – 97 % for analysis EMPARTA® ACS	101833	11
Sulfuric acid 95 – 97 % for analysis EMSURE® ISO	100731	10
	100731	11
	100731	27
Sulfuric acid 96 % for the determination of viscosity acc. to DIN EN ISO 307	108131	11
Sulfuric acid 96 % Suprapur®	100714	22
Sulfuric acid 96 % Ultrapur	101516	22
Sulfuric acid 98 % for analysis EMSURE®	112080	10
	112080	11
Sulfuric acid 98 % for the determination of nitrogen	100748	11
Sulfuric acid $c(H_2SO_4) = 0.05 \text{ mol/I} (0.1 \text{ N}) \text{ Titripur}^{(0)}$	109074	14
Sulfuric acid $c(H_2SO_4) = 0.25 \text{ mol/I} (0.5 \text{ N}) \text{ Titripur}^{\otimes}$	109073	14
Sulfuric acid $c(H_2SO_4) = 0.5 \text{ mol/l} (1 \text{ N}) \text{ Titripur}^{\otimes}$	109072	14
Sulfuric acid $c(H_2SO_4) = 2.5 \text{ mol/l} (5 \text{ N}) \text{ Titripur}^{(8)}$	480364	14
Sulfuric acid for 1000 ml, $c(H_2SO_4) = 0.005 \text{ mol/l} (0.01 \text{ N}) \text{ Titrisol}^{\textcircled{m}}$	109982	14
Sulfuric acid for 1000 ml, $c(H_2SO_4) = 0.05 \text{ mol/l} (0.1 \text{ N}) \text{ Titrisol}^{\textcircled{B}}$	109984	14
Sulfuric acid for 1000 ml, $c(H_2SO_4) = 0.5 \text{ mol/l} (1 \text{ N}) \text{ Titrisol}^{\circ}$	109981	9
	109981	14
Sulfuric acid fuming 65 % SO $_3$ extra pure	100720	11
Sulfurous acid 5 – 6 % SO $_2$ for analysis EMSURE®	100761	11
Tantalum ICP standard traceable to SRM from NIST (NH <sub>4</sub> ) <sub>2</sub> TaF <sub>7</sub> in H <sub>2</sub> O 1000 mg/I Ta Certipur®	170356	18
Tellurium ICP standard traceable to SRM from NIST $H_6$ TeO <sub>6</sub> in HNO <sub>3</sub> 2 – 3 % 1000 mg/l Te Certipur®	170357	18
Tellurium standard solution traceable to SRM from NIST $H_6$ TeO <sub>6</sub> in HNO <sub>3</sub> 0.5 mol/l 1000 mg/l Te Certipur®	119514	18
Terbium ICP standard traceable to SRM from NIST Tb(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 % 1000 mg/l Tb Certipur <sup>®</sup>	170358	18
Tetrachloroauric(III) acid trihydrate 99.5 % for analysis EMSURE®	101582	12
Test Kit for water determination according to Karl Fischer Apura®	188025	15
Test Kit Solvent for water determination acc. to Karl Fischer Apura®	188026	15
Test Kit Titrant for water determination acc. to Karl Fischer Apura®	188027	15
Tetramethylammonium hydroxide solution in 2-propanol / methanol	108124	14

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Tetra-n-butylammonium hydroxide solution in 2-propanol / methanol for titrations in nonaqueous media $c[(C_4H_9)_4NOH] = 0.1 mol/l (0.1 N) Titripur®$	109162	14
Tetraphenylarsonium chloride monohydrate for analysis EMSURE®	108150	12
tetra-Sodium diphosphate decahydrate for analysis EMSURE® ACS, Reag. Ph Eur	106591	7
Thallium ICP standard traceable to SRM from NIST TINO3 in HNO3 2 – 3 % 1000 mg/l Tl Certipur®	170359	18
Thallium standard solution traceable to SRM from NIST TINO $_3$ in HNO $_3$ 0.5 mol/l 1000 mg/l Tl Certipur®	119801	18
Thionine (acetate) [C.I. 52000] for microscopy Certistain®	115929	29
Thorin indicator for sulfate titration	108294	29
Thorium ICP standard traceable to SRM from NIST Th(NO <sub>3</sub> ) <sub>4</sub> in HNO <sub>3</sub> 2 – 3 % 10 mg/l Th Certipur <sup>®</sup>	170391	18
Thulium ICP standard traceable to SRM from NIST Tm(NO3)3 in HNO3 2 – 3 % 1000 mg/l Tm Certipur®	170361	18
Thymol blue indicator ACS, Reag. Ph Eur	108176	29
Thymolphthalein indicator ACS, Reag. Ph Eur	108175	29
Tin fine powder pure (particle size <71 μm)	107807	12
Tin foil about 0.04 mm thick	107826	12
Tin granulated for analysis (particle size about 4 mm) EMSURE® Reag. Ph Eur	107806	12
Tin ICP standard traceable to SRM from NIST SnCl $_4$ in HCl 7 $\%$ 1000 mg/l Sn Certipur®	170362	18
Tin standard solution traceable to SRM from NIST SnCl $_4$ in HCl 2 mol/l 1000 mg/l Sn Certipur $^{\circledast}$	170242	18
Tin(II) chloride dihydrate for analysis (max. 0.000001 % Hg) EMSURE®	107814	8
	107814	ę
Tin(II) chloride dihydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	107815	8
Tin(II) sulfate for analysis EMSURE®	107823	ε
Tin(IV) oxide extra pure	107818	12
Titan yellow [C.I. 19540] indicator Reag. Ph Eur	101307	29
Titanium ICP standard traceable to SRM from NIST (NH <sub>4</sub> ) <sub>2</sub> TiF <sub>6</sub> in H <sub>2</sub> O 1000 mg/l Ti Certipur <sup>®</sup>	170363	18
Titanium standard 1000 mg Ti, $(NH_4)_2 TiF_6$ in $H_2 O$ Titrisol®	109829	18
Titanium standard solution traceable to SRM from NIST (NH <sub>4</sub> ) <sub>2</sub> TiF <sub>6</sub> in H <sub>2</sub> O 1000 mg/l Ti Certipur <sup>®</sup>	170243	18
Titanium(IV) oxide for analysis EMSURE® Reag. Ph Eur	100808	12
Titrant 2, titrant for volumetric Karl Fischer titration with two component reagents 1 ml / approx. 2 mg $\rm H_2O$ Apura®	188011	15
Titrant 5, titrant for volumetric Karl Fischer titration with two component reagents 1 ml / approx. 5 mg H₂O Apura®	188010	15
Titriplex <sup>®</sup> III solution for 1000 ml, c(Na <sub>2</sub> -EDTA 2 $H_2O$ ) = 0.01 mol/l Titrisol <sup>®</sup>	108446	14
Titriplex <sup>®</sup> III solution for 1000 ml, c(Na <sub>2</sub> -EDTA 2 H <sub>2</sub> 0) = 0.1 mol/l Titrisol <sup>®</sup>	109992	14
Titriplex <sup>®</sup> III solution for metal titration c(Na <sub>2</sub> -EDTA $\cdot$ 2 H <sub>2</sub> O) = 0.1 mol/l Titripur <sup>®</sup>	108431	14
Titriplex® solution A for the determination of alkaline earth metals in water 1 ml / approx. 56 mg CaO / I using 100 ml of water	108419	14
Titriplex® solution B for the determination of alkaline earth metals in water 1 ml / approx. 10 mg CaO / I using 100 ml of water Apura®	108420	14
TOC standard solution acc. to EN 1484-H3 / DIN 38409-H3 as potassium hydrogen phthalate in water, stabilized 1000 mg/l Certipur®	109017	21
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т	Trichloroacetic acid for analysis EMSURE® ACS, Reag. Ph Eur	100807	113
	Trifluoromethanesulfonic acid in anhydrous acetic acid $c(CF_3SO_3H) = 0.1 mol/l Titripur®$	108450	145
	Tris(hydroxymethyl)aminomethane volumetric standard, secondary reference material for acidimetry, traceable to NIST Standard Reference Material (SRM) Certipur®	102408	207
	tri-Sodium citrate dihydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	106448	78
ŕ	tri-Sodium phosphate dodecahydrate for analysis EMSURE®	106572	80
ŕ	tri-Sodium phosphate dodecahydrate for analysis EMSURE® ACS, Reag. Ph Eur	106578	80
-	Triton® X-100 for analysis	108603	285
	Tungsten ICP standard traceable to SRM from NIST (NH <sub>4</sub> ) $_2$ WO $_4$ in H $_2$ O 1000 mg/I W Certipur®	170364	181
	Tungsten standard solution traceable to SRM from NIST (NH <sub>4</sub> ) <sub>2</sub> WO <sub>4</sub> in H <sub>2</sub> O 1000 mg/I W Certipur <sup>®</sup>	170244	186
	Tungstophosphoric acid hydrate cryst. extra pure	100582	113
	Tungstophosphoric acid hydrate for analysis EMSURE®	100583	113
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U	Uranine AP [C.I. 45350] concentrated for the examination of subterranean waters	108462	300
	Uranium ICP standard traceable to SRM from NIST UO <sub>2</sub> (NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 % 10 mg/l U Certipur <sup>®</sup>	170360	181
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1	UV-VIS Standard 6: Holmium oxide solution reference material for the wavelength according to Ph Eur	108166	205
v	Vanadium ICP standard traceable to SRM from NIST NH₄VO $_3$ in HNO $_3$ 15 % 10000 mg/l V Certipur®	170388	181
	Vanadium ICP standard traceable to SRM from NIST NH <sub>4</sub> VO <sub>3</sub> in HNO <sub>3</sub> 2 – 3 % 1000 mg/I V Certipur <sup>®</sup>	170366	181
,	Vanadium standard 1000 mg V, (VOSO <sub>4</sub> in 8.6 % $H_2SO_4$ ) Titrisol®	109994	185
,	Vanadium standard solution traceable to SRM from NIST NH $_4$ VO $_3$ in HNO $_3$ 0.5 mol/l 1000 mg/l V Certipur®	170245	186
,	Vanadium(V) oxide extra pure	100824	129
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,	Water standard 0.1 % Standard for coulometric Karl Fischer titration 1 g / approx. 1 mg H <sub>2</sub> O Apura®	188051	171
1	Water standard 1 % Standard for volumetric Karl Fischer titration 1 g / approx. 10 mg H <sub>2</sub> O Apura®	188052	171
1	Water standard 5 mg / ml (1 ml contains 5 mg $H_2$ 0) Apura®	109259	171
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	Yttrium ICP standard traceable to SRM from NIST Y(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 % 1000 mg/l Y Certipur <sup>®</sup>	170368	181
	Yttrium oxide 99+	112412	129
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	Zinc coarse powder for analysis suitable for filling of reductors, particle size about 0.3 – 1.5 mm EMSURE® Reag. Ph Eur	108756	129
	Zinc dust particle size <63 μm	108774	129
	Zinc granular extra pure particle size about 3 – 8 mm	108755	129
	Zinc granular for analysis, particle size about 3 – 8 mm EMSURE® ISO	108780	129
	Zinc ICP standard traceable to SRM from NIST Zn(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 % 1000 mg/l Zn Certipur®	170369	181
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	Zinc oxide for analysis EMSURE® ACS, Reag. Ph Eur	108849	129
	Zinc powder for analysis particle size <45 μm EMSURE®	108789	129
	Zinc standard 1000 mg Zn, (ZnCl $_2$ in 0.06 % HCl) Titrisol $^{\odot}$	109953	185
	Zinc standard solution traceable to SRM from NIST Zn(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 0.5 mol/l 1000 mg/l Zn Certipur®	119806	186
	Zinc sticks, triangular cross section about 8 mm for analysis EMSURE®	108782	129
	Zinc sulfate heptahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	108883	81
	Zinc sulfate solution $c(ZnSO_4) = 0.1 \text{ mol/l Titripur}^{\textcircled{B}}$	108879	145
	Zinc sulfate solution for 1000 ml, c(ZnSO <sub>4</sub> ) = 0.1 mol/l (0.1 M) Titrisol <sup>®</sup>	109991	145
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	Zincon for the photometric determination of copper and zinc	108739	299
	Zirconium ICP standard traceable to SRM from NIST, ZrOCl $_2$ in HCl 7 $\%$ 1000 mg/l Zr Certipur®	170370	181
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# Classical inorganic analysis

Merck Millipore offers an extensive range of acids, salts, caustic alkalis and special inorganic reagents for analytical use, mainly in wet chemical labs. EMSURE® is our designation for analytical reagents, where we place the greatest importance on the high quality of each individual product. Thus, quality assurance laboratories benefit from the international acceptance of our reagents, as well as safe analytical laboratory auditing. With our comprehensive product range, you enjoy the convenience of obtaining all your basic reagents from one trusted source.









# Salts

EMSURE<sup>®</sup> Salts for analytical use are manufactured under strictly controlled conditions at Merck Millipore facilities in Darmstadt, Germany. The key feature of these salts is their outstanding analytical purity. The product range contains a wide assortment of inorganic salts for use in qualitative and quantitative analysis.

# Acids

EMSURE® | EMPARTA® Merck Millipore's acids for analysis offer the highest possible quality, the greatest safety and most optimized packaging. Our products undergo strict quality checks using the most sensitive analytical instruments and methods. That is why you can rely on Merck Millipore's acids for analysis for a multitude of applications.

# Caustics and bases

EMSURE<sup>©</sup> Our premium quality caustic alkalis and bases are produced using specially selected raw materials. The product range includes sodium and potassium hydroxide pellets and the corresponding solutions, as well as ammonia solutions in various concentrations and quality grades. Please find the right caustic and base for your specific use.

# Metals and metal oxides

Merck Milipore's metal salts, metals and noble metals renowned high quality and purity are used in a multitude of applications in R&D laboratories, production departments and for quality control. These products are available in the grades yextra pure or y for analysis EMSURE® c.



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# Salts for analysis EMSURE®

EMSURE® inorganic salts are manufactured under strictly controlled conditions at Merck Millipore facilities in Darmstadt, Germany. Their outstanding analytical purity makes them the ideal choice for both qualitative and quantitative analysis. The product range offers a wide assortment of inorganic salts that are suitable for analyzing various substances and substance mixtures.



# Salts

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# Advantages

- Comprehensive product range
- Standardized high quality premium grade for reagents
- High batch-to-batch consistency
- Extensive Certificate of Analysis (CoA)
- Compliance with international standards, such as ISO, ACS and Reag. Ph Eur
- Salts are specified for pharmacopoeia analysis
- Reliability, flexibility and safety



www.merckmillipore.com/salts

# Characteristics



#### Safety and environment

In 2010, Merck Millipore established a new production facility for inorganic salts, enabling us to increase capacity by 50 percent. This expansion and modernization is our response to a growing global demand for superior quality and product safety. The new plant, is also our commitment to developing Darmstadt into a powerful center for Merck Millipore's chemical production.

#### Quality management

Merck Millipore is certified according to the DIN EN ISO 9001 quality management systems, as well as the ISO 14001 environmental management system.

#### Improved flowability

Many inorganic salts tend to cake naturally, forming hard lumps. This makes it difficult to extract the product from the pack and process it. Exact dosing is also very important for most analytical tests in order to obtain precise and reproducible results. Since chemical purity may not be altered, we looked for a solution that neither changed the promised specifications nor involved any anti-caking materials. Our answer: a new and improved production process that substantially reduces the clumping of substances.

Suprapur® For analysis EMSURE® extra pure

# Specification

- ACS Standards of the American Chemical Society
- ISO Standards of the International Organization for Standardization

Reag. Ph Eur Requirements for reagents specified by the European Pharmacopoeia

# Quality grades for your individual needs

#### Salts Suprapur®

- For instrumental analysis we offer a range of high purity salts Suprapur®
- For further information please have a look in chapter »High purity salts«

#### Salts for analysis EMSURE® with ACS declaration

ACS standards are currently the most recognized international regulations and specifications for analytical reagents. This is why Merck Millipore tests and releases reagents based on the specifications and analytical regulations of the ACS – in addition to other criteria. The 10th edition of ACS was released in 2006. Our reagent declaration always conforms to the latest version of ACS. Updates are published regularly online.

#### Salts for analysis EMSURE® specified for pharmacopoeia analysis

Production and quality control in the pharmaceutical industry are strictly regulated. With the product designation »for analysis EMSURE® ACS, Reag. Ph Eur« Merck Millipore offers an extensive range of reagents including salts for pharmacopoeia analysis. These products conform fully to the specifications described in the reagents section of the European Pharmacopoeia as well as that of the U.S. Pharmacopoeia. Thus, quality assurance laboratories in the pharmaceutical industry benefit from the international acceptance of these reagents and the security of analytical laboratory auditing.

#### Salts extra pure

# Categories

# Categories A-0

	Categories A-0	Definition				
	Category	Definition	Area of application			
Α	Acetates	Salts of acetic acid	Buffer substance			
	Antimonates	Antimony compounds with antimony in oxidation state +5	Special oxidizing agent			
В	Bismuthates	Bismuth compounds with bismuth in oxidation state +5	Special oxidizing agent			
	Bromates	Salts of bromic acid only stable in aqueous solution	Special oxidizing agent			
	Bromides	Salts of hydrobromic acid	Technical applications			
С	Carbamates	Salts of carbamic acid – stable »carbonate«	Neutralization of acids			
	Carbonates	Salts of carbonic acid	Buffer substance, application in neutralizing reactions			
	Chlorates	Salts of chloric acid	Special oxidizing agent			
	Chlorides	Salts of aqueous hydrogen chloride (hydrochloric acid)	Very important inorganic compounds in laboratory and production			
	Chromates/dichromates	Chromium compounds with chromium in oxidation state +6	Special oxidizing agent			
	Citrates	Salts of citric acid	Buffer substance			
	Cyanides	Salts of hydrogen cyanide	Complexing agent, cyaniding			
D	Difluorides	Acid salts of hydrofluoric acid	Etchant			
	Diiodates	Acid salts of iodic acid	Special oxidizing agent			
	Disulfates	Salts of thiosulfurous acid	Dissolution agent			
	Disulfites	Salts of pyrosulfurous acid	Special reducing agent			
	Dithionites	Salts of dithionous acid	Special reducing agent			
F	Fluorides	Salts of hydrofluoric acid	Reagent in complexometry, etchant			
	Formates	Salts of formic acid	Special reducing agent			
н	Hexacyanoferrates	Stable cyano complexes with iron(II) and iron(III)	Iron(II) / iron(III) detection			
	Hexanitrocobaltates	Stable cyano complex with cobalt	Potassium detection			
	Hydroxide	alkaline compounds	Starting substances for alkalis, neutralization of acids, etchants, dissolution agents			
I.	lodates	Salts of iodic acid	Special oxidizing agent, iodometry			
	lodides	Salts of hydriodic acid	lodometry and photography			
Μ	Molybdates	Molybdenum compounds with molybdenum in oxidation state +6	Phosphate detection			
Ν	Nitrates	Salts of nitric acid	Special oxidizing agent			
	Nitrites	Salts of nitrous acid	Laboratory reagent			
0	Oxalates	Salts of oxalic acid	Reducing agent in permanganometry			
Ρ	Pentacyanonitrosylferrates	Cyano complex with nitrosyl ion	Laboratory reagent			
	Perchlorates	Salts of perchloric acid	Special oxidizing agent, potassium detection			
	Periodates	Salts of periodic acid	Special oxidizing agent			
	Permanganates	Salts of unstable permanganic acid	Special oxidizing agent, permanganometry			
	Peroxides	Salts with peroxo groups	Special oxidizing agent			
	Peroxodisulfates	Disulfates with peroxo groups	Special oxidizing agent			
	Phosphates	Salts of phosphoric acid	Often used buffer substances			
	Phthalates	Salts of phthalic acid	Alkalimetry			

# Categories P-Z

	Category	Definition	Area of application				
S	Salicylates	Salts of salicylic acid	Laboratory reagent				
	Selenites	Salts of selenous acid	Special reducing agent, technical applications (e.g. trace element, free-radical scavenger)				
	Sulfates	Salts of sulfuric acid	Very important inorganic compound in laboratory and production				
	Sulfides	Salts of hydrosulfuric acid	H <sub>2</sub> S development, heavy metal precipitation				
	Sulfites	Salts of sulfurous acid	Special reducing agent				
Т	Tartrates	Salts of tartaric acid	Special marker in complexometry, buffer substance				
	Thiocyanates	Salts of unstable thiocyanic acid	Iron(III) detection				
	Thiosulfates	Salts of unstable thiosulfuric acid	lodometry and photography				
	Tungstates	Tungsten compounds with tungsten in oxidation state +6	Special oxidizing agent, Folin-Ciocalteus's phenol determination				
V	Vanadates	Vanadium compounds with vanadium in oxidation state +5	Special oxidizing agent, phosphate determination				



# Storage instructions

#### **Recommended storage instructions**

We recommend storing all inorganic salts – with the exception of the compounds listed here – in airtight containers in a dry place at room temperature when possible. Further storage instructions are included in our Safety Data Sheets.

#### Higher water of crystallization content

In case of temperature fluctuations, products with a higher water of crystallization content have a greater tendency to cake than anhydrous salts. In addition, many products tend to decompose when exposed to heat. These compounds should therefore be stored in a dry place that is as cool as possible or at room temperature (max. 25°C). Please follow the storage instructions on the labels of the products.

#### Store in a cool place [below 25°C]

Product	Cat. No.
Ammonium carbamate for analysis EMSURE®	101134
Ammonium fluoride for analysis EMSURE® ACS	101164
Ammonium iron(III) sulfate dodecahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	103776
Ammonium iron(II) sulfate hexahydrate for analysis EMSURE® ISO	103792
Ammonium peroxodisulfate for analysis EMSURE® ACS, Reag. Ph Eur	101201
Iron(III) chloride hexahydrate for analysis EMSURE® ACS, Reag. Ph Eur	103943
Iron(II) sulfate heptahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	103965
Magnesium hydroxide carbonate for analysis EMSURE®	105827
Manganese(II) sulfate tetrahydrate for analysis EMSURE®	102786
Sodium carbonate decahydrate for analysis EMSURE® ISO, Reag. Ph Eur	106391
di-Sodium hydrogen phosphate anhydrous for analysis particle size about 0.2 – 1 mm (~ 18 – 80 mesh ASTM) EMSURE®	106559
di-Sodium hydrogen phosphate dodecahydrate for analysis EMSURE® ISO, Reag. Ph Eur	106579
di-Sodium hydrogen phosphate heptahydrate for analysis EMSURE® ACS	106575
Sodium peroxidisulfate for analysis EMSURE®	106609
Sodium sulfate decahydrate for analysis EMSURE® ACS, Reag. Ph Eur	106648
Tin(II) chloride dihydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	107815
Tin(II) chloride dihydrate for analysis (max. 0.000001 % Hg) EMSURE®	107814

# Salts for analysis FAQ's

What enables Merck Millipore to provide highly specified salts for analytical use?

Our state-of-the-art production facilities in Darmstadt, and close cooperation with our customers.

Are Merck Millipore reagents for analytical use ACS certified?

Yes, Merck Millipore offers ACS validated reagents.

What kind of anti-caking additives does Merck Millipore use?

None. Due to the analytical purity of EMSURE® salts it is not possible to use anti-caking agents.

How do Merck Millipore salts achieve improved flowability?

Improved flowability is attained through optimization during the production process.



# Ordering information Salts | EMSURE®

### Salts A

Product	CAS No.	Chemical formula	Molar weight	Content	Packaging	Ord. No.
Aluminium ammonium sulfate dodeca- hydrate for analysis EMSURE® ACS	7784-26-1	$NH_4AI(SO_4)_2 \cdot 12 H_2O$	453.33 g/mol	500 g	Plastic bottle	1.01031.0500
Aluminium nitrate nonahydrate for	7784-27-2	AI(NO <sub>3</sub> ) · 9 H <sub>2</sub> O	375.13 g/mol	500 g	Plastic bottle	1.01063.0500
analysis EMSURE®		1	5.	50 kg	Fibre carton	1.01063.9050
Aluminium potassium sulfate	7784-24-9	KAI(SO <sub>4</sub> ) <sub>2</sub> · 12 H <sub>2</sub> O	474.39 g/mol	1 kg	Plastic bottle	1.01047.1000
dodecahydrate for analysis EMSURE® ACS, Reag. Ph Eur			-	25 kg	Fibre carton	1.01047.9025
Ammonium acetate for analysis	631-61-8	CH <sub>3</sub> COONH <sub>4</sub>	77.08 g/mol	500 g	Plastic bottle	1.01116.0500
EMSURE <sup>®</sup> ACS, Reag. Ph Eur				1 kg	Plastic bottle	1.01116.1000
				5 kg	Plastic bottle	1.01116.5000
				50 kg	Fibre carton	1.01116.9050
Ammonium amidosulfonate for analysis EMSURE® ACS, Reag. Ph Eur	7773-06-0	$H_2NSO_3NH_4$	114.13 g/mol	100 g	Plastic bottle	1.01220.0100
Ammonium bromide for analysis	12124-97-9	NH₄Br	97.94 g/mol	1 kg	Plastic bottle	1.01125.1000
EMSURE® ACS				25 kg	Fibre carton	1.01125.9025
Ammonium carbamate for analysis EMSURE®	1111-78-0	$H_2NCOONH_4$	78.07 g/mol	500 g	Plastic bottle	1.01134.0500
Ammonium carbonate for analysis	10361-29-2	$CH_6N_2O_2 \cdot CH_5NO_3$	157.13 g/mol	250 g	Plastic bottle	1.59504.0250
EMSURE® ACS, Reag. Ph Eur			_	1 kg	Plastic bottle	1.59504.1000
Ammonium cerium(IV) nitrate for	16774-21-3	$(NH_4)_2[Ce(NO_3)_6]$	548.22 g/mol	100 g	Plastic bottle	1.02276.0100
analysis EMSURE® ACS, Reag. Ph Eur			_	1 kg	Plastic bottle	1.02276.1000
Ammonium cerium(IV) sulfate dihydrate for analysis EMSURE® ACS	10378-47-9	$(NH_4)_4Ce(SO_4)_4 \cdot 2 H_2O$	632.55 g/mol	100 g	Plastic bottle	1.02273.0100
Ammonium chloride for analysis	12125-02-9	NH₄CI	53.49 g/mol	500 g	Plastic bottle	1.01145.0500
EMSURE® ACS, ISO, Reag. Ph Eur				1 kg	Plastic bottle	1.01145.1000
				5 kg	Plastic bottle	1.01145.5000
				25 kg	Fibre carton	1.01145.9025
				50 kg	Fibre carton	1.01145.9050
Ammonium dihydrogen phosphate for	7722-76-1	(NH <sub>4</sub> )H <sub>2</sub> PO <sub>4</sub>	115.02 g/mol	500 g	Plastic bottle	1.01126.0500
analysis EMSURE® ACS, Reag. Ph Eur				50 kg	Fibre carton	1.01126.9050
Ammonium fluoride for analysis	12125-01-8	NH₄F	37.04 g/mol	250 g	Plastic bottle	1.01164.0250
EMSURE® ACS				1 kg	Plastic bottle	1.01164.1000
				25 kg	Fibre carton	1.01164.9025
Ammonium heptamolybdate	12054-85-2	(NH <sub>4</sub> ) <sub>6</sub> Mo <sub>7</sub> O <sub>24</sub> · 4 H <sub>2</sub> O	1.235.86 g/mol	250 g	Plastic bottle	1.01182.0250
tetrahydrate for analysis EMSURE®	12001 00 2		5.	1 kg	Plastic bottle	1.01182.1000
ACS, ISO, Reag. Ph Eur				5 kg	Plastic bottle	1.01182.5000
di-Ammonium hydrogen citrate for	3012-65-5	$C_{s}H_{s}O_{7} \cdot 2 NH_{3}$	226.19 g/mol	500 g	Plastic bottle	1.01154.0500
analysis EMSURE® ACS, Reag. Ph Eur				2.5 kg	Plastic bottle	1.01154.2500
-				50 kg	Fibre carton	1.01154.9050
, 5 1 1	7783-28-0	(NH <sub>4</sub> ) <sub>2</sub> HPO <sub>4</sub>	132.05 g/mol	500 g	Plastic bottle	1.01207.0500
analysis EMSURE® ACS, Reag. Ph Eur	7700 00 7		400.10	50 kg	Fibre carton	1.01207.9050
Ammonium iron(III) sulfate	7783-83-7	$(NH_4)Fe(SO_4)_2 \cdot 12 H_2O$	482.19 g/mol	500 g	Plastic bottle	1.03776.0500
dodecahydrate for analysis EMSURE®				1 kg	Plastic bottle	1.03776.1000
ACS, ISO, Reag. Ph Eur				5 kg	Plastic bottle	1.03776.5000
				50 kg	Fibre carton	1.03776.9050

#### Salts A-C

Product	CAS No.	Chemical formula	Molar weight	Content	Packaging	Ord. No.
Ammonium iron(II) sulfate hexahydrate	7783-85-9	(NH <sub>4</sub> ) <sub>2</sub> Fe(SO <sub>4</sub> ) <sub>2</sub> · 6 H <sub>2</sub> O	392.14 g/mol	500 g	Plastic bottle	1.03792.0500
for analysis EMSURE® ISO				1 kg	Plastic bottle	1.03792.1000
				5 kg	Plastic bottle	1.03792.5000
				50 kg	Fibre carton	1.03792.9050
Ammonium nitrate	6484-52-2	NH <sub>4</sub> NO <sub>3</sub>	80.04 g/mol	500 g	Plastic bottle	1.01188.0500
for analysis EMSURE® ACS			-	1 kg	Plastic bottle	1.01188.1000
				5 kg	Plastic bottle	1.01188.5000
di-Ammonium oxalate monohydrate	6009-70-7	$(NH_4)_2C_2O_4 \cdot H_2O$	142.11 g/mol	250 g	Plastic bottle	1.01192.0250
for analysis EMSURE® ACS, ISO,			-	1 kg	Plastic bottle	1.01192.1000
Reag. Ph Eur						
Ammonium peroxodisulfate	7727-54-0	$(NH_4)_2S_2O_8$	228.19 g/mol	500 g	Plastic bottle	1.01201.0500
for analysis EMSURE® ACS,				1 kg	Plastic bottle	1.01201.1000
Reag. Ph Eur				5 kg	Plastic bottle	1.01201.5000
				25 kg	Fibre carton	1.01201.9025
				50 kg	Fibre carton	1.01201.9050
Ammonium sulfate for analysis	7783-20-2	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	132.14 g/mol	100 g	Plastic bottle	1.01217.0100
EMSURE® ACS, ISO, Reag. Ph Eur			5.	1 kg	Plastic bottle	1.01217.1000
-				5 kg	Plastic bottle	1.01217.5000
				25 kg	Fibre carton	1.01217.9025
				50 kg	Fibre carton	1.01217.9050
Ammonium thiocyanate for analysis	1762-95-4	NH₄SCN	76.11 g/mol	500 g	Plastic bottle	1.01213.0500
EMSURE® ACS, ISO, Reag. Ph Eur			gr	25 kg	Fibre carton	1.01213.9025
Barium acetate for analysis EMSURE® ACS	543-80-6	Ba(CH <sub>3</sub> COO) <sub>2</sub>	255.42 g/mol	500 g	Plastic bottle	1.01704.0500
Barium carbonate for analysis	513-77-9	BaCO <sub>3</sub>	197.34 g/mol	250 g	Plastic bottle	1.01714.0250
EMSURE <sup>®</sup> ACS, Reag. Ph Eur		5	5.	1 kg	Plastic bottle	1.01714.1000
Barium chloride dihydrate	10326-27-9	$BaCl_2 \cdot 2 H_2O$	244.28 g/mol	500 g	Plastic bottle	1.01719.0500
for analysis EMSURE® ACS, ISO,			5.	1 kg	Plastic bottle	1.01719.1000
Reag. Ph Eur				5 kg	Plastic bottle	1.01719.5000
				50 kg	Fibre carton	1.01719.9050
Barium hydroxide octahydrate	12230-71-6	$Ba(OH)_2 \cdot 8 H_2O$	315.48 g/mol	500 g	Plastic bottle	1.01737.0500
for analysis EMSURE® ACS, ISO, Reag. Ph Eur						
Barium nitrate for analysis EMSURE®	10022-31-8	Ba(NO <sub>3</sub> ) <sub>2</sub>	261.34 g/mol	500 g	Plastic bottle	1.01729.0500
ACS				50 kg	Fibre carton	1.01729.9050
Barium perchlorate anhydrous	13465-95-7	Ba(ClO <sub>4</sub> ) <sub>2</sub>	336.23 g/mol	250 g	Metal can	1.01738.0250
for analysis EMSURE®			-	1 kg	Metal can	1.01738.1000
Bismuth(III) nitrate alkaline for analysis EMSURE® Reag. Ph Eur	1304-85-4	Bi <sub>5</sub> O(OH) <sub>9</sub> (NO <sub>3</sub> ) <sub>4</sub>	1461.99 g/mol	100 g	Plastic bottle	1.01878.0100
Cadmium acetate dihydrate for analysis EMSURE®	5743-04-4	$(CH_3COO)_2Cd \cdot 2 H_2O$	266.52 g/mol	500 g	Plastic bottle	1.02003.0500
Cadmium sulfate hydrate for analysis EMSURE® ACS	7790-84-3	$3 \text{ CdSO}_4 \cdot 8 \text{ H}_2\text{O}$	769.51 g/mol	100 g	Plastic bottle	1.02027.0100
Calcium carbonate precipitated	471-34-1	CaCO <sub>3</sub>	100.09 g/mol	250 g	Plastic bottle	1.02066.0250
for analysis EMSURE® Reag. Ph Eur			-	1 kg	Plastic bottle	1.02066.1000
				50 kg	Fibre carton	1.02066.9050
Calcium carbonate precipitated for analysis of silicates EMSURE®	471-34-1	CaCO <sub>3</sub>	100.09 g/mol	500 g	Plastic bottle	1.02067.0500

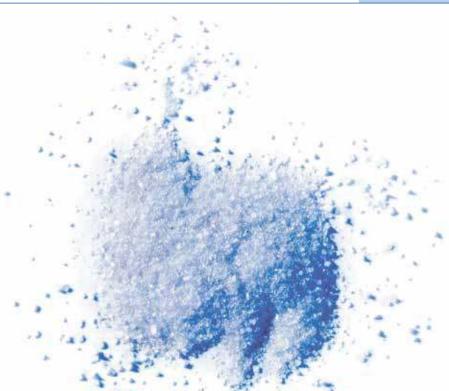
# Ordering information Salts | EMSURE®

### Salts C-H

Product	CAS No.	Chemical formula	Molar weight	Content	Packaging	Ord. No.
Calcium chloride dihydrate for analysis	10035-04-8	$CaCl_2 \cdot 2 H_2O$	147.02 g/mol	250 g	Plastic bottle	1.02382.0250
EMSURE® ACS, Reag. Ph Eur				500 g	Plastic bottle	1.02382.0500
				1 kg	Plastic bottle	1.02382.1000
				5 kg	Plastic bottle	1.02382.5000
				25 kg	Fibre carton	1.02382.9025
				50 kg	Fibre carton	1.02382.9050
Calcium hydroxide for analysis	1305-62-0	Ca(OH) <sub>2</sub>	74.09 g/mol	500 g	Plastic bottle	1.02047.0500
EMSURE <sup>®</sup> ACS, Reag. Ph Eur				1 kg	Plastic bottle	1.02047.1000
				50 kg	Fibre carton	1.02047.9050
Calcium nitrate tetrahydrate	13477-34-4	$Ca(NO_3)_2 \cdot 4 H_2O$	236.15 g/mol	500 g	Plastic bottle	1.02121.0500
for analysis EMSURE® ACS				50 kg	Fibre carton	1.02121.9050
Calcium sulfate dihydrate precipitated for analysis EMSURE®	10101-41-4	$\text{CaSO}_4 \cdot 2 \text{ H}_2\text{O}$	172.17 g/mol	500 g	Plastic bottle	1.02161.0500
Cerium(IV) sulfate tetrahydrate	10294-42-5	$Ce(SO_4)_2 \cdot 4 H_2O$	404.30 g/mol	25 g	Plastic bottle	1.02274.0025
for analysis EMSURE®		L L	5.	100 g	Plastic bottle	1.02274.0100
-				250 g	Plastic bottle	1.02274.0250
Chromium(III) nitrate nonahydrate for analysis EMSURE®	7789-02-8	$Cr(NO_3)_3 \cdot 9 H_2O$	400.15 g/mol	250 g	Plastic bottle	1.02481.0250
Chromium(III) potassium sulfate	7788-99-0	$KCr(SO_4)_2 \cdot 12 H_2O$	499.41 g/mol	250 g	Plastic bottle	1.01036.0250
dodecahydrate for analysis EMSURE® ACS, Reag. Ph Eur				200 9		
Cobalt(II) acetate tetrahydrate for analysis EMSURE® ACS	6147-53-1	$(CH_{3}COO)_{2}Co \cdot 4 H_{2}O$	249.08 g/mol	100 g	Plastic bottle	1.02529.0100
Cobalt(II) chloride hexahydrate for	7791-13-1	$CoCl_2 \cdot 6 H_2O$	237.93 g/mol	100 g	Plastic bottle	1.02539.0100
analysis EMSURE® ACS, Reag. Ph Eur				250 g	Plastic bottle	1.02539.0250
Cobalt(II) nitrate hexahydrate for	10026-22-9	$Co(NO_3)_2 \cdot 6 H_2O$	291.04 g/mol	100 g	Plastic bottle	1.02536.0100
analysis EMSURE®				250 g	Plastic bottle	1.02536.0250
Cobalt(II) nitrate hexahydrate for	10026-22-9	$Co(NO_3)_2 \cdot 6 H_2O$	291.04 g/mol	50 g	Plastic bottle	1.02554.0050
analysis (max. 0.001 % Ni) EMSURE® ACS, Reag. Ph Eur				250 g	Plastic bottle	1.02554.0250
Cobalt(II) sulfate heptahydrate	10026-24-1	$CoSO_4 \cdot 7 H_2O$	281.10 g/mol	100 g	Plastic bottle	1.02556.0100
for analysis EMSURE®				250 g	Plastic bottle	1.02556.0250
Copper(II) acetate monohydrate for analysis EMSURE® ACS	6046-93-1	$(CH_3COO)_2Cu \cdot H_2O$	199.65 g/mol	250 g	Plastic bottle	1.02711.0250
Copper(I) chloride	7758-89-6	CuCl	99 g/mol	250 g	Plastic bottle	1.02739.0250
for analysis EMSURE® ACS			-	5		
Copper(II) chloride dihydrate for	10125-13-0	$CuCl_2 \cdot 2 H_2O$	170.48 g/mol	250 g	Plastic bottle	1.02733.0250
analysis EMSURE <sup>®</sup> ACS, Reag. Ph Eur			5.	1 kg	Plastic bottle	1.02733.1000
Copper(II) nitrate trihydrate	10031-43-3	$Cu(NO_3)_2 \cdot 3 H_2O$	241.60 g/mol	250 g	Plastic bottle	1.02753.0250
for analysis EMSURE®		- 0-2 2	5.	1 kg	Plastic bottle	1.02753.1000
-				25 kg	Fibre carton	1.02753.9025
Copper(II) sulfate anhydrous	7758-98-7	CuSO <sub>4</sub>	159.61 g/mol	250 g	Plastic bottle	1.02791.0250
for analysis EMSURE®		+		1 kg	Plastic bottle	1.02791.1000
Copper(II) sulfate pentahydrate	7758-99-8	$CuSO_4 \cdot 5 H_2O$	249.68 g/mol	250 g	Plastic bottle	1.02790.0250
for analysis EMSURE® ACS, ISO,			9,	200 g	Plastic bottle	1.02790.1000
Reag. Ph Eur				5 kg	Plastic bottle	1.02790.5000

## Salts I-L

	Product	CAS No.	Chemical formula	Molar weight	Content	Packaging	Ord. No.
	Iron(III) chloride hexahydrate	10025-77-1	$\text{FeCl}_3 \cdot 6 \text{ H}_2 0$	270.33 g/mol	250 g	Plastic bottle	1.03943.0250
	for analysis EMSURE® ACS,				1 kg	Plastic bottle	1.03943.1000
	Reag. Ph Eur				25 kg	Plastic drum	1.03943.9025
	Iron(II) chloride tetrahydrate	13478-10-9	$\text{FeCl}_2 \cdot 4 \text{ H}_2 0$	198.83 g/mol	250 g	Plastic bottle	1.03861.0250
	for analysis EMSURE®				1 kg	Plastic bottle	1.03861.1000
					50 kg	Plastic drum	1.03861.9050
	Iron(III) nitrate nonahydrate	7782-61-8	$Fe(NO_3)_3 \cdot 9 H_2O$	404.00 g/mol	250 g	Plastic bottle	1.03883.0250
	for analysis EMSURE® ACS,				1 kg	Plastic bottle	1.03883.1000
	Reag. Ph Eur				50 kg	Steel drum	1.03883.9050
	Iron(III) phosphate for analysis calcined	10045-86-0	FePO <sub>4</sub>	150.82 g/mol	100 g	Plastic bottle	1.03935.0100
	(max. 0.001 % SO <sub>4</sub> ) EMSURE®				500 g	Plastic bottle	1.03935.0500
	lron(II) sulfate heptahydrate	7782-63-0	$FeSO_4 \cdot 7 H_2O$	278.02 g/mol	100 g	Plastic bottle	1.03965.0100
	for analysis EMSURE® ACS, ISO,				500 g	Plastic bottle	1.03965.0500
	Reag. Ph Eur				1 kg	Plastic bottle	1.03965.1000
					5 kg	Plastic bottle	1.03965.5000
					25 kg	Plastic drum	1.03965.9025
L	Lead(II) acetate trihydrate for analysis	6080-56-4	$(CH_{3}COO)_{2}Pb \cdot 3 H_{2}O$	379.34 g/mol	250 g	Plastic bottle	1.07375.0250
	EMSURE® ACS, Reag. Ph Eur				1 kg	Plastic bottle	1.07375.1000
	Lead(II) carbonate	598-63-0	PbCO <sub>3</sub>	267.21 g/mol	250 g	Plastic bottle	1.07381.0250
	for analysis EMSURE® ACS						
	Lead(II) hydroxide acetate anhydrous,	51404-69-4	$(CH_3COO)_2Pb \cdot Pb(OH)_2$	566.50 g/mol	1 kg	Plastic bottle	1.07414.1000
	for the analysis of sugar acc. to Horne EMSURE® ACS				30 kg	Fibre carton	1.07414.9030
	Lead(II) nitrate for analysis	10099-74-8	Pb(NO <sub>a</sub> ) <sub>a</sub>	331.2 g/mol	100 g	Plastic bottle	1.07398.0100
	EMSURE® ACS, Reag. Ph Eur			97	1 kg	Plastic bottle	1.07398.1000
	Lithium carbonate for analysis	554-13-2	Li <sub>2</sub> CO <sub>3</sub>	73.89 g/mol	250 g	Plastic bottle	1.05680.0250
	EMSURE® ACS, Reag. Ph Eur	-	<u> 2</u> 3	- <u>10</u> -			
	Lithium chloride for analysis	7447-41-8	LiCl	42.39 g/mol	100 g	Plastic bottle	1.05679.0100
	EMSURE® ACS, Reag. Ph Eur				250 g	Plastic bottle	1.05679.0250
	Lithium sulfate monohydrate for	10102-25-7	$Li_2SO_4 \cdot H_2O$	127.96 g/mol	250 g	Plastic bottle	1.05694.0250
	analysis EMSURE® ACS, Reag. Ph Eur						



# Ordering information Salts | EMSURE®

### Salts M

Product	CAS No.	Chemical formula	Molar weight	Content	Packaging	Ord. No.
Magnesium acetate tetrahydrate for	16674-78-5	$(CH_3COO)_2Mg \cdot 4 H_2O$	214.46 g/mol	250 g	Plastic bottle	1.05819.0250
analysis EMSURE® ACS, Reag. Ph Eur				1 kg	Plastic bottle	1.05819.1000
				50 kg	Fibre carton	1.05819.9050
Magnesium chloride hexahydrate	7791-18-6	MgCl <sub>2</sub> · 6 H <sub>2</sub> O	203.30 g/mol	250 g	Plastic bottle	1.05833.0250
for analysis EMSURE® ACS, ISO,			5.	1 kg	Plastic bottle	1.05833.1000
Reag. Ph Eur				5 kg	Plastic bottle	1.05833.5000
				25 kg	Fibre carton	1.05833.9025
				50 kg	Fibre carton	1.05833.9050
Magnesium hydroxide carbonate	12125-28-9	~ 4 MgCO <sub>3</sub> · Mg(OH) <sub>2</sub> ·	485 g/mol	250 g	Plastic bottle	1.05827.0250
for analysis EMSURE®	12120 20 0	5 H <sub>2</sub> O	ioo gimoi	1 kq	Plastic bottle	1.05827.1000
Magnesium nitrate hexahydrate for	13446-18-9	$Mg(NO_3)_2 \cdot 6 H_2O$	256.41 g/mol	500 g	Plastic bottle	1.05853.0500
analysis EMSURE® ACS, Reag. Ph Eur	13440-10-3		230.41 g/mor	25 kg	Plastic drum	1.05853.9025
Magnesium perchlorate hydrate [about	64010 42 0	$Mg(ClO_4)_2 \cdot x H_2O$		100 g	Metal can	1.05874.0100
83 % Mg(ClO <sub>4</sub> ) <sub>2</sub> ] for analysis EMSURE <sup>®</sup>			-		Metal can	
		Maco	120.27 - /	500 g		1.05874.0500
Magnesium sulfate anhydrous for analysis EMSURE®	7487-88-9	MgSO₄	120.37 g/mol	1 kg	Glass bottle	1.06067.1000
•	10024 00 0	M-50 740	246.40	25 kg	Plastic drum	1.06067.9025
Magnesium sulfate heptahydrate for	10034-99-8	$MgSO_4 \cdot 7 H_2O$	246.48 g/mol	500 g	Plastic bottle	1.05886.0500
nalysis EMSURE® ACS, Reag. Ph Eur				1 kg	Plastic bottle	1.05886.1000
				5 kg	Plastic bottle	1.05886.5000
				50 kg	Fibre carton	1.05886.9050
Manganese(II) chloride dihydrate	20603-88-7	$MnCl_2 \cdot 2 H_2O$	161.87 g/mol	100 g	Plastic bottle	1.05934.0100
for analysis EMSURE®				1 kg	Plastic bottle	1.05934.1000
Manganese(II) chloride tetrahydrate	13446-34-9	$MnCl_2 \cdot 4 H_2O$	197.91 g/mol	100 g	Plastic bottle	1.05927.0100
for analysis EMSURE® ACS				1 kg	Plastic bottle	1.05927.1000
Manganese(II) nitrate tetrahydrate	20694-39-7	$Mn(NO_3)_2 \cdot 4 H_2O$	251.01 g/mol	500 g	Plastic bottle	1.05940.0500
for analysis EMSURE®				1 kg	Plastic bottle	1.05940.1000
				5 kg	Plastic bottle	1.05940.5000
Manganese(II) sulfate monohydrate	10034-96-5	$MnSO_4 \cdot 4 H_2O$	169.02 g/mol	250 g	Plastic bottle	1.05941.0250
spray-dried for analysis EMSURE®				25 kg	Fibre carton	1.05941.9025
ACS, Reag. Ph Eur						
Manganese(II) sulfate tetrahydrate	10101-68-5	$MnSO_4 \cdot H_2O$	223.06 g/mol	1 kg	Plastic bottle	1.02786.1000
for analysis EMSURE®				25 kg	Fibre carton	1.02786.9025
Mercury extra pure	7439-97-6	Hg	200.59 g/mol	250 g	Plastic bottle	1.04401.0250
				1 kg	Plastic bottle	1.04401.1000
Mercury for analysis and for	7439-97-6	Нд	200.59 g/mol	250 g	Plastic bottle	1.04403.0250
polarography EMSURE®				1 kg	Plastic bottle	1.04403.1000
Mercury(II) acetate for analysis	1600-27-7	Hg(CH <sub>3</sub> COO) <sub>2</sub>	318.68 g/mol	50 g	Plastic bottle	1.04410.0050
EMSURE <sup>®</sup> ACS, Reag. Ph Eur			-	250 g	Plastic bottle	1.04410.0250
Mercury(II) bromide for analysis	7789-47-1	HgBr <sub>2</sub>	360.39 g/mol	50 g	Plastic bottle	1.04421.0050
EMSURE® ACS, Reag. Ph Eur		<i>.</i>	<b>5.</b> -	250 g	Plastic bottle	1.04421.0250
Mercury(II) chloride extra pure fine	7487-94-7	HgCl <sub>2</sub>	271.5 g/mol	100 g	Plastic bottle	1.04417.0100
cryst.		J2	,	1 kq	Plastic bottle	1.04417.1000
Mercury(II) chloride for analysis	7487-94-7	HgCl <sub>2</sub>	271.5 g/mol	50 g	Plastic bottle	1.04419.0050
EMSURE® ACS, Reag. Ph Eur	. 10/ 07/		2, 1.0 g/mor	250 g	Plastic bottle	1.04419.0250
				-	Plastic bottle	1.04419.1000
Marauru(II) indida rad for analysis	7774 00 0	Hal	454 20 almal	1 kg		
Mercury(II) iodide red, for analysis	7774-29-0	Hgl <sub>2</sub>	454.39 g/mol	50 g	Plastic bottle	1.04428.0050
EMSURE® ACS, Reag. Ph Eur			15100 1 1	250 g	Plastic bottle	1.04428.0250
Mercury(II) iodide red, extra pure	7774-29-0	Hgl <sub>2</sub>	454.39 g/mol	100 g	Plastic bottle	1.04420.0100
				1 kg	Plastic bottle	1.04420.1000

#### Salts M-P

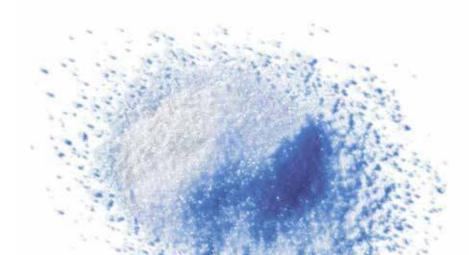
Product	CAS No.	Chemical formula	Molar weight	Content	Packaging	Ord. No.
Mercury(II) nitrate monohydrate for	7783-34-8	$Hg(NO_3)_2 \cdot H_2O$	342.62 g/mol	50 g	Plastic bottle	1.04439.0050
analysis EMSURE® ACS, Reag. Ph Eur			-	250 g	Plastic bottle	1.04439.0250
Mercury(II) oxide red extra pure	21908-53-2	HgO	216.58 g/mol	100 g	Plastic bottle	1.04465.0100
Mercury(II) oxide red, for analysis	21908-53-2	HgO	216.58 g/mol	50 g	Plastic bottle	1.04466.0050
EMSURE®				250 g	Plastic bottle	1.04466.0250
Mercury(II) sulfate extra pure	7783-35-9	HgSO₄	296.65 g/mol	100 g	Plastic bottle	1.04481.0100
		-	-	250 g	Plastic bottle	1.04481.0250
				1 kg	Plastic bottle	1.04481.1000
Mercury(II) sulfate for analysis	7783-35-9	HgSO₄	296.65 g/mol	50 g	Plastic bottle	1.04480.0050
EMSURE® ACS		-	-	250 g	Plastic bottle	1.04480.0250
Mercury(II) thiocyanate for analysis	592-85-8	Hg(SCN) <sub>2</sub>	316.76 g/mol	25 g	Glass bottle	1.04484.0025
EMSURE <sup>®</sup> , Reag. Ph Eur		-	-	100 g	Glass bottle	1.04484.0100
Nickel(II) chloride hexahydrate	7791-20-0	$NiCl_2 \cdot 6 H_2O$	237.70 g/mol	250 g	Plastic bottle	1.06717.0250
for analysis EMSURE® ACS			-	1 kg	Plastic bottle	1.06717.1000
Nickel(II) nitrate hexahydrate	13478-00-7	$Ni(NO_3)_2 \cdot 6 H_2O$	290.81 g/mol	100 g	Plastic bottle	1.06721.0100
for analysis EMSURE® ACS			-	250 g	Plastic bottle	1.06721.0250
				1 kg	Plastic bottle	1.06721.1000
Nickel(II) sulfate hexahydrate	10101-97-0	NiSO₄ · 6 H₂O	262.86 g/mol	100 g	Plastic bottle	1.06727.0100
for analysis EMSURE® ACS			-	250 g	Plastic bottle	1.06727.0250
				1 kg	Plastic bottle	1.06727.1000
Potassium bromate for analysis	7758-01-2	KBrO <sub>3</sub>	167 g/mol	100 g	Metal can	1.04912.0100
EMSURE® ACS, ISO, Reag. Ph Eur		-	-	250 g	Metal can	1.04912.0250
Potassium bromide for analysis	7758-02-3	KBr	119.00 g/mol	500 g	Plastic bottle	1.04905.0500
EMSURE® ACS, Reag. Ph Eur			-	-		
Potassium carbonate for analysis	584-08-7	K <sub>2</sub> CO <sub>3</sub>	138.21 g/mol	500 g	Plastic bottle	1.04928.0500
EMSURE® ACS, ISO, Reag. Ph Eur				1 kg	Plastic bottle	1.04928.1000
				50 kg	Fibre carton	1.04928.9050
Potassium chlorate	3811-04-9	KCIO <sub>3</sub>	122.55 g/mol	100 g	Metal can	1.04944.0100
for analysis EMSURE®				500 g	Metal can	1.04944.0500
Potassium chloride for analysis	7447-40-7	KCI	74.55 g/mol	500 g	Plastic bottle	1.04933.0500
(≤0.005 % Br) EMSURE® ACS, ISO,						
Reag. Ph Eur						
Potassium chloride	7447-40-7	KCI	74.55 g/mol	250 g	Plastic bottle	1.04936.0250
for analysis EMSURE®				500 g	Plastic bottle	1.04936.0500
				1 kg	Plastic bottle	1.04936.1000
				5 kg	Plastic bottle	1.04936.5000
				50 kg	Fibre carton	1.04936.9050
Potassium chromate for analysis	7789-00-6	$K_2CrO_4$	194.19 g/mol	250 g	Plastic bottle	1.04952.0250
EMSURE® ACS, Reag. Ph Eur				1 kg	Plastic bottle	1.04952.1000
Potassium cyanide for analysis	151-50-8	KCN	65.12 g/mol	100 g	Plastic bottle	1.04967.0100
EMSURE® ACS, Reag. Ph Eur				250 g	Plastic bottle	1.04967.0250
				1 kg	Plastic bottle	1.04967.1000
Potassium dichromate for analysis	7778-50-9	$K_2 Cr_2 O_7$	294.19 g/mol	500 g	Plastic bottle	1.04864.0500
EMSURE® ACS, ISO, Reag. Ph Eur			-	1 kg	Plastic bottle	1.04864.1000
				5 kg	Plastic bottle	1.04864.5000

# Ordering information Salts | EMSURE®

Product	CAS No.	Chemical formula	Molar weight	Content	Packaging	Ord. No.
Product Potassium dichromate						
for analysis (max. 0.000001 % Hg) EMSURE® ACS, ISO	7778-50-9	$K_2Cr_2O_7$	294.19 g/mol	500 g	Glass bottle	1.04865.0500
Potassium dihydrogen phosphate	7778-77-0	KH <sub>2</sub> PO <sub>4</sub>	136.08 g/mol	1 kg	Plastic bottle	1.04877.1000
for analysis (≤0.005 % Na)				12 kg	Plastic container	1.04877.9012
EMSURE® ACS, ISO, Reag. Ph Eur				25 kg	Fibre carton	1.04877.9025
Potassium dihydrogen phosphate	7778-77-0	KH <sub>2</sub> PO <sub>4</sub>	136.08 g/mol	250 g	Plastic bottle	1.04873.0250
for analysis EMSURE® ISO	///0 // 0		130.00 g/mor	230 g	Plastic bottle	1.04873.1000
				5 kg	Plastic bottle	1.04873.5000
				25 kg	Fibre carton	1.04873.9025
				2		
	7 60 60 7	KEO	254.22 almal	50 kg	Fibre carton	1.04873.9050
Potassium disulfate	7790-62-7	$K_2S_2O_7$	254.33 g/mol	1 kg	Plastic bottle	1.05107.1000
(Potassium pyrosulfate)				5 kg	Plastic bottle	1.05107.5000
for analysis EMSURE® ACS	40704	K C O		50 kg	Fibre carton	1.05107.9050
Potassium disulfite	16731-55-8	$K_2S_2O_5$	222.33 g/mol	500 g	Plastic bottle	1.05057.0500
for analysis EMSURE®				1 kg	Plastic bottle	1.05057.1000
				2.5 kg	Plastic bottle	1.05057.2500
Potassium fluoride	7789-23-3	KF	58.1 g/mol	250 g	Plastic bottle	1.04994.0250
for analysis EMSURE® ACS				1 kg	Plastic bottle	1.04994.1000
Potassium hexacyanoferrate(III)	13746-66-2	K <sub>3</sub> [Fe(CN) <sub>6</sub> ]	329.25 g/mol	100 g	Plastic bottle	1.04973.0100
for analysis EMSURE®				250 g	Plastic bottle	1.04973.0250
ACS, Reag. Ph Eur				1 kg	Plastic bottle	1.04973.1000
				50 kg	Fibre carton	1.04973.9050
Potassium hexacyanoferrate(II)	14459-95-1	$K_4[Fe(CN)_6] \cdot 3 H_2O$	422.39 g/mol	100 g	Plastic bottle	1.04984.0100
trihydrate for analysis EMSURE®				500 g	Plastic bottle	1.04984.0500
ACS, ISO, Reag. Ph Eur				50 kg	Fibre carton	1.04984.9050
Potassium hexahydroxoantimonate(V) cryst. for analysis EMSURE®	12208-13-8	K[Sb(OH) <sub>6</sub> ]	262.9 g/mol	100 g	Plastic bottle	1.05110.0100
Potassium hydrogen carbonate for analysis EMSURE® ACS	298-14-6	KHCO <sup>3</sup>	100.12 g/mol	500 g	Plastic bottle	1.04854.0500
Potassium hydrogen diiodate for analysis EMSURE®	13455-24-8	KH(IO <sub>3</sub> ) <sub>2</sub>	389.91 g/mol	50 g	Glass bottle	1.04867.0050
di-Potassium hydrogen phosphate	7758-11-4	K <sub>2</sub> HPO <sub>4</sub>	174.18 g/mol	1 kg	Plastic bottle	1.05104.1000
anhydrous for analysis EMSURE®			-	25 kg	Fibre carton	1.05104.9025
				50 kg	Fibre carton	1.05104.9050
di-Potassium hydrogen phosphate trihydrate buffer substance	16788-57-1	K2HPO4 * 3 H <sub>2</sub> O	228.23 g/mol	250 g	Glass bottle	1.19754.0250
for chromatography LiChropur®						
di-Potassium hydrogen phosphate	16788-57-1	$\rm K_{2}HPO_{4}\cdot 3\ H_{2}O$	228.23 g/mol	250 g	Plastic bottle	1.05099.0250
trihydrate for analysis EMSURE®				1 kg	Plastic bottle	1.05099.1000
				5 kg	Plastic bottle	1.05099.5000
				50 kg	Fibre carton	1.05099.9050
Potassium hydrogen phthalate	877-24-7	C <sub>8</sub> H <sub>5</sub> KO <sub>4</sub>	204.22 g/mol	250 g	Plastic bottle	1.04874.0250
for analysis EMSURE® Reag. Ph Eur				1 kg	Plastic bottle	1.04874.1000
				12 kg	Plastic container	1.04874.9012
Potassium hydrogen sulfate	7646-93-7	KHSO₄	136.17 g/mol	500 g	Plastic bottle	1.04885.0500
			5.	2.5 kg	Plastic bottle	1.04885.2500
for analysis EMSURE® Reag. Ph Eur						
Potassium iodate for analysis	7758-05-6	KIO <sub>3</sub>	214 g/mol	100 g	Plastic bottle	1.05051.0100

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	Product	CAS No.	Chemical formula	Molar weight	Content	Packaging	Ord. No.
	Potassium iodide for analysis EMSURE®		KI	166.00 g/mol	250 g	Plastic bottle	1.05043.0250
	ISO, Reag. Ph Eur	7001 11 0		100.00 g/mor	500 g	Plastic bottle	1.05043.0500
					1 kg	Plastic bottle	1.05043.1000
					2.5 kg	Plastic bottle	1.05043.2500
					2.5 kg	Fibre carton	1.05043.9025
	Potassium nitrate for analysis	7757-79-1	KNO,	101.10 g/mol	500 g	Plastic bottle	1.05063.0500
	EMSURE® ISO, Reag. Ph Eur	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	1.110 <sub>3</sub>	101.10 g/mor	1 kg	Plastic bottle	1.05063.1000
					5 kg	Plastic bottle	1.05063.5000
					25 kg	Fibre carton	1.05063.9025
	Potassium nitrite cryst. for analysis	7758-09-0	KNO <sub>2</sub>	85.11 g/mol	25 kg	Plastic bottle	1.05067.0250
-	EMSURE® ACS	7750-05-0	KNO <sub>2</sub>	65.11 g/mor	230 g	Plastic bottle	1.05067.1000
	di-Potassium oxalate monohydrate	6487-48-5	$K_2C_2O_4 \cdot H_2O$	184.24 g/mol	250 g	Plastic bottle	1.05073.0250
	for analysis EMSURE® ACS	0407-40-5	$R_2 C_2 O_4 + H_2 O_4$	104.24 g/mor	230 g	Plastic bottle	1.05073.1000
	Potassium perchlorate	7778-74-7	KClO₄	138.55 g/mol	250 g	Metal can	1.05076.0250
	for analysis EMSURE® ACS	///0-/4-/	KCIO <sub>4</sub>	136.55 g/mor	250 y	Metal can	1.05076.1000
	Potassium permanganate for analysis	7722-64-7	KMnO₄	158.03 g/mol	250 g	Glass bottle	1.05082.0250
	EMSURE® ACS, Reag. Ph Eur	//22-04-/	KWIIO <sub>4</sub>	156.05 g/mor	230 g	Glass bottle	1.05082.1000
	Potassium permanganate for analysis	7722-64-7	KMn0₄	158.03 g/mol	1 kg	Glass bottle	1.05082.1000
	(max. 0.000005 % Hg) EMSURE® ACS	//22-04-/	KIVITIO <sub>4</sub>	156.03 g/moi	тку	Glass Gottle	1.05084.1000
	Potassium peroxodisulfate	7727-21-1	K <sub>2</sub> S <sub>2</sub> O <sub>8</sub>	270.32 g/mol	250 g	Plastic bottle	1.05092.0250
	for analysis (≤0.001 % N)						
	EMSURE® ACS, Reag. Ph Eur						
	Potassium peroxodisulfate for analysis	7727-21-1	K <sub>2</sub> S <sub>2</sub> O <sub>8</sub>	270.32 g/mol	250 g	Plastic bottle	1.05091.0250
	EMSURE <sup>®</sup>				1 kg	Plastic bottle	1.05091.1000
	Potassium sodium tartrate	6381-59-5	$C_4H_4KNaO_6\cdot 4H_2O$	282.23 g/mol	500 g	Plastic bottle	1.08087.0500
	tetrahydrate for analysis EMSURE®				1 kg	Plastic bottle	1.08087.1000
	ACS, ISO, Reag. Ph Eur				5 kg	Plastic bottle	1.08087.5000
					12 kg	Plastic container	1.08087.9012
					50 kg	Fibre carton	1.08087.9050
	Potassium sulfate for analysis	7778-80-5	K <sub>2</sub> SO <sub>4</sub>	174.26 g/mol	500 g	Plastic bottle	1.05153.0500
	EMSURE® ACS, ISO, Reag. Ph Eur				1 kg	Plastic bottle	1.05153.1000
					5 kg	Plastic bottle	1.05153.5000
					25 kg	Fibre carton	1.05153.9025
	Potassium sulfide small lumps	39365-88-3	-	-	250 g	Plastic bottle	1.05134.0250
	for analysis EMSURE®				1 kg	Plastic bottle	1.05134.1000
	Potassium thiocyanate for analysis	333-20-0	KSCN	97.18 g/mol	250 g	Plastic bottle	1.05125.0250
	EMSURE <sup>®</sup> ACS, ISO, Reag. Ph Eur			-	1 kg	Plastic bottle	1.05125.1000
	-				50 kg	Fibre carton	1.05125.9050
					5		



# Ordering information Salts | EMSURE®

## Salts S

Product	CAS No.	Chemical formula	Molar weight	Content	Packaging	Ord. No.
Silver nitrate	7761-88-8	AgNO <sub>3</sub>	169.87 g/mol	25 g	Plastic bottle	1.01512.0025
for analysis EMSURE® ACS,				100 g	Plastic bottle	1.01512.0100
ISO, Reag. Ph Eur				250 g	Plastic bottle	1.01512.0250
				1 kg	Plastic bottle	1.01512.1000
Sodium acetate anhydrous	127-09-3	CH <sub>3</sub> COONa	82.03 g/mol	250 g	Plastic bottle	1.06268.0250
for analysis EMSURE®				1 kg	Plastic bottle	1.06268.1000
ACS, Reag. Ph Eur				2.5 kg	Plastic bottle	1.06268.2500
				12 kg	Plastic container	1.06268.9012
				25 kg	Fibre carton	1.06268.9025
Sodium acetate trihydrate for analysis	6131-90-4	NaCH <sub>3</sub> COO · 3 H <sub>2</sub> O	136.08 g/mol	500 g	Plastic bottle	1.06267.0500
indifferent to potassium permanganate		5 2	5.	1 kg	Plastic bottle	1.06267.1000
EMSURE® ACS, ISO, Reag. Ph Eur				5 kg	Plastic bottle	1.06267.5000
				50 kg	Fibre carton	1.06267.9050
Sodium ammonium hydrogen phosphate tetrahydrate for analysis EMSURE®	7783-13-3	$NaNH_4HPO_4 \cdot 4 H_2O$	209.07 g/mol	1 kg	Plastic bottle	1.06682.1000
, Sodium carbonate anhydrous,	497-19-8	Na <sub>2</sub> CO <sub>3</sub>	105.99 g/mol	1 kg	Plastic bottle	1.06393.1000
for analysis EMSURE®	437 13 0	142203	105.55 g/mor	50 kg	Fibre carton	1.06393.9050
ACS, ISO, Reag. Ph Eur				50 Kg		1.00353.5050
Sodium carbonate anhydrous	497-19-8	Na <sub>2</sub> CO <sub>3</sub>	105.99 g/mol	500 g	Plastic bottle	1.06392.0500
for analysis EMSURE® ISO				1 kg	Plastic bottle	1.06392.1000
				5 kg	Plastic bottle	1.06392.5000
				25 kg	Fibre carton	1.06392.9025
				50 kg	Fibre carton	1.06392.9050
Sodium carbonate decahydrate	6132-02-1	$Na_2CO_3 \cdot 10 H_2O$	286.14 g/mol	1 kg	Plastic bottle	1.06391.1000
for analysis EMSURE®				5 kg	Plastic bottle	1.06391.5000
ISO, Reag. Ph Eur				25 kg	Fibre carton	1.06391.9025
Sodium chloride	7647-14-5	NaCl	58.44 g/mol	500 g	Plastic bottle	1.06404.0500
for analysis EMSURE®				1 kg	Plastic bottle	1.06404.1000
ACS, ISO, Reag. Ph Eur				5 kg	Plastic bottle	1.06404.5000
				12 kg	Plastic container	1.06404.9012
				25 kg	Fibre carton	1.06404.9025
				50 kg	Fibre carton	1.06404.9050
tri-Sodium citrate dihydrate	6132-04-3	$C_6H_5Na_3O_7 \cdot 2 H_2O$	294.10 g/mol	500 g	Plastic bottle	1.06448.0500
for analysis EMSURE®		-05-3-7 2-		1 kg	Plastic bottle	1.06448.1000
ACS, ISO, Reag. Ph Eur				5 kg	Plastic bottle	1.06448.5000
				25 kg	Fibre carton	1.06448.9025
Sodium dichromate dihydrate	7789-12-0	$Na_2Cr_2O_7 \cdot 2 H_2O$	298.00 g/mol	250 g	Plastic bottle	1.06336.0250
for analysis EMSURE® ACS	7703 12 0		230.00 g/mor	230 g	Plastic bottle	1.06336.1000
Sodium dihydrogen phosphate	13472-35-0	$NaH_2PO_4 \cdot 2 H_2O$	156.02 g/mol	250 g	Plastic bottle	1.06342.0250
dihydrate for analysis EMSURE®	13772-33-0		130.02 y/mor	250 y	Plastic bottle	1.06342.0250
Reag. Ph Eur					Plastic bottle	1.06342.1000
-	10040 21 5		137.99 g/mol	2.5 kg	Plastic bottle	
Sodium dihydrogen phosphate monohydrate for analysis EMSURE®	10049-21-5	$NaH_2PO_4 \cdot H_2O$	137.99 g/moi	500 g		1.06346.0500
ACS, Reag. Ph Eur				1 kg	Plastic bottle	1.06346.1000
Aco, heag. In Eur				12 kg	Plastic container	1.06346.9012
				25 kg	Fibre carton	1.06346.9025
				50 kg	Fibre carton	1.06346.9050
tetra-Sodium diphosphate decahydrate	13472-36-1	$Na_4P_2O_7 \cdot 10 H_2O$	446.06 g/mol	500 g	Plastic bottle	1.06591.0500
for analysis EMSURE® ACS,				2.5 kg	Plastic bottle	1.06591.2500
Reag. Ph Eur				50 kg	Fibre carton	1.06591.9050

## Salts S

Salts S						
Product	CAS No.	Chemical formula	Molar weight	Content	Packaging	Ord. No.
Sodium disulfite	7681-57-4	$Na_2S_2O_5$	190.11 g/mol	100 g	Plastic bottle	1.06528.0100
(sodium metabisulfite)				500 g	Plastic bottle	1.06528.0500
for analysis EMSURE®				1 kg	Plastic bottle	1.06528.1000
ACS, Reag. Ph Eur				5 kg	Plastic bottle	1.06528.5000
				50 kg	Fibre carton	1.06528.9050
Sodium dithionite for analysis	7775-14-6	$Na_2S_2O_4$	174.11 g/mol	500 g	Metal can	1.06507.0500
				2.5 kg	Metal can	1.06507.2500
Sodium fluoride for analysis EMSURE®	7681-49-4	NaF	41.98 g/mol	250 g	Plastic bottle	1.06449.0250
ACS, ISO, Reag. Ph Eur				1 kg	Plastic bottle	1.06449.1000
				50 kg	Fibre carton	1.06449.9050
Sodium formate for analysis EMSURE®	141-53-7	HCOONa	68.01 g/mol	500 g	Plastic bottle	1.06443.0500
ACS, Reag. Ph Eur				50 kg	Fibre carton	1.06443.9050
Sodium hexanitrocobaltate(III)	13600-98-1	$Na_3[Co(NO_2)_6]$	403.93 g/mol	25 g	Plastic bottle	1.02521.0025
[sodium cobalt(III)nitrite] for analysis EMSURE® ACS, Reag. Ph Eur				100 g	Plastic bottle	1.02521.0100
Sodium hydrogen carbonate	144-55-8	NaHCO <sub>3</sub>	84.01 g/mol	500 g	Plastic bottle	1.06329.0500
for analysis EMSURE®		-	5.	1 kg	Plastic bottle	1.06329.1000
ACS, Reag. Ph Eur				5 kg	Plastic bottle	1.06329.5000
				12 kg	Plastic container	1.06329.9012
				25 kg	Plastic drum	1.06329.9025
				50 kg	Fibre carton	1.06329.9050
di-Sodium hydrogen phosphate	7558-79-4	Na <sub>2</sub> HPO <sub>4</sub>	141.96 g/mol	500 g	Plastic bottle	1.06586.0500
anhydrous for analysis EMSURE®			5.	1 kg	Plastic bottle	1.06586.1000
ACS, Reag. Ph Eur				2.5 kg	Plastic bottle	1.06586.2500
				12 kg	Plastic container	1.06586.9012
				50 kg	Fibre carton	1.06586.9050
di-Sodium hydrogen phosphate	7558-79-4	Na₂HPO₄	141.96 g/mol	500 g	Plastic bottle	1.06559.0500
anhydrous for analysis particle size about 0.2–1 mm (~18–80 mesh ASTM) EMSURE®			5	25 kg	Fibre carton	1.06559.9025
di-Sodium hydrogen phosphate dihydrate buffer substance for chromatography LiChropur®	10028-24-7	$Na_2HPO_4 \cdot 2 H_2O$	177.99 g/mol	250 g	Glass bottle	1.19753.0250
di-Sodium hydrogen phosphate	10028-24-7	$Na_2HPO_4 \cdot 2 H_2O$	177.99 g/mol	500 g	Plastic bottle	1.06580.0500
dihydrate for analysis EMSURE®		2 4 - 2		1 kg	Plastic bottle	1.06580.1000
. ,				5 kg	Plastic bottle	1.06580.5000
				25 kg	Fibre carton	1.06580.9025
				50 kg	Fibre carton	1.06580.9050
di-Sodium hydrogen phosphate	10039-32-4	Na,HPO₄ · 12 H₂O	358.14 g/mol	500 g	Plastic bottle	1.06579.0500
dodecahydrate for analysis EMSURE®		-24	5/1101	1 kg	Plastic bottle	1.06579.1000
ISO, Reag. Ph Eur				5 kg	Plastic bottle	1.06579.5000
-				25 kg	Fibre carton	1.06579.9025
di-Sodium hydrogen phosphate	7782-85-6	$Na_2HPO_4 \cdot 7 H_2O$	268.03 g/mol	23 kg	Plastic bottle	1.06575.1000
heptahydrate for analysis EMSURE® ACS	,, <u>0</u> 2-0 <b>3-</b> 0	1142111 04 7 1120	200.00 y/mor	25 kg	Fibre carton	1.06575.9025
Sodium hydrogen sulfate monohydrate for analysis EMSURE®	10034-88-5	$NaHSO_{4}\cdot H_{2}O$	138.07 g/mol	500 g	Plastic bottle	1.06352.0500
Sodium iodate for analysis EMSURE®	7681-55-2	NalO <sub>3</sub>	197.89 g/mol	100 g	Glass bottle	1.06525.0100
	7681-82-5	Nal	149.89 g/mol	100 g	Plastic bottle	1.06523.0100
Sodium iodide for analysis EMSURE®	/001-02-5	INdi	149.69 y/moi	100 g	Thus are obtained	
Sodium iodide for analysis EMSURE® Reag. Ph Eur, ACS	7001-02-5	Nai	149.69 g/moi	250 g	Plastic bottle	1.06523.0250

# Ordering information Salts | EMSURE®

## Salts S

Product	CAS No.	Chemical formula	Molar weight	Content	Packaging	Ord. No.
Sodium metaperiodate for analysis	7790-28-5	NalO <sub>4</sub>	213.89 g/mol	50 g	Plastic bottle	1.06597.0050
EMSURE® ACS, Reag. Ph Eur				250 g	Plastic bottle	1.06597.0250
				1 kg	Plastic bottle	1.06597.1000
Sodium molybdate dihydrate	10102-40-6	$Na_2MoO_4 \cdot 2 H_2O$	241.95 g/mol	100 g	Plastic bottle	1.06521.0100
for analysis EMSURE®				250 g	Plastic bottle	1.06521.0250
				1 kg	Plastic bottle	1.06521.1000
Sodium nitrate for analysis EMSURE®	7631-99-4	NaNO <sub>3</sub>	84.99 g/mol	500 g	Plastic bottle	1.06537.0500
ACS, ISO, Reag. Ph Eur				1 kg	Plastic bottle	1.06537.1000
				12 kg	Plastic container	1.06537.9012
				25 kg	Fibre carton	1.06537.9025
Sodium nitrite for analysis EMSURE®	7632-00-0	NaNO <sub>2</sub>	69.00 g/mol	100 g	Plastic bottle	1.06549.0100
ACS, Reag. Ph Eur		-	5.	500 g	Plastic bottle	1.06549.0500
-				12 kg	Plastic container	1.06549.9012
di-Sodium oxalate	62-76-0	$Na_2C_2O_4$	134 g/mol	250 g	Plastic bottle	1.06557.0250
for analysis EMSURE®		2 2 7	5.	1 kg	Plastic bottle	1.06557.1000
Sodium perchlorate monohydrate	7791-07-3	$NaClO_4 \cdot H_2O$	140.46 g/mol	100 g	Metal can	1.06564.0100
for analysis EMSURE®				500 g	Metal can	1.06564.0500
,				25 kg	Steel drum	1.06564.9025
Sodium peroxidisulfate	7775-27-1	Na <sub>2</sub> S <sub>2</sub> O <sub>8</sub>	238.11 g/mol	500 g	Plastic bottle	1.06609.0500
for analysis EMSURE®		11020208	200111 gj01	1 kg	Plastic bottle	1.06609.1000
				12 kg	Plastic container	1.06609.9012
				25 kg	Fibre carton	1.06609.9025
tri-Sodium phosphate dodecahydrate	10101-89-0	$Na_{3}PO_{4} \cdot 12 H_{2}O$	380.18 g/mol	23 kg	Plastic bottle	1.06572.1000
for analysis EMSURE®	10101 05 0	11031 04 12 1120	500.10 g/mor	5 kg	Plastic bottle	1.06572.5000
				25 kg	Fibre carton	1.06572.9025
tri-Sodium phosphate dodecahydrate	10101_89_0	$Na_3PO_4 \cdot 12 H_2O$	380.18 g/mol	23 kg 1 kg	Plastic bottle	1.06578.1000
for analysis EMSURE®	10101-03-0		500.10 g/mor	5 kg	Plastic bottle	1.06578.5000
ACS, Reag. Ph Eur				12 kg	Plastic container	1.06578.9012
				50 kg	Fibre carton	1.06578.9050
Sodium coliculate for analysis	E4 01 7		160 10 g/mal		Plastic bottle	
Sodium salicylate for analysis EMSURE®	54-21-7	$C_7H_5NaO_3$	160.10 g/mol	250 g	Plastic bottle	1.06601.0250
	7757 00 0	Νο 50	142.04 almal	1 kg		1.06601.1000
Sodium sulfate anhydrous, coarse granules for analysis 0.63-2.0 mm	7757-82-6	$Na_2SO_4$	142.04 g/mol	500 g	Plastic bottle	1.06637.0500
EMSURE® ACS				1 kg	Plastic bottle	1.06637.1000
	7757 00 0	N- 50	110.01 - /	25 kg	Fibre carton	1.06637.9025
Sodium sulfate anhydrous for analysis	7757-82-6	$Na_2SO_4$	142.04 g/mol	500 g	Plastic bottle	1.06649.0500
EMSURE® ACS, ISO, Reag. Ph Eur				1 kg	Plastic bottle	1.06649.1000
				5 kg	Plastic bottle	1.06649.5000
				25 kg	Fibre carton	1.06649.9025
Sodium sulfate anhydrous granulated for organic trace analysis EMSURE®	7757-82-6	$Na_2SO_4$	142.04 g/mol	500 g	Glass bottle	1.06639.0500
Sodium sulfate decahydrate for	7727-73-3	$Na_2SO_4 \cdot 10 H_2O$	322.19 g/mol	1 kg	Plastic bottle	1.06648.1000
analysis EMSURE® ACS, Reag. Ph Eur				25 kg	Fibre carton	1.06648.9025
Sodium sulfite anhydrous for analysis	7757-83-7	$Na_2SO_3$	126.04 g/mol	500 g	Plastic bottle	1.06657.0500
EMSURE® Reag. Ph Eur				1 kg	Plastic bottle	1.06657.1000
				5 kg	Plastic bottle	1.06657.5000
				50 kg	Fibre carton	1.06657.9050
di-Sodium tartrate dihydrate	6106-24-7	$C_4H_4Na_2O_6 \cdot 2H_2O$	230.08 g/mol	250 g	Plastic bottle	1.06663.0250
for analysis EMSURE®			5.	1 kg	Plastic bottle	1.06663.1000

#### Salts S-Z

Product	CAS No.	Chemical formula	Molar weight	Content	Packaging	Ord. No.
Sodium thiosulfate pentahydrate	10102-17-7	$Na_2S_2O_3 \cdot 5 H_2O$	248.21 g/mol	500 g	Plastic bottle	1.06516.0500
for analysis EMSURE® ACS, ISO,				1 kg	Plastic bottle	1.06516.1000
Reag. Ph Eur				5 kg	Plastic bottle	1.06516.5000
				25 kg	Fibre carton	1.06516.9025
Sodium tungstate dihydrate	10213-10-2	$Na_2WO_4 \cdot 2 H_2O$	329.86 g/mol	250 g	Plastic bottle	1.06673.0250
for analysis EMSURE®				1 kg	Plastic bottle	1.06673.1000
Strontium chloride hexahydrate	10025-70-4	$SrCl_2 \cdot 6 H_2O$	266.62 g/mol	250 g	Plastic bottle	1.07865.0250
for analysis EMSURE® ACS				1 kg	Plastic bottle	1.07865.1000
Strontium nitrate	10042-76-9	Sr(NO <sub>3</sub> ) <sub>2</sub>	211.63 g/mol	250 g	Plastic bottle	1.07872.0250
for analysis EMSURE®						
Tin(II) chloride dihydrate for analysis	10025-69-1	$SnCl_2 \cdot 2 H_2O$	225.63 g/mol	100 g	Glass bottle	1.07815.0100
EMSURE® ACS, ISO, Reag. Ph Eur				250 g	Glass bottle	1.07815.0250
				1 kg	Glass bottle	1.07815.1000
Tin(II) chloride dihydrate for analysis	10025-69-1	$SnCl_2 \cdot 2 H_2O$	225.63 g/mol	250 g	Glass bottle	1.07814.0250
(max. 0.000001 % Hg) EMSURE®				2.5 kg	Glass bottle	1.07814.2500
Tin(II) sulfate for analysis EMSURE®	7488-55-3	SnSO₄	214.77	250 g	Plastic bottle	1.07823.0250
Zinc acetate dihydrate	5970-45-6	(CH <sub>3</sub> COO) <sub>2</sub> Zn · 2 H <sub>2</sub> O	219.49	250 g	Plastic bottle	1.08802.0250
for analysis EMSURE® ACS				1 kg	Plastic bottle	1.08802.1000
Zinc chloride for analysis EMSURE®	7646-85-7	ZnCl <sub>2</sub>	136.30	250 g	Plastic bottle	1.08816.0250
ACS, ISO, Reag. Ph Eur				1 kg	Plastic bottle	1.08816.1000
				25 kg	Plastic drum	1.08816.9025
Zinc iodide for analysis EMSURE®	10139-47-6	Znl <sub>2</sub>	319.18	25 g	Glass bottle	1.08828.0025
				100 g	Glass bottle	1.08828.0100
Zinc nitrate tetrahydrate	19154-63-3	$Zn(NO_3)_2 \cdot 4 H_2O$	261.44	1 kg	Plastic bottle	1.08833.1000
for analysis EMSURE®						
Zinc sulfate heptahydrate for analysis	7446-20-0	$ZnSO_4 \cdot 7 H_2O$	287.54	500 g	Plastic bottle	1.08883.0500
EMSURE® ACS, ISO, Reag. Ph Eur				1 kg	Plastic bottle	1.08883.1000
				5 kg	Plastic bottle	1.08883.5000
				50 kg	Fibre carton	1.08883.9050



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#### Salts A-H

	Product	Chemical formula
А	Aluminum ammonium sulfate dodecahydrate for analysis EMSURE®	$NH_4AI(SO_4)_2 \cdot 12 H_2O$
	Aluminum nitrate nonahydrate for analysis EMSURE®	$AI(NO_3) \cdot 9 H_2O$
	Aluminum potassium sulfate dodecahydrate for analysis EMSURE® ACS	$KAI(SO_4)_2 \cdot 12 H_2O$
	Ammonium acetate for analysis EMSURE® ACS, Reag. Ph Eur	CH₃COONH₄
	Ammonium bromide for analysis EMSURE® ACS	NH₄Br
	Ammonium carbamate for analysis EMSURE®	H <sub>2</sub> NCOONH <sub>4</sub>
	Ammonium carbonate for analysis EMSURE® ACS, Reag. Ph Eur	$CH_6N_2O_2 \cdot CH_5NO_3$
	Ammonium cerium(IV) nitrate for analysis EMSURE® ACS, Reag. Ph Eur	(NH <sub>4</sub> ) <sub>2</sub> [Ce(NO <sub>3</sub> ) <sub>6</sub> ]
	Ammonium chloride for analysis EMSURE® ACS, ISO	NH₄CI
	Ammonium dihydrogen phosphate for analysis EMSURE® ACS, Reag. Ph Eur	(NH <sub>4</sub> )H <sub>2</sub> PO <sub>4</sub>
	Ammonium fluoride for analysis EMSURE® ACS	NH₄F
	Ammonium heptamolybdate tetrahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	(NH <sub>4</sub> ) <sub>6</sub> Mo <sub>7</sub> O <sub>24</sub> · 4 H <sub>2</sub> O
	di-Ammonium hydrogen phosphate for analysis EMSURE® ACS, Reag. Ph Eur	(NH <sub>4</sub> ) <sub>2</sub> HPO <sub>4</sub>
	Ammonium iodide for analysis EMSURE® ACS	NH₄I
	Ammonium iron(III) sulfate dodecahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	(NH <sub>4</sub> )Fe(SO <sub>4</sub> ) <sub>2</sub> · 12 H <sub>2</sub> O
	Ammonium iron(II) sulfate hexahydrate for analysis EMSURE®, ISO	$(NH_4)_2 Fe(SO_4)_2 \cdot 6 H_2O$
	Ammonium nitrate for analysis EMSURE® ACS, ISO	$NH_4NO_3$
	di-Ammonium oxalate monohydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	$(NH_4)_2C_2O_4 \cdot H_2O$
	Ammonium peroxodisulfate for analysis EMSURE® ACS, Reag. Ph Eur	(NH <sub>4</sub> ) <sub>2</sub> S <sub>2</sub> O <sub>8</sub>
	Ammonium sulfate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>
	Ammonium thiocyanate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	NH₄SCN
В	Barium acetate for analysis EMSURE® ACS	Ba(CH <sub>3</sub> COO) <sub>2</sub>
	Barium carbonate for analysis EMSURE® ACS, Reag. Ph Eur	BaCO <sub>3</sub>
	Barium chloride dihydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	$BaCl_2 \cdot 2 H_2O$
	Barium hydroxide octahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	$Ba(OH)_2 \cdot 8 H_2O$
	Barium nitrate for analysis EMSURE® ACS	Ba(NO <sub>3</sub> ) <sub>2</sub>
	Barium perchlorate anhydrous for analysis EMSURE®	Ba(CIO <sub>4</sub> ) <sub>2</sub>
С	Cadmium sulfate hydrate for analysis EMSURE® ACS, Reag. Ph Eur	$3 \text{ CdSO}_4 \cdot 8 \text{ H}_2\text{O}$
	Calcium carbonate precipitated for analysis EMSURE®, Reag. Ph Eur	
	Calcium chloride dihydrate cryst. for analysis EMSURE® ACS, Reag. Ph Eur	$CaCl_2 \cdot 2 H_2O$
	Calcium hydroxide for analysis EMSURE®, Reag. Ph Eur	Ca(OH) <sub>2</sub>
	Calcium nitrate tetrahydrate for analysis EMSURE® ACS	$Ca(NO_3)_2 \cdot 4 H_2O$
	Calcium sulfate dihydrate precipitated for analysis EMSURE® ACS	$CaSO_4 \cdot 2 H_2O$
	Cerium(II) sulfate tetrahydrate for analysis EMSURE®	$Ce(SO_4)_2 \cdot 4 H_2O$
	Chromium(III) nitrate nonahydrate for analysis EMSURE®	$Cr(NO_3)_3 \cdot 9 H_2O$
	Chromium(III) potassium sulfate dodecahydrate for analysis EMSURE® ACS, Reag. Ph Eur	$\text{KCr}(\text{SO}_4)_2 \cdot 12 \text{ H}_2\text{O}$
	Cobalt(II) acetate tetrahydrate for analysis EMSURE® ACS	$(CH_{3}COO)_{2}Co \cdot 4 H_{2}O$
	Cobalt(II) chloride hexahydrate for analysis EMSURE® ACS, Reag. Ph Eur	$CoCl_2 \cdot 6 H_2O$
	Cobalt(II) nitrate hexahydrate for analysis EMSURE®	$Co(NO_3)_2 \cdot 6 H_2O$
	Cobalt(II) nitrate hexahydrate for analysis EMSURE® (max. 0.001 % Ni), ACS, Reag. Ph Eur	$Co(NO_3)_2 \cdot 6 H_2O$
	Cobalt(II) sulfate heptahydrate for analysis EMSURE®	$CoSO_4 \cdot 7 H_2O$
	Copper(II) acetate monohydrate for analysis EMSURE® ACS	$(CH_3COO)_2Cu \cdot H_2O$
	Copper(I) chloride for analysis EMSURE® ACS	CuCl
	Copper(II) chloride dihydrate for analysis EMSURE® ACS, Reag. Ph Eur	$CuCl_2 \cdot 2 H_2O$
	Copper(II) nitrate trihydrate for analysis EMSURE®	$Cu(NO_3)_2 \cdot 3 H_2O$
	Copper(II) sulfate anhydrous for analysis EMSURE®	CuSO₄
	Copper(II) sulfate pentahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	$CuSO_4 \cdot 5 H_2O$
	- means: no values available	

 Solubility Fal						Soturated and	ution [20°C]	Cot No	Dogo
Solubility [g/ 0°C	100 g H <sub>2</sub> 0] at 20°C	40°C	60°C	80°C	100°C	Saturated sol	ution [20°C] Density	Cat. No.	Page
2.6	6.6	12.4	21.1	35.2	100 C 109.2 (95°C)	6.2	1.0459 (15.5°C)	101031	70
61	75.4	89	108	-	-	43	-	101063	70
2.96	6.01	13.6	33.3	72	109.0 (90°C)	5.67	1.053	101003	70
2.90 148.0 (4°C)	-	-	-	-	-	-	-	101047	70
60.6	- 75.5	91.1	107.8	126.7	- 145.6	43.9	_	1011125	70
-	73.5	-	-	-	-	-	_	101123	70 (68)
-	32.0	-	_	-	-	_	_	159504	70 (00)
_	141	-	_	_	-	-	-	102276	70
- 29.7	37.6	46	55.3	65.6	77.3	27.3	1.075	102276	70
23.7	36.8	40 56.7	82.9	120.7	174	26.9	-	101145	70
100	-	-	-		-	-	-	101120	70 (68)
-	- 40	-	_	-	-	-	_	101182	70 (88)
- 57.5	40 68.6	- 81.8	- 97.6	- 115.5		- 40.7	– 1.3436 (14.5°C)	101182	70
					-				
154.2 -	172.3 124.0 (25°C)	190.5	208.9	228.8	250.3 -	63.3 -	-	101173	70
		- 20 E						103776	70 (68)
17.8	26.9	38.5	53.4	72	-	21.2	1.18	103792	71 (68)
118.5 2 F	187.7	283	415	610	1000	65	1.308	101188	71
2.5	4.4	8	-	-	-	-	-	101192	71
58.2	-	-	-	-	-	-	-	101201	71 (68)
70.4	75.4	81.2	87.4	94.1	102	43	1.247	101217	71
115	163	235	347	-	-	62	-	101213	71
58	72	79	74	74	74	-	-	101704	71
-	0.002	-	-	-	-	-	-	101714	71
30.7	35.7	40.8	46.4	52.5	58.7	26.3	1.28	101719	71
1.5 -	3.5	8.2	21	-	-	3.4	1.04	101737	71
5	9.1	14.4	20.3	27.2	34.2	8.3	1.069	101729	71
-	198.5 (25°C)	-	-	-	-	-	-	101738	71
75.5	76.7	79.3	81.9	84.6	-	43.4	1.616	102027	71
-	0.0014	-	-	-	-	-	-	102066	71
-	-	128.1	136.8	147	159	-	-	102382	72
-	0.012 (18°C)	-	-	-	-	-	-	102047	72
101	129.4	196	-	-	-	56.4	-	102121	72
0.18	0.2	0.21	0.2	0.19	0.16	0.2	1.001	102161	72
-	-	3.8 (50°C)	-	-	-	-	-	102274	72
-	81	-	-	-	-	-	-	102481	72
-	25 (25°C)	-	-	-	-	-	-	101036	72
-	38	-	-	-	-	-	-	102529	72
41.9	53.6	69.5	-	-	-	38.4	1.49	102539	72
83.5	100	126	169.5	-	-	-	-	102536	72
-	0.2	-	-	-	-	-	-	102554	72
25.5	36.3	49.9	-	-	-	34.9	-	102556	72
-	7.2	-	-	-	-	-	-	102711	72
-	1.5 (25°C)	-	-	-	-	1.497 (25°C)	-	102739	72
70.6	77	83.8	91.2	99.2	107.9	43.5	1.55	102733	72
-	-	160	179	208	-	-	-	102753	72
25.5	36.2	48	60	70	83	-	_	102791	72

Salts I-P

	Product	Chemical formula
	Iron(III) chloride hexahydrate for analysis EMSURE® ACS, Reag. Ph Eur	$FeCl_3 \cdot 6 H_2O$
	Iron(II) chloride tetrahydrate for analysis EMSURE®	$FeCl_2 \cdot 4 H_2O$
	Iron(II) sulfate heptahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	FeSO₄ · 7 H₂O
L	Lead(II) acetate trihydrate for analysis EMSURE® ACS, Reag. Ph Eur	(CH <sub>3</sub> COO) <sub>2</sub> Pb · 3 H <sub>2</sub> O
	Lead(II) nitrate for analysis EMSURE® ACS, Reag. Ph Eur	Pb(NO <sub>3</sub> ) <sub>2</sub>
	Lithium carbonate for analysis EMSURE® ACS, Reag. Ph Eur	Li <sub>2</sub> CO <sub>3</sub>
	Lithium chloride for analysis EMSURE® ACS, Reag. Ph Eur	LiCl
	Lithium sulfate monohydrate for analysis EMSURE® ACS, Reag. Ph Eur	$Li_2SO_4 \cdot H_2O$
М	Magnesium acetate tetrahydrate for analysis EMSURE® ACS, Reag. Ph Eur	$(CH_3COO)_2Mg \cdot 4 H_2O$
	Magnesium chloride hexahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	$MgCl_2 \cdot 6 H_2O$
	Magnesium nitrate hexahydrate for analysis EMSURE® ACS, Reag. Ph Eur	$Mg(NO_3)_2 \cdot 6 H_2O$
	Magnesium perchlorate hydrate [about 83 % Mg(ClO <sub>4</sub> ) <sub>2</sub> ] for analysis EMSURE®	$Mg(CIO_4)_2 \cdot x H_2O$
	Magnesium sulfate anhydrous for analysis EMSURE®	MgSO₄
	Magnesium sulfate heptahydrate for analysis EMSURE® ACS, Reag. Ph Eur	$MgSO_4 \cdot 7 H_2O$
	Manganese(II) chloride dihydrate for analysis EMSURE®	$MnCl_2 \cdot 2 H_2O$
	Manganese(II) chloride tetrahydrate for analysis EMSURE® ACS	$MnCl_2 \cdot 4 H_2O$
	Manganese(II) nitrate tetrahydrate for analysis EMSURE®	$Mn(NO_3)_2 \cdot 4 H_2O$
	Manganese(II) sulfate tetrahydrate for analysis EMSURE®	$MnSO_4 \cdot 4 H_2O$
	Manganese(II) sulfate monohydrate spray-dried for analysis EMSURE® ACS, Reag. Ph Eur	$MnSO_4 \cdot H_2O$
	Mercury(II) acetate for analysis EMSURE® ACS, Reag. Ph Eur	Hg(CH <sub>3</sub> COO) <sub>2</sub>
	Mercury(II) bromide for analysis EMSURE® ACS, Reag. Ph Eur	HgBr₂
	Mercury(II) chloride extra pure fine cryst.	HgCl₂
	Mercury(II) chloride for analysis EMSURE® ACS, Reag. Ph Eur	HgCl₂
	Mercury(II) iodide red, for analysis EMSURE® ACS, Reag. Ph Eur	Hgl <sub>2</sub>
	Mercury(II) iodide red, extra pure	Hgl <sub>2</sub>
	Mercury(II) sulfate extra pure	HgSO₄
	Mercury(II) sulfate for analysis EMSURE® ACS	HgSO₄
	Mercury(II) thiocyanate for analysis EMSURE®, Reag. Ph Eur	Hg(SCN) <sub>2</sub>
Ν	Nickel(II) chloride hexahydrate for analysis EMSURE® ACS	$NiCl_2 \cdot 6 H_2O$
	Nickel(II) nitrate hexahydrate for analysis EMSURE® ACS	$Ni(NO_3)_2 \cdot 6 H_2O$
	Nickel(II) sulfate hexahydrate for analysis EMSURE® ACS	$NiSO_4 \cdot 6 H_2O$
Р	Potassium bromate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	KBrO <sub>3</sub>
	Potassium bromide for analysis EMSURE® ACS, Reag. Ph Eur	KBr
	Potassium carbonate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	K <sub>2</sub> CO <sub>3</sub>
	Potassium chlorate for analysis EMSURE®	KCIO <sub>3</sub>
	Potassium chloride for analysis EMSURE® (max. 0.005 % Br), ACS, ISO, Reag. Ph Eur	KCI
	Potassium chloride for analysis EMSURE®	KCI
	Potassium chromate for analysis EMSURE®	$K_2CrO_4$
	Potassium cyanide for analysis EMSURE® ACS, ISO, Reag. Ph Eur	KCN
	Potassium dichromate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub>
	Potassium dichromate for analysis EMSURE® (max. 0.000001 % Hg) ACS, ISO	K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub>
	Potassium dihydrogen phosphate for analysis EMSURE® (max. 0.005 $\%$ Na) ACS, ISO, Reag. Ph Eur	KH₂PO₄
	Potassium disulfite for analysis EMSURE®	$K_2S_2O_5$
	Potassium fluoride for analysis EMSURE® ACS	KF
	Potassium hexacyanoferrate(III) for analysis EMSURE® ACS, Reag. Ph Eur	K <sub>3</sub> [Fe(CN) <sub>6</sub> ]
	Potassium hexacyanoferrate(II) trihydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	$K_4[Fe(CN)_6] \cdot 3 H_2O$
	– means: no values available	

Colubility I de	00 - 11 01 - (					Calmenterland	4: [20%C]	Cat No	Demo
Solubility [g/1 0°C	00 g H <sub>2</sub> 0] at 20°C	40°C	60°C	80°C	100°C	Saturated solu Percentage	tion [20°C] Density	Cat. No.	Page
83.5	100	40 C	169.5 (56°C)	- 00 C	- 100 C	50	–	103943	73 (68)
-	62.4	68.6	78.3	-	-	-	_	103343	73 (00)
15.6	26.6	40.3	47.6	-	-	21	1.225	103965	73 (68)
-	44.3	-	-	_	_	-	-	107375	73
36.4	52.2	69.4	88	107.5	127.3	34.3	1.4	107398	73
-	1.3	-	-	-	-	1.31	-	105680	73
-	-	-	_	-	133	_	-	105679	73
36.2	34.8	33.5	32.3	31.5	31	25.6	1.23	105694	73
-	120.0 (15°C)	-	-	-	-	-	-	105819	74
52.8	54.6	57.5	60.7	65.9	72.7	41.2	1.388 (25°C)	105833	74
63.9	70.5	81.8	93.7	110.9	-	-	-	105853	74
-	50.0 (25°C)	-	-	-	_	_	_	105874	74
-	30.0	_	_	-	-	_	_	106067	74
-	35.6	45.4	-	-	-	26.25	1.31	105886	74
-	-	-	108.6	110.5	115	-	-	105934	74
63.6	73.6	88.7	106.0 (58.1°C)		-	42.4	1.499	105927	74 (68)
-	380.0	-	-	-	-	-	-	105940	74
63.4 (15°C)	65.2 (25°C)	70.7 (45°C)	-	-	_	_	_	105941	74
-	-	60	58.6	45.5	35.5	-	-	102786	74 (68)
-	40.0	-	-	-	-	-	-	104410	74
-	0.6 (25°C)	1	1.7	2.8	4.9	0.62 (25°C)	-	104421	74
4.3	6.6	9.6	13.9	24.2	54.1	6.2	1.052	104417	74
4.3	6.6	9.6	13.9	24.2	54.1	6.2	1.052	104419	74
-	0.005 (25°C)	-	-	-	-	-	-	104428	74
-	0.005 (25°C)	_	_	-	-	-	-	104420	75
-	0.04 (25°C)	_	_	-	-	_	-	104481	75
-	0.04 (25°C)	-	-	-	-	-	-	104480	75
-	0.07 (25°C)	-	-	-	-	-	-	104484	75
51.7	55.3	-	-	-	-	35.6	1.46	106717	75
79.2	94.1	118.8	-	-	-	48.5	-	106721	75
-	-	-	57	-	-	-	-	106727	75
3.1	6.8	13.1	22	33.9	49.7	6.4	1.048	104912	75
54	65.8	76.1	85.9	95.3	104.9	39.7	1.37	104905	75
105.5	111.5	117	127	140	156	-	-	104928	75
3.3	7.3	14.5	25.9	39.7	56.2	6.8	1.042	104944	75
28.1	34.2	40.3	45.6	51	56.2	25.5	1.174	104933	75
-	34.7	-	-	-	-	-	-	104936	75
59	63.7	67	70.9	75.1	79.2	38.9	1.378	104952	75
63	71.6	-	81	95	122	41.73	-	104967	75
4.7	12.5	26.3	45.6	73	103	11.1	1.077	104864	75
-	13.0	-	-	-	-	-	-	104865	76
14.3	22.7	33.9	48.6	68	-	18.5	-	104877	76
27.5	44.9	63.9	85	108	133	30.99	-	105057	76
	00.0		_	-	-	-	-	104994	76
15.0	28.9	-	-					101001	
15.0 29.9	28.9 46	- 59.5	70.9	81.8	91.6	31.5	1.18	104973	76

#### Salts P-S

	Surger 5	
	Product	Chemical formula
Р	Potassium hexahydroantimonate(V) cryst. for analysis EMSURE®	K[Sb(OH),]
	Potassium hydrogen carbonate for analysis EMSURE® ACS	KHCO3
	Potassium hydrogen diiodate for analysis EMSURE®	KH(IO <sub>3</sub> ) <sub>2</sub>
	di-Potassium hydrogen phosphate anhydrous for analysis EMSURE®	K₂HPO₄
	di-Potassium hydrogen phosphate trihydrate for analysis EMSURE®	$K_2HPO_4 \cdot 3 H_2O$
	Potassium hydrogen phthalate for analysis EMSURE® Reag. Ph Eur	C <sub>8</sub> H <sub>5</sub> KO <sub>4</sub>
	Potassium hydrogen sulfate for analysis EMSURE® Reag. Ph Eur	KHSO₄
	Potassium iodate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	KIO <sup>3</sup>
	Potassium iodide for analysis EMSURE® ACS, ISO, Reag. Ph Eur	KI
	Potassium nitrate for analysis EMSURE® ISO, Reag. Ph Eur	KNO <sub>3</sub>
	Potassium nitrite cryst. for analysis EMSURE® ACS	KNO <sub>2</sub>
	di-Potassium oxalate monohydrate for analysis EMSURE® ACS	$K_2C_2O_4 \cdot H_2O$
	Potassium perchlorate for analysis EMSURE® ACS	KClO₄
	Potassium permanganate for analysis EMSURE® ACS, Reag. Ph Eur	KMnO₄
	Potassium permanganate for analysis EMSURE® (max. 0.000005 % Hg) ACS	KMnO₄
	Potassium peroxodisulfate for analysis EMSURE® (max. 0.001 % N), ACS, Reag. Ph Eur	K <sub>2</sub> S <sub>2</sub> O <sub>8</sub>
	Potassium peroxodisulfate for analysis EMSURE® ACS, Reag. Ph Eur	K <sub>2</sub> S <sub>2</sub> O <sub>8</sub>
	Potassium sodium tartrate tetrahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	$C_4H_4KNaO_6 \cdot 4H_2O$
	Potassium sulfate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	K₂SO₄
	Potassium sulfide about 44 % small lumps, for analysis EMSURE®	-
	Potassium thiocyanate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	KSCN
S	Silver nitrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	AgNO <sub>3</sub>
	Sodium acetate anhydrous for analysis EMSURE® ACS, Reag. Ph Eur	CH <sub>3</sub> COONa
	Sodium acetate trihydrate for analysis EMSURE® indifferent to potassium permanganate ACS, ISO, Reag. Ph Eur	$NaCH_{3}COO \cdot 3 H_{2}O$
	Sodium ammonium hydrogen phosphate tetrahydrate for analysis EMSURE®	$NaNH_4HPO_4 \cdot 4 H_2O$
	Sodium carbonate anhydrous for analysis EMSURE® ACS, ISO, Reag. Ph Eur	Na <sub>2</sub> CO <sub>3</sub>
	Sodium carbonate anhydrous for analysis EMSURE®, ISO	Na <sub>2</sub> CO <sub>3</sub>
	Sodium carbonate decahydrate for analysis EMSURE®, ISO, Reag. Ph Eur	$Na_2CO_3 \cdot 10 H_2O$
	Sodium chloride for analysis EMSURE® ACS, ISO, Reag. Ph Eur	NaCl
	tri-Sodium citrate dihydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	$C_6H_5Na_3O_7 \cdot 2 H_2O$
	Sodium dichromate dihydrate for analysis EMSURE® ACS	$Na_2Cr_2O_7 \cdot 2 H_2O$
	Sodium dihydrogen phosphate dihydrate for analysis EMSURE® Reag. Ph Eur	$NaH_2PO_4 \cdot 2 H_2O$
	Sodium dihydrogen phosphate monohydrate for analysis EMSURE® ACS, Reag. Ph Eur	$NaH_2PO_4 \cdot H_2O$
	tetra-Sodium diphosphate decahydrate for analysis EMSURE® ACS, Reag. Ph Eur	$Na_4P_2O_7 \cdot 10 H_2O$
	Sodium disulfite (sodium metabisulfite) for analysis EMSURE® ACS, Reag. Ph Eur	$Na_2S_2O_5$
	Sodium dithionite for analysis	$Na_2S_2O_4$
	Sodium fluoride for analysis EMSURE® ACS, ISO, Reag. Ph Eur	NaF
	Sodium formate for analysis EMSURE® ACS, Reag. Ph Eur	HCOONa
	Sodium hexanitrocobaltate(III) [sodium cobalt(III)nitrite] for analysis EMSURE® ACS, Reag. Ph Eur	$Na_{3}[Co(NO_{2})_{6}]$
	Sodium hydrogen carbonate decahydrate for analysis EMSURE® ACS, Reag. Ph Eur	NaHCO <sub>3</sub>
	di-Sodium hydrogen phosphate anhydrous for analysis EMSURE® ACS, Reag. Ph Eur	Na <sub>2</sub> HPO <sub>4</sub>
	di-Sodium hydrogen phosphate anhydrous for analysis EMSURE®, particle size about 0.2-1 mm (~ 18-80 mesh ASTM)	$Na_2HPO_4$
	di-Sodium hydrogen phosphate dihydrate for analysis EMSURE®	$Na_2HPO_4 \cdot 2 H_2O$
	di-Sodium hydrogen phosphate dodecahydrate for analysis EMSURE® ISO, Reag. Ph Eur	$Na_2HPO_4 \cdot 12 H_2O$
	di-Sodium hydrogen phosphate heptahydrate for analysis EMSURE® ACS, Reag. Ph Eur	$Na_2HPO_4 \cdot 7 H_2O$
	Sodium hydrogen sulfate monohydrate for analysis EMSURE®	$NaHSO_4 \cdot H_2O$
	Sodium iodate for analysis EMSURE®	NalO <sub>3</sub>
	- means: no values available	

_	C. L. L. W. C. C.	ibility [g/100 g H <sub>2</sub> 0] at Saturated solution [20°C]							D	
			1000	C010	0000	100%			Cat. No.	Page
	0°C	20°C	40°C	60°C	80°C	100°C	Percentage	Density	105110	76
	-	2.0	-	-	-	-		- 1 10	105110	76
	22.6	33.3	45.3	60	-	-	24.98	1.18	104854	76
	-	1.3	-	-	-	-	-	-	104867	76
	-	-	-	266	-	-	-	-	105104	76
	-	159	212.5	-	-	-	61.4	-	105099	76
	-	8.0	-	-	-	-	-	-	104874	76
	36.3	51.4	76.3	-	-	121.6	33.95	-	104885	76
	4.7	8.1	12.9	18.5	24.8	32.3	7.5	1.064	105051	76
	127.8	144.5	161	176.2	191.5	208	59.1	1.71	105043	77
	13.3	31.7	63.9	109.9	169	245.2	24.1	1.16	105063	77
	278.8	298.4	334.9	-	-	412.8	-	-	105067	77
	-	36.0	-	-	-	-	-	-	105073	77
	0.76	1.7	3.6	7.2	13.4	22.2	1.7	1.008	105076	77
	2.8	6.4	12.6	22.4	-	-	6	1.04	105082	77
	-	6.4	-	-	-	-	6	1.04	105084	77
	-	0.5	1.1	-	-	-	0.468	-	105092	77
	0.18	0.5	1.1	-	-	-	0.468	-	105091	77
	-	63.0	-	-	-	-	-	-	108087	77
	7.3	11.1	14.8	18.2	21.3	24.1	10	1.0807	105153	77
	_	50.0	-	-	-	-	-	-	105134	77
	177	218	-	-	-	-	68.55	1.42	105125	77
	115	219.2	334.8	471	652	1024	68.6	2.18	101512	78
	-	36.5	-	-	-	-	-	-	106268	78
	36.3	46.4	65.4	138.0 (58°C)	-	-	31.7	1.17	106267	78
	-	16.7	-	-	-	-	-	_	106682	78
	7.1	21.4	48.5	46.5	45.8	45.5	-	_	106393	78
	_	22.0	_	-	-	_	_	_	106392	78
	6.9	21.7	-	-	-	-	17.8	1.1941	106391	78 (68)
	-	35.9	36.4	37.1	38.1	39.2	26.4	1.201	106404	78
	_	72.0	-	-	-	-	_	_	106448	78
	163.2	180.2	220.5	283	385	-	64.3	_	106336	78
	-	85.0	-	-	-	_	-	_	106342	78
	_	-	158.6 (50°C)	_	_	-	_	_	106346	78
	2.7	5.5	12.5	21.9	30	40.3	5.2	1.05	106591	78
	-	5.5 65.3	71.1	79.9	30 88.7	40.3	5.2 39.5	-	106528	78
	-	25.0	-	-	-	-	- -	-	106528	79
	- 3.6	4.1		-	-		- 3.94		106507	79 79
			-			-		1.04		
	43.8	85.3	107	131	198	150	-	-	106443	79
	-	72.0	-	-	-	-	-	-	102521	79
	6.9	9.6	12.7	16	19.7	23.6	8.76	1.08	106329	79
	-	-	-	-	-	104.1	-	-	106586	79
	-	7.7	-	-	-	-	-	-	106559	79 (68)
	-	-	-	83	92.4	-	-	-	106580	79
	1.63	7.7	-	-	-	-	7.2	1.08	106579	79 (68)
	-	-	55	-	-	-	-	-	106575	79 (68)
	-	108.0	-	-	-	-	-	-	106352	79
	2.5	9.1		23	27				106525	79

#### Salts S-Z

	Product	Chemical formula
S	Sodium iodide for analysis, Reag. Ph Eur	Nal
	Sodium metaperiodate for analysis EMSURE® ACS, Reag. Ph Eur	NalO₄
	Sodium molybdate dihydrate for analysis EMSURE®	$Na_2MoO_4 \cdot 2 H_2O$
	Sodium nitrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	NaNO <sub>3</sub>
	Sodium nitrite for analysis EMSURE® ACS, Reag. Ph Eur	NaNO <sub>2</sub>
	di-Sodium oxalate for analysis EMSURE®	NaOOCCOONa
	Sodium perchlorate monohydrate for analysis EMSURE®	$NaClO_4 \cdot H_2O$
	Sodium peroxydisulfate for analysis EMSURE®	Na <sub>2</sub> S <sub>2</sub> O <sub>8</sub>
	tri-Sodium phosphate dodecahydrate for analysis EMSURE®	$Na_{3}PO_{4} \cdot 12 H_{2}O$
	tri-Sodium phosphate dodecahydrate for analysis EMSURE® ACS, Reag. Ph Eur	$Na_{3}PO_{4} \cdot 12 H_{2}O$
	Sodium salicylate for analysis EMSURE®	$C_7H_5NaO_3$
	Sodium sulfate anhydrous for analysis EMSURE® ACS, ISO, Reag. Ph Eur	Na <sub>2</sub> SO <sub>4</sub>
	Sodium sulfate decahydrate for analysis EMSURE® ACS, Reag. Ph Eur	$Na_2SO_4 \cdot 10 H_2O$
	Sodium sulfite anhydrous for analysis EMSURE® ACS, Reag. Ph Eur	Na <sub>2</sub> SO <sub>3</sub>
	di-Sodium tartrate dihydrate for analysis EMSURE®	$C_4H_4Na_2O_6\cdot 2 H_2O$
	Sodium thiosulfate pentahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	$Na_2S_2O_3 \cdot 5 H_2O$
	Sodium tungstate dihydrate for analysis EMSURE®	$Na_2WO_4 \cdot 2 H_2O$
	Strontium chloride hexahydrate for analysis EMSURE® ACS, ISO	$SrCl_2 \cdot 6 H_2O$
	Strontium nitrate for analysis EMSURE®	Sr(NO <sub>3</sub> ) <sub>2</sub>
Т	Tin(II) chloride dihydrate for analysis EMSURE® ACS, Reag. Ph Eur	$SnCl_2 \cdot 2 H_2O$
	Tin(II) chloride dihydrate for analysis EMSURE® (max. 0.000001 % Hg)	$SnCl_2 \cdot 2 H_2O$
Ζ	Zinc acetate dihydrate for analysis EMSURE® ACS	$(CH_{3}COO)_{2}Zn \cdot 2 H_{2}O$
	Zinc chloride for analysis EMSURE® ACS, ISO, Reag. Ph Eur	ZnCl <sub>2</sub>
	Zinc iodide for analysis EMSURE®	Znl <sub>2</sub>
	Zinc nitrate tetrahydrate for analysis EMSURE®	$Zn(NO_3)_2 \cdot 4 H_2O$
	Zinc sulfate heptahydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	$ZnSO_4 \cdot 7 H_2O$
	- means: no values available	

 Colubility	Solubility [g/100 g H₂0] at					Coturotod	ution [20°C]	Cot No	Page
							ution [20°C]	Cat. No.	Page
0°C	20°C	40°C	60°C	80°C	100°C	Percentage	Density		
-	-	-	-	295	303	-	-	106523	79
-	9.1	-	-	-	-	-	-	106597	80
-	84.0	-	-	-	-	-	-	106521	80
70.7	88.3	104.9	124.7	148	176	46.8	1.38	106537	80
73	84.5	95.7	112.3	135.5	163	45.8	1.33	106549	80
-	3.7	-	-	-	-	-	-	106557	80
167	181	243	-	-	-	64.4	1.757	106564	80
-	54.5	-	-	-	-	-	-	106609	80 (68)
-	28.5	-	-	-	-	-	-	106572	80
1.5	12.1	31	55	81	108	10.8	1.106	106578	80
-	100	-	-	-	-	-	-	106601	80
-	-	48.1	45.3	43.1	42.3	-	-	106649	80
4.56	19.2	-	-	-	-	16.1	1.15	106648	80 (68)
-	-	37	33.2	29	26.6	-	-	106657	80
(29)	29.0	-	-	-	-	_	-	106663	80
52.5	70.1	102.6	-	-	-	41.2	1.39	106516	80
-	73	73.9	-	90.2	96.9	-	-	106673	81
44.1	53.8	66.6	85.2	-	-	35	1.39	107865	81
-	-	91.2	94.2	97.2	101.2	-	-	107872	81
83.9	269.8 (15°C)	-	-	-	-	-	-	107815	81 (68)
-	118.7	-	-	-	-	-	-	107814	81 (68)
-	43	-	-	-	-	-	-	108802	81
-	-	453	488	541	-	-	-	108816	81
429.4	-	445.2	467.2	490	510.5	-	-	108828	81
-	-	211.5	-	_	-	-	_	108833	81
41.6	53.8	-	-	_	-	35	1.47	108883	81



# Determination of mercury

The long discussed question of the toxicity of mercury is dependent on the type of binding involved: liquid, metallic mercury is practically non-toxic; however, mercury vapor, especially that released by organo-metallic compounds which can readily be formed in the biosphere, is extremely toxic and is an ever-present danger for humans and animals. For this reason the determination of mercury in environment foodstuffs is extremely important.

#### Digestion

As mercury occurs in numerous binding forms. The total material to be analyzed must first be chemically digested and the mercury converted into inorganic compounds before the mercury content in different samples can be determined.

#### Methods

Due to the different compositions of the sample materials three common digestion methods are normally employed:

- Digestion with sulfuric acid / nitric acid under reflux
- Digestion with nitric acid at high temperature and under pressure in a digestion vessel
- Digestion by UV irradiation

The mercury compound is dissolved or digested using oxidation processes involving oxidants or digestion agents. The excess oxidant or digestion agent is then removed. At the same time the mercury is reduced. Elemental mercury is then transferred to the measuring cell using inert gas.







### Mercury analysis

The most common methods for the determination of mercury are:

Mercury content	Procedure
<0.5 μg/l	Amalgam
0.2 – 10 μg/l	Cold vapor AAS
>10 µg/l	Dithizone

As most mercury contamination is with the range 0.2 to 10  $\mu$ g/l, the most usual procedure is that involving flameless atomic absorption spectrometry – the cold vapor AAS process.



# Reagents for routine determination of mercury

	Mercury analysis A-R	Digestion preparation	Digestion with sulfuric / nitric acid under reflux	Digestion with nitric acid in a digestion bomb	Digestion by UV irradiation	Cold vapor AAS	Cold vapor AAS	Cold vapor AAS subsequent to UV irradiation	Cold vapor AAS subsequent to wet-chemical digestion	Dithizone method	Dithizone method	Content	Packaging	Ord. No.
С	Calcium chloride fused, granular about						•	•	•			250 g	PE bottle	1.02083.0250
	0.5-2.0 mm for elementary analysis Chemizorb® Hg Reagents and accessories											1 kg 1 set	PE bottle PE bottle	1.02083.1000 1.12576.0001
	for absorbent for mercury						•	•	•		•	i see		1120/010001
D	Dithizone for analysis											5 g	Glass bottle	1.03092.0005
	(1,5-Diphenylthiocarbazone) Reag. Ph Eur										•	25 g	Glass bottle	1.03092.0025
Н	Hydrochloric acid fuming 37 % for analysis (max. 0.001 ppm Hg) EMSURE®		•				•	•	•			2.5	Glass bottle	1.13386.2500
	Hydroxylammonium chloride for analysis (≤0.000001 % Hg) ACS, ISO						•	•	•		•	250 g	Glass bottle	1.04619.0250
Μ	Magnesium perchlorate hydrate [about 83 %							•				100 g	Metal can	1.05875.0100
	$Mg(CIO_4)_2$ ] for elementary analysis						•	•				500 g	Metal can	1.05875.0500
	Mercury ICP standard traceable to SRM from NIST Hg(NO $_3$ ) $_2$ in HNO $_3$ 10 % 1000 mg/l Hg Certipur®						•	•	•			100 ml	PE bottle	1.70333.0100
	Mercury standard solution traceable to SRM from											100 g	PE bottle	1.70226.0100
	NIST Hg(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 mol/l 1000 mg/l Hg Certipur®						•	•	•		•	500 g	PE bottle	1.70226.0500
Ν	Nitric acid 65 % for analysis (max. 0.005 ppm		•	•			•	•	•			1	Glass bottle	1.00452.1000
	Hg) EMSURE® ISO											2.5 l	Glass bottle	1.00452.2500
Р	Perchloric acid 70 % for analysis (max. 0.0000005 % Hg) EMSURE® ACS, ISO, Reag. Ph Eur		•									1	Glass bottle	1.00514.1000
	Potassium dichromate for analysis (max. 0.000001 % Hg) EMSURE® ACS, ISO		•	•	•		•	•	•			500 g	Glass bottle	1.04865.0500
	Potassium permanganate for analysis (max. 0.000005 % Hg) EMSURE® ACS		•									1 kg	Glass bottle	1.05084.1000
	Potassium peroxodisulfate for analysis											250 g	PE bottle	1.05092.0250
	(≤0.001 % N) EMSURE® ACS, Reag. Ph Eur											5 kg	PE bottle	1.05092.5000

	Mercury analysis S-Z	Digestion preparation	Digestion with sulfuric / nitric acid under reflux	Digestion with nitric acid in a digestion bomb	Digestion by UV irradiation	Cold vapor AAS	Cold vapor AAS	Cold vapor AAS subsequent to UV irradiation	Cold vapor AAS subsequent to wet-chemical digestio	Dithizone method	Dithizone method	Content	Packaging	Ord. No.
S	Silica gel beads, desiccant ~ 2 – 5 mm						•	•	•			1 kg	PE bottle	1.07735.1000
	Silica gel with indicator (orange gel),											1 kg	PE bottle	1.01969.1000
	granulate ~ 1 – 3 mm						•	•	•			5 kg	PE bottle	1.01969.5000
	Silicon anti-foaming agent							•				100 ml	PE bottle	1.07743.0100
								•	•			500 ml	PE bottle	1.07743.0500
	Sodium borohydride for analysis						•		•			100 g	Glass bottle	1.06371.0100
	Sulfuric acid 95 – 97 % for analysis											1	Glass bottle	1.00732.1000
	(max. 0.005 ppm Hg) EMSURE® ACS, ISO,		•									2.5	Glass bottle	1.00732.2500
	Reag. Ph Eur											2.5 l	Safebreak bottle	1.00732.2510
	Sulfuric acid for 1000 ml,								•			1	Plastic ampoule	1.09981.0001
	$c(H_2SO_4) = 0.5 \text{ mol/l } (1 \text{ N}) \text{ Titrisol}^{\otimes}$													
Т	Tin(II) chloride dihydrate for analysis											250 g	Glass bottle	1.07814.0250
	(max. 0.000001 % Hg) EMSURE®							-				2.5 kg	Glass bottle	1.07814.2500
W	Water for analysis EMSURE®		•	•	•		•	•	•			5	PE bottle	1.16754.5000

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# Acids for analysis EMSURE<sup>®</sup> | EMPARTA<sup>®</sup>

# High-quality, pure acids in optimized packaging

Your responsibility is our incentive! Merck Millipore's acids for analysis offer the highest possible quality, the greatest safety and most optimized packaging. Our products undergo strict quality checks in ultra-modern laboratories using the latest and most sensitive analytical instruments. That is why you can rely on Merck Millipore's acids for analysis for your most demanding applications.



# Acids

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# Advantages

- Wide and comprehensive product range
- Outstanding, application-oriented quality
- Highly reliable and reproducible results through batch-to-batch consistency
- Extensive specification and Certificate of Analysis (CoA)
- Most acids meet or exceed the requirements of ACS, ISO and / or Reag. Ph Eur
- Excellent cost / performance ratio thanks to extensive specifications



www.merckmillipore.com/acids

# High standards of purity, safety and environmental protection

#### Quality management

For decades, our sophisticated quality management system has proven to be the best basis for precise and reproducible analytical applications. From purchase and storage of raw materials, through production, filling and analysis of the acids, to storage and shipment of the final product, every step is subject to stringent monitoring and documentation.

Thanks to our highly modern production chain, particulate impurities and cross-contamination of acids from other products can be ruled out.

#### **Reliable results**

Acids from Merck Millipore are analyzed according to extensive and stringent specifications. Most of our acids for analysis with the EMSURE® quality grade are specified in compliance with international standards such as ACS, ISO and / or Reag. Ph Eur. Thus, you can always be confident of the results of your analysis made with Merck Millipore reagents.





Working with hazardous products like acids requires particular attention in order to protect users and the environment. Our acids are delivered to you in highquality packaging which offers the greatest safety during transportation, storage and handling. As a certified independent packaging institute, we develop and test our own packaging materials to fulfill our strict quality standards and, of course, your individual requirements.

Please see the chapter »Packaging« to find out more about the efforts we take to offer the most appropriate and safest packaging for every single product.

#### Special packaging concepts

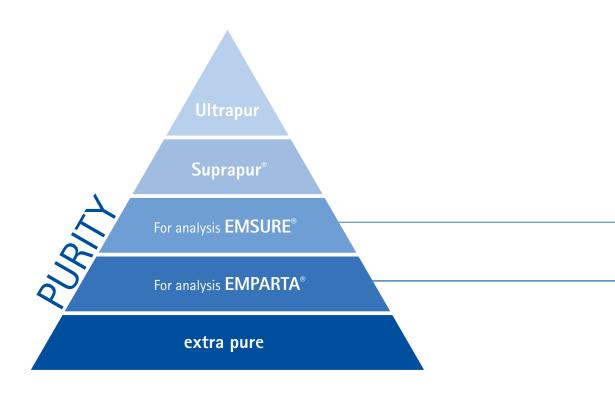
Particularly for acids, we offer packaging concepts that take into account the potential hazards of the products. You can learn more about our Safebreak bottle, SafetyCap and special bottle for hydrofluoric acid on the following pages.

Take advantage of our leading role in the field to reach your targets swiftly and safely.



# Different grades to meet your needs

Whenever you are using an acid, you have to consider your application, your target and, of course, your budget. Each application is different and therefore we offer a wide range of acids in various grades, packaging sizes and materials. Whether you need to perform a highly critical analysis or a basic lab application, you will always find the most suitable reagent among our brand names.



# Your advantages

- The right product quality for each individual application
- The right pack size for your specific demand
- The right pricing according to the real product performance

# Specification

- ACS Standards of the American Chemical Society
- **ISO** Standards of the International Organization for Standardization
- Reag. Ph Eur Reagents specified by the European Pharmacopoeia

#### Ultrapur and Suprapur<sup>®</sup> acids

- For instrumental analysis, we offer a range of high purity acids under the brand names Suprapur<sup>®</sup> and Ultrapur
- For further information please see the chapter »High purity acids and bases«

Acids for analysis	EMSURE®	EMPARTA <sup>®</sup>
Purity	99.7 – 99.9 %	99.0 – 99.5 %
Number of specified	Up to 50 parameters and more	>10
parameters		
Specified regulations	ACS, ISO, Reag. Ph Eur	ACS
Pack sizes	500 ml/g, 1 l/kg, 2.5 l/kg, 5 l/kg, 25 l/kg, 50 l/kg	2.5 l bottle
	and many more	
Packaging materials	Glass, HDPE, PE-coated glass, container, barrels etc.	HDPE
Segments	All regulated industries	Non-regulated industries
Applications	Quality control (QC)	Quality control (QC)
	Critical lab applications	Basic lab applications
	If you need to be absolutely sure in your lab	If you need only a part of the
	application – use EMSURE®	EMSURE <sup>®</sup> performance – use EMPARTA <sup>®</sup>

#### extra pure acids

- For preparative lab applications and cleaning purposes
- Specifications with the most important parameters
- Available in different pack sizes

# Why choose Merck Millipore acids for analysis?

- Wide and comprehensive product range
- Outstanding, application-oriented quality
- Great variety of packaging materials and pack sizes
- Sophisticated quality management system:
  - Every batch is subject to stringent analysis
  - Analysis of each single parameter batch by batch!
  - Use of the latest and most sensitive analytical instruments
- Large number of specified parameters for each acid
- Extremely low limiting values
- Most acids for analysis EMSURE<sup>®</sup> meet or exceed the requirements of ACS, ISO and / or Reag. Ph Eur
- Acids for analysis EMPARTA® meet the requirements of ACS



## Your advantages

- Reliable and reproducible results
- Confidence in analytical results and right conclusions
- Fulfilling expectations of your customers
- Maximum certainty during analytical and preparative operations
- High safety during daily lab work
- Very convenient use of acids
- One product usable for many applications and all over the world
- Less validation efforts
- Excellent cost / performance ratio



# Useful information about ...

## Available concentrations of some acids

Name	Weight	Density	Concentration
	0⁄0	<b>g/cm</b> ³ (20°C)	mol/l *
Acetic acid	96	1.05	17
	60	1.06	10.5
Acetic acid (glacial acetic acid)	99 – 100	1.05	17.5
Acetic acid, dilute	30	1.04	5
Formic acid	98 – 100	1.22	26
Hydrobromic acid	47	1.49	8.5
Hydrochloric acid	25	1.12	7.5
	32	1.16	10
Hydrochloric acid, fuming	37	1.19	12
Hydrofluoric acid	48	1.16	28
	40	1.13	22.5
Hydroiodic acid	57	1.7	7.5
Hydrogen peroxide	30	1.11	10
Nitric acid	65	1.39	14.5
	69	1.41	15.5
Nitric acid, fuming	100	1.51	24
Perchloric acid	70	1.68	11.5
	60	1.53	9
ortho-Phosphoric acid	85	1.71	15
Sulfuric acid	95 – 97	1.84	18
	25	1.18	3

\* rounded off

# Variety of packaging materials

# Safe and environmentally friendly packaging for every application

Acids for analysis from Merck Millipore are always delivered to you in the most suitable packaging. Our packaging materials are exactly fitted to the product. This not only secures the purity of our high-quality reagents, but also protects users and the environment. For detailed information about our packaging, please see the chapter »Packaging«.



# Overview of the most common packages for acids

Glass bottles from 25 ml to 2500 ml Our bestseller: excellent glass quality with very low traces of impurities and long shelf life

HDPE bottles for liquids from 250 ml to 2500 ml Easy to handle, very low weight

HDPE bottles for solids from 25 g to 5000 g Easy dosage thanks to the wide opening

Safebreak bottles (PE-coated glass bottles) for 2500 ml Exceptionally safe glass bottle for especially dangerous acids

**PE containers from 5 I to 50 I, PE/metal barrels for 25 I and PE barrels up to 200 I** Convenient handling of even large quantities

and much more ...

Hazardous acids need safe packaging! Employee safety plays a decisive role in your lab – that is why Merck Millipore provides innovative packaging concepts especially made for acids.

#### More information

▶ Turn the page to explore our packaging concepts for acids. ▶ See chapter »Packaging material« on page 18

# Merck Millipore Safebreak bottle. Just in case!

Glass bottles are still the most preferred packaging for reagents. As containers for all manner of chemicals, they remain a valuable inert material for everyday use in the laboratory. However, if they are to be safely used, the same care has to be taken as with glassware in private households. A problem occurs when a bottle containing a hazardous liquid breaks just when and where it shouldn't ...



#### But if there is a breakage ...

We always wonder how could it happen after an accident. Despite the measures taken to avoid breakage, it is something that simply has to be reckoned with in every laboratory. How harmful breakage is in a laboratory depends on the circumstances. Fortunately, in most cases, all that is required is proper clearing up of the spilled contents and glass splinters. However, serious injury, contamination or consequential damage may result in some cases.

#### ... we have the solution.

At Merck Millipore, we have carefully examined the risks involved with acids in glass bottles and have come up with an effective solution: the »Safebreak« bottle. This glass bottle is coated with polyethylene, which provides maximum safety in case of breakage. Should the bottle fall and break, the liquid and glass splinters are reliably contained within the PE coating.

Each Safebreak bottle is fitted with a screw cap made of polypropylene that has an integrated PTFE component. This S40 cap renders the bottle absolutely airtight so that no liquid or vapor can escape. Also, it can be disposed of with conventional glass; during incineration, the PE is burnt off without affecting the environment.

#### **More information**

# Safebreak bottles

The right bottle for your peace of mind. Exclusively available from Merck Millipore, the computer-designed Safebreak bottle combines several advantages in one solution:

- Meets all safety requirements
- Ensures that you receive exactly the same quality of contents as that dispatched from Merck Millipore
- Can be incorporated in all logistic systems
- Can be easily and ecologically disposed of and reused

# Ordering information

#### Acids in Safebreak bottles

Product	Content	Packaging	Ord. No.
Acetic acid (glacial) 100 % anhydrous for analysis EMSURE® ACS, ISO, Reag. Ph Eur	2.5 l	Safebreak bottle	1.00063.2510
Formic acid 98 – 100 % for analysis EMSURE® ACS, Reag. Ph Eur	2.5 l	Safebreak bottle	1.00264.2510
Hydrochloric acid fuming 37 % for analysis EMSURE® ACS, ISO, Reag. Ph Eur	2.5 l	Safebreak bottle	1.00317.2510
Nitric acid 65 % for analysis EMSURE® ISO	2.5 l	Safebreak bottle	1.00456.2510
Nitric acid 69 % for analysis EMSURE® ACS, Reag. Ph Eur	2.5 l	Safebreak bottle	1.01799.2510
Perchloric acid 70 – 72 % for analysis EMSURE® ACS, ISO, Reag. Ph Eur	2.5 l	Safebreak bottle	1.00519.2510
ortho-Phosphoric acid 85 % for analysis EMSURE® ACS, ISO, Reag. Ph Eur	2.5 l	Safebreak bottle	1.00573.2510
Sulfuric acid 95 – 97 % for analysis EMSURE® ISO	2.5 l	Safebreak bottle	1.00731.2510
Sulfuric acid 95 – 97 % for analysis (max. 0.005 ppm Hg) EMSURE® ACS, ISO, Reag. Ph Eur	2.5 l	Safebreak bottle	1.00732.2510
Sulfuric acid 98 % for analysis EMSURE®	2.5 l	Safebreak bottle	1.12080.2510

#### The right bottle from Merck Millipore

All our experience in dealing with the potential hazards of acids in glass bottles has been incorporated in the specifications of the Merck Millipore Safebreak bottle:

- The bottle is able to withstand considerable impact force
- Should breakage occur, the acid and any glass splinters are reliably contained
- Even after frequent opening and closing, the screw cap remains perfectly intact
- The bottle is just as recyclable as a conventional glass bottle

# Merck Millipore SafetyCap

# Leakproof against liquids – allows excess pressure to be released

Reagents such as sodium hypochlorite solution or hydrogen peroxide that are capable of generating excess pressure through chemical reactions are often supplied in bottles with a special screw cap. This cap has a valve that allows the gas formed to be released, hence preventing the build-up of pressure. However, on occasion, traces of reagent also diffused through the valve. For safety purposes, these bottles are additionally packaged in a PE bag.

#### SafetyCap – the intelligent closure

In order to completely avoid contamination, Merck Millipore supplies all such reagents fitted with the SafetyCap. This innovative cap allows absolutely no reagent to leak – even if the bottle is tipped. The PTFE membrane sintered onto the inside of the cap allows gas to be released but is absolutely leakproof against liquids. This has been proven in numerous warehouse and stress tests at our package testing facility at all temperatures and in all positions.

#### Small improvements - huge effects

The PTFE membrane incorporated in the SafetyCap protects our reagents. It allows neither gas nor liquid to enter the bottle, thus completely eliminating contamination. In addition, the SafetyCap has no protruding parts that can break off.



# SafetyCap

- Allows gas to be released and the internal pressure to be decreased
- Allows no liquid to escape, thus protecting the environment from contamination
- Allows neither gas nor liquid to enter the bottle, thus protecting the contents from contamination

# Ordering information

### Acids delivered with SafetyCap

Product	Content	Packaging	Ord. No.
Hydrogen peroxide 30 % (Perhydrol®) for analysis EMSURE® ISO	250 ml	Plastic bottle	1.07209.0250
	500 ml	Plastic bottle	1.07209.0500
	11	Plastic bottle	1.07209.1000
	2.5 l	Plastic bottle	1.07209.2500
Hydrogen peroxide 30 % (Perhydrol®)	250 ml	Plastic bottle	1.07210.0250
(stabilized for higher storage temperature) for analysis $EMSURE^{\circledast}$ ISO	1	Plastic bottle	1.07210.1000
	2.5 l	Plastic bottle	1.07210.2500
Hydrogen peroxide 30 % H <sub>2</sub> O <sub>2</sub> Suprapur <sup>®</sup>	250 ml	Plastic bottle	1.07298.0250
	500 ml	Plastic bottle	1.07298.0500
	1	Plastic bottle	1.07298.1000
Hydrogen peroxide 35 % technical grade	25 I	Plastic container	1.08556.9025
Perhydrit® tablets 1 g (Hydrogen peroxide – Urea)	100 g	Plastic bottle	1.07201.0100
	4 kg	Plastic bottle	1.07201.4000
Sodium hypochlorite solution (6 – 14 % active chlorine)	2.5 l	Plastic bottle	1.05614.2500
	25 l	Plastic container	1.05614.9025

#### More information

www.merckmillipore.com/safetycap

# Merck Millipore HDPE dosage bottle for hydrofluoric acid

#### Hydrofluoric acid

Hydrofluoric acid is one of the most dangerous acids. Even small quantities can cause severe injuries and poisoning. To avoid such fatal accidents, Merck Millipore has developed a pouring aid that is specially suited to the characteristics of hydrofluoric acid.

All 250 and 500 ml bottles are provided with this innovative and safe pouring aid. It allows drop-by-drop withdrawal of the acid and the last drop stays reliably in the bottle.

Furthermore, our exclusive S40 closure system ensures that the bottle is completely airtight.







# HDPE dosage bottle

- 250 and 500 ml bottle with a withdrawal system especially constructed for this hazardous acid
- Allows drop-by-drop withdrawal and the last drop stays reliably in the bottle
- Special density function of our exclusive S40 screwing system
- Allows transportation and storage in cardboard boxes



## Ordering information

### Hydrofluoric acid in HPDE dosage bottle with special pouring system

Product	Content	Packaging	Ord. No.
Hydrofluoric acid 40 % for analysis EMSURE® ISO, Reag. Ph Eur	500 ml	Plastic bottle	1.00338.0500
Hydrofluoric acid 40 % Suprapur®	250 ml	Plastic bottle	1.00335.0250
	500 ml	Plastic bottle	1.00335.0500
Hydrofluoric acid 48 % for analysis EMSURE® ACS, ISO, Reag. Ph Eur	500 ml	Plastic bottle	1.00334.0500

# Ordering information Acids

Acids A-G

Product	Content	Packaging	Ord. No.
Acetic acid 30 % for analysis Reag. Ph Eur	500 ml	Glass bottle	1.59166.0500
Acetic acid 96 % for analysis EMSURE®	11	Glass bottle	1.00062.1000
	11	Plastic bottle	1.00062.1011
	2.5 l	Glass bottle	1.00062.2500
	2.5 l	Plastic bottle	1.00062.2511
	25 l	Plastic container	1.00062.9025
Acetic acid (glacial) 100 % for analysis EMPARTA® ACS	2.5 l	Plastic bottle	1.01830.2500
Acetic acid (glacial) 100 % anhydrous for analysis EMSURE® ACS, ISO, Reag. Ph Eur	11	Glass bottle	1.00063.1000
	11	Plastic bottle	1.00063.1011
	2.5 l	Glass bottle	1.00063.2500
	2.5 l	Safebreak bottle	1.00063.2510
	2.5 l	Plastic bottle	1.00063.2511
	25 l	Plastic container	1.00063.9026
Acetic anhydride for analysis EMSURE® ACS, ISO, Reag. Ph Eur	1	Glass bottle	1.00042.1000
	2.5 l	Glass bottle	1.00042.2500
	25 l	Plastic container	1.00042.9025
Amidosulfuric acid extra pure		Plastic bottle	1.00219.2500
	25 kg	Fibre carton	1.00219.9025
Amidosulfuric acid for analysis EMSURE®		Plastic bottle	1.00103.0100
	250 g	Plastic bottle	1.00103.0250
L(+)-Ascorbic Acid for analysis EMSURE® ACS, ISO, Reag. Ph Eur	100 g	Plastic bottle	1.00468.0100
	500 g	Plastic bottle	1.00468.0500
	1 kg	Plastic bottle	1.00468.1000
Barbituric acid for analysis EMSURE®	25 g	Plastic bottle	1.00132.0025
	100 g	Plastic bottle	1.00132.0100
Benzoic acid for analysis EMSURE® Reag. Ph Eur	100 g	Plastic bottle	1.00136.0100
	250 g	Plastic bottle	1.00136.0250
	1 kg	Plastic bottle	1.00136.1000
Boric acid for analysis EMSURE® ACS, ISO, Reag. Ph Eur	100 g	Plastic bottle	1.00165.0100
	500 g	Plastic bottle	1.00165.0500
	1 kg	Plastic bottle	1.00165.1000
	5 kg	Plastic bottle	1.00165.5000
	12 kg	Plastic container	1.00165.9012
	25 kg	Fibre carton	1.00165.9025
Citric acid monohydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur	500 g	Plastic bottle	1.00244.0500
	1 kg	Plastic bottle	1.00244.1000
	5 kg	Plastic bottle	1.00244.5000
	12 kg	Plastic container	1.00244.9012
	25 kg	Fibre carton	1.00244.9026
Formic acid 89 – 91 % for analysis EMSURE® ACS	11	Glass bottle	1.00253.1000
Formic acid 90 % for determination of viscosity acc. to DIN EN ISO 307	2.5 l	Glass bottle	1.10854.2500
Formic acid 98 - 100 % for analysis EMSURE® ACS, Reag. Ph Eur	100 ml	Glass bottle	1.00264.0100
	11	Glass bottle	1.00264.1000
	1	Plastic bottle	1.00264.1011
	2.5 l	Glass bottle	1.00264.2500
	2.5 l	Safebreak bottle	1.00264.2510
	2.5 l	Plastic bottle	1.00264.2511
	25 I	Plastic container	1.00264.9026
Glycolic acid for analysis EMSURE®	100 g	Plastic bottle	1.04106.0100

### Acids H

Product	Content	Packaging	Ord. No.
Hydrobromic acid 47 % extra pure	500 ml	Glass bottle	1.00304.0500
	2.5 l	Glass bottle	1.00304.2500
	20	Glass btl. pl. coat.	1.00304.9020
Hydrobromic acid 47 % for analysis EMSURE® ACS, ISO	500 ml	Glass bottle	1.00307.0500
	1	Glass bottle	1.00307.1000
Hydrochloric acid 25 % for analysis EMSURE®	1	Glass bottle	1.00316.1000
	11	Plastic bottle	1.00316.1011
	2.5 l	Glass bottle	1.00316.2500
	2.5 l	Plastic bottle	1.00316.2511
	25 I	Plastic container	1.00316.9025
Hydrochloric acid 32 % extra pure	2.5	Glass bottle	1.00313.2500
	25 I	Plastic container	1.00313.9025
	180 I	Plastic barrel	1.00313.9180
Hydrochloric acid 32 % for analysis EMSURE®	11	Glass bottle	1.00319.1000
	11	Plastic bottle	1.00319.1011
	2.5	Glass bottle	1.00319.2500
	2.5	Plastic bottle	1.00319.2511
	25	Plastic container	1.00319.9025
Hydrochloric acid fuming 37 % for analysis EMPARTA® ACS	2.5	Plastic bottle	1.01834.2500
Hydrochloric acid fuming 37 % for analysis EMSURE® ACS, ISO, Reag. Ph Eur	1	Glass bottle	1.00317.1000
nyurochioric aciu tunning 37 % for analysis elvisore" ACS, 150, Reag. Fil Eur	2.5	Glass bottle	1.00317.2500
	2.5	Safebreak bottle	1.00317.2510
	2.5	Plastic bottle	1.00317.2501
	25	Plastic container	1.00317.9026
Hydrochloric acid fuming 37 % for analysis max. 0.001 ppm Hg EMSURE®	2.5	Glass bottle	1.13386.2500
Hydrofluoric acid 38 – 40 %	11	Plastic bottle	1.00329.1000
	2.5	Plastic bottle	1.00329.2500
Hydrofluoric acid 38 – 40 % extra pure	11	Plastic bottle	1.00337.1000
	2.5	Plastic bottle	1.00337.2500
	25	Plastic container	1.00337.9025
Hydrofluoric acid 40 % for analysis EMSURE® ISO, Reag. Ph Eur	500 ml	Plastic bottle	1.00338.0500
	11	Plastic bottle	1.00338.1000
	2.5	Plastic bottle	1.00338.2500
Hydrofluoric acid 48 % for analysis EMSURE® ACS, ISO, Reag. Ph Eur	500 ml	Plastic bottle	1.00334.0500
	11	Plastic bottle	1.00334.1000
	2.5 l	Plastic bottle	1.00334.2500
	5 l	Plastic container	1.00334.5000
Hydrogen peroxide 30 % (Perhydrol®) for analysis EMSURE® ACS, ISO	250 ml	Plastic bottle	1.07209.0250
	500 ml	Plastic bottle	1.07209.0500
	1	Plastic bottle	1.07209.1000
	2.5 l	Plastic bottle	1.07209.2500
Hydrogen peroxide 30 % (Perhydrol®)	250 ml	Plastic bottle	1.07210.0250
(stabilized for higher storage temperature) for analysis EMSURE® ACS, ISO	1	Plastic bottle	1.07210.1000
	2.5 l	Plastic bottle	1.07210.2500
Hydrogen peroxide 35 % technical grade	25 I	Plastic container	1.08556.9025
Hydroiodic acid 57 % extra pure	250 ml	Glass bottle	1.00341.0250
	22	Carboy	1.00341.9022
Hydroiodic acid 57 % for analysis EMSURE®	50 ml	Glass bottle	1.00344.0050
,	250 ml	Glass bottle	1.00344.0250

# Ordering information Acids

Acids H-P

Product	Content	Packaging	Ord. No.
Hydroiodic acid 67 % for analysis EMSURE®	250 ml	Glass bottle	1.00345.0250
Hypophosphorous acid 50 % for analysis EMSURE®	100 ml	Glass bottle	1.04633.0100
	500 ml	Glass bottle	1.04633.0500
Molybdatophosphoric acid hydrate for analysis EMSURE® ACS, Reag. Ph Eur		Glass bottle	1.00532.0025
woryouatophosphoric actu nyurate for analysis Ewisone Acis, heag, fit Eur	25 g 100 g	Glass bottle	1.00532.0100
Molybdic acid about 85 % MoO <sub>3</sub> (containing ammonium molybdate)	1 kg	Plastic bottle	1.00400.1000
Nitric acid 65 % extra pure	1 kg	Glass bottle	
Nitric actu 65 % extra pure	2.5	Glass bottle	1.00443.1000
	2.5 1		1.00443.2500
Nitric acid 65 % for analysis EMSURE® ISO	1	Barrel PE/met Glass bottle	1.00443.9025
Nitric acid 65 % for analysis EMSORE® ISO			1.00456.1000
	2.5	Glass bottle	1.00456.2500
	2.5	Safebreak bottle	1.00456.2510
Nitric acid 65 % for analysis (max. 0.005 ppm Hg) $EMSURE^{\circledast}$ ISO	11	Glass bottle	1.00452.1000
	2.5	Glass bottle	1.00452.2500
Nitric acid 69 % for analysis EMPARTA® ACS	2.5	Glass bottle	1.01832.2500
Nitric acid 69 % for analysis EMSURE® ACS, Reag. Ph Eur	11	Glass bottle	1.01799.1000
	2.5 l	Glass bottle	1.01799.2500
	2.5	Safebreak bottle	1.01799.2510
Nitric acid fuming 100 % extra pure	11	Glass btl. pl. coat.	1.00450.1000
Nitric acid fuming 100 % for analysis EMSURE® ACS, Reag. Ph Eur	11	Glass btl. pl. coat.	1.00455.1000
Oxalic acid dihydrate extra pure	1 kg	Plastic bottle	1.00492.1000
	5 kg	Plastic bottle	1.00492.5000
	50 kg	Fibre carton	1.00492.9050
Oxalic acid dihydrate for analysis EMSURE® ACS, ISO, Reag. Ph Eur		Plastic bottle	1.00495.0100
		Plastic bottle	1.00495.0500
	1 kg	Plastic bottle	1.00495.1000
	25 kg	Fibre carton	1.00495.9025
Perchloric acid 60 % for analysis EMSURE® ACS	1	Glass bottle	1.00518.1001
	6 x 1 l	Glass bottle	1.00518.1016
	2.5 l	Glass bottle	1.00518.2501
	4 x 2.5 l	Glass bottle	1.00518.2514
Perchloric acid 70 % for analysis (max. 0.0000005 % Hg) EMSURE® ACS, ISO, Reag. Ph Eur	11	Glass bottle	1.00514.1000
,	6 x 1 l	Glass bottle	1.00514.1006
Perchloric acid 70 – 72 % for analysis EMSURE® ACS, ISO, Reag. Ph Eur	11	Glass bottle	1.00519.1001
······································	6 x 1 l	Glass bottle	1.00519.1016
	2.5	Glass bottle	1.00519.2501
	4 x 2.5 l	Glass bottle	1.00519.2514
	2.5	Safebreak bottle	1.00519.2510
Periodic acid for analysis EMSURE®	2.5 r	Glass bottle	1.00524.0025
	100 g	Glass bottle	1.00524.0100
meta-Phosphoric acid pieces for analysis (stabilized with sodium metaphosphate) EMSURE®		Metal can	1.00546.0100
אוינים י איסאווטויני מכום אויניניש זיטי מוומוצאיש (אנטוווצכם אונוו שטטועווו וווכנמאווטגאוומנכ) בMOUNE	100 g		1.00546.0500
artha Dhaanharia aaid OE 04 far analusis ENCLIDE® ACC ISO Daar Dh. Fur	500 g	Plastic bottle	
ortho-Phosphoric acid 85 % for analysis EMSURE® ACS, ISO, Reag. Ph Eur	11	Plastic bottle	1.00573.1000
	2.5	Plastic bottle	1.00573.2500
	2.5	Safebreak bottle	1.00573.2510
	25 l	Plastic container	1.00573.9025
ortho-Phosphoric acid 99 % cryst. for analysis EMSURE®	500 g	Plastic bottle	1.00565.0500

### Acids S-T

Product	Content	Packaging	Ord. No.
Succinic acid for analysis EMSURE®	250 g	Plastic bottle	1.00682.0250
Sulfuric acid 25 % for analysis EMSURE®	1	Plastic bottle	1.00716.1000
Sulfuric acid 40 % for determination of gas metabolism acc. to Knipping	2.5 l	Glass bottle	1.09286.2500
Sulfuric acid 62 % for analysis, for the determination of fat in cheese (d 1.52)	1	Plastic bottle	4.80531.1000
	2.5 l	Plastic bottle	4.80531.2500
Sulfuric acid 90 – 91 % for Gerber fat determination and determination of nitrates in milk	500 ml	Glass bottle	1.00729.0500
	2.5 l	Glass bottle	1.00729.2500
	25 I	Plastic container	1.00729.9025
Sulfuric acid 95 – 97 % for analysis EMPARTA® ACS	2.5 l	Plastic bottle	1.01833.2500
Sulfuric acid 95 – 97 % for analysis EMSURE <sup>®</sup> ISO	11	Glass bottle	1.00731.1000
	11	Plastic bottle	1.00731.1011
	2.5 I	Glass bottle	1.00731.2500
	2.5	Safebreak bottle	1.00731.2510
	2.5	Plastic bottle	1.00731.2511
	2.3 I 25 I	Plastic container	1.00731.9025
Sulfuric acid 95 – 97 % for analysis (max. 0.005 ppm Hg) EMSURE® ACS, ISO, Reag. Ph Eur	1	Glass bottle	1.00732.1000
Summe actu 35 – 37 % for analysis (max. 0.005 ppm mg) Eivisones Acs, iso, neag. Fil Eur		Glass bottle	
	2.5		1.00732.2500
	2.5	Safebreak bottle	1.00732.2510
Sulfuric acid 96 % for the determination of viscosity acc. to DIN EN ISO 307	11	Plastic bottle	1.08131.1000
	2.5	Plastic bottle	1.08131.2500
Sulfuric acid 98 % for analysis EMSURE®	11	Glass bottle	1.12080.1000
	2.5 l	Glass bottle	1.12080.2500
	2.5 l	Safebreak bottle	1.12080.2510
	25 I	Plastic container	1.12080.9025
Sulfuric acid 98 % for the determination of nitrogen	500 ml	Glass bottle	1.00748.0500
	2.5 l	Glass bottle	1.00748.2500
Sulfuric acid 100 % for conductivity measurements	11	Glass bottle	1.12223.1000
Sulfuric acid fuming 65 % SO $_3$ extra pure	1	Glass btl. pl. coat.	1.00720.1000
Sulfurous acid 5 – 6 % SO <sub>2</sub> for analysis EMSURE <sup>®</sup>	1	Glass bottle	1.00761.1000
	2.5 l	Glass bottle	1.00761.2500
L(+)-Tartaric acid for analysis EMSURE® ACS, ISO, Reag. Ph Eur	250 g	Plastic bottle	1.00804.0250
	1 kg	Plastic bottle	1.00804.1000
	5 kg	Plastic bottle	1.00804.5000
	50 kg	Fibre carton	1.00804.9050
Toluene-4-sulfonic acid monohydrate for analysis EMSURE® ACS	100 g	Plastic bottle	1.09613.0100
	500 g	Plastic bottle	1.09613.0500
Trichloroacetic acid for analysis EMSURE® ACS, Reag. Ph Eur	100 g	Glass bottle	1.00807.0100
-	250 g	Glass bottle	1.00807.0250
	1 kg	Glass bottle	1.00807.1000
Tungstophosphoric acid hydrate cryst. extra pure	100 g	Plastic bottle	1.00582.0100
<b>.</b> . , , , , ,	1 kg	Plastic bottle	1.00582.1000
	25 kg	Fibre carton	1.00582.9025
	100 g	Plastic bottle	1.00583.0100
Tungstophosphoric acid hydrate for analysis EMSURF®	.00 9		
Tungstophosphoric acid hydrate for analysis EMSURE®	250 a	Plastic hottle	1.00583 0250
Tungstophosphoric acid hydrate for analysis EMSURE® Tungstosilicic acid hydrate for analysis EMSURE®	250 g 25 g	Plastic bottle Glass bottle	1.00583.0250

# Caustic alkalis and bases for analysis EMSURE®

## Caustic alkalis and bases of high purity and defined quality

Caustic alkalis and bases are, along with salts and acids, the most important inorganic chemicals. In order to provide our customers with specified products for all types of wet chemical labs, Merck Millipore produces these chemicals itself using high-quality raw materials.



# Caustic alkalis and bases

# ContentsPage• Unique production technique116• Quality classification and branding118• Ordering information120

### Advantages

- Defined outstanding quality
- High reliability and reproducibility of results through batch-to-batch consistency and extremely low limiting values of impurities
- Most caustic alkalis and bases for analysis EMSURE<sup>®</sup> meet or exceed the requirements of ACS, ISO and / or Reag. Ph Eur
- Uniform pellets of defined size



www.merckmillipore.com/caustics-bases

# Unique production technique leads to unique quality

Merck Millipore's product range for caustic alkalis and bases for analysis includes sodium and potassium hydroxide pellets and the corresponding solutions, as well as ammonia solutions in various concentrations and quality grades.

Caustic alkalis and bases are produced in a dedicated plant at Merck Millipore in Darmstadt, Germany. Product-specific production lines anticipate cross-contamination between the different materials. Our unrivaled production techniques and sophisticated quality management system enable us to offer caustic alkalis and bases of outstanding quality. Our products are even qualified for demanding lab applications in regulated markets, such as the pharmaceutical industry. Furthermore, most caustic alkalis and bases for analysis EMSURE® meet or exceed the requirements of international standards such as ACS, ISO and / or Reag. Ph Eur.



### Our speciality: Pellets with very low values of impurities

Thanks to our unique production process and extensive experience, Merck Millipore is able to offer pellets with extremely low values of impurities. Thus, you can expect sodium hydroxide pellets with very low potassium impurities, and potassium hydroxide pellets with equally low sodium impurities. The specified maximum values for potassium and sodium for these products are indicated in the product name as well as in the Certificate of Analysis.

# Why choose Merck Millipore caustic alkalis and bases for analysis?

## Your advantages

- Comprehensive product range
- Defined outstanding quality for all applications
- High reliability and reproducibility of results through batch-to-batch consistency and extremely low limiting values of impurities
- Most caustic alkalis and bases for analysis EMSURE® meet or exceed the requirements of ACS, ISO and / or Reag. Ph Eur
- Uniform pellets of defined size thanks to unique production technique
- Made using specially selected raw materials
- Complete analysis of each batch with specific data on content or other limits provided in the Certificate of Analysis
- Sophisticated quality management system with audits, thus comprehensively certificated data
- Fulfill individual customer requests

## Useful information about ...

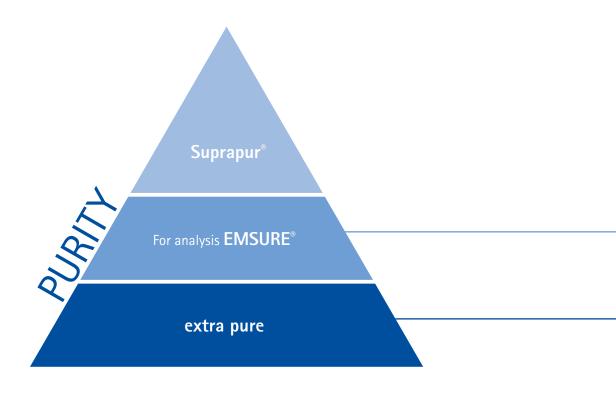
### Available concentrations of some bases

Name	Weight	Density	Concentration
	%	g/cm³ [20°C]	mol/l *
Ammonia solution	32	0.88	16.5
	30	0.9	15.5
	25	0.91	13.5
Potassium hydroxide solution	30	1.48	7
Sodium hydroxide solution	10	1.11	3
	27	1.3	9
	32	1.35	11
	45	1.48	16.5
	50	1.53	19

\* rounded off

# The ideal quality grade for each application

Each application is different and therefore we offer a wide range of caustic alkalis and bases in various grades and packaging sizes. Whether you need to perform a highly critical analysis or prepare basic lab applications, you will always find the most suitable reagent with our quality grades and brand names.



## Specification

ACS	Standards of the American Chemical Society
ISO	Standards of the International Organization for Standardization
Reag. Ph Eur	Reagents specified by the European Pharmacopoeia

### Suprapur<sup>®</sup> caustics and bases

- Suprapur<sup>®</sup> reagents are ideal for digestion in trace analysis. They have extremely low contents of all cations
- For more information on Suprapur® reagents please see the chapter »High purity acids and bases«

### Caustics and bases for analysis EMSURE® | ACS, ISO, Reag. Ph Eur

- Our premium grade for all regulated and highly demanding lab applications
- Worldwide best and most extensive product specifications
- Widest range of pack sizes

### Caustics and bases for analysis EMSURE® are used:

- For sample preparation
- As neutralization and digestion agents
- For the production of alkaline salts
- For the production of buffer solutions
- For the regeneration of anionic ion exchangers
- For use in refining and precipitation processes
- As alkaline cleansing agents

### Caustics and bases extra pure

- For preparative lab applications and cleaning purposes
- Specifications with the most important parameters
- Available in different pack sizes

#### Caustics and bases extra pure are used:

- For the production of sodium and potassium hydroxide solutions
- For adjusting the pH of liquids
- As alkaline cleansing agents
- As neutralization agents for acids

# Ordering information Caustics and bases for analysis EMSURE®

### **Caustics and bases** Pellets

Product	Content	Packaging	Ord. No.
Potassium hydroxide pellets	500 g	Plastic bottle	1.05033.0500
for analysis EMSURE®	1 kg	Plastic bottle	1.05033.1000
	5 kg	Plastic bottle	1.05033.5000
	25 kg	Fibre carton	1.05033.9025
	50 kg	Fibre carton	1.05033.9050
Potassium hydroxide pellets (max. 0.002 % Na)	250 g	Plastic bottle	1.05021.0250
for analysis EMSURE® ACS, ISO, Reag. Ph Eur	1 kg	Plastic bottle	1.05021.1000
	5 kg	Plastic bottle	1.05021.5000
	25 kg	Fibre carton	1.05021.9025
Potassium hydroxide pellets (max. 0.05 % Na)	1 kg	Plastic bottle	1.05029.1000
for analysis EMSURE® ACS, Reag. Ph Eur	12 kg	Plastic container	1.05029.9012
	50 kg	Fibre carton	1.05029.9050
Sodium hydroxide pellets	500 g	Plastic bottle	1.06498.0500
for analysis EMSURE® ISO	1 kg	Plastic bottle	1.06498.1000
	5 kg	Plastic bottle	1.06498.5000
	25 kg	Fibre carton	1.06498.9025
	50 kg	Fibre carton	1.06498.9050
Sodium hydroxide pellets (max. 0.0002 % K)	250 g	Plastic bottle	1.06495.0250
for analysis EMSURE® ACS, Reag. Ph Eur	1 kg	Plastic bottle	1.06495.1000
	5 kg	Plastic bottle	1.06495.5000
Sodium hydroxide pellets (max. 0.02 % K)	1 kg	Plastic bottle	1.06469.1000
for analysis EMSURE® ACS, Reag. Ph Eur	5 kg	Plastic bottle	1.06469.5000
	12 kg	Plastic container	1.06469.9012
	50 kg	Fibre carton	1.06469.9050



### **Caustics and bases** Solutions

Product	Content	Packaging	Ord. No.
Ammonia solution 25 %	1	Glass bottle	1.05432.1000
for analysis EMSURE®	1	Plastic bottle	1.05432.1011
	2.5 I	Glass bottle	1.05432.2500
	5	Plastic bottle	1.05432.5000
	25 I	Plastic container	1.05432.9025
Ammonia solution 28 – 30 %	11	Glass bottle	1.05423.1000
or analysis EMSURE® ACS, Reag. Ph Eur		Glass bottle	1.05423.2500
	25 I	Plastic container	1.05423.9025
Ammonia solution 32 %	1	Glass bottle	1.05426.1000
extra pure	2.5 I	Glass bottle	1.05426.2500
Potassium hydroxide solution 47 %	1	Plastic bottle	1.05545.1000
for analysis EMSURE®	25 I	Plastic container	1.05545.9025
Sodium hydroxide solution min. 10 % (1.11)	1	Plastic bottle	1.05588.1000
for analysis EMSURE®	10 I	Plastic container	1.05588.9010
Sodium hydroxide solution min. 27 % (1.30) (for the determination of nitrogen)	2.5	Plastic bottle	1.05591.2500
for analysis EMSURE®	25 I	Plastic container	1.05591.9025
Sodium hydroxide solution about 32 %	2.5	Plastic bottle	1.05587.2500
extra pure	5	Plastic bottle	1.05587.5000
	25 I	Plastic container	1.05587.9025
	200 I	Plastic barrel	1.05587.9200
Sodium hydroxide solution about 32 % (for the determination of nitrogen)	2.5	Plastic bottle	1.05590.2500
for analysis EMSURE®	25 I	Plastic container	1.05590.9025
Sodium hydroxide solution min. 45 %	2.5	Plastic bottle	1.11360.2500
for analysis EMSURE®	25 I	Plastic container	1.11360.9025
Sodium hydroxide solution 50 %	1	Plastic bottle	1.58793.1000
for analysis EMSURE®	25 I	Plastic container	1.58793.9025



# Metals and metal oxides

### You can count on our experience

Merck Millipore produces chemicals of internationally renowned high quality and purity. All over the world we supply laboratories and production departments across almost every industry with the products and services they need. The fact that we are in constant dialog with our customers enables us to supply very specific industrial sectors with top-class products.



# Metals and metal oxides

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### Advantages

- Top-level quality
- Excellent batch-to-batch consistency
- Application-led purity and specs
- Extensive choice of products and package sizes
- Comprehensive and individual support



www.merckmillipore.com/metals

# Metal salts, metals and noble metals from Merck Millipore

### Highest quality standards

Our extensive range of premium metal salts, metals and noble metals is used in a multitude of applications in R&D laboratories, production departments and in quality control. As customer of Merck Millipore's metal salts, metals and noble metals you can be confident that you are purchasing products of the highest quality, no matter whether you are involved in the chemical industry, in the manufacture of ceramics or electronic components, or in food analytics.

# Selected raw materials are the cornerstone of our products

Using meticulous production techniques, our highly qualified workforce ensures that, batch for batch, they manufacture constantly premium products under tightly controlled conditions to provide you with consistently high purity and specs you need for your everyday work.

# Quality management and service – total satisfaction

We want all our customers to be always totally satisfied with the products they obtain from us, which is why we also place such emphasis on other aspects such as product documentation, traceability and certification.

### Purity grades meeting your demands

Merck Millipore sells products in purity grades consistent with what the market demands. Our metal salts, metals and noble metals come in »for analysis EMSURE®« and »extra pure« grades, so you can usually choose a product to suit your precise needs.

For analysis **EMSURE**®

extra pure

## Typical applications











**Chemical research** Wood's alloy is used as heating bath medium. Synthesis Many noble metals are used as catalysts.

Laboratory Cesium chloride is used as reagent for density gradient centrifugation. **Piezoelectric ceramics** Titanium(IV) oxide, antimony(III) oxide, bismuth(III) oxide and lead(II) oxide are used during synthesis. Electroplating

Various metals are used for deposition onto items such as bathroom fittings.

## Our purity grades

### for analysis EMSURE®

- high purity (generally >99.0 %)
- tested for trace impurities
- conforming to international standards such as ACS, ISO, Reag. Ph Eur

### extra pure

- extra purity (generally >99.0 %)
- decisive parameters listed

# Ordering information Metal salts, metals and noble metals

### Noble metals A-Z

Product	CAS No.	Chemical formula	Content	Packaging	Ord. No.
Hexachloroplatinic(IV) acid solution about 10 %	-	Pt	5 ml	Glass bottle	1.07341.0005
(3.8 % Pt) for analysis EMSURE®			25 ml	Glass bottle	1.07341.0025
Palladium powdered 99+	7440-05-3	Pd	1 g	Glass bottle	1.19225.0001
			5 g	Glass bottle	1.19225.0005
Platinum powder 99+	7440-06-4	Pt	1 g	Glass bottle	1.19232.0001
Platinum black 98+	7440-06-4	Pt	5 g	Glass bottle	1.19233.0005
Potassium hexachloroplatinate(IV) 99+	16921-30-5	$K_2PtCI_6$	1 g	Glass bottle	1.19238.0001
			5 g	Fibre case	1.19238.0005
Potassium tetrachloroplatinate(II) 99+	10025-99-7	Pt	5 g	Glass bottle	1.19235.0005
Ruthenium(III) chloride hydrate	14898-67-0	$RuCl_3 \cdot x H_2O$	5 g	Glass bottle	1.19247.0005
			25 g	Glass bottle	1.19247.0025
Silver wool for elemental analysis	7440-22-4	Ag	10 g	Glass bottle	1.01506.0010
Silver chloride 99+	7783-90-6	AgCl	25 g	Plastic bottle	1.19203.0025
			100 g	Plastic bottle	1.19203.0100
Silver diethyldithiocarbamate for analysis (reagent for arsenic and antimony) EMSURE® ACS, Reag. Ph Eur	1470-61-7	$C_5H_{10}AgNS_2$	5 g	Glass bottle	1.01515.0005
Silver oxide 99+	20667-12-3	Ag <sub>2</sub> O	25 g	Plastic bottle	1.19208.0025
			100 g	Plastic bottle	1.19208.0100
Tetrachloroauric(III) acid trihydrate 99.5 %	16961-25-4	$AuCl_4H \cdot 3 H_2O$	1 g	EPS box	1.01582.0001
for analysis EMSURE®			5 g	Fibre case	1.01582.0005



### Metals and metal oxides A-D

Product	CAS No.	Chemical formula	Content	Packaging	Ord. No.
Alloy acc. to Wood, melting point about 73°C bar diameter about 1 cm	8049-22-7	Bi/Pb/Sn/Cd	1 kg	Plastic bag	1.06001.1000
Aluminium (foil) for analysis	7429-90-5	AI	250 g	Plastic box	1.01057.0250
0.3 mm thickness, 30 mm width EMSURE®			1 kg	Plastic bag	1.01057.1000
Aluminium fine powder,	7429-90-5	AI	250 g	Metal can	1.01056.0250
stabilized about 2 % fat			1 kg	Metal can	1.01056.1000
Ammonium thiosulfate 98 %+	7783-18-8	(NH <sub>4</sub> ) <sub>2</sub> S <sub>2</sub> O <sub>3</sub>	1 kg	Plastic bottle	1.01208.1000
			25 kg	Fibre carton	1.01208.9025
Antimony powder for analysis EMSURE® particle size <150 μm	7440-36-0	Sb	25 g	Plastic bottle	1.07832.0025
Antimony(III) chloride for analysis EMSURE®	10025-91-9	SbCl <sub>3</sub>	250 g	Glass bottle	1.07838.0250
			1 kg	Glass bottle	1.07838.1000
Antimony(III) oxide extra pure	1309-64-4	Sb <sub>2</sub> O <sub>3</sub>	2.5 kg	Plastic bottle	1.07835.2500
			50 kg	Fibre carton	1.07835.9050
Antimony(III) oxide for analysis EMSURE®	1309-64-4	Sb <sub>2</sub> O <sub>3</sub>	100 g	Plastic bottle	1.07836.0100
			1 kg	Fibre carton	1.07836.1000
Arsenic lumps for analysis (protective gas: nitrogen) EMSURE®	7440-38-2	As	25 g	Metal can	1.00115.0025
Bismut(III)-oxide extra pure	1304-76-3	Bi <sub>2</sub> O <sub>3</sub>	25 kg	Fibre carton	1.01862.9025
Bromine extra pure	7726-95-6	Br <sub>2</sub>	250 g	Glass bottle	1.01945.0250
			1 kg	Glass bottle	1.01945.1000
Bromine for analysis EMSURE® ACS, ISO,	7726-95-6	Br <sub>2</sub>	50 g	Glass bottle	1.01948.0050
Reag. Ph Eur			250 g	Glass bottle	1.01948.0250
			1 kg	Glass btl. pl. coat.	1.01948.1000
Calcium granular,	7440-70-2	Са	100 g	Glass bottle	1.02053.0100
particle size about 2–6 mm			500 g	Glass bottle	1.02053.0500
Cadmium granular, for analysis EMSURE® particle size about 3-6 mm	7440-43-9	Cd	250 g	Metal can	1.02004.0250
Cadmium coarse powder, for analysis EMSURE® and	7440-43-9	Cd	250 g	Metal can	1.02001.0250
for filling reductors particle size about 0.3 - 1.6 mm			1 kg	Metal can	1.02001.1000
Cesium chloride extra pure	7647-17-8	CICs	1 kg	Plastic bottle	1.02041.1000
Cesium chloride for analysis EMSURE®	7647-17-8	CICs	25 g	Glass bottle	1.02038.0025
			100 g	Plastic bottle	1.02038.0100
Cesium nitrate 99+	7789-18-6	CsNO <sub>3</sub>	25 g	Glass bottle	1.02856.0025
		-	1 kg	Plastic bottle	1.02856.1000
Chromium(VI) oxide extra pure	1333-82-0	CrO <sub>3</sub>	2.5 kg	Glass bottle	1.00227.2500
Chromium(VI) oxide for analysis EMSURE®	1333-82-0	CrO <sub>3</sub>	250 g	Glass bottle	1.00229.0250
Copper foil about 0.1 mm thickness	7440-50-8	Cu	250 g	Fibre case	1.02700.0250
Copper fine powder for analysis EMSURE®	7440-50-8	Cu	250 g	Plastic bottle	1.02703.0250
particle size <63 μm (>230 mesh ASTM)			1 kg	Plastic bottle	1.02703.1000
Copper(II) oxide granular for analysis EMSURE®	1317-38-0	CuO	500 g	Plastic bottle	1.02768.0500
Copper(II) oxide powder extra pure	1317-38-0	CuO	500 g	Plastic bottle	1.02761.0500
•			25 kg	Plastic drum	1.02761.9025
Copper(II) oxide powder for analysis EMSURE®	1317-38-0	CuO	100 g	Plastic bottle	1.02766.0100
- *			5	Plastic bottle	1.02766.0500
			500 g		11027 0010000
Devarda's alloy for analysis EMSURE®	8049-11-4	Cu/Al/Zn	250 g	Plastic bottle	1.05341.0250

# Ordering information Metals and metal oxides

### Metals and metal oxides I-T

	Product	CAS No.	Chemical formula	Content	Packaging	Ord. No.
I	Iron for analysis reduced EMSURE®,	7439-89-6	Fe	100 g	Plastic bottle	1.03819.0100
	particle size 10 µm			500 g	Plastic bottle	1.03819.0500
	di-lodine pentoxide for analysis EMSURE® granular 0.5 – 2.5 mm	12029-98-0	I <sub>2</sub> 0 <sub>5</sub>	100 g	Glass bottle	1.00358.0100
	Iodine sublimated for analysis EMSURE®	7553-56-2	l <sub>2</sub>	100 g	Glass bottle	1.04761.0100
	ACS, ISO, Reag. Ph Eur			500 g	Glass bottle	1.04761.0500
L	Lanthanum(III) oxide	1312-81-8	$La_2O_3$	100 g	Plastic bottle	1.12220.0100
				500 g	Plastic bottle	1.12220.0500
	Lead foil for analysis EMSURE® about 0.25 mm thick	7439-92-1	Рb	500 g	Plastic bag	1.07365.0500
	Lead(II) oxide extra pure	1317-36-8	РЬО	1 kg	Plastic bottle	1.05658.1000
				5 kg	Plastic bottle	1.05658.5000
				50 kg	Plastic drum	1.05658.9050
	Lead(II) oxide for analysis EMSURE®	1317-36-8	РьО	250 g	Plastic bottle	1.07401.0250
				1 kg	Plastic bottle	1.07401.1000
	Lead(IV) oxide for analysis EMSURE®	1309-60-0	PbO <sub>2</sub>	250 g	Plastic bottle	1.07407.0250
				1 kg	Plastic bottle	1.07407.1000
	Lithium hydroxide about 98 %+	1310-65-2	LiOH	100 g	Plastic bottle	1.05691.0100
				1 kg	Plastic bottle	1.05691.1000
Μ	Magnesium foil 0.15 – 0.30 mm thickness, 3 mm wide	7439-95-4	Mg	1	Fibre case	1.05812.0001
	Magnesium powder particle size about 0.06 – 0.3 mm	7439-95-4	Mg	1 kg	Metal can	1.05815.1000
	Magnesium oxide for analysis EMSURE®	1309-48-4	MgO	100 g	Plastic bottle	1.05865.0100
				500 g	Plastic bottle	1.05865.0500
	Magnesium oxide for analysis EMSURE®	1309-48-4	MgO	100 g	Plastic bottle	1.05866.0100
	(max. 0.001 % SO <sub>4</sub> ) ACS			500 g	Plastic bottle	1.05866.0500
	Manganese(IV) oxide powder	1313-13-9	MnO <sub>2</sub>	1 kg	Glass bottle	1.05957.1000
	Molybdenum(VI) oxide extra pure	1313-27-5	MoO <sub>3</sub>	250 g	Plastic bottle	1.00401.0250
	Molybdenum(VI) oxide for analysis EMSURE®	1313-27-5	MoO <sub>3</sub>	100 g	Plastic bottle	1.00403.0100
				500 g	Plastic bottle	1.00403.0500
Ν	Niobium(V) oxide 99+	1313-96-8	Nb <sub>2</sub> O <sub>5</sub>	100 g	Plastic bottle	1.06868.0100
Р	Phenylhydrazinium chloride for analysis EMSURE®	59-88-1	$C_6H_9CrN_2$	100 g	Glass bottle	1.07253.0100
	Reag. Ph Eur			250 g	Glass bottle	1.07253.0250
R	Rubidium chloride for analysis EMSURE®	7791-11-9	RbCl	25 g	Glass bottle	1.07615.0025
S	Selenium black 99+	7782-49-2	Se	50 g	Plastic bottle	1.07714.0050
				250 g	Plastic bottle	1.07714.0250
				1 kg	Plastic drum	1.07714.1000
	Sodium, rod diameter 2.5 cm	7440-23-5	Na	250 g	Glass bottle	1.06260.0250
	(protective liquid: paraffin oil)	770		1 kg	Glass bottle	1.06260.1000
	Sulfur for external use Ph Eur, BP	7704-34-9	S	2.5 kg	Plastic bottle	1.07983.2500
÷	Total and an and a second s	101170 10 0		25 kg	Fibre carton	1.07983.9025
Т	Tetraphenylarsonium chloride monohydrate for analysis EMSURE®	104170-16-3	$C_{24}H_{20}AsCI \cdot H_2O$	5 g	Glass bottle	1.08150.0005
	Tin fine powder pure (particle size <71 $\mu$ m)	7440-31-5	Sn	250 g	Plastic bottle	1.07807.0250
				1 kg	Plastic bottle	1.07807.1000

### Metals and metal oxides T-Z

Product	CAS No.	Chemical formula	Content	Packaging	Ord. No.
Tin foil about 0.04 mm thick	7440-31-5	Sn	200 strips	Plastic box	1.07826.0001
Tin granulated for analysis EMSURE®	7440-31-5	Sn	200 strips	Plastic bottle	1.07806.0250
(particle size about 4 mm) Reag. Ph Eur	7440-31-5	211	3	Plastic bottle	
Tin(IV) oxide extra pure	7440-31-5	Sn	1 kg	Plastic bottle	1.07806.1000
lin(iv) oxide extra pure	7440-31-5	Sn	250 g		1.07818.0250
			1 kg	Plastic bottle	1.07818.1000
			5 kg	Plastic bottle	1.07818.5000
	10100 07 7		25 kg	Fibre carton	1.07818.9025
Titanium(IV) oxide for analysis EMSURE® Reag. Ph Eur	13463-67-7	TiO <sub>2</sub>	1 kg	Plastic bottle	1.00808.1000
			25 kg	Fibre carton	1.00808.9025
			50 kg	Fibre carton	1.00808.9050
Vanadium(V) oxide extra pure	1314-62-1	V <sub>2</sub> O <sub>5</sub>	250 g	Plastic bottle	1.00824.0250
			1 kg	Plastic bottle	1.00824.1000
Yttrium oxide 99+	1314-36-9	Y <sub>2</sub> O <sub>3</sub>	25 g	Plastic bottle	1.12412.0025
Zinc granular extra pure	7440-66-6	Zn	1 kg	Plastic bottle	1.08755.1000
particle size about 3 – 8 mm			10 kg	Plastic bottle	1.08755.9010
Zinc granular for analysis EMSURE®	7440-66-6	Zn	500 g	Plastic bottle	1.08780.0500
particle size about 3 – 8 mm ISO			1 kg	Plastic bottle	1.08780.1000
Zinc coarse powder for analysis suitable for filling	7440-66-6	Zn	250 g	Metal can	1.08756.0250
of reductors, particle size about 0.3 – 1.5 mm EMSURE® Reag. Ph Eur			1 kg	Metal can	1.08756.1000
Zinc powder for analysis,	7440-66-6	Zn	500 g	Metal can	1.08789.0500
particle size <45 µm EMSURE®			1 kg	Metal can	1.08789.1000
Zinc sticks, triangular cross section about 8 mm	7440-66-6	Zn	500 g	Plastic bag	1.08782.0500
for analysis EMSURE®					
Zinc dust particle size <63 μm	7440-66-6	Zn	1 kg	Metal can	1.08774.1000
			50 kg	Steel drum	1.08774.9050
Zinc oxide for analysis EMSURE®	1314-13-2	ZnO	500 g	Plastic bottle	1.08849.0500
ACS, Reag. Ph Eur			1 kg	Plastic bottle	1.08849.1000
			25 kg	Fibre carton	1.08849.9025
Zirconium(IV) oxide chloride octahydrate for analysis EMSURE®	13520-92-8	$ZrOCl_2 \cdot 8 H_2O$	100 g	Plastic bottle	1.08917.0100



# Instrumental inorganic analysis

Merck Millipore reagents for instrumental analysis are produced from high purity materials and analyzed using the most sophisticated methods. Choose from our broad range of reagents to achieve the most reliable results!



# Volumetric solutions

Titripur® volumetric solutions - the standard for precise titration results. With Titripur® grade we ensure that your analyses always meet the highest demands.

## Karl Fischer reagents and standards

Apura® - Karl Fischer reagents and standards for sophisticated, precise and consistent water determination. Excellent quality distinguishes our Apura® range.

# **Reference** materials

Certipur® reference materials for calibration and qualification in instrumental analysis ensure the most reliable results. Our certified reference materials come with a comprehensive Certificate of Analysis.

# Fluxes for XRF

Spectromelt<sup>®</sup> materials for fluxing digestion or pressing of powder tablets are the perfect aids for your X-ray fluorescence analysis. The purity of Spectromelt® borates provides the necessary degree of precision and our grinding aid materials are specially developed for automated sample preparation.

## High purity acids and bases

Suprapur® | Ultrapur acids offer the most suitable purity of wet digestion materials. High purity acids from Merck Millipore have low impurities during the entire minimum shelf life.

# High purity salts

Suprapur® salts for sample preparation in instrumental analysis. Be sure to prepare your samples with the highest purity!

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# Volumetric solutions Titripur<sup>®</sup> | Titripac<sup>®</sup> | Titrisol<sup>®</sup> | Titriplex<sup>®</sup>

Sophisticated and correct analyses require precisely adjusted, certified volumetric solutions. Whether you're titrating reducing or oxidizing substances, acids, bases or complexing reagents in either aqueous or non-aqueous solutions: Merck Millipore offers the right solutions for every application.

# The measure of all things – volumetric solutions from Merck Millipore.



# Volumetric solutions

## Contents

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## Advantages

- Certified and consistently high level of quality
- Innovative and optimum packaging systems
- Reliable and precise analyses
- Measured in a DIN EN ISO / IEC 17025 accredited laboratory

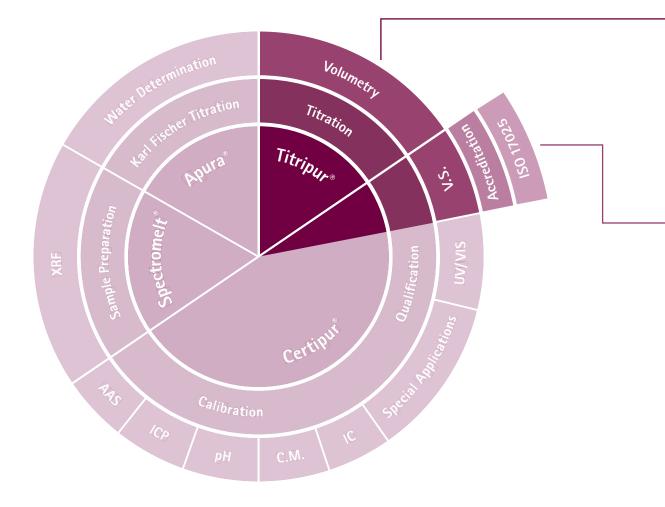


www.merckmillipore.com/titration

# Titripur<sup>®</sup> – our quality standard in titration

### Titripur<sup>®</sup> – precision and quality

Like all Merck Millipore products, Titripur<sup>®</sup> volumetric solutions are subject to stringent manufacturing and testing requirements. From the selection of the raw materials and packaging on up to quality control, the highest demands are placed on purity and quality. All Titripur<sup>®</sup> volumetric solutions are measured in our DIN EN ISO / IEC 17025 accredited laboratory.



### Specification / Traceability

NIST	National Institute of Standards and Technology, Gaithersburg,
	Maryland, USA
Reag. Ph Eur	Reagents specified by the European Pharmacopoeia
USP	United States Pharmacopoeia requirements for reagents

### Titripur<sup>®</sup> quality for your individual needs

### Titripur<sup>®</sup> | Volumetric solutions

Under the brand name Titripur<sup>®</sup>, Merck Millipore offers you first-rate volumetric solutions. Volumetric solution described in the European and the US Pharmacopeia are in accordance with the reagents chapter of the Pharmacopeia. For each solution, titer determination is performed under optimum and standardized conditions. All volumetric solutions are traceable to certified secondary reference materials, which in turn are directly traceable to standard reference materials from the NIST. Consequently, all volumetric solutions from Merck Millipore are traceable to NIST standard reference material and measured in our DIN EN ISO / IEC 17025 accredited laboratory.

### Certipur<sup>®</sup> | Volumetric standard [V.S.]

Volumetric standards are used for the standardization – titer determination – of volumetric solutions and for the qualification of the titration system. Influencing factors such as temperature, instrument variances, different methods of handling, weighing errors etc. and the volumetric solution itself can impact the titration result. To compensate for these factors, titer determination under working conditions is necessary in the respective laboratory. Merck Millipore is in the scope of the DIN EN ISO / IEC 17025 accreditation for mass fraction of titrimetric standards. Secondary standard reference materials for titration are traceable to standard reference material from NIST.

#### More information

See chapter »Standard reference materials and buffers« on page 172

# Titripur<sup>®</sup> – for reliable quality all the time

The standard for precise titration results

# With the Titripur<sup>®</sup> grade, you can be sure that your analyses always meet the highest demands.

Specification	i l	
1.09141.1000 Spdium hydroxide	solution c(NaCH) = 0.1 molit (0.1 N) Titrpurit	
lainah shika isan na shi kape ka	period for Pranticeport Ph Tatl	
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### That's Merck Millipore: Quality assurance throughout the entire manufacturing and quality control process

The manufacturing and quality control process for Titripur® solutions meets the highest standards. The Certificate of Analysis contains all the information important for quality management documentation. Furthermore, it includes information on traceability and data on the NIST batch used for standardization. Additionally it gives the information about the measurement in the ISO 17025 accredited laboratory.

### Titripur<sup>®</sup> – volumetric solutions in accordance with the reagents part of European and US pharmacopoeias

The Titripur<sup>®</sup> line also includes volumetric solutions that are used specifically for analytical purposes in the pharmaceutical industry and correspond to the European and US Pharmacopoeia for analytical applications. For available solutions, this is indicated in the certificate.

# Titripur<sup>®</sup> – volumetric solutions made from raw materials in accordance with Ph Eur

The Titripur<sup>®</sup> line also includes volumetric solutions made from raw materials in accordance with the European Pharmacopoeia. The relevant information can be found in the certificate.

### Titriplex<sup>®</sup> – for complexometric titration

For chelate ligands, Merck Millipore offers the well known Titriplex<sup>®</sup> brand to determine metal ions by complexometric titration. In addition to solid substances, ready-to-use solutions are also available in PE bottles and Titripac<sup>®</sup>.



# Innovative packaging – optimized for every application

In order to ensure high product quality, packaging is needed to protect the solution from impurities and contamination. Merck Millipore therefore offers a wide range of high-quality packaging options for volumetric solutions and only uses glass and plastic grades that do not alter any of the product's characteristics. All our packaging materials have been tested for their quality and atmospheric permeability, thereby ensuring the purity of the solutions, also during storage up to their minimum shelf life for sealed, original containers.

### Titrisol® - concentrates for greater flexibility

For users who regularly perform different types of analyses, we also offer most volumetric solutions as Titrisol<sup>®</sup> concentrates.

Every ampoule contains a precisely defined quantity of substance that is normally filled to a final volume of 1 liter. However, any other concentrations can be produced by diluting correspondingly.



### Titrisol<sup>®</sup> advantages

- Space-saving
- Flexible and easy to use
- Individual concentrations can be prepared



# Titripur<sup>®</sup> – ready-to-use solutions reliable and precise

Merck Millipore offers ready-to-use volumetric solutions for daily laboratory analyses. These can be connected directly to a titrator via an adapter.

Depending on the amount required, volumetric solutions are available in 0.5 l, 1 l, 2.5 l, 5 l, 10 l and 25 l PE bottles or canisters. In addition, volumetric solutions, such as acetic acid, which cannot be stored in a stable manner for an extended period of time in PE containers, are available in 1 l and 2.5 l glass bottles.

Merck Millipore also offers Titripur<sup>®</sup> grade ready-to-use solutions in 4 and 10 liter Titripac<sup>®</sup>.

### Polyethylene bottles and canisters

- Pack diversity for every different requirement
- Unbreakable and contamination-free material
- Direct connection to the titrator is possible
- The stability of the solution is ensured for the entire shelf life

### **Glass bottles**

• Ensuring the stability of special volumetric solutions

# Titripac<sup>®</sup> – the innovative packaging solution that saves costs and time

# Titripac<sup>®</sup> – reliable solution from the first to the last drop

Titripac<sup>®</sup> from Merck Millipore is an innovative and safe packaging option for high-quality, ready-to-use volumetric solutions. Its economical and ecological advantages enable you to optimize your working processes. The consistent quality of a volumetric solution is ensured from the first to the last drop. A hermetically sealed package system makes this possible. Contamination caused by air, carbon dioxide or microorganisms is excluded. Titripac<sup>®</sup> saves you the trouble of time-consuming process of checking the solutions as well as the costintensive disposal of contaminated residual amounts. The carton can simply be disposed of together with paper and the internal liner can be easily folded together prior to disposal.

Titripac<sup>®</sup> is extremely easy to use. The integrated spout appears simply by pressing on the pack. By opening the tap, liquid can be withdrawn at any time – conveniently and without the risk of contamination. In addition, Titripac<sup>®</sup> can be connected directly to the titrator via an adapter.



Titripac<sup>®</sup> is available in 4 liter and 10 liter sizes.





## Titripac<sup>®</sup> advantages

- No contamination: Hermetically sealed pack
- Easy to use: Integrated withdrawal tap, direct connection to a titrator is possible
- Saves costs and time: No unnecessary titer determinations, no contaminated residual amounts
- Environmentally friendly disposal: Reduced package waste, as carton and internal liner can be disposed of separately



Precise analyses require precisely adjusted volumetric solutions. With Titripac<sup>®</sup> you can be sure that you've got a consistent solution up to the very last drop. A direct connection to the titrator via an adaptor makes lab work easier and helps to avoid contamination.

# Ordering information Titripur<sup>®</sup> volumetric solutions

			Ready-to-use solutions	
Product	Concentration	Titrisol <sup>®</sup> Ampoules	Glass bottles,	
		for 1 l solution	PE bottles/canisters	Titripac®
		Ord. No.	Ord. No.	Ord. No.
Acetic acid	0.1 mol/l (0.1 N)	1.09944.0001	-	-
	1 mol/l (1 N)	1.09951.0001	-	-
	1 mol/l (1 N) prepared from	-	25 l: 1.99061.9025	-
	raw materials acc. to Ph Eur			
Ammonium cer(IV)nitrate solution	0.1 mol/l (0.1 N)	-	1 l: 1.02277.1000 <sup>1</sup>	-
Ammonium iron(II) sulfate solution	0.1 mol/l (0.1 N)	1.09864.0001	-	-
Ammonium thiocyanate solution	0.1 mol/l (0.1 N)	1.09900.0001	1 l: 1.09079.1000 <sup>1,2</sup>	-
Barium chloride solution	0.05 mol/l (0.1 N)	1.09962.0001	-	-
Barium perchlorate solution	0.005 mol/l in 2-propanol/ water (80 : 20)	-	1 l: 1.09086.1000 <sup>2</sup>	-
Bromide bromate solution	0.05 mol/l (0.1 N)	1.09905.0001	-	-
Cer (IV) sulfate solution	0.1 mol/l (0.1 N)	-	1 l: 1.09092.1000 <sup>1,2</sup>	-
Copper sulfate solution	0.1 mol/l	-	1 l: 1.02784.1000	-
Copper-di-ammonium Titriplex <sup>®</sup> solution	0.1 mol/l	-	0.5 l: 1.05217.0500	-
Hanus solution (lodomonobromide solution)	0.1 mol/l in acetic acid	-	1 l: 1.09164.1000	-
Hyamine, 1622 solution for the determination of anionic tensides	0.004 mol/l	-	1 l: 1.15480.1000	-
Hydrochloric acid	0.01 mol/l (0.01 N)	1.09974.0001	-	-
	0.1 mol/l	1.09973.0001	1 l: 1.09060.1000 <sup>1,2</sup>	4 l: 1.09060.4000
			5 l: 1.09060.5000 <sup>1,2</sup>	10 l: 1.09060.9010
			25 l: 1.09060.9025 <sup>1,2</sup>	
	0.1 mol/l (0.1 N) in 2-propanol	-	1  : 1.00326.1000	-
	0.357 mol/l (1/2.8 N)	-	-	10 l: 1.13136.9010
	0.5 mol/l (0.5 N)	1.09971.0001	1 l: 1.09058.1000 <sup>1,2</sup>	4 l: 1.09058.4000
			5 l: 1.09058.5000 <sup>1,2</sup>	
			25 l: 1.09058.9025 <sup>1,2</sup>	
	1 mol/l (1 N)	1.09970.0001	1 l: 1.09057.1000 <sup>1,2</sup>	4 l: 1.09057.4000
			2.5 l: 1.09057.2500 1.2	10 l: 1.09057.9010
			5 l: 1.09057.5000 1.2	
			25 l: 1.09057.9025 <sup>1,2</sup>	
	1 mol/l (1 N) prepared from	-	25 l: 1.99070.9025	-
	raw materials acc. to Ph Eur			
	2 mol/l (2 N)	-	1 l: 1.09063.1000	-
			25 l: 1.09063.9025	
	3.571 mol/l (1/0.28 N)	-	-	10 l: 1.13134.9010
	5 mol/l (5 N)	-	1 l: 1.09911.1000	-
lodide-iodate solution	1/128 mol/l   2 (1/64 N)	1.09914.0001	-	-
lodine solution	0.05 mol/l (0.1 N)	1.09910.0001	1 l: 1.09099.1000 <sup>1,2</sup>	-
	0.5 mol/l (1 N)	-	1 l: 1.09098.1000 <sup>1</sup>	-
Mercury(II)nitrate solution	0.05 mol/l (0.1 N)	-	1 l: 1.09143.1000 <sup>2</sup>	-
Nitric acid	0.1 mol/l	1.09964.0001	-	-
	1 mol/l (1 N)	1.09966.0001	-	-
	10 mol/l (10 N)	-	1 l: 1.00630.1000	-

1: Solution in accordance with the reagents chapter of Pharm. Eur (European Pharmacopoeia) | 2: Solution in accordance with the reagents chapter of USP (United States Pharmacopeia)

			Ready-to-use solutions	
Product	Concentration	Titrisol® Ampoules for 1 I solution Ord. No.	Glass bottles, PE bottles/canisters Ord. No.	Titripac® Ord. No.
Perchloric acid	0.1 mol/l (0.1 N) in water-free acetic acid	-	1 l: 1.09065.1000 <sup>1,2</sup>	-
Potassium bromate solution	1/60 mol/l (0.1 N)	1.09925.0001	-	-
Potassium dichromate solution	1/60 mol/l (0.1 N)	1.09928.0001	-	-
	1/24 mol/l (0.25 N)	-	1 l: 1.09118.1000	-
	0.020 mol/l	-	1 l: 1.09119.1000	-
Potassium hydroxide solution	0.1 mol/l (0.1 N)	1.09921.0001	1 l: 1.09112.1000 <sup>1</sup>	-
	0.1 mol/l (0.1 N) in ethanol	-	1 l: 1.09115.1000 <sup>1,2</sup> 2.5 l: 1.09115.2500 <sup>1,2</sup>	-
	0.1 mol/l (0.1 N) in methanol	-	1 l: 1.11587.1000	-
	0.1 mol/l (0.1 N) in 2-propanol	-	1 l: 1.05544.1000	-
	0.5 mol/l (0.5 N)	1.09919.0001	5 l: 1.11586.5000 <sup>2</sup>	_
	0.5 mol/l (0.5 N) in ethanol	-	1 l: 1.09114.1000 <sup>1,2</sup> 2.5 l: 1.09114.2500 <sup>1,2</sup>	-
	0.5 mol/l (0.5 N) in methanol	-	1 l: 1.09351.1000	_
	1 mol/l (1 N)	1.09918.0001	1 l: 1.09108.1000 <sup>1,2</sup>	-
	1 mol/l (1 N) max. 0.4 ppm Ca	-	1 l: 1.09107.1000	-
	2.0 mol/l (2 N) in methanol	-	2.5 l: 1.11787.2500 <sup>2</sup>	-
Potassium iodate solution	1/60 mol/l (0.1 N)	1.09917.0001	-	-
Potassium permanganate solution	0.002 mol/l (0.01 N)	1.09930.0001	-	-
	0.02 mol/l (0.1 N)	1.09935.0001	-	-
	0.02 mol/l (0.1 N) standardized with sodium thiosulfate	-	1 l: 1.09121.1000 <sup>1</sup>	-
	0.02 mol/l (0.1 N) standardized with oxalate	-	1 l: 1.09122.1000 <sup>2</sup>	-
	0.05 mol/l (0.25 N)	-	2.5 l: 4.80160.2500	-
Silver nitrate solution	0.05 mol/l (0.05 N)	-	1 l: 1.11718.1000	-
	0.1 mol/l (0.1 N)	1.09990.0001	1 l: 1.09081.1000 <sup>1,2</sup> 2.5 l: 1.09081.2500 <sup>1,2</sup>	4 l: 1.09081.4000 <sup>1,2</sup> 10 l: 1.09081.9010 <sup>1,2</sup>
	1 mol/l (1 N)	-	1 l: 1.09080.1000	-
Sodium arsenite solution	0.05 mol/l (0.1 N)	-	1 l: 1.06277.1000 <sup>2</sup>	-
Sodium carbonate solution	0.05 mol/l (0.1 N)	1.09940.0001	_	-

1: Solution in accordance with the reagents chapter of Pharm. Eur (European Pharmacopoeia) | 2: Solution in accordance with the reagents chapter of USP (United States Pharmacopeia)

# Ordering information Titripur<sup>®</sup> volumetric solutions

			Ready-to-use solutions	
Product	Concentration	Titrisol <sup>®</sup> Ampoules	Glass bottles,	
		for 1 I solution	PE bottles/canisters	Titripac®
		Ord. No.	Ord. No.	Ord. No.
Sodium hydroxide solution	0.005 mol/l (0.005 N) in methanol	-	10 l: 4.80621.9010	-
	0.01 mol/l (0.01 N)	1.09961.0001	-	-
	0.02 mol/l (0.02 N)	-	0.5 l: 1.09142.0500	_
	0.1 mol/l (0.1 N)	1.09959.0001	1 l: 1.09141.1000 <sup>1,2</sup>	4 l: 1.09141.4000
			5 l: 1.09141.5000 1,2	10 l: 1.09141.9010
			25 l: 1.09141.9025 1,2	
	0.111 mol/l (0.111 N)	-	25 l: 1.10822.9025	-
	0.2 mol/l (0.2 N)	-	1 l: 1.09140.1000	10  : 1.09140.9010
	0.25 mol/l (0.25 N)	1.09958.0001	1  : 1.09139.1000	10 l: 1.09139.9010
	0.33 mol/l (0.33 N)	-	1 l: 1.05595.1000	10 l: 1.05595.9010
	0.5 mol/l (0.5 N)	1.09957.0001	1  : 1.09138.1000	4 l: 1.09138.4000
			25 l: 1.09138.9025	10 l: 1.09138.9010
	1 mol/l (1 N)	1.09956.0001	1 l: 1.09137.1000 <sup>1,2</sup>	4 l: 1.09137.4000
			2.5 l: 1.09137.2500 1.2	10 l: 1.09137.9010
			25 l: 1.09137.9025 <sup>1,2</sup>	
	1 mol/l (1 N) prepared from raw materials acc. to Ph Eur	-	25 l: 1.99060.9025	-
	2 mol/l (2 N)	_	1 l: 1.09136.1000	_
			25 l: 1.09136.9025	
	4 mol/l (4 N)	-	5 l: 1.11584.5000	-
	5 mol/l (5 N)	-	1 l: 1.09913.1000	-
	6 mol/l (1 N) prepared from raw materials acc. to Ph Eur	-	25 l: 1.99062.9025	-
Sodium thiosulfate solution	0.01 mol/l (0.01 N)	1.09909.0001	-	-
	0.1 mol/l (0.1 N)	1.09950.0001	1 l: 1.09147.1000 <sup>1,2</sup>	4 l: 1.09147.4000
			25  : 1.09147.9025 <sup>1,2</sup>	10 l: 1.09147.9010
Sulfuric acid	0.005 mol/l (0.01 N)	1.09982.0001	-	-
	0.05 mol/l (0.1 N)	1.09984.0001	1 l: 1.09074.1000 <sup>1</sup>	4 l: 1.09074.4000
			5 l: 1.09074.5000 1	10 l: 1.09074.9010
	0.25 mol/l (0.5 N)	-	1 l: 1.09073.1000	4 l: 1.09073.4000
				10 l: 1.09073.9010
	0.5 mol/l (1 N)	1.09981.0001	1 l: 1.09072.1000 <sup>1,2</sup>	4 l: 1.09072.4000
			5 l: 1.09072.5000 <sup>1,2</sup>	10 l: 1.09072.9010
	2.5 mol/l (5 N)	1.09912.0001	1 l: 4.80364.1000	-
			25 l: 4.80364.9025	
Tetra-n-butyl-ammonium hydroxide	0.1 mol/l (0.1 N)	-	0.5 l: 1.09162.0500 <sup>1,2</sup>	-
solution	in 2-propanol/methanol		1 l: 1.09162.1000 <sup>1,2</sup>	
Tetramethylammonium hydroxide	0.1 mol/l (0.1 N)	-	0.25 l: 1.08124.0250 <sup>2</sup>	
solution	in 2-propanol/methanol		1 l: 1.08124.1000	
Titriplex <sup>®</sup> solution A	50 mg CaO/l = 1 ml	1.08419.1000	-	_
Titriplex <sup>®</sup> solution B	10 mg CaO/l = 1 ml	1.08420.1000	5 l: 1.08420.5000	10 l: 1.08420.9010
Titriplex <sup>®</sup> , III solution (Na <sub>2</sub> -EDTA)	0.01 mol/l	1.08446.0001	-	-
	0.1 mol/l	1.09992.0001	1 l: 1.08431.1000 <sup>1</sup>	4 l: 1.08431.4000
				10 l: 1.08431.9010

1: Solution in accordance with the reagents chapter of Pharm. Eur (European Pharmacopoeia) | 2: Solution in accordance with the reagents chapter of USP (United States Pharmacopeia)

				Ready-to-use solutions	
	Product	Concentration	Titrisol® Ampoules for 1 l solution Ord. No.	Glass bottles, PE bottles/canisters Ord. No.	Titripac <sup>®</sup> Ord. No.
Т	Trifluoromethanesulfonic acid in anhydrous acetic acid	0.1 mol/l	-	1 l: 1.08050.1000	-
	Titriplex <sup>®</sup> , IV solution (Na <sub>2</sub> -DCTA)	0.1 mol/l	-	1 l: 1.08447.1000	4 l: 1.08447.4000
W	Wijs solution (iodomonochloride in acetic acid)	0.1 mol/l	-	1 l: 1.09163.1000 2.5 l: 1.09163.2500	-
Z	Zinc sulfate solution	0.1 mol/l	1.09991.0001	2.5 l: 1.08879.1000 <sup>1</sup>	-

1: Solution in accordance with the reagents chapter of Pharm. Eur (European Pharmacopoeia)

### Titripur® - made from raw materials in accordance with the European Pharmacopoeia

For some pharmaceutical industry applications, it's important to work with solutions made from raw materials in accordance with the European Pharmacopoeia (Pharm. Eur). Even the water<sup>3</sup> used for this purpose has been tested in accordance with Pharm. Eur. The relevant information can be found in the certificate.

## Ordering information Titripur® prepared from raw materials acc. to Ph Eur

			Ready-to-use solutions	
	Product	Concentration	Glass bottles, PE bottles/canisters Ord. No.	Titripac® Ord. No.
A	Acetic acid	1 mol/l (1 N) prepared from raw materials acc. to Ph Eur	25 l: 1.99061.9025	-
Н	Hydrochloric acid	1 mol/l (1 N) prepared from raw materials acc. to Ph Eur	25 l: 1.99070.9025	-
S	Sodium hydroxide solution	1 mol/l (1 N) prepared from raw materials acc. to Ph Eur	25 l: 1.99060.9025	-
		6 mol/l (1 N) prepared from raw materials acc. to Ph Eur	25 l: 1.99062.9025	-

Other volumetric solutions made from raw materials in accordance with Pharm. Eur are available on request. | 3: Purified water

# Ordering information Titriplex<sup>®</sup> solid substances

	Product	Content	Packaging	Ord. No.
I	Titriplex® I for analysis (nitrilotriacetic acid)	250 g	Plastic bottle	1.08416.0250
II	riplex® II for analysis (ethylenedinitrilotetraacetic acid, EDTA) ACS, Reag. Ph Eur		Plastic bottle	1.08417.0100
		250 g	Plastic bottle	1.08417.0250
		1 kg	Plastic bottle	1.08417.1000
		5 kg	Plastic bottle	1.08417.5000
III	itriplex® Ⅲ for analysis (ethylenedinitrilotetraacetic acid disodium salt dihydrate, ƊTA disodiumsalt dihydrate) ACS, ISO, Reag. Ph Eur	100 g	Plastic bottle	1.08418.0100
		250 g	Plastic bottle	1.08418.0250
		1 kg	Plastic bottle	1.08418.1000
		5 kg	Plastic bottle	1.08418.5000
		10 kg	Fibre carton	1.08418.9010
		25 kg	Fibre carton	1.08418.9025
V	Titriplex® IV for analysis (1,2-cyclohexylenedinitrilotetraacetic acid monohydrate)	25 g	Plastic bottle	1.08424.0025
		100 g	Plastic bottle	1.08424.0100
/	Titriplex® V for analysis (dietheylenetriaminepentaacetic acid)	100 g	Plastic bottle	1.08426.0100
٧I	Titriplex <sup>®</sup> VI for analysis (3,6 –Dioxaoctamethylenedinitriloacetic acid)	25 g	Plastic bottle	1.08435.0025
		100 g	Plastic bottle	1.08435.0100

# Karl Fischer reagents Apura®

Using the Karl Fischer titration together with our Apura<sup>®</sup> reagents from Merck Millipore, the water content of gases, liquids and solids can be determined easily and with a high degree of accuracy. Not without reason is no other method as widespread as this one: Determination of water content according to Karl Fischer is rapid, accurate and reliable; it is thus the method of choice in quality and in-process control, production and research & development.

# Brilliant results with Apura® for highest standards.

# Karl Fischer reagents and standards

### Content

Page • Precise water determination every time 148 Quality standard 150 • The right method for any application 152 Volumetric method and reagents 154 • Coulometric method and reagents 164 Standards for volumetric and coulometric titration 168

## Advantages

- High accuracy and excellent precise
- Rapid and reproducible titration results
- Large water capacity
- No crystallization
- Innovative formulations for special applications
- Comprehensive product line
- Wide range of water standards



www.merckmillipore.com/apura

# Apura<sup>®</sup> – precise water determination every time ...

### The standard for reproducible results

Since we apply the highest standards to production processes and stringent testing requirements, Apura<sup>®</sup> reagents and standards are distinguished by excellent quality. The measuring results are therefore always reliable and accurate. From the selection of raw materials, packaging on up to quality control of the finished reagents we apply the strictest standards. This ensures a sophisticated consistency, quality and thus comparable and transparent results. The certificates of analysis for reagents and standards contain all the information important for quality management documentation.



# ... with 30 years of experience

# A proven method, modified according to the latest research

Brand quality	We use only high-quality and approved raw materials. This ensures batch-to-batch consistency and comparable analytical results.
Competence	You benefit from our high scientific standards and our long international experience.
Safety	Merck Millipore in particular places great importance on safety. For this reason, the Apura® Karl Fischer reagents contain no toxic components or, if this is not possible, they are limited to a minimum.
Service	Our international sales network and local offices ensure safe delivery and service – worldwide.
Support	Whenever you need help, we can supply it via our application and development laboratories. Here we can advise you on applications, offer technical support, and assist you in validation projects. Contact us at apura@merckgroup.com or visit our application finder: www.merckmillipore.com/applicationfinder-apura
Transparency	Our quality control is carried out according to the most stringent standards. Analyses are carried out using recognized international standards. This renders the results achieved completely transparent.

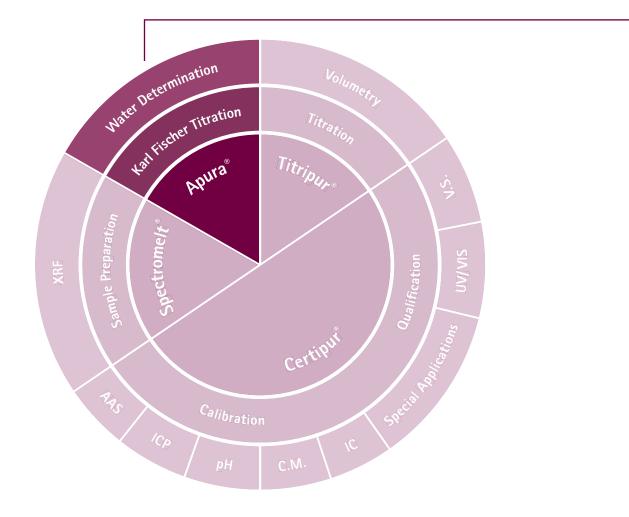


### More information

www.merckmillipore.com/applicationfinder-apura

# Apura<sup>®</sup> – our quality standard for water determination

The significance of Karl Fischer titration is emphasized by the fact that it has been included in the most important Pharmacopoeias, ASTM (American Standard Methods), German industrial norm DIN EN ISO 9001/2008 and other guidelines. Together with our Apura® standards the Karl Fischer product line from Merck Millipore is always reliable, rapid, accurate and globally available to facilitate international product transfer.



### Specification / Traceability

- ASTM American Standard Methods
  - **ISO** International Organization for Standardization
  - NIST National Institute of Standards and Technology, USA

### Apura® quality for your individual needs

#### Apura®

Karl Fischer reagents and standards for sophisticated, precise and consistent water determination. Excellent quality distinguishes our Apura® range.

### Karl Fischer titration

Since their development in the 1930s, Karl Fischer titrations have found use worldwide. With this method the water contents of a large variety of materials can be determined over a wide concentration range from 1 ppm to 100 %. In contrast to other methods, the Karl Fischer titration is based on a chemical reaction with water, so that water is exclusively determined. The Karl Fischer titration has a wide spectrum of applications and is used in a broad range of fields, for example water determination in foodstuffs, chemicals, pharmaceuticals, cosmetics and mineral oils.

#### Water determination | Volumetry | Coulometry | Standards

Two methods are available for the determination of water using the Karl Fischer method: a volumetric and a coulometric method. The method selected depends primarily on the amount of water to be determined.

**Volumetry** | In the case of higher levels of water (0.1 – 100 %), the volumetric method should be used. For volumetric titration, we supply both one component and two component reagents.

**Coulometry** | For very low levels of water (<1 %) or for water determination of very expensive substances with small sample amount, the coulometric method should be used. In this case,

a difference is made between cells with and without a diaphragm. For both cell types, the appropriate Apura® reagents are available.

**Standards** | The Apura<sup>®</sup> range is rounded off with water standards for monitoring and qualification of Karl Fischer instruments, for checking measuring results and titer determination.

# The right method for any application

## Benefit from our experience



Volumetry See page 154 0.1 – 100 % water level

<1 % water level

For further information and details please visit www.merckmillipore.com/applicationfinder-apura and have a look at our application finder.

For support please contact us: apura@merckgroup.com

# Coulometry

See page 164

Standards See page 168

## Titrants

CombiTitrant 5 [188005]
<b>Titrant 5</b> [188010]
CombiTitrant 5 Keto [188006]
CombiTitrant 2 [188002]
<b>Titrant 2</b> [188011]
CombiTitrant 1 [188001]
CombiCoulomat frit [109255]
CombiCoulomat fritless [109257]
Standards

To meet the requirements of modern laboratories Merck Millipore supplies a broad range of different Karl Fischer reagents in its Apura<sup>®</sup> line. It covers the whole range of volumetric and coulometric Karl Fischer reagents together with water standards.

Thus the user can choose the Karl Fischer reagents most suitable for his purposes, depending on his requirements and the available Karl Fischer instrument. The choice of the »right« reagent is a decisive factor in obtaining correct and reproducible results. It depends on the type of sample and the matrix in which the water is encapsulated. Important is that the sample is dissolved or dispersed completely in the Karl Fischer solvent or to make sure that all the water is extracted. The overview of the table offers advice regarding the correct choice and combination of reagents.

# Solvents

Samples dissolving in Ethanol	Samples dissolving in Methanol	Samples dissolving in Methanol	Samples containing aldehydes + ketones	Mineral oils	Oils + fats in food	Samples of long-chain hydrocarbons
CombiSolvent [188008]	CombiMethanol [188009]	<b>Solvent</b> [188015]	CombiSolvent Keto [188007]	CombiSolvent oils [188020]	CombiSolvent fats [188021]	Solvent oils & fats [188016]
•	•	•		•	•	•
		•				•
			•			
•	•	•		•	•	•
		•				•
•	•	•		•	•	•

Can be used for anode and cathode compartment in cells with diaphragm

Can be used for cells with and without diaphragm

Water standard: 0.01 % / 0.1 % / 1 % / oven 1 % / oil 15 - 30 ppm / 5 mg/ml | Lactose standard 5 % | Sodium tartrate dihydrate 15.66 %

# Apura<sup>®</sup> Water determination according to Karl Fischer Volumetric titration

# Apura<sup>®</sup> volumetric reagents

<ul> <li>One component reagents</li> </ul>	156
<ul> <li>Two component reagents</li> </ul>	158
<ul> <li>Solvents for oils and fats</li> </ul>	160
<ul> <li>Reagents for aldehydes and ketones</li> </ul>	162
Buffer solutions	163

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## Advantages

- High titration rate
- $\bullet$  For water content from 0.1 % 100 %
- Accurate and reliable results



# Volumetry

This method is preferably used for water contents in samples between 0.1 and 100 %. The unknown water content of the sample is determined by measuring the required volume of Karl Fischer Titrant to reach the titration endpoint. The sample is completely dissolved or dispersed in a suitable solvent with strict exclusion of atmospheric moisture and titrated with the Karl Fischer Titrant. The titration end point is indicated by an excess of iodine and is determined potentiometrically.

# One component reagents

# CombiTitrant | CombiSolvent | CombiMethanol

#### CombiTitrant

In one component volumetric Karl Fischer analysis, the titrant contains all ingredients required by the Karl Fischer reaction: iodine, base, sulfur dioxide, and an alcohol. Apura<sup>®</sup> CombiTitrants and CombiSolvent are free of toxic alcohols and available in packaging compatible with all major brands of titrators. The range includes titrants with nominal titer values of 1, 2, 5 mg H<sub>2</sub>O/ml to cover a wide range of practical applications.

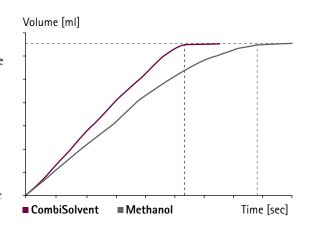
### CombiSolvent

The Apura® CombiSolvent is an innovative methanol-free solvent for one component volumetric titration. This enthanol-based solvent is also without toxic alcohols and used in conjunction with the Apura® CombiTitrants allows a non toxic titration. This means more safety for the laboratory staff as well as easy storage of the reagents.

**Titration curve:** The Apura® CombiSolvent is a methanol-free solvent for the one component system. This solvent brings about a higher titration rate and a more distinct endpoint.

#### CombiMethanol

Methanol – with a low water content – is the solvent typically used in one component titration for standard applications. The quality of methanol used for the Karl Fischer Titration is critical to minimize both pretitration, and interfering side reactions. Apura® CombiMethanol is optimized specifically for use in Karl Fischer titrations by strictly limited key impurities and water content.





## One component reagents advantages

- High titration rate for fast analyses
- Distinct endpoint to ensure confidence in results
- Ensured constant and consistent high quality for comparable results
- No toxic alcohols in CombiTitrant and CombiSolvent for more safety for lab assistants
- More flexibility the solvent can be suited to the sample matrix
- Unlimited water capacity more titrations are possible than with two component reagents

## Ordering information

Product	Content	Packaging	Ord. No.
Apura <sup>®</sup>   CombiTitrant 5   approx. 5 mg H <sub>2</sub> 0/ml	500 ml	Glass bottle	1.88005.0500
	1	Glass bottle	1.88005.1000
	2.5 l	Glass bottle	1.88005.2500
Apura®   CombiTitrant 2   approx. 2 mg H <sub>2</sub> 0/ml	1	Glass bottle	1.88002.1000
Apura <sup>®</sup>   CombiTitrant 1   approx. 1 mg H <sub>2</sub> 0/ml	1	Glass bottle	1.88001.1000
Apura®   CombiSolvent   methanol-free solvent	1	Glass bottle	1.88008.1000
	2.5 l	Glass bottle	1.88008.2500
Apura®   CombiMethanol   dried methanol for Karl Fischer titration max. 0.01 % water	11	Glass bottle	1.88009.1000
	2.5 l	Glass bottle	1.88009.2500

For water determinations in oils and fats we recommend CombiSolvent oils, (188020) and CombiSolvent fats (188021), for more information see page 161.

# Two component reagents

# Titrant | Solvent

Unlike one component Karl Fischer reagents, in two component reagent systems, the titrant contains only iodine and methanol, while the solvent contains the other Karl Fischer reaction components – sulfur dioxide and a suitable base dissolved in methanol. It is used as the working medium in the Karl Fischer cell. Apura<sup>®</sup> Titrants are available with concentrations of 2 and 5 mg  $H_2O/ml$  together with Apura<sup>®</sup> Solvent.

Two component reagents are distinguished by greater long term stability and a slightly faster titration time in comparison to one component reagents. This is due to the fact that the Karl Fischer reaction components are divided in between the two reagents making each of the reagent less susceptible to slow side reactions over time. Faster titration rates result from the initial availability of sulfur dioxide an base in the solvent to which the sample to be analyzed is added.



### Two component reagents advantages

- Slightly faster titration in comparison to one component titration
- High accuracy for reliable results
- Better buffer capacity
- Higher titer stability than one component reagents
- Ensured consistent high quality for comparable results

### Ordering information

Product	Content	Packaging	Ord. No.
Apura®   Titrant 5   approx. 5 mg H <sub>2</sub> 0/ml	500 ml	Glass bottle	1.88010.0500
	11	Glass bottle	1.88010.1000
	2.5 l	Glass bottle	1.88010.2500
Apura®   Titrant 2   approx. 2 mg H <sub>2</sub> 0/ml	11	Glass bottle	1.88011.1000
Apura®   Solvent   solvent for the two component titration	1	Glass bottle	1.88015.1000
	2.5 l	Glass bottle	1.88015.2500
Karl Fischer Test Kit for Karl Fischer Tiration without instruments			
Apura®   Test Kit for water determination according to Karl Fischer   Set	1 set	Fibre carton	1.88025.0001
1 set consisting of: 100 ml Titrant (Glass bottle), 500 ml Solvent (Glass bottle), 1 Syringe, 1 Measuring	flask		
Apura®   Test Kit Titrant for water determination acc. to Karl Fischer   Refill pack	100 ml	Glass bottle	1.88027.0001
Apura®   Test Kit Solvent for water determination acc. to Karl Fischer   Refill pack	500 ml	Glass bottle	1.88026.0001

For water determinations in oils and fats we recommend Solvent oils & fats (188016) which has been developed especially for this application. See page 161.

# Solvents for oils and fats

# CombiSolvent oils | CombiSolvent fats | Solvent oils & fats

When determining the exact water content in oils and fats it is important that the samples are completely dissolved or dispersed. If the sample is not dissolved or dispersed completely, the water is not all extracted during determination. Depending on the type of oil or fat to be determined, different solvents are suited for this purpose.



### CombiSolvent oils | CombiSolvent fats

Mineral oils are primarily a mixture of long-chain hydrocarbons with aromatic compounds. Fats consist primarily of glycerol esters of higher fatty acids. The two classes of substance have different dissolution properties requiring different dissolution aids. We have therefore developed appropriate solvents for one component titration in both applications: CombiSolvent oils for mineral oils and CombiSolvent fats for fats in foodstuffs. Both are used in combination with CombiTitrants.

### Solvent oils & fats

Solvent oils & fats is a universal solvent for long-chain, nonpolar substances as well as light fats and oils. It can be used with two component titrants, as well as with CombiTitrants.

### Solvents for oils and fats advantages

- · Good solubility of oils and fats in the corresponding solvent
- No addition of dissolution aids necessary immediately ready for use
- Constant and consistently high quality
- Without chloroform

### Ordering information

Product	Content	Packaging	Ord. No.
Apura®   Solvent oils & fats   for long-chain substances	1	Glass bottle	1.88016.1000
Apura®   CombiSolvent oils   for mineral oils	1	Glass bottle	1.88020.1000
Apura®   CombiSolvent fats   for fats in foodstuffs	1	Glass bottle	1.88021.1000

Solvent oils & fats can be used in combination with two component Titrant (188010), but also in combination with CombiTitrant (188005). For titration of oils and fats

with a very low water content we recommend our titrants with the factor 2 and 1, respectively: CombiTitrant 1 (188001), CombiTitrant 2 (188002) and Titrant 2 (188011).

# Reagents for aldehydes and ketones

## CombiTitrant 5 Keto | CombiSolvent Keto

CombiTitrant 5 Keto and CombiSolvent Keto are two products of the Apura<sup>®</sup> line. When performing water determinations with the Karl Fischer method in samples containing aldehydes and ketones it must be taken into account that these determinations are influenced by side reactions if methanol is used as a solvent.

### CombiTitrant 5 Keto

CombiTitrant 5 Keto is a one component titrant – all reactive components required for the Karl Fischer reaction:  $SO_2$ ,  $I_2$  and the base dissolved in a long-chain alcohol, are contained in a single reagent system.

### **CombiSolvent Keto**

CombiSolvent Keto is a solvent based upon a mixture of alcohol suppressing to a large extent the impact of side reactions with aldehydes and ketones. At the same time, its composition ensures optimum conditions for titration according to the Karl Fischer method with respect to stoichiometry, reaction rate and indication of the endpoint.



## Reagents for aldehydes and ketones advantages

- Fast titration rate
- No toxic alcohol contained
- Improved accuracy and reproducibility of the results
- Suppression of side reactions
- Consistently high quality

### Ordering information

Product	Content	Packaging	Ord. No.
Apura <sup>®</sup>   CombiTitrant 5 Keto   approx. 5 mg H <sub>2</sub> 0/ml for aldehydes and ketones	1	Glass bottle	1.88006.1000
Apura®   CombiSolvent Keto   methanol-free solvent for aldehydes and ketones	1	Glass bottle	1.88007.1000

CombiTitrant 5 Keto and CombiSolvent Keto must always be used in combination as both components are specifically complementary to one another.

# **Buffer solutions**

Of decisive importance for the water determination according to Karl Fischer is on the one hand the complete dissolution of the sample and on the other hand the pH optimum between pH 5–7. When strong acids or alkalis are used as samples, the pH value can be out of equilibrium. By the addition of special buffer solutions the Karl Fischer reaction can proceed in the correct pH range. The two ready-to-use buffer solutions are used in the Karl Fischer titration method, particularly in connection with samples that may disturb the pH optimum for the Karl Fischer reaction. Normally the buffering of the reaction medium by the Karl Fischer reagent is sufficient. With some samples, however, such as strong bases or strong acids, the pH is shifted into the acid or alkaline range. In such cases additional buffer solution must be added to maintain the pH in the optimum range.

# Ordering information

Product	Content	Packaging	Ord. No.
Apura <sup>®</sup>   Buffer solution for strong acids   additive to the Karl Fischer solvent for titration of strong acids	500 ml	Glass bottle	1.88035.0500
Apura®   Buffer solution for strong bases   additive to the Karl Fischer solvent for titration of strong bases	500 ml	Glass bottle	1.88036.0500

# Apura<sup>®</sup> Water determination according to Karl Fischer Coulometric titration

Apura <sup>®</sup> coulometric reagents	Page
CombiCoulomat frit	166
CombiCoulomat fritless	166

## Advantages

- $\bullet$  For low water contents below 1 %
- Highly accurate and reliable reproducible results
- Same reagents for anode and cathode cell usable



# Coulometry

The coulometric Karl Fischer titration is preferred for water contents below 1 %. In coulometric Karl Fischer titration the iodine required for reaction is produced in the titration vessel itself by anodic oxidation at the generator electrode. The water content is accurately determined by calculation from the amount of current used over a specific time period. The measuring cell for coulometric determination consists of anode and cathode compartments which can be separated by a diaphragm.

# Coulometric reagents

# CombiCoulomat frit | CombiCoulomat fritless

Our combined coulometric reagents have long since proved to be excellently suited for water determinations according to the coulometric Karl Fischer method. They have the advantage that even in diaphragm cells only one reagent is required; this considerably simplifies the application and excludes mistaking an anolyte for a catholyte.

#### CombiCoulomat frit | CombiCoulomat fritless

The CombiCoulomat fritless can be used universally for cells with or without diaphragm. Both reagents, CombiCoulomat frit and CombiCoulomat fritless, are, of course, free from chlorinated hydrocarbons. The coulometric reagents Apura® CombiCoulomat frit and CombiCoulomat fritless are comparable to the volumetric one component reagents. All reaction components are contained in one solution. This makes handling easier for the user. Compared with volumetric reagents, coulometric reagents do not contain iodine, but iodide. The iodine required for the Karl Fischer reaction to take place develops through electrochemical oxidation. The coulometric method is primarily used for the determination of low water contents (<1 %).



# Coulometric reagents advantages

- No confusion between anode and cathode cell solutions, since only one solution is required
- Very accurate and reproducible results
- Cell very rapidly ready for use after the first filling
- For samples with low water contents below 1 %

### Ordering information

Product	Content	Packaging	Ord. No.
Apura®   CombiCoulomat frit   for cells with diaphragm	500 ml	Glass bottle	1.09255.0500
Apura®   CombiCoulomat fritless   for cells with and without diaphragm	500 ml	Glass bottle	1.09257.0500
Apura®   CombiCoulomat fritless   for cells with and without diaphragm	2.5 l	Glass bottle	1.09257.2500

We recommend CombiCoulomat frit for coulometric devices with a diaphragm cell. For users working without a diaphragm or who frequently switch from cells with diaphragm

to cells without diaphragm we recommend CombiCoulomat fritless. It can be used for both cell types.

# $Apura^{\ensuremath{\mathbb{B}}}$ Water determination according to Karl Fischer Standards

# Apura<sup>®</sup> standards

<ul> <li>Water standards in ampoules</li> </ul>	170
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<ul> <li>Water standard oil 15–30 ppm</li> </ul>	170
Lactose standard 5 %	171
<ul> <li>Sodium tartrate dihydrate 15.66 %</li> </ul>	171
<ul> <li>Water standard 5 mg/ml</li> </ul>	171

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## Advantages

- Batch-specific Certificate of Analysis in the package for QM documentation
- Reliable and correct results

Within the European Union and even worldwide, increasing pressure demands that analytical results become more transparent and comparable. This is particularly true for water determinations according to Karl Fischer. In order to ensure this, reliable reference materials are necessary. The Apura® line offers a series of excellent standards. In addition to their use in monitoring Karl Fischer equipment and performing titer determination of volumetric Karl Fischer reagents, they can also be used for checking measuring results in order to evaluate their accuracy. To ensure the highest quality, our Apura® standards are manufactured under the strictest control and accurately measured using validated procedures.



# A series of excellent standards

#### Water standards in ampoules

The Apura® range offers water standards in ampoules with different water contents. The standards consist of solvent mixtures with a defined water content. They are tested against standard reference material from NIST – National Institute of Standards and Technology, Gaithersburg, USA. Each package contains a batch-specific certificate with the exact measured water content of the batch, uncertainty data, measuring method, the NIST batch, and minimum shelf-life.

Water standard 1 % is recommended for the use of titer determination with the volumetric method with one component and two component reagents. The two water standards 0.01 % and 0.1 % are used for the coulometric method.

#### Water standard oven 1 %

This standard is an entirely solid standard for the Karl Fischer oven method. In contrast to other solid standards like lactose, citrate or tartrate, this standard has a considerably lower water content of only 1 % and can be used over a wide temperature range from 140-400°C. The composition of the solid standard is based on an inorganic substance, which is stable even at high temperatures. In contrast, with organic substances decomposition reactions with formation of water may already set in at temperatures above 150°C, leading to erroneous results. The low water content of 1 % is particularly suited when using the Karl Fischer oven with a coulometer. Each package contains a batch-specific certificate with the exact measured water content of the batch, uncertainty data, measuring methods - Karl Fischer method and thermogravimetry - and minimum shelf-life.

#### Water standard oil 15 - 30 ppm

This standard is especially designed for users who need



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#### Lactose standard 5 %

The Apura® lactose standard is a further solid standard with a water content of about 5 %. The batch-specific Certificate of Analysis states the exact value measured in the respective batch. The lactose standard can be applied universally. Due to its ease of solubility in methanol and its water content, this standard is suitable as a solid standard for coulometry, and also for volumetric Karl Fischer titrations. It is especially suitable if the working medium used consists of solvent mixtures in which sodium tartrate dihydrate is only slightly soluble. But it can also be used as an oven standard in a temperature range preferably from 140–160°C.

### Sodium tartrate dihydrate 15.66 %

Sodium tartrate dihydrate is the volumetric standard for Karl Fischer titration. Under normal conditions it is stable and non-hygroscopic. Sodium tartrate dihydrate has a stoichiometric water content of 15.66 % and is primarily used for titer determination in volumetry.

### Water standard 5 mg/ml

This liquid water standard is based on a long-chain alcohol. Under normal conditions the adjusted water content is maintained after opening the bottle. This water standard 5 mg/ml can be used for daily titer control, but is not recommended for exact titer determinations.

### Ordering information

Product	Content	Packaging	Ord. No.
Apura <sup>®</sup>   Water standard 0.01 %   1 g contains 0.10 mg H <sub>2</sub> 0	10 x 8 ml	Glass ampoule	1.88050.0010
Apura <sup>®</sup>   Water standard 0.1 %   1 g contains 1.0 mg H <sub>2</sub> 0	10 x 8 ml	Glass ampoule	1.88051.0010
Apura <sup>®</sup>   Water standard 1 %   1 g contains 10 mg H <sub>2</sub> 0	10 x 8 ml	Glass ampoule	1.88052.0010
Apura®   Water standard oven 1 %   solid standard for KF oven method	5 g	Glass bottle	1.88054.0005
Apura®   Water standard oil   standard for oil samples for coulometric Karl Fischer titration (15–30 ppm)	10 x 8 ml	Glass ampoule	1.88055.0010
Apura®   Lactose standard 5 %   for volumetry and KF oven method	10 g	PE bottle	1.12939.0010
Apura®   Sodium tartrate dihydrate   volumetric standard for water determination acc. to Karl Fischer   contains 15.66 % H <sub>2</sub> O	100 g	PE bottle	1.06664.0100
Apura®   Water standard 5 mg/ml   1 ml contains 5 mg water	250 ml	Glass bottle	1.09259.0250

# Reference materials Certipur<sup>®</sup>

In order to obtain accurate analytical results, it is essential to calibrate all of the instruments used for analysis before commencing. However, such calibration work is only meaningful if reliable reference materials are employed. With Merck Millipore's Certipur® reference materials you can have always confidence in your analytical results. All Certipur® reference materials and standards come with a comprehensive Certificate of Analysis (CoA), where all batch-specific parameters and important inspection-relevant data are documented: used analytical method, uncertainty data, traceability including batch number, date of release, minimum shelf life and the responsible laboratory head.

# Not all reference materials are the same ...

# **Reference** materials

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### Advantages

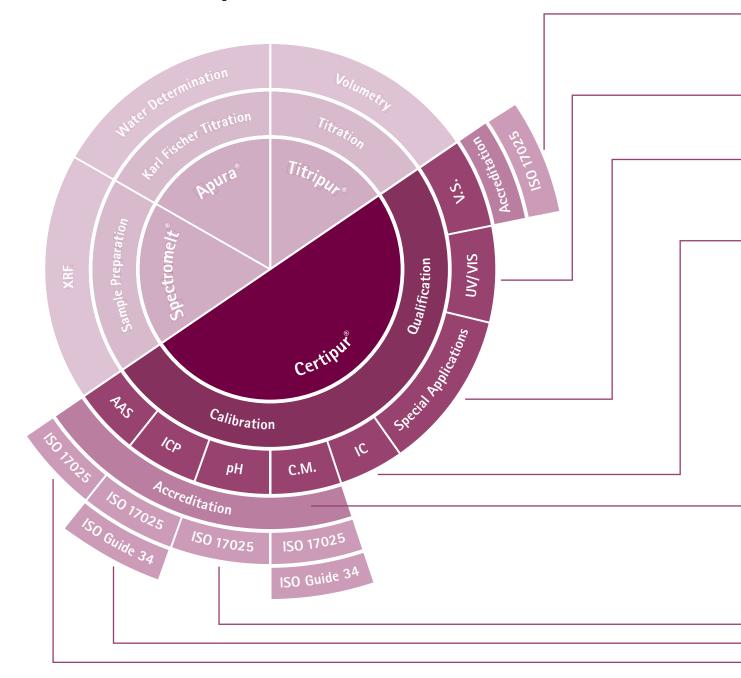
- Outstanding, application-oriented quality
- Stringent quality management system
- Always reliable and reproducible results
- Accreditation according to ISO 17025
- Accreditation according to ISO Guide 34
- Traceability to standard reference material from NIST and PTB
- Strict specifications



www.merckmillipore.com/certipur

# Certipur<sup>®</sup> – our quality standard in instrumental analysis

Merck Millipore supplies a range of first-class reference materials for a huge range of analytical methods under the trade name Certipur<sup>®</sup>, so as to fulfill your quality management requirements. For each standard provided, the optimal analytical method has been selected; not only that, but the methods involved are constantly being refined. Merck Millipore invests substantially in equipping its own laboratories and optimizes the quality and accuracy of its products with every new instrument generation.



### Specification / Traceability

ISO	International Organization for Standardization
NIST	National Institute of Standards and Technology, USA
PTB	Physical Technical Institute, Germany
Reag. Ph Eur	Reagents specified by the European Pharmacopoeia
USP	United States Pharmacopoeia requirements for reagents

### Volumetric standards for titration [V.S.]

Volumetric standards | Traceable to standard reference material from NIST and specified according to Reag. Ph Eur and USP. Measured in Merck KGaA's accredited laboratory for mass fraction according to DIN EN ISO / IEC 17025.

#### UV-VIS spectroscopy [UV-VIS]

UV-VIS standards | UV-VIS standards are specified according to Reag. Ph Eur

### Special applications

Refractive Index standards | Barium sulfate white standard | TOC standard | Hazen reference solution | Reference colour solutions according to Ph Eur | Refractive index Brix standards

#### Ion chromatography [IC]

Ion chromatography standards | Traceable to standard reference material from NIST

#### Conductivity measurement [C.M.]

Conductivity standards | Conductivity standards traceable to standard reference material from NIST and PTB | Measured in Merck Millipore's accredited laboratory for pH and conductivity measurement according to DIN EN ISO / IEC 17025 and produced according to ISO Guide 34

#### pH measurement [pH]

Buffer solutions and substances and buffer concentrates | Traceable to standard reference material from NIST and PTB | Measured in Merck KGaA's accredited laboratory for pH and conductivity measurement according to DIN EN ISO / IEC 17025

#### Inductively coupled plasma spectroscopy [ICP]

ICP single-element standards | ICP multi-element standards | Traceable to standard reference material from NIST and PTB | Measured in Merck KGaA's accredited laboratory for mass fraction according to DIN EN ISO / IEC 17025 and produced according to ISO Guide 34

#### Atomic absorption spectroscopy [AAS]

AAS standards: ready-to-use solutions and concentrates in ampoules | AAS standards dissolved in oil | Traceable to standard reference material from NIST | Measured in Merck KGaA's accredited laboratory for mass fraction according to DIN EN ISO / IEC 17025

# Certipur<sup>®</sup> – accreditation and ISO certification

With the worldwide globalization, analytical results have to become more comparable and transparent. Quality management especially for reference material is gaining in importance with respect to the accuracy and precision of analytical measurements. Using high sophisticated reference materials leads to avoid repeat analysis and safe qualitatively costs.

### Accreditation

Accreditation and certification is a process in which certification of competency, authority, or credibility is presented. The basis of our accreditation is our adherence to DIN EN ISO / IEC 17025. In contrast to certification according to DIN EN ISO 9001:2008 which is based on the presence of a company-wide quality management system, accreditation is based on the presence of a competent testing laboratory where not only qualified staff have to be employed, but where each and every analysis carried out has to correspond to specified criteria.

All instrumentation used is regularly checked by officially authorized calibration laboratories. In addition, all relevant parameters used in measurement, e.g. weight and temperature, have to correspond to national and international standards.

Within the framework of accreditation EMD Millipore is required to make a comprehensive statistical uncertainty budget according to GUM (Guide to the expression of Uncertainty in Measurement): all errors that can influence measurement are assessed and included in a so-called uncertainty budget. The accredited laboratory of Merck KGaA, Darmstadt, Germany, is also obliged to participate regularly in an external international laboratory testing program. Merck KGaA, Darmstadt, Germany is accredited by the German Accreditation Authority, DAkkS and registered as a calibration laboratory according to DIN EN ISO / IEC 17025.





#### ISO 9001

With our DIN EN ISO 9001:2008 certification we ensure customer satisfaction and improve our efficiency. It helps to sponsor our business and organization as best in class. Our commitment for continues improvements are exhibit and it provides confidence to the supply chain.

### ISO 17025

DIN EN ISO / IEC 17025:2005 is one of the most popular quality standards for all testing and calibration laboratories. It is for use by laboratories to design their Quality Management System, administrative and operative procedures and specifies the general requirements for the competence to carry out tests and / or calibration methods. All measurements and decisions should be based on accurate, repeatable, verifiable, reliable and correct measurements and procedures. Customers should ensure that all data and opinions upon which they rely could be traced back through an unbroken chain of data based on DIN EN ISO / IEC 17025.

### ISO Guide 34

Certified reference materials according to ISO Guide 34 are for analytical labs that work to regulatory requirements. ISO Guide 34 is an 'umbrella' accreditation that consists of general guidelines for a producer of reference materials. It describes a set of stringent requirements that ensures all aspects of the production of our reference materials are carried out with measurable and traceable quality and competence. The guide covers the chain from the selection of the raw materials, the production, the characterization - which is the measurement method by the lab - to the storage and transportation. Furthermore, Merck Millipore is responsible for the assessment of the homogeneity and stability data. ISO Guide 34 means quality from start to finish. You can calibrate and monitor your equipment with complete confidence.

#### **Reference material**

Material, sufficiently homogeneous and stable with respect to one or more specified properties, which has been established to be fit for its intended use in a measurement process [ISO Guide 30].

### Certified reference material

Reference material characterized by a metrologically valid procedure for one or more specified properties, accompanied by a certificate that provides the values of the specified property, its associated uncertainty, and a statement of metrological traceability [ISO Guide 30].

- Primary reference material It is directly measured and certified by a certifying body.
- Secondary reference material It is directly measured against primary reference material.

#### Traceability

Traceability means to verify a reference material to a stated international reference material of an official authority body like NIST and PTB through an unbroken chain of comparisons.

# Certipur<sup>®</sup> reference materials for reliable calibration in atomic absorption spectroscopy

# Certipur<sup>®</sup> products

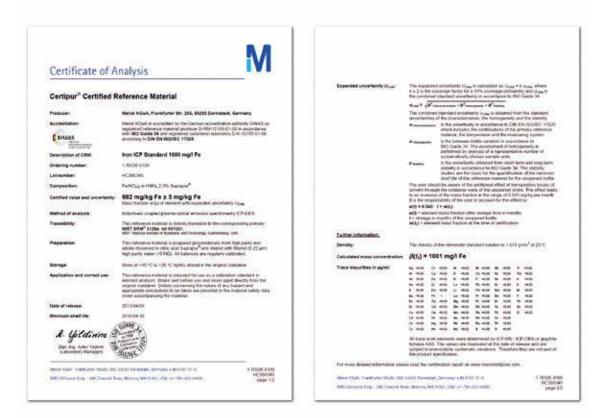
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### Advantages

- ICP standards according to ISO 17025 and ISO Guide 34
- Standards are traceable to primary standards
- ICP standards with an extensive Certificate of Analysis



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### Quality management

ICP and AAS standards are analyzed using ICP / OES and ICP / MS methods. The true value is certified by an accredited calibration laboratory according to ISO / IEC 17025.

For this analysis, a method has been developed for all 68 element standards. The advantage is obvious: a high degree of precision; this is reflected in the uncertainty data, which can vary, depending on the chemical nature of the element concerned.

### Traceability

ICP and AAS standards are directly traceable to an international standard. Direct traceability means that the measuring instrument used in quality control is calibrated prior to each measurement using a primary standard obtained from an international institution such as NIST. Traceability to the primary standard is documented in the Certificate of Analysis with indication of the batch used.

# ICP single-element standards

The quality control of our ICP single-element standards is carried out by the accredited calibration laboratory according to DIN EN ISO / IEC 17025. They are directly traceable to primary reference material from NIST. A Certificate of Analysis is enclosed in each package. It includes exact data on content including uncertainty budget, trace element impurities, composition, traceability, date of release and minimum shelf life. **Our ICP single-element standards are produced according to ISO Guide 34.** 



( DALAS



### **Certipur® ICP standards A-M**

Designation	Element	Composition	ICP 1000 mg/l	ICP 10000 mg/l
			Ord. No. [100 ml]	Ord. No. [100 ml]
Aluminium	AI	$AI(NO_3)_3$ in $HNO_3 2 - 3 \%$	1.70301.0100	1.70371.0100
Antimony	Sb	$Sb_2O_3$ in HCl 7 %	1.70302.0100	-
Arsenic	As	$H_3AsO_4$ in $HNO_3$ 2 – 3 %	1.70303.0100	-
Barium	Ba	$Ba(NO_3)_2$ in $HNO_3 2 - 3 \%$	1.70304.0100	-
Beryllium	Ве	$Be_4O(C_2H_3O_2)_6$ in $HNO_3 2 - 3 \%$	1.70305.0100	-
Bismuth	Bi	Bi(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 %	1.70306.0100	-
Boron	В	$H_3BO_3$ in water	1.70307.0100	-
Cadmium	Cd	Cd(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 %	1.70309.0100	-
Calcium	Ca	$Ca(NO_3)_2$ in $HNO_3 2 - 3 \%$	1.70308.0100	1.70373.0100
Cerium	Ce	Ce(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 %	1.70311.0100	-
Cesium	Cs	$CsNO_3$ in $HNO_3$ 2 – 3 %	1.70310.0100	-
Chromium	Cr	$Cr(NO_3)_3$ in $HNO_3$ 2 – 3 %	1.70312.0100	1.70374.0100
Cobalt	Со	$Co(NO_3)_2$ in $HNO_3 2 - 3 \%$	1.70313.0100	1.70375.0100
Copper	Cu	Cu(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 %	1.70314.0100	1.70378.0100
Dysprosium	Dy	Dy <sub>2</sub> O <sub>3</sub> in HNO <sub>3</sub> 2 – 3 %	1.70315.0100	-
Erbium	Er	$Er_2O_3$ in HNO <sub>3</sub> 2 – 3 %	1.70316.0100	-
Europium	Eu	$Eu_2O_3$ in $HNO_3 2 - 3 \%$	1.70317.0100	-
Gadolinium	Gd	$Gd_2O_3$ in HNO <sub>3</sub> 2 – 3 %	1.70318.0100	-
Gallium	Ga	$Ga(NO_3)_3$ in $HNO_3 2 - 3 \%$	1.70319.0100	-
Germanium	Ge	$(NH_4)_2GeF_6$ in water	1.70320.0100	-
Gold	Au	H(AuCl₄) in HCl 7 %	1.70321.0100	-
Hafnium	Hf	HfOCl₂ in HCI 7 %	1.70322.0100	-
Holmium	Но	$Ho_2O_3$ in $HNO_3 2 - 3 \%$	1.70323.0100	-
Indium	In	In(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 %	1.70324.0100	-
Iridium	lr	IrCl <sub>3</sub> in HCl 7 %	1.70325.0100	-
Iron	Fe	$Fe(NO_3)_3$ in $HNO_3$	1.70326.0100	1.70376.0100
Lanthanum	La	$La(NO_3)_3$ in $HNO_3 2 - 3 \%$	1.70327.0100	-
Lead	Pb	Pb(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 %	1.70328.0100	1.70372.0100
Lithium	Li	$LiNO_3$ in $HNO_3$ 2 – 3 %	1.70329.0100	-
Luthetium	Lu	$Lu_2O_3$ in HNO <sub>3</sub> 2 – 3 %	1.70330.0100	-
Magnesium	Mg	$Mg(NO_3)_2$ in $HNO_3 2 - 3 \%$	1.70331.0100	1.70379.0100
Manganese	Mn	$Mn(NO_3)_2$ in $HNO_3 2 - 3 \%$	1.70332.0100	1.70380.0100
Mercury	Hg	$Hg(NO_3)_2$ in $HNO_3$ 10 %	1.70333.0100	1.70384.0100
Molybdenum	Мо	$(NH_4)_6 Mo_7 O_{24}$ in water	1.70334.0100	-

#### Certipur<sup>®</sup> ICP standards N-Z

Designation	Element	Composition	ICP 1000 mg/l	ICP 10000 mg/l
			Ord. No. [100 ml]	Ord. No. [100 ml]
Neodymium	Nd	$Nd_{2}O_{3}$ in $HNO_{3} 2 - 3 \%$	1.70335.0100	-
Nickel	Ni	Ni(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 %	1.70336.0100	1.70382.0100
Niobium	Nb	NH₄NbF₀ in water	1.70337.0100	-
Osmium	Os	(NH <sub>4</sub> ) <sub>2</sub> OsCl <sub>6</sub> in HCl 7 %	1.70338.0100	-
Palladium	Pd	Pd(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 %	1.70339.0100	-
Phosporous	Р	$H_3PO_4$ in water	1.70340.0100	1.70383.0100
Platinum	Pt	$H_2$ PtCl <sub>6</sub> in HCl 7 %	1.70341.0100	-
Potassium	К	$KNO_3$ in $HNO_3$ 2 – 3 %	1.70342.0100	1.70377.0100
Praseodymium	Pr	$Pr(NO_3)_3$ in $HNO_3 2 - 3 \%$	1.70343.0100	-
Rhenium	Re	NH₄ReO₄ in water	1.70344.0100	-
Rhodium	Rh	Rh(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 %	1.70345.0100	-
Rubidium	Rb	RbNO <sub>3</sub> in HNO <sub>3</sub> 2 – 3 %	1.70346.0100	-
Ruthenium	Ru	RuCl₃ in HCl 7 %	1.70347.0100	-
Samarium	Sm	$Sm_2O_3$ in $HNO_3 2 - 3 \%$	1.70348.0100	-
Scandium	Sc	$Sc_2O_3$ in $HNO_3$ 7 %	1.70349.0100	-
Selenium	Se	$SeO_2$ in $HNO_3 2 - 3 \%$	1.70350.0100	-
Silicon	Si	$SiO_2$ in NaOH 2 %	1.70365.0100	1.70386.0100
Silver	Ag	AgNO <sub>3</sub> in HNO <sub>3</sub> 2 – 3 %	1.70352.0100	-
Sodium	Na	$NaNO_3$ in $HNO_3$ 2 – 3 %	1.70353.0100	1.70381.0100
Sulfur	S	$H_2SO_4$ in water	1.70355.0100	1.70385.0100
Strontium	Sr	Sr(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 %	1.70354.0100	-
Tantalum	Та	$(NH_4)_2$ TaF <sub>7</sub> in water	1.70356.0100	-
Tellurium	Те	$H_6 TeO_6$ in $HNO_3 2 - 3 \%$	1.70357.0100	-
Terbium	Тb	Tb(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 %	1.70358.0100	-
Thallium	TI	$TINO_3$ in $HNO_3$ 2 – 3 %	1.70359.0100	-
Thulium	Tm	Tm(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 %	1.70361.0100	-
Tin	Sn	SnCl₄ in HCl 7 %	1.70362.0100	-
Titanium	Ti	$(NH_4)_2 TiF_6$ in water (trace HF)	1.70363.0100	-
Tungsten	W	$(NH_4)_2WO_4$ in water	1.70364.0100	-
Vanadium	V	$NH_4VO_3$ in $HNO_3$	1.70366.0100	1.70388.0100
Ytterbium	Yb	$Yb_2O_3$ in $HNO_3 2 - 3 \%$	1.70367.0100	-
Yttrium	Y	Y(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 2 – 3 %	1.70368.0100	-
Zinc	Zn	Zn(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 %	1.70369.0100	1.70389.0100
Zirconium	Zr	ZrCl₄ in HCl 7 %	1.70370.0100	1.70390.0100

#### Certipur<sup>®</sup> standards 10 mg/l

	Designation	Element	Composition	ICP 10 mg/l
				Ord. No. [100 ml]
М	Mercury*	Hg	Hg(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 %	1.08623.0100
R	Rhodium, internal standard for ICP	Rh	$Rh(NO_{3})_{3}$ in $HNO_{3} 2 - 3 \%$	1.08525.0100
Т	Thorium	Th	$Th(NO_3)_4$ in $HNO_3 2 - 3 \%$	1.70391.0100
U	Uranium	U	UO <sub>2</sub> (NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 – 3 %	1.70360.0100
*	and the model of an entropy of and VVI (1.00.400)			

\*suppl. to multi-element standard XXI (1.09498)

# ICP multi-element standards

Multi-element calibration standards are traceable to standard reference material from NIST. A Certificate of Analysis is enclosed in each package. It includes data on content, composition, traceability, date of release and minimum shelf life.

	Certipur® Multi-elemer Designation	Element	ICP calibration standards Multi-element standard I 19 elements, different concentrations Ord. No. 1.15474.0100	Multi-element standard IV 23 elements, 1000 mg/l Ord. No. 1.11355.0100	Multi-element standard VIII 24 elements, 100 mg/l Ord. No. 1.09492.0100	Multi-element standard IX toxic elements, Chromium VI Ord. No. 1.09494.0100	Multi-element standard X for surface water, equivalent to NIST SRM 1643d   Ord. No. 1.09493.0100	Multi-element standard XI for sewage sludge Ord. No. 1.09491.0100
Α	Aluminium	Al	100 mg/l	1000 mg/l	100 mg/l	-	-	-
	Arsenic	As	-	-	-	100 mg/l	50 μg/l	-
В	Barium	Ba	5 mg/l	1000 mg/l	100 mg/l	-	50 μg/l	-
	Beryllium	Be	1 mg/l	-	100 mg/l	100 mg/l	20 μg/l	-
	Bismuth	Bi	200 mg/l	1000 mg/l	100 mg/l	-	10 μg/l	-
с	Boron Cadmium	B Cd	15 mg/l	1000 mg/l	100 mg/l	-	100 μg/l	-
C			20 mg/l	1000 mg/l	100 mg/l	100 mg/l	20 μg/l	10 mg/l
	Calcium Chromium	Ca	-	1000 mg/l	100 mg/l	-	35000 μg/l	-
	Cobalt	Cr Co	25 mg/l	1000 mg/l	100 mg/l	100 mg/l	20 μg/l	900 mg/l
	Copper	Co	20 mg/l 20 mg/l	1000 mg/l 1000 mg/l	100 mg/l	-	25 μg/l	
G	Gallium	Ga	150 mg/l	1000 mg/l	100 mg/l 100 mg/l	-	20 μg/l	800 mg/l
I	Indium	In	200 mg/l	1000 mg/l	100 mg/i		_	
	Iron	Fe	15 mg/l	1000 mg/l	- 100 mg/l	_	– 100 μg/l	
L	Lead	Рь	200 mg/l	1000 mg/l	100 mg/l	100 mg/l	25 μg/l	900 mg/l
-	Lithium	Li	-	1000 mg/l	100 mg/l		20 µg/i	
М	Magnesium	Mg	_	1000 mg/l	100 mg/l	_	15000 μg/l	_
	Manganese	Mn	5 mg/l	1000 mg/l	100 mg/l	_	30 μg/l	_
	Mercury	Hg	-	_		100 mg/l	-	8 mg/l
	Molybdenum	Mo	-	-	-	_	100 μg/l	_
Ν	Nickel	Ni	50 mg/l	1000 mg/l	100 mg/l	100 mg/l	50 μg/l	200 mg/l
Р	Potassium	К	-	1000 mg/l	100 mg/l	-	3000 μg/l	-
S	Selenium	Se	-	-	100 mg/l	100 mg/l	10 μg/l	-
	Silver	Ag	50 mg/l	1000 mg/l	-	-	-	-
	Sodium	Na	-	1000 mg/l	100 mg/l	-	8000 μg/l	-
	Strontium	Sr	1 mg/l	1000 mg/l	100 mg/l	-	100 μg/l	-
Т	Tellurium	Te	_	-	100 mg/l	-	-	-
	Thallium	TI	400 mg/l	1000 mg/l	100 mg/l	100 mg/l	10 μg/l	-
V	Vanadium	V	-	-	-	-	50 μg/l	-
Ζ	Zinc	Zn	20 mg/l	1000 mg/l	100 mg/l	-	50 μg/l	2500 mg/l
	Matrix		1 mol/l HNO <sub>3</sub>	1 mol/l HNO <sub>3</sub>	1 mol/l HNO <sub>3</sub>	$1 \text{ mol/I HNO}_3$	$1 \text{ mol/l HNO}_3$	1 mol/l HNO <sub>3</sub>

	Certipur® Multi-elemer	nt standards Element	ICP calibration standards Multi-element standard XIII 15 elements Ord. No. 1.09480.0100	Multi-element standard XVI 21 elements Ord. No. 1.09487.0100	Multi-element standard XVII HCl soluble elements Ord. No. 1.09495.0100	ICP/MS calibration standards	Multi-element standard VI calibration in ICP MS, 30 elements Ord. No. 1.10580.0100	Multi-element standard XXI set, contains 274473 and 108623 Ord. No. 1.09498.0001	Tuning solution	Multi-element standard XXIV tuning solution 700 ES Ord. No. 1.09411.0500
А	Aluminium	Al	500 mg/l	2 7 0	2 ± 0			ے م 10 mg/l		
^	Antimony	Sb	500 mg/i	- 100 mg/l	- 100 mg/l		10 mg/l	TO mg/i		50 mg/l
	Artimony	As					100 mg/l	- 10 mg/l		- -
В	Barium	Ba	100 mg/l	100 mg/l	-		100 mg/l	10 mg/l		50 mg/l
Б		Ве	-	-	-		10 mg/l	10 mg/l		50 mg/l
	Beryllium Bismuth	Bi	100 mg/l	100 mg/l	-		100 mg/l	10 mg/l		-
			-	-	-		10 mg/l	10 mg/l		-
0	Boron Cadmium	B	-	-	-		100 mg/l	-		-
С		Cd	25 mg/l	100 mg/l	-		10 mg/l	10 mg/l		50 mg/l
	Calcium	Ca	-	100 mg/l	-		1000 mg/l	10 mg/l		-
	Cesium	Cs	-	-	-		-	10 mg/l		-
	Chromium	Cr	100 mg/l	100 mg/l	-		10 mg/l	10 mg/l		50 mg/l
	Cobalt	Co	100 mg/l	100 mg/l	-		10 mg/l	10 mg/l		50 mg/l
	Copper	Cu	100 mg/l	100 mg/l	-		10 mg/l	10 mg/l		50 mg/l
G	Gallium	Ga	-	-	-		10 mg/l	10 mg/l		-
H	Hafnium	Hf	-	-	100 mg/l		-	-		-
I	Indium	In	-	-	-		-	10 mg/l		-
	Iridium	lr	-	-	100 mg/l		-	-		-
	Iron	Fe	100 mg/l	100 mg/l	-		100 mg/l	10 mg/l		-
L	Lead	Pb	100 mg/l	100 mg/l	-		10 mg/l	10 mg/l		50 mg/l
	Lithium	Li	-	100 mg/l	-		10 mg/l	10 mg/l		-
М	Magnesium	Mg	-	100 mg/l	-		10 mg/l	10 mg/l		-
	Manganese	Mn	100 mg/l	100 mg/l	-		10 mg/l	10 mg/l		50 mg/l
	Mercury	Hg	5 mg/l	-	-		-	* 10 mg/l		-
	Molybdenum	Mo	-	100 mg/l	-		10 mg/l	-		50 mg/l
Ν	Nickel	Ni	100 mg/l	100 mg/l	-		10 mg/l	10 mg/l		50 mg/l
Р	Potassium	К	-	-	-		10 mg/l	10 mg/l		500 mg/l
R	Rubidium	Rb	-	-	-		10 mg/l	10 mg/l		-
S	Selenium	Se	25 mg/l	100 mg/l	-		100 mg/l	10 mg/l		50 mg/l
	Silver	Ag	-		-		10 mg/l	10 mg/l		-
	Sodium	Na	-	-	-		10 mg/l	10 mg/l		-
	Strontium	Sr	-	100 mg/l	-		10 mg/l	10 mg/l		50 mg/l
Т	Tantalum	Та	-	-	100 mg/l		-	-		-
	Tellurium	Te	-	-	-		10 mg/l	-		-
	Thallium	TI	-	100 mg/l	-		10 mg/l	10 mg/l		-
	Tin	Sn	-	-	100 mg/l		-	-		-
	Titanium	Ti	-	100 mg/l	100 mg/l		-	-		-
U	Uranium	U	-	-	-		10 mg/l	10 mg/l		-
V	Vanadium	V	250 mg/l	100 mg/l	-		10 mg/l	10 mg/l		-
Z	Zinc	Zn	100 mg/l	100 mg/l	-		100 mg/l	10 mg/l		50 mg/l
	Zirconium	Zr	-	-	100 mg/l			_		-
	Matrix * = Hg in separate bottle		5 % HNO <sub>3</sub>	$5 \% HNO_{3}$	15 % HCI		1 mol/l HNO <sub>3</sub>	5 % HNO <sub>3</sub>		1 % HNO <sub>3</sub>

\* = Hg in separate bottle (1.08623.0100)

	Certipur® Multi-elemen		Multi-element standard XXIII for mass calibration Ord. No. 1.09410.0500	Graphit furnace AAS stand.	Multi-element standard XVIII for calibration Ord. No. 1.09500.0100	Wavelength calibration stand.	Multi-element standard V HCl soluble elements Ord. No. 1.10714.0500	Multi-element standard XIV HCI soluble elements Ord. No. 1.09481.0500
	Designation	Element	or Mr	G	ML for <b>Or</b>	3	Mr HC <b>O</b> re	Dre Mr
Α	Aluminium	Al	-		100 mg/l		20 mg/l	-
	Antimony	Sb	-		100 mg/l		-	-
	Arsenic	As	-		100 mg/l		20 mg/l	20 mg/l
В	Barium	Ва	1 μg/l		50 mg/l		2 mg/l	-
	Beryllium	Ве	-		5 mg/l		1 mg/l	-
	Boron	В	1 μg/l		-		2 mg/l	-
С	Cadmium	Cd	-		5 mg/l		2 mg/l	-
	Calcium	Са	-		-		10 mg/l	-
	Cerium	Ce	-		-		-	-
	Chromium	Cr	-		20 mg/l		2 mg/l	-
	Cobalt	Co	1 μg/l		50 mg/l		-	-
	Copper	Cu	-		50 mg/l		2 mg/l	-
G	Gallium	Ga	1 μg/l		-		-	-
	Germanium	Ge	-		-		-	-
T	Indium	In	1 μg/l		-		-	-
	Iron	Fe	1 μg/l		20 mg/l		2 mg/l	-
L	Lanthanum	La	-		-		-	20 mg/l
	Lead	Рb	-		100 mg/l		20 mg/l	-
	Lithium	Li	1 μg/l		-		2 mg/l	20 mg/l
	Luthetium	Lu	1 μg/l		-		-	-
М	Magnesium	Mg	-		-		1 mg/l	-
	Manganese	Mn	-		20 mg/l		1 mg/l	20 mg/l
	Mercury	Hg	-		-		5 mg/l	-
	Molybdenum	Mo	-		-		-	20 mg/l
N	Nickel	Ni	-		50 mg/l		5 mg/l	20 mg/l
Р	Phosporous	P	-		-		10 mg/l	100 mg/l
P	Potassium	K	1 μg/l		-		100 mg/l	100 mg/l
R	Rhodium	Rh	1 μg/l		-		- 1	-
S	Scandium	Sc	1 μg/l				1 mg/l	20 mg/l
	Selenium	Se	-		100 mg/l		20 mg/l	-
	Silver Sodium	Ag Na	-		10 mg/l		-	-
			1 µg/l		-		20 mg/l	20 mg/l
	Strontium Sulfur	Sr S	-		-		1 mg/l _	– 100 mg/l
т	Tellurium	Te	-		-			100 mg/i
	Terbium	Tb	-		-		20 mg/l	-
	Thallium	TI	-		- 100 mg/l		-	-
	Titanium	Ti	1 μg/l		100 mg/l			
		U	-		-		2 mg/l	-
U	Uranium Xttrium	Y	1 μg/l		-		- 1	
Y 7	Yttrium Zinc		1 μg/l		-		1 mg/l	-
Z		Zn	- F 06 HNO		- F 06 HNO		2 mg/l	-
	Matrix		5 % HNO <sub>3</sub>		5 % HNO <sub>3</sub>		5 % HCl	2 % HCl

# Titrisol<sup>®</sup> standards

Titrisol<sup>®</sup> standards are traceable to standard reference material from NIST.

A batch specific Certificate of Analysis is available via internet (www.merckmillipore.com).

#### Dilution of Titrisol<sup>®</sup> standards

In preparing a dilution series, the ICP- and AAS standards should be diluted with the same concentration of acid as is in the standard itself. Tritisol® standards should be diluted to 1 I with 0.1 mol/I hydrochloric acid in the first step. Alternatively, 30 ml of 30 % hydrochloric acid can be placed in a volumetric flask and made up to 1 I with distilled water. Further dilutions should then be made with distilled water.

Minimum shelf life of standard solutions				
Unopened Titrisol® ampoules	5 years			
Solution of 1000 mg/l	Recommended max. 12 months			
Solution of 10–100 mg/l	Recommended 1 month			
Solution of 1–10 mg/l	Recommended max. 1 week			
Solution of <1 mg/l	Recommended 1 day			

As exceptions, the following Titrisol <sup>®</sup> standards should be diluted as indicated:				
Gold	Place 500 ml 30 % hydrochloric acid in a volumetric flask and			
	make up to 1 I with distilled water.			
Lead,   Selenium   Silver	Dilute to 1 l with 0.1 mol/l or place 5 ml 65 % nitric acid in a			
	volumetric flask and make up to the mark.			
Molybdenum	Place 10 ml 25 % ammonia solution in a volumetric flask and			
	make up to 1 l with distilled water.			
Silicon	Make up to the mark with distilled water.			
Vanadium	Place 5 ml 96 % sulfuric acid in a volumetric flask and make up			
	to 1 l with distilled water.			

#### **Titrisol® standards**

	Designation	Element	Composition	Ord. No.
			[1000 mg of element per ampoule]	
4	Aluminium	AI	$AICI_3$ in water	1.09967.0001
	Arsenic	As	$As_2O_5$ in water	1.09939.0001
	Barium	Ва	BaCl <sub>2</sub> in 7 % HCl	1.09968.0001
	Cadmium	Cd	CdCl <sub>2</sub> in water	1.09960.0001
	Calcium	Ca	CaCl <sub>2</sub> in 6.5 % HCl	1.09943.0001
	Chloride	CI	HCl in water	1.09871.0001
	Chromium	Cr	CrCl <sub>3</sub> in 4.2 % HCl	1.09948.0001
	Cobalt	Co	CoCl <sub>2</sub> in water	1.09986.0001
	Copper	Cu	CuCl <sub>2</sub> in water	1.09987.0001
	Fluoride	F	KF in water	1.09869.0001
i	Gold	Au	H(AuCl₄) in 12.72 % HCl	1.09868.0001
	Iron	Fe	FeCl <sub>3</sub> in 15 % HCl	1.09972.0001
	Lead	Рb	$Pb(NO_3)_2$ in water	1.09969.0001
	Lithium	Li	LiCl in water	1.09934.0001
١	Magnesium	Mg	$MgCl_2$ in 6 % HCl	1.09949.0001
	Manganese	Mn	MnCl₂ in water	1.09988.0001
	Molybdenum	Мо	(NH <sub>4</sub> ) <sub>6</sub> Mo <sub>7</sub> O <sub>24</sub> in 0.7 % NH <sub>4</sub> OH	1.09926.0001
	Nickel	Ni	NiCl <sub>2</sub> in water	1.09989.0001
	Nitrite	NO <sub>2</sub>	NaNO <sub>2</sub> in water	1.09866.0001
	Phosphate	PO₄	$H_3PO_4$ in water	1.09870.0001
	Potassium	К	KCl in water	1.09924.0001
	Selenium	Se	$SeO_2$ in 6.3 % $HNO_3$	1.09915.0001
	Silicon	Si	SiCl₄ in 14 % NaOH	1.09947.0001
	Silver	Ag	$AgNO_3$ in 5 % $HNO_3$	1.09906.0001
	Sodium	Na	NaCl in water	1.09927.0001
	Strontium	Sr	SrCl <sub>2</sub> in 7 % HCl	1.09993.0001
	Sulfate	SO₄	$H_2SO_4$ in water	1.09872.0001
	Titanium	Ti	$(NH_4)_2 TiF_6$ in water	1.09829.0001
	Vanadium	V	$VOSO_4$ in 8.6 % $H_2SO_4$	1.09994.0001
	Zinc	Zn	ZnCl <sub>2</sub> in 0.06 % HCl	1.09953.0001

# AAS standards

Atomic absorption spectroscopy standards are traceable to standard reference material from NIST. Analysis of content is carried out by ICP-OES in Merck KGaA, Darmstadt, Germany, accredited laboratory for mass fraction according to DIN EN ISO / IEC 17025. A batch specific Certificate of Analysis is available via internet (www.merckmillipore.com).

#### Certipur® AAS standards

	Designation	Element	Composition	Concentration 1000 mg/l	Concentration 1000 mg/l
	2 co.g. a cion			Ord. No. [100 ml]	Ord. No. [500 ml]
A	Aluminium	AI	$AI(NO_3)_3$ in $HNO_3 0.5$ mol/l	1.19770.0100	1.19770.0500
	Antimony	Sb	$Sb_2O_3$ in HCl 2 mol/l	1.70204.0100	1.70204.0500
	Arsenic	As	H₃AsO₄ in HNO₃ 0.5 mol/l	1.19773.0100	1.19773.0500
3	Barium	Ва	Ba(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 0.5 mol/l	1.19774.0100	1.19774.0500
	Beryllium	Ве	$Be_4O(C_2H_3O_2)_6$ in $HNO_3 0.5$ mol/l	1.70207.0100	1.70207.0500
	Bismuth	Ві	$Bi(NO_3)_3$ in $HNO_3$ 0.5 mol/l	1.19804.0100	1.19804.0500
	Boron	В	$H_{3}BO_{3}$ in water	1.19500.0100	1.19500.0500
2	Cadmium	Cd	Cd(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 0.5 mol/l	1.19777.0100	1.19777.0500
	Calcium	Ca	$Ca(NO_3)_2$ in $HNO_3$ 0.5 mol/l	1.19778.0100	1.19778.0500
	Cesium	Cs	$CsNO_3$ in $HNO_3$ 0.5 mol/l	1.70212.0100	-
	Chromium	Cr	$Cr(NO_3)_3$ in HNO <sub>3</sub> 0.5 mol/l	1.19779.0100	1.19779.0500
	Cobalt	Со	Co(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 0.5 mol/l	1.19785.0100	1.19785.0500
	Copper	Cu	$Cu(NO_3)_2$ in $HNO_3$ 0.5 mol/l	1.19786.0100	1.19786.0500
ì	Gold	Au	H(AuCl₄) in HCl 2 mol/l	1.70216.0100	1.70216.0500
	Indium	In	$In(NO_3)_3$ in $HNO_3$ 0.5 mol/l	1.19504.0100	-
	Iron	Fe	Fe(NO <sub>3</sub> ) <sub>3</sub> in HNO <sub>3</sub> 0.5 mol/l	1.19781.0100	1.19781.0500
	Lead	Pb	$Pb(NO_3)_2$ in $HNO_3$ 0.5 mol/l	1.19776.0100	1.19776.0500
	Lithium	Li	LiNO <sub>3</sub> in HNO <sub>3</sub> 0.5 mol/l	1.70223.0100	1.70223.0500
Λ	Magnesium	Mg	$Mg(NO_3)_2$ in $HNO_3$ 0.5 mol/l	1.19788.0100	1.19788.0500
	Manganese	Mn	Mn(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 0.5 mol/l	1.19789.0100	1.19789.0500
	Mercury	Hg	Hg(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 2 mol/l	1.70226.0100	1.70226.0500
	Molybdenum	Мо	$(NH_4)_6 Mo_7 O_{24}$ in water	1.70227.0100	1.70227.0500
	Nickel	Ni	Ni(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 0.5 mol/l	1.19792.0100	1.19792.0500
	Palladium	Pd	$Pd(NO_3)_2$ in $HNO_3$ 0.5 mol/l	1.14282.0100	1.14282.0500
	Platinum	Pt	H₂PtCl₀ in HCl 2 mol/l	1.70219.0100	1.70219.0500
	Potassium	К	$KNO_3$ in $HNO_3$ 0.5 mol/l	1.70230.0100	1.70230.0500
	Scandium	Sc	$Sc_2O_3$ in HNO <sub>3</sub> 1 mol/l	1.19513.0100	1.19513.0500
	Selenium	Se	$SeO_2$ in $HNO_3$ 0.5 mol/l	1.19796.0100	1.19796.0500
	Silicon	Si	$(NH_4)_2SiF_6$ in water	1.12310.0100	1.12310.0500
	Silicon	Si	SiO <sub>2</sub> in NaOH 0.5 mol/l	1.70236.0100	1.70236.0500
	Silver	Ag	AgNO <sub>3</sub> in HNO <sub>3</sub> 0.5 mol/l	1.19797.0100	1.19797.0500
	Sodium	Na	NaNO <sub>3</sub> in HNO <sub>3</sub> 0.5 mol/l	1.70238.0100	1.70238.0500
	Strontium	Sr	Sr(NO <sub>3</sub> ) <sub>2</sub> in HNO <sub>3</sub> 0.5 mol/l	1.19799.0100	1.19799.0500
	Tellurium	Te	H <sub>6</sub> TeO <sub>6</sub> in HNO <sub>3</sub> 0.5 mol/l	1.19514.0100	-
	Thallium	TI	$TINO_3$ in $HNO_3$ 0.5 mol/l	1.19801.0100	1.19801.0500
	Tin	Sn	SnCl₄ in HCl 2 mol/l	1.70242.0100	1.70242.0500
	Titanium	Ti	$(NH_4)_2 TiF_6$ in water	1.70243.0100	1.70243.0500
	Tungsten	W	$(NH_4)_2WO_4$ in water	1.70244.0100	1.70244.0500
/	Vanadium	V	$NH_4VO_3$ in $HNO_3$ 0.5 mol/l	1.70245.0100	1.70245.0500
,	Yttrium	Y	$Y(NO_3)_3$ in $HNO_3$ 0.5 mol/l	1.19809.0100	1.19809.0500
	Zinc	Zn	$Zn(NO_3)_2$ in HNO <sub>3</sub> 0.5 mol/l	1.19806.0100	1.19806.0500

# AAS standards dissolved in oil

Standards dissolved in oil are ready-to-use calibration standards for the analysis of non-aqueous matrices. For AAS, they are diluted to the required concentration with Merck Millipore standard oil or with non-polar solvents.

#### Certipur® standards, dissolved in oil

Designation	Raw material	Concentration	Ord. No. [100 ml]
Calcium	2-Ethylhexanoic acid-Ca salt	1 g/kg	1.15053.0100
Cobalt	Cyclohexanebutyric acid-Co(II) salt	1 g/kg	1.15061.0100
Phosphorus	Triphenyl phosphate	1 g/kg	1.15072.0100
Potassium	Cyclohexanebutyric acid-K salt	1 g/kg	1.15054.0100
Sodium	Cyclohexanebutyric acid-Na salt	1 g/kg	1.15058.0100

#### Certipur® multi-element standards, Oil dissolved calibration standards

Designation	Composition	Ord. No. [100 ml]
Multi-element standard I	4 elements, different concentrations	1.15075.0100
	Ba (8 g/kg), Ca (4 g/kg), Mg (1 g/kg), Zn (1.6 g/kg)	
Multi-element standard II	21 elements, 100 ppm	1.09469.0100
	Ag, Al, B, Ba, Ca, Cd, Cr, Cu, Fe, Mg, Mn, Mo, Na, Ni, P, Pb, Si, Sn, Ti, V, Zn	
Multi-element standard III	21 elements, 900 ppm	1.09479.0100
	Ag, Al, B, Ba, Ca, Cd, Cr, Cu, Fe, Mg, Mn, Mo, Na, Ni, P, Pb, Si, Sn, Ti, V, Zn	

#### **Auxiliaries**

Designation	Ord. No. [2.5 l]
Standard oil for AAS (30–40 mPas)	1.13898.2500



# Certipur<sup>®</sup> reference materials for reliable calibration in pH measurement

# Certipur<sup>®</sup> products

<ul> <li>Certified secondary standard reference materials</li> </ul>	192
<ul> <li>Certified buffer solutions</li> </ul>	193
<ul> <li>Titrisol<sup>®</sup> buffer concentrates</li> </ul>	195
<ul> <li>Certified buffer solutions in sachets</li> </ul>	196

## Advantages

- Measured and certified in Merck Millipore's ISO 17025 accredited lab
- Conducts always reliable and accurate results
- Traceable to NIST and PTB
- Available in convenient and safe packaging



Page

The measurement of pH value is the most commonly used analytical method. This is because the determination itself and the subsequent maintenance of pH value is of critical importance in numerous chemical, biochemical and biological processes, especially those in quality assurance and in-process-control.



#### Certipur® reference materials for pH measurement

Merck Millipore offers a distinguished range of buffer solutions and substances for the calibration, monitoring and qualifying of pH instruments and buffer solutions. All buffer solutions and substances are measured and certified in our DIN EN ISO / IEC 17025 accredited laboratory for pH measurement. Buffer reference materials have a direct influence on the accuracy of measurement results and therefore they are important for modern quality management systems. Our laboratory for pH measurement operates a comprehensive quality management system and is accredited by the German accreditation body DAkkS (Deutsche Akkreditierungsstelle) to the current DIN EN ISO / IEC 17025 standard. For our customer, accreditation ensures state recognition of the reliability of our testing and calibration results, worldwide comparability of the results; independence and impartiality, international approval of our competence; high-level quality and transparency of our services.

# Quality control

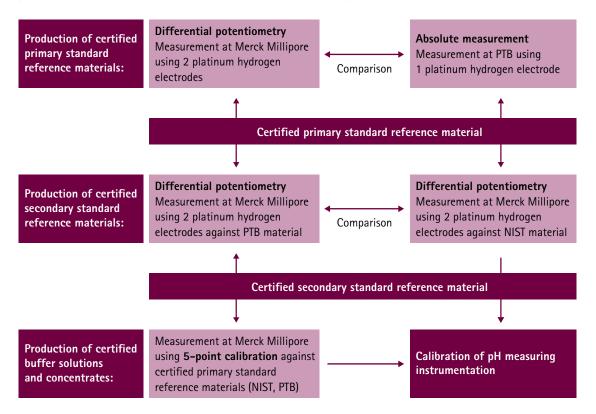
# Merck Millipore's laboratory for pH measurement is accredited according to DIN EN ISO / IEC 17025

For certification of our secondary reference materials Merck Millipore produces its own primary reference materials, which are certified and characterized by the German Metrological Institute PTB (Physical Technical Institute, Braunschweig, Germany). Additionally our laboratory for pH measurement is not only an accredited accreditation laboratory but also an accredited testing laboratory with the permission to act as an independent authority to test samples and materials.



#### Quality control of certified secondary reference buffers

Merck Millipore offers certified secondary standard reference buffer solution and substances which are at most used for extremely precise calibration, qualifying and monitoring of pH instruments and testing other buffer solutions. These certified secondary reference materials are tested with two platinum hydrogen electrodes using differential potentiometry. The uncertainty of this measuring method is +/- 0.003 pH units.

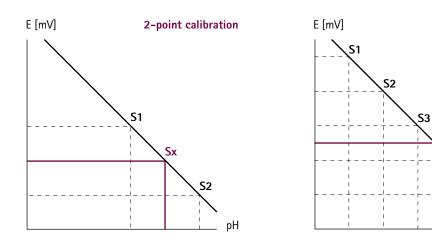


#### Quality control of ready-to-use buffer solutions for the daily calibration

For the precise measurement of our ready-to-use reference buffer solutions for the regularly calibration of pH instruments, Merck Millipore uses a 5-point calibration. The 5-point calibration is more accurate than a 2- or 3-point calibration.

5-point calibration

рΗ



# Certified secondary standard reference buffer solutions / substances

Certified secondary Reference buffers are employed for precisely calibrating and monitoring of pH measurement instruments and buffer solutions. The uncertainty of those buffer solutions and substances is +/- 0.003 pH units. Each buffer is accompanied by a Certificate of Analysis showing the relevant data for every quality management system and requirements. The quality control is carried out in the Merck Millipores's accredited calibration laboratory suing differential potentiometry with two specially developed platinium hydrogen electrodes. This method allows the comparison of the certified secondary reference buffers to the corresponding primary reference materials by PTB Physical Technical Institute, Germany) and NIST (National Institute of Standards and Technology, USA).

#### Certipur® certified secondary standard reference buffer solutions [25°C]

Designation	pH value [25°C]	Package size	Ord. No.
Certipur® certified secondary standard reference buffer solution pH 1.68,	1.68 <sub>1</sub>	5 x 100 ml	1.07204.0105
Potassium tetroxalat dihydrate			
Certipur® certified secondary standard reference buffer solution pH 4.00 <sub>5</sub>	4.005	5 x 100 ml	1.07200.0105
Potassium hydrogen phtalate			
Certipur <sup>®</sup> certified secondary standard reference buffer solution pH 6.86 <sub>3</sub>	6.86 <sub>3</sub>	5 x 100 ml	1.07202.0105
Potassium dihydrogen phosphate / di-Sodium hydrogen phosphate			
Certipur <sup>®</sup> certified secondary standard reference buffer solution pH 7.41 <sub>6</sub>	7.41 <sub>6</sub>	5 x 100 ml	1.07205.0105
Potassium dihydrogen phosphate / di-Sodium hydrogen phosphate			
Certipur <sup>®</sup> pH certified secondary standard reference buffer solution 9.18 <sub>4</sub>	9.18 <sub>4</sub>	5 x 100 ml	1.07203.0105
di-Sodium tetraborate decahydrate			

#### Certipur® certified secondary reference buffer substances [25°C]

Designation	pH value [25°C]	Package size	Ord. No.
Certipur® certified secondary reference buffer substance	1.68 <sub>1</sub>	25 g	1.01961.0025
Potassium tetroxalate dihydrate			
Certipur® certified secondary reference buffer substance	3.63 <sub>9</sub>	25 g	1.01963.0025
Potassium hydrogen tartate			
Certipur® certified secondary reference buffer substance	4.005	25 g	1.01965.0025
Potassium hydrogene phthalate			
Certipur® certified secondary reference buffer substance	6.86 <sub>3</sub> / 7.41 <sub>6</sub>	2 x 25 g	1.01960.0001
Potassium hydrogene phosphate / di-Sodium hydrogene phosphate			
Certipur® certified secondary reference buffer substance	9.18 <sub>4</sub>	25 g	1.01964.0025
di-Sodium tetraborate decahydrate			
Certipur® certified secondary reference buffer substance	10.014	2 x 25 g	1.01962.0001
Sodium hydrogene carbonate / Sodium carbonate			



# Ready-to-use buffer solutions

All Certipur<sup>®</sup> buffer solutions are traceable to standard reference materials from NIST and PTB. They are analyzed in our own independent accredited calibration laboratory according to DIN EN ISO / IEC 17025 with a combined glass electrode and 5-point calibration according to DIN 19268 using reference material for the preparation of standard buffer solutions according to DIN 19266.

#### Certipur<sup>®</sup> buffer solutions [20°C]

Designation	pH value [20°C]	Composition	Content	Packaging	Ord. No.
Certipur <sup>®</sup> buffer solution	1.00	Glycine, Sodium chloride, Hydrogen chloride	1000 ml	PE bottle	1.09432.1000
Certipur <sup>®</sup> buffer solution	2.00	Citric acid, Sodium hydroxide, Hydrogen chloride	1000 ml	PE bottle	1.09433.1000
			4	Titripac®	1.09433.4000
			10 I	Titripac®	1.09433.9010
Certipur <sup>®</sup> buffer solution	3.00	Citric acid, Sodium hydroxide, Hydrogen chloride	1000 ml	PE bottle	1.09434.1000
Certipur <sup>®</sup> buffer solution	4.00	Citric acid, Sodium hydroxide, Hydrogen chloride	1000 ml	PE bottle	1.09435.1000
			4	Titripac®	1.09435.4000
			10 I	Titripac®	1.09435.9010
Certipur <sup>®</sup> buffer solution	4.66	Acetic acid, Sodium acetate	1000 ml	PE bottle	1.07827.1000
Certipur <sup>®</sup> buffer solution	5.00	Citric acid, Sodium hydroxide	1000 ml	PE bottle	1.09436.1000
Certipur <sup>®</sup> buffer solution	6.00	Citric acid, Sodium hydroxide	1000 ml	PE bottle	1.09437.1000
			4	Titripac®	1.09437.4000
Certipur <sup>®</sup> buffer solution	6.88	di-Sodium hydrogen phosphate,	1000 ml	PE bottle	1.07294.1000
		Potassium dihydrogen phosphate			
Certipur <sup>®</sup> buffer solution	7.00	di-Sodium hydrogen phosphate,	1000 ml	PE bottle	1.09439.1000
		Potassium dihydrogen phosphate	4	Titripac®	1.09439.4000
			10 I	Titripac®	1.09439.9010
Certipur <sup>®</sup> buffer solution	8.00	Boric acid, Sodium hydroxide, Hydrogen chloride	1000 ml	PE bottle	1.09460.1000
			4	Titripac®	1.09460.4000
Certipur <sup>®</sup> buffer solution	9.00	Boric acid, Sodium hydroxide, Potassium chloride	1000 ml	PE bottle	1.09461.1000
			4	Titripac®	1.09461.4000
			10 I	Titripac®	1.09461.9010
Certipur <sup>®</sup> buffer solution	9.22	di-Sodium tetraborate	1000 ml	PE bottle	1.01645.1000
Certipur <sup>®</sup> buffer solution	10.00	Boric acid, Sodium hydroxide, Potassium chloride	1000 ml	PE bottle	1.09438.1000
			4	Titripac®	1.09438.4000
			10 I	Titripac®	1.09438.9010
Certipur <sup>®</sup> buffer solution	11.00	Boric acid, Sodium hydroxide, Potassium chloride	1000 ml	PE bottle	1.09462.1000

#### Certipur<sup>®</sup> buffer solutions [25°C]

Designation	pH value [25°C]	Composition	Content	Packaging	Ord. No.
Certipur <sup>®</sup> buffer solution	1.00	Glycine, Sodium chloride, Hydrogen chloride	500 ml	PE bottle	1.09441.0500
			4	Titripac®	1.09441.4000
Certipur <sup>®</sup> buffer solution	2.00	Citric acid, Sodium hydroxide, Hydrogen chloride	500 ml	PE bottle	1.09442.0500
			4	Titripac <sup>®</sup>	1.09442.4000
Certipur <sup>®</sup> buffer solution	3.00	Citric acid, Sodium hydroxide, Hydrogen chloride	500 ml	PE bottle	1.09444.0500
			4	Titripac®	1.09444.4000
Certipur <sup>®</sup> buffer solution	4.00	Citric acid, Sodium hydroxide, Hydrogen chloride	500 ml	PE bottle	1.09445.0500
			4	Titripac®	1.09445.4000
Certipur <sup>®</sup> buffer solution	4.01	Potassium hydrogen phtalate	500 ml	PE bottle	1.09406.0500
			1	PE bottle	1.09406.1000
			4	Titripac®	1.09406.4000
Certipur <sup>®</sup> buffer solution	5.00	Citric acid, Sodium hydroxide	500 ml	PE bottle	1.09446.0500
			4	Titripac®	1.09446.4000
Certipur <sup>®</sup> buffer solution	6.00	Citric acid, Sodium hydroxide	500 ml	PE bottle	1.99036.4000
			4	Titripac®	1.99036.0500
Certipur <sup>®</sup> buffer solution	7.00	di-Sodium hydrogen phosphate,	500 ml	PE bottle	1.09407.0500
		Potassium dihydrogen phosphate	1	PE bottle	1.09407.1000
			4	Titripac <sup>®</sup>	1.09407.4000
Certipur <sup>®</sup> buffer solution	8.00	Boric acid, Sodium hydrogen, Hydrogen chloride	500 ml	PE bottle	1.99038.0500
			4	Titripac <sup>®</sup>	1.99038.4000
Certipur <sup>®</sup> buffer solution	9.00	Boric acid, Sodium hydrogen, Potassium chloride	500 ml	PE bottle	1.09408.0500
			1	PE bottle	1.09408.1000
			4	Titripac®	1.09408.4000
Certipur <sup>®</sup> buffer solution	10.00	Boric acid, Sodium hydrogen, Potassium chloride	500 ml	PE bottle	1.09409.0500
			1	PE bottle	1.09409.1000
			4	Titripac®	1.09409.4000
Certipur <sup>®</sup> buffer solution	11.00	Boric acid, Sodium hydrogen, Potassium chloride	500 ml	PE bottle	1.99041.0500
			4	Titripac®	1.99041.4000
Certipur <sup>®</sup> buffer solution	12.00	di-Sodium hydrogen phosphate, Sodium hydroxide	4	Titripac®	1.99022.4000

#### Certipur<sup>®</sup> buffer solutions [20°C] – color coded

Designation	pH value [20°C]	Composition	Content	Packaging	Ord. No.
Certipur <sup>®</sup> buffer solution	4.00	Citric acid, Sodium hydroxide, Hydrogen chloride	500 ml	PE bottle	1.09475.0500
	[red]		4	Titripac®	1.09475.4000
			10 I	Titripac®	1.09475.9010
Certipur <sup>®</sup> buffer solution	7.00	di-Sodium hydrogen phosphate, Potassium dihydrogen	500 ml	PE bottle	1.09477.0500
	[green]	phosphate	4	Titripac®	1.09477.4000
			10 I	Titripac®	1.09477.9010
Certipur <sup>®</sup> buffer solution	9.00	Boric acid, Potassium chloride, Sodium hydroxide	500 ml	PE bottle	1.09476.0500
	[blue]	]	4	Titripac®	1.09476.4000
			10 I	Titripac®	1.09476.9010
Certipur <sup>®</sup> buffer solution	10.00	Boric acid, Potassium chloride, Sodium hydroxide	500 ml	PE bottle	1.09400.0500
	[yellow]		4	Titripac®	1.09400.4000
			10 I	Titripac®	1.09400.9010

Designation	pH value [25°C]	Composition	Content	Packaging	Ord. No.
Certipur <sup>®</sup> buffer solution	4.00	Citric acid, Sodium hydroxide, Hydrogen chloride	500 ml	PE bottle	1.99054.0500
	[red]		4	Titripac®	1.99054.4000
Certipur <sup>®</sup> buffer solution	7.00		500 ml	PE bottle	1.99057.0500
	[yellow]		4	Titripac®	1.99057.4000
Certipur <sup>®</sup> buffer solution	10.00	Boric acid, Potassium chloride, Sodium hydrogen	500 ml	PE bottle	1.99050.0500
	[blue]		4	Titripac <sup>®</sup>	1.99050.4000

#### Certipur<sup>®</sup> buffer solutions [25°C] - color coded





## Titrisol<sup>®</sup> buffer concentrates

All Titrisol® buffer concentrates are traceable to standard reference materials from NIST and PTB. They are analyzed in our own independent accredited calibration laboratory with a combined glass electrode and 5-point calibration according to DIN 19268 using reference material for the preparation of standard buffer solutions according to DIN 19266.

#### Titrisol® buffer concentrate for 500 ml buffer solution [20°C]

Designation	pH value [20°C]	Composition	Package size	Ord. No.
Buffer Titrisol®	1.00	Glycine, Sodium chloride, Hydrogen chloride	1 ampoule	1.09881.0001
Buffer Titrisol®	2.00	Citric acid, Sodium hydroxide, Hydrogen chloride	1 ampoule	1.09882.0001
Buffer Titrisol®	3.00	Citric acid, Sodium hydroxide, Hydrogen chloride	1 ampoule	1.09883.0001
Buffer Titrisol®	4.00	Citric acid, Sodium hydroxide, Hydrogen chloride	1 ampoule	1.09884.0001
Buffer Titrisol®	5.00	Citric acid, Sodium hydroxide	1 ampoule	1.09885.0001
Buffer Titrisol®	6.00	Citric acid, Sodium hydroxide	1 ampoule	1.09886.0001
Buffer Titrisol®	7.00	di-Sodium hydrogen phosphate, Potassium dihydrogen phosphate	1 ampoule	1.09887.0001
Buffer Titrisol® acc. to Weise	7.20	di-Sodium hydrogen phosphate, Potassium dihydrogen phosphate	1 ampoule	1.09879.0001
Buffer Titrisol®	8.00	Boric acid, Sodium hydroxide, Hydrogen chloride	1 ampoule	1.09888.0001
Buffer Titrisol®	9.00	Boric acid, Sodium hydroxide, Hydrogen chloride	1 ampoule	1.09889.0001
Buffer Titrisol®	10.00	Boric acid, Sodium hydroxide, Hydrogen chloride	1 ampoule	1.09890.0001
Buffer Titrisol®	11.00	Boric acid, Sodium hydroxide, Potassium chloride	1 ampoule	1.09880.0001
Buffer Titrisol®	12.00	di-Sodium hydrogen phosphate, Sodium hydroxide	1 ampoule	1.09892.0001
Buffer Titrisol®	13.00	Potassium chloride, Sodium hydroxide	1 ampoule	1.09893.0001

# Certipur<sup>®</sup> buffer sachets

Ready-to-use buffer solutions, packed in practical small quantities in sealed sachets. All Certipur® buffer sachets are traceable to standard reference materials from NIST and PTB. They are analyzed in our own independent accredited calibration laboratory with a combined glass electrode and 5-point calibration according to DIN 19268 using reference material for the preparation of standard buffer solutions according to DIN 19266.

#### Advantages of Certipur® buffer sachets

- Always fresh and ready-to-use
- Always available in suitable quantities
- Precise pH value
- No risk of contamination
- No residues
- Easy to use
- Includes batch-specific Certificate of Analysis
- Traceable to NIST and PTB



#### Certipur<sup>®</sup> buffer solutions in sachets [25°C]

Designation	pH value [25°C]	Composition	Package size	Ord. No.
Certipur <sup>®</sup> buffer solution	2.00	Citric acid, Sodium hydroxide, Hydrogen chloride	30 sachets x 30 ml	1.99012.0001
Certipur <sup>®</sup> buffer solution	4.00	Citric acid, Sodium hydroxide, Hydrogen chloride	30 sachets x 30 ml	1.99064.0001
Certipur <sup>®</sup> buffer solution	4.01	Potassium hydrogen phthalate	30 sachets x 30 ml	1.99001.0001
Certipur <sup>®</sup> buffer solution	6.00	Citric acid, Sodium hydroxide	30 sachets x 30 ml	1.99016.0001
Certipur <sup>®</sup> buffer solution	7.00	Potassium dihydrogen phosphate, di-Sodium hydrogen phosphate	30 sachets x 30 ml	1.99002.0001
Certipur <sup>®</sup> buffer solution	9.00	Boric acid, Sodium hydroxide, Potassium chloride	30 sachets x 30 ml	1.99003.0001
Certipur <sup>®</sup> buffer solution	9.18	di-Sodium tetraborate	30 sachets x 30 ml	1.99019.0001
Certipur <sup>®</sup> buffer solution	10.00	Boric acid, Sodium hydroxide, Potassium chloride	30 sachets x 30 ml	1.99004.0001
Certipur <sup>®</sup> buffer solution	11.00	Boric acid, Sodium hydroxide, Potassium chloride	30 sachets x 30 ml	1.99021.0001
Certipur <sup>®</sup> buffer solution	12.00	di-Sodium hydrogen phosphate, Sodium hydroxide	30 sachets x 30 ml	1.99022.0001
Certipur <sup>®</sup> buffer solution	4.01	-	3 x 10 sachets x 30 ml	1.99005.0001
Set I	7.00			
	9.00			
Certipur <sup>®</sup> buffer solution	4.01	-	3 x 10 sachets x 30 ml	1.99006.0001
Set II	7.00			
	10.00			



# Certipur<sup>®</sup> reference materials for reliable calibration in conductivity measurement

# Certipur<sup>®</sup> products

• Certified reference material for conductivity measurement	200
<ul> <li>Certipur<sup>®</sup> conductivity sachets</li> </ul>	201

## Advantages

#### Certipur<sup>®</sup> conductivity sachets

- Solutions are always fresh and ready-to-use
- Available in suitable quantities
- Precise conductivity values
- No risk of contamination
- No residues
- Easy to use
- Includes batch-specific Certificate of Analysis
- Traceable to PTB and NIST



Page

For precise conductivity measurement, Merck Millipore offers a distinguished range of conductivity standards. In conductivity measurement the reliable calibration of your instruments is the first indispensable step. Our laboratory for conductivity measurement operates a comprehensive quality management system and is accredited by the German accreditation body to the current DIN EN ISO / IEC 17025 standard. Additionally we now also have the ISO Guide 34 accreditations as reference material producer for conductivity standards.

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# Certified Secondary Reference Materials for conductivity measurement

# Merck Millipore's calibration standards for conductivity measurement have a double accreditation according to DIN EN ISO / IEC 17025 and according to ISO Guide 34

For certification of our secondary reference materials for conductivity measurement Merck Millipore produces its own primary reference materials which are certified an characterized by the German Metrological Institute PTB (Physical Technical Institute, Braunschweig, Germany). Additionally the secondary conductivity standards are measured against the certified primary reference materials from PTB Physical Technical Institute, Germany) and NIST (National Institute of Standards and Technology, USA).



#### Certipur® reference material for conductivity measurement

Designation	Nominal conductivity [mS/cm] at 25°C	Package size	Ord. No. [500 ml]
Conductivity water	0	5 PE bottles x 100 ml	1.01810.0105
Potassium chloride solution (0.0001 mol/l)	0.015	5 PE bottles x 100 ml	1.01811.0105
Potassium chloride solution (0.001 mol/l)	0.147	1 PE bottle x 500 ml	1.01557.0500
Potassium chloride solution (0.01 mol/l)	1.41	1 PE bottle x 500 ml	1.01203.0500
Potassium chloride solution (0.1 mol/l)	12.8	1 PE bottle x 500 ml	1.01254.0500
Potassium chloride solution (1 mol/l)	111	1 PE bottle x 500 ml	1.01255.0500





## Certipur<sup>®</sup> conductivity sachets

Ready-to-use conductivity sachets can be used in laboratories or outside in the field. The package is convenient and easy to use and ensures accurate and reliable results. They are measured as well in our own accredited laboratory. In environmental analysis it is often necessary to perform the measurement directly at the river or at similar places. Our sachets were designed especially for this purpose. They are reliable, easy-to-use and always fresh. Safe and neat packages with small portions suitable for laboratory and external use prevent the solutions from contamination through micro-organisms,  $CO_2$  or other foreign substances. The conductivity sachets are ready-to-use and therefore ideal for mobile analysis. In principle, you do not even need a beaker. To measure the electrode can simply be inserted into the sachet.

#### Certipur<sup>®</sup> conductivity solutions in sachets

Designation	Nominal conductivity [mS/cm] at 25°C	Package size	Ord. No. [500 ml]
Potassium chloride solution (0.001 mol/l)	0.147	30 sachets à 30 ml	1.01586.0001
Potassium chloride solution (0.01 mol/l)	1.41	30 sachets à 30 ml	1.01553.0001
Potassium chloride solution (0.1 mol/l)	12.8	30 sachets à 30 ml	1.01554.0001

# Certipur<sup>®</sup> reference materials for reliable calibration in ion chromatography

# Certipur<sup>®</sup> products

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<ul> <li>Ion chromatography multi standards, cationic</li> </ul>	203

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## Advantages

- Single-element standards as ready-to-use standards or as concentrates
- Multi-element standards available as anionic and cationic solution
- Traceable to NIST





# lon chromatography

lon chromatography requires the constant use of calibration solutions. Merck Millipore has a wide range of single- and multi-element solutions for this purpose. Recent additions are the multi-element solutions. All ion chromatography standards are traceable to NIST standard reference materials.

#### Certipur<sup>®</sup> ion chromatography standards

Designation	ltem	Composition	Concentration	Package size	Ord. No.
Ammonium	Ready-to-use solution	NH₄CI in water	1000 mg/l	500 ml	1.19812.0500
Bromide	Ready-to-use solution	NaBr in water	1000 mg/l	500 ml	1.19896.0500
Chloride	Ready-to-use solution	NaCl in water	1000 mg/l	500 ml	1.19897.0500
	Titrisol®	HCl in water	1000 mg	1 ampoule	1.09871.0001
Chromate	Ready-to-use solution	$K_2CrO_4$ in water	1000 mg/l	500 ml	1.19780.0500
Cyanide	Ready-to-use solution	$K_2[Zn(CN)_4]$ in water	1000 mg/l	500 ml	1.19533.0500
Fluoride	Ready-to-use solution	NaF in water	1000 mg/l	500 ml	1.19814.0500
	Titrisol®	KF in water	1000 mg	1 ampoule	1.09869.0001
Nitrate	Ready-to-use solution	$NaNO_3$ in water	1000 mg/l	500 ml	1.19811.0500
Nitrite	Ready-to-use solution	NaNO <sub>2</sub> in water	1000 mg/l	500 ml	1.19899.0500
	Titrisol®	$NaNO_2$ in water	1000 mg	1 ampoule	1.09866.0001
Phosphate	Ready-to-use solution	$KH_2PO_4$ in water	1000 mg/l	500 ml	1.19898.0500
	Titrisol®	$H_{3}PO_{4}$ in water	1000 mg	1 ampoule	1.09870.0001
Sodium	Ready-to-use solution	$NaNO_3$ in water	1000 mg/l	500 ml	1.19507.0500
Sulfate	Ready-to-use solution	$Na_2SO_4$ in water	1000 mg/l	500 ml	1.19813.0500
	Titrisol®	$H_2SO_4$ in water	1000 mg	1 ampoule	1.09872.0001

#### Certipur® ion chromatography multi-element standards, anionic

Designation	Composition	Matrix	Package size	Ord. No.
Anion multi-element standard I	1000 mg/l: F, Br, PO₄	Water	500 ml	1.11437.0500
Anion multi-element standard II	1000 mg/I: CI, NO <sub>3</sub> , SO <sub>4</sub>	Water	500 ml	1.11448.0500
IC multi-element standard I	100 mg/l: F / 250 mg/l: Cl / 500 mg/l: $NO_{3}$ /	Water	500 ml	1.70398.0500
	500 mg/l: SO <sub>4</sub> / 1000 mg/l: PO <sub>4</sub>			
IC multi-element standard V	10 mg/l: F, Br / 50 mg/l: NO <sub>3</sub> , PO <sub>4</sub> /	Water	100 ml	1.09032.0100
	100 mg/I: Cl / 200 mg/I: SO₄			

#### Certipur® ion chromatography multi-element standards, cationic

Designation	Composition	Matrix	Package size	Ord. No.
IC multi-element standard VII	100 mg/l: Li, Na, K, NH₄, Mn, Ca, Mg, Sr, Ba	0.001 mol/l $HNO_3$	100 ml	1.10322.0100
acc. to EN ISO 14911				
IC multi-element standard VI	10 mg/l: NH₄ / 50 mg/l: K /	$0.01 \text{ mol/l HNO}_3$	100 ml	1.09036.0100
	100 mg/l: Na, Ca, Mg			

# Certipur<sup>®</sup> reference materials for reliable calibration in UV-VIS spectroscopy

# Certipur<sup>®</sup> products

• UV-VIS standards

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Advantages

- Ready-to-use solutions
- Solutions are according to Reag. Ph Eur
- Stable solutions in glass ampoules





# UV-VIS spectroscopy

UV-VIS spectroscopy is indispensable in the modern analytical laboratory. It is a method that is well-established and is reliable and accurate. However, UV-VIS spectrometers must be checked regularly for correctness and uniformity of results and function and documented as such. This is obligatory when working according to Ph Eur, but also when working according to GLP, GMP, USP, ASTM and DIN EN ISO 9001:2008.

#### The following parameters can be determined using UV-VIS standards according to Ph Eur:

- Absorbance
- Stray light
- Spectral resolution power
- Wavelength accuracy

#### **Certipur® UV–VIS standards**

Designation	Content	Package size	Ord. No.
UV-VIS standard 1	Potassium dichromate solution for the absorbance	$2 \ x \ 10 \ ml \ K_2 Cr_2 O_7 \ 60.06 \ mg/l \ in \ H_2 SO_4 \ 0.01 \ N \ and$	1.08160.0001
	acc. to Ph Eur	$6 \times 10 \text{ ml } \text{H}_2\text{SO}_4 0.01 \text{ N}$	
UV-VIS standard 1A	Potassium dichromate solution for the absorbance	2 x 10 ml K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub> 600.6 mg/l in H <sub>2</sub> SO <sub>4</sub> 0.01 N and	1.04660.0001
	at 430 nm acc. to Ph Eur	$6 \times 10 \text{ ml H}_2\text{SO}_4 0.01 \text{ N}$	
UV-VIS standard 2	Sodium nitrite solution for straylight testing	3 x 10 ml NaNO <sub>2</sub> 50 g/l in H <sub>2</sub> 0	1.08161.0001
	acc. to Ph Eur		
UV-VIS standard 3	Sodium iodide solution for straylight testing	$3 \times 10$ ml Nal 10 g/l in H <sub>2</sub> O	1.08163.0001
	acc. to Ph Eur		
UV-VIS standard 4	Potassium chloride solution for straylight testing	3 x 10 ml KCl 12 g/l in H <sub>2</sub> 0	1.08164.0001
	acc. to Ph Eur		
UV-VIS standard 5	Toluene solution in hexane for testing	2 x 10 ml 0.02 % (v/v) Toluol in n-Hexan and	1.08165.0001
	the spectral resolution power acc. to Ph Eur	6 x 10 ml n-Hexan	
UV-VIS standard 6	Holmium oxide solution reference material	$3 \times 10 \text{ ml Ho}_2O_3 40 \text{ g/l in HClO}_4 (10 \% \text{ v/v})$	1.08166.0001
	for the wavelength acc. to Ph Eur		

#### Auxiliaries

Designation	Content	Package size	Ord. No.
Rectangular cells	Quartz 10 mm Spectroquant®	2 cells	1.00784.0001

# Certipur<sup>®</sup> reference materials for reliable calibration in volumetry

# Certipur<sup>®</sup> products

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# Advantages

- Secondary reference material for accurate titer determination
- Highly pure materials traceable to NIST
- In accordance to the reagents part of Pharmacopeias
- Measured in the ISO 17025 accredited laboratory



Page

# Certipur<sup>®</sup> secondary reference materials in volumetry

Merck Millipore's laboratory for mass fraction is accredited according to DIN EN ISO / IEC 17025. Certipur® primary substances are directly traceable to standard reference materials from the NIST (National Institute for Standard and Technology, Gaithersburg, Maryland, United States). Proper titer determination is an important prerequisite for accurate and comparable analysis in the titration laboratory. Influential factors such as temperature, instrument variances, different methods of handling, weighing errors, etc. and the volumetric solution itself can impact the titration results. To compensate for these factors, titer determination under working conditions is necessary in the respective laboratory. This is where Certipur® – secondary reference materials (volumetric standards) comes in. These are very pure, high-grade and stable solid substances. To ensure their high standard of quality, they are manufactured under the strictest control and measured with the highest possible precision in our accredited laboratory according to DIN EN ISO / IEC 17025.

#### **Volumetric standards**

Analysis	Designation	Package size	Ord. No.
Acidimetry	Sodium carbonate 1	80 g	1.02405.0080
	Tris(hydroxymethyl)aminomethane <sup>2</sup>	80 g	1.02408.0080
Alkalimetry	Potassium hydrogen phthalate 1,2	80 g	1.02400.0080
	Benzoic acid <sup>1,2</sup>	60 g	1.02401.0060
Argentometry	Sodium chloride <sup>1,2</sup>	80 g	1.02406.0080
Complexometry	Zinc <sup>1</sup>	100 g	1.02409.0100
	Calcium carbonate <sup>2</sup>	50 g	1.02410.0050
lodometry	Potassium iodate	100 g	1.02404.0100
Redox titration	Iron(II)ethylenediammonium sulfate	80 g	1.02402.0080
	Potassium dichromate <sup>2</sup>	80 g	1.02403.0080
	di-Sodium oxalate <sup>2</sup>	60 g	1.02407.0060

1: Solution according to Reag. Ph Eur | 2: Solution according to the reagents chapter of USP

# Certipur<sup>®</sup> reference materials for reliable calibration in special applications

# Certipur<sup>®</sup> products

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## Advantages

- Ready-to-use solutions
- Solutions according to international norms
- Stable solutions with long minimum shelf lives



# Reference materials for refractometry

These standards are for the calibration of refractometers. We offer a range of different refractive index standards in order to perform calibration in the same range as the final measurement. These Certipur<sup>®</sup> standards are traceable to SRM from NIST and to PTB.

#### Certipur® refractive index standards acc. to Ph Eur

Designation	Composition	Refractive index (20°C)	Package size	Ord. No.
Refractive index standard Kit 1	(2,2,4-Trimethyl-pentane/water)	1.3915	1 box	1.08962.0001
			with 5 x 8 ml ampoules	
Refractive index standard Kit 2	(Toluene/water)	1.4969	1 box	1.08961.0001
			with 5 x 8 ml ampoules	
Refractive index standard Kit 3	(1-Methyl-naphthalene/water)	1.6160	1 box	1.08963.0001
			with 5 x 8 ml ampoules	

#### NEW: Brix Standards Certipur® for refractometer calibration

Brix Standards Certipur<sup>®</sup> are high quality sucrose calibration standards that can be used for the determination of the Brix value by all types of refractometers. Featuring extended certification as well as extended shelf life, they ensure unrivalled reliability and efficiency.

#### Certipur® refractive index standards traceable to SRM from NIST and PTB

Designation	Package size	Ord. No.
Refractive index standard traceable to SRM from NIST and PTB	1 box	5.00400.0001
0° Brix	with 5 x 8 ml ampoules	
Refractive index standard traceable to SRM from NIST and PTB	1 box	5.00405.0001
5° Brix	with 5 x 8 ml ampoules	
Refractive index standard traceable to SRM from NIST and PTB	1 box	5.00410.0001
10° Brix	with 5 x 8 ml ampoules	
Refractive index standard traceable to SRM from NIST and PTB	1 box	5.00420.0001
20° Brix	with 5 x 8 ml ampoules	
Refractive index standard traceable to SRM from NIST and PTB	1 box	5.00430.0001
30° Brix	with 5 x 8 ml ampoules	
Refractive index standard traceable to SRM from NIST and PTB	1 box	5.00440.0001
40° Brix	with 5 x 8 ml ampoules	
Refractive index standard traceable to SRM from NIST and PTB	1 box	5.00450.0001
50° Brix	with 5 x 8 ml ampoules	
Refractive index standard traceable to SRM from NIST and PTB	1 box	5.00460.0001
60° Brix	with 5 x 8 ml ampoules	

# Ready-to-use colour comparison solutions according to the Ph Eur

Analysis of the color of a liquid in the range brown – yellow – red is described in European pharmacopoeia. However, the preparation of the solutions required is complex and time-consuming. Certipur<sup>®</sup> reference solutions save time and money: all color standards – B, BY, Y, GY and R – are available as a ready-to-use kit. To perform the analysis, the sample is simply placed in the empty cuvette contained in the kit and placed alongside the color comparison solutions in the rack provided.

#### Certipur® colour reference solutions acc. to Ph Eur

Designation	For examination of the color of solutions	Package size	Ord. No.
Colour reference solution B	According to Ph Eur B1-B9	1 set of cuvettes	1.00265.0001
Colour reference solution BY	According to Ph Eur BY1-BY7	1 set of cuvettes	1.00266.0001
Colour reference solution Y	According to Ph Eur Y1-Y7	1 set of cuvettes	1.00267.0001
Colour reference solution GY	According to Ph Eur GY1-GY7	1 set of cuvettes	1.00268.0001
Colour reference solution R	According to Ph Eur R1-R7	1 set of cuvettes	1.00269.0001
Empty cells with screw cap		25 units	1.14724.0001



# TOC analysis according to DIN EN 1484-H3

Potassium hydrogen phthalate standard. The European standard has come into force for the analysis of TOC (Total Organic Compound): EN 1484-H3 (which replaces DIN 38409-H3). This standard regulates the procedures for the determination of the concentration of organic carbon in drinking-, ground-, surface-, lake- and waste water. Our Certipur® TOC standard facilitates the calibration of TOC instruments according to the new standard, hence creating a reliable basis for your analyses. The potassium hydrogen phthalate solution is available as a ready-to-use standard in a concentration of 1000 mg/l in water. The standard is stabilized and protected from light in brown glass bottles. It has a minimum shelf life of 3 years for sealed, properly stored products.

#### Certipur<sup>®</sup> TOC standard

Designation	Package size	Ord. No.
TOC standard solution acc. to EN 1484 / DIN 38409-H3 as potassium hydrogen phthalate in water,	100 ml	1.09017.0100
stabilized, 1000 mg/l		

### Color measurement and photometry according to DIN 5033, part 9

Color measurement according to this standard is possible using the primary reference material provided by Merck Millipore. This has been certified by the PTB. The Certificate of Analysis of barium sulfate white standard includes values for spectral density (reflection from 350 – 800 nm), reflection factor and standard color values.

#### Certipur<sup>®</sup> white standard

Designation	Package size	Ord. No.
Barium sulfate white standard acc. to DIN 5033		1.01748.0250

### Platinum Cobalt color reference solution (Hazen)

Color measurement according to DIN EN ISO 7887, ASTM D 1209-05, DIN EN ISO 6271-1 and APHA 2120 B + C can be done with the ready-to-use Hazen 500 reference solution.

#### Certipur® Hazen color reference solution

Designation	Package size	Ord. No.
Hazen 500 color reference solution	250 ml	1.00246.0250

# Fluxes for XRF Spectromelt<sup>®</sup>

Spectromelt<sup>®</sup> materials for fluxing digestion or pressing of powder tablets are the perfect assistent for your X-ray fluorescence analysis (XRF). The purity of Spectromelt<sup>®</sup> borates gives the necessary degree of precision and our grinding aid materials are specially developed for automated sample preparation.



# Fluxes for XRF

### Contents

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## Advantages

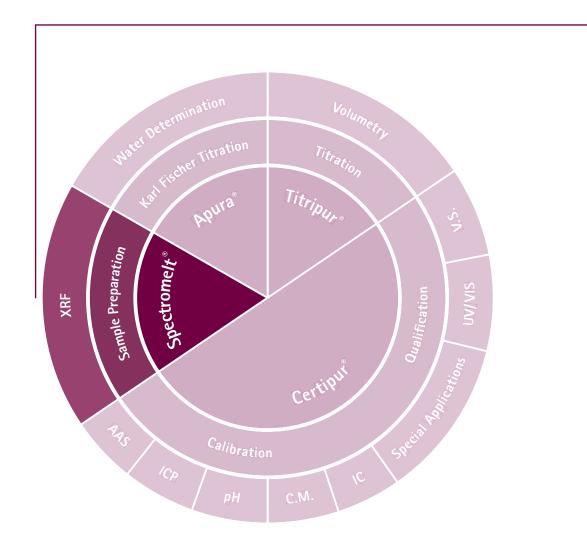
- Spectromelt<sup>®</sup> for fluxing digestion as granulate or as fused material
- Spectromelt<sup>®</sup> tablets as grinding and tabletting aid
- Spectromelt<sup>®</sup> tablets can be used for automated sample preparation
- High purity materials
- Batch-to-batch consistency
- Extensive specification of trace impurities



www.merckmillipore.com/spectromelt

# Spectromelt<sup>®</sup> – our quality standard in instrumental analysis

Choose Spectromelt<sup>®</sup> for your sample preparation and get the best results in XRF. Spectromelt<sup>®</sup> is suitable for fluxing and grinding techniques.



# Specification

ISO Standards of the International Organization for Standardization

# Quality grades for your individual needs

#### Spectromelt<sup>®</sup>

 ${\sf Spectromelt}^{\circledast}$  fluxes for XRF | High purity borates for fluxing | Tablets for automated sample preparation



# Reagents for sample preparation in X-ray fluorescence (XRF) analysis

Two techniques are used for XRF sample preparation: fluxing or grinding.

#### Fluxing

Spectromelt <sup>®</sup> A-series is based on lithium borates, whereas Spectromelt <sup>®</sup> B-series is based on sodium borate. Fusion with our borates gives a high homogeneity of the sample. Interference of trace impurities in the sample can be reduced by using a high purity flux. All Spectromelt <sup>®</sup> fluxes with their extensive specifications are analyzed for inorganic trace impurities and have a high batch-to-batch consistency. This reduces necessary adjustments of the instruments and increases the accuracy of the measurement. We offer Spectromelt<sup>®</sup> in powder and fused quality. The fused particles are spherical, have a small particle size distribution and a higher bulk density. The material is easy to handle and has a low loss on ignition. Fluxes are also available with integrated non-wetting-agent, such as lithium bromide.



### Grinding

Tabletting the sample is an alternative to fusion. The Spectromelt<sup>®</sup> C-series is especially developed as a grinding and tabletting aid. In automated sample preparation of powder samples speed of analysis is an important factor. Spectromelt<sup>®</sup> C10, C15 and C20 are offered as tablets with a weight of 333 mg each. The tablets are robust, so the automated sample preparation will not be disturbed. Tablets made with Spectromelt<sup>®</sup> C10 are very stable for analysis and storage. Spectromelt<sup>®</sup> C10 is developed to have distinguishable adhesive properties. Spectromelt<sup>®</sup> C20 is compared to C10 less hygroscopic because of an adjusted composition. C20 is preferable for damp climate and its properties effect fewer residues in the grinder. The new Spectromelt<sup>®</sup> C15 combines the adhesive properties of C10 and the lower hygroscopy of C20. The improved homogenization of the sample gives more precise results in analysis. The new Spectromelt<sup>®</sup> C15 MP micropowder can be used as a grinding and tabletting aid. Hoechst Wax C can only be used as a tabletting aid.

# Ordering information Fluxes for XRF

# Spectromelt® A series Lithium borates

Product	Composition	Content	Packaging	Ord. No.			
Spectromelt <sup>®</sup> A 10	Lithium tetraborate	1 kg	PE bottle	1.10783.1000			
		5 kg	PE bottle	1.10783.5000			
		25 kg	PE drum	1.10783.9025			
Spectromelt® A 12 *	66 % Lithium tetraborate, 34 % lithium metaborate	1 kg	PE bottle	1.11802.1000			
		5 kg	PE bottle	1.11802.5000			
Spectromelt® A 14	51 % Lithium tetraborate, 27 % lithium metaborate,	1 kg	PE bottle	1.11724.1000			
	12 % lanthanum (III) oxide, 10 % lithium fluoride						
Spectromelt® A 20 *	Lithium metaborate	1 kg	PE bottle	1.12996.1000			
Spectromelt <sup>®</sup> A 22 *	22 parts lithium metaborate, 12 parts lithium	1 kg	PE bottle	1.24001.1000			
	tetraborate						
Spectromelt <sup>®</sup> A 50 *	50 % Lithium tetraborate, 50 % lithium metaborate	1 kg	PE bottle	1.17057.1000			
Spectromelt® A 100 *	Lithium tetraborate	1 kg	PE bottle	1.12630.1000			
		5 kg	PE bottle	1.12630.5000			
Spectromelt® A 1000	Lithium tetraborate with 0.07 % lithium bromide	1 kg	PE bottle	1.13175.1000			
		5 kg	PE bottle	1.13175.5000			
* = Spectromelt <sup>®</sup> fused quality	During the production process of the Spectromelt <sup>®</sup> the material is	s molten to a	chieve high densi	ty			
	particles. The particles are spherical and have a small particle size distribution.						
	The fused material has a higher density than the granulated mate						
	advantage is that the fused material is more easy to handle in the	e laboratory a	nd has a				
	lower loss on ignition.						

# Spectromelt® B series Sodium borates

Product	Composition	Content	Packaging	Ord. No.
Spectromelt® B 10	di-Sodium tetraborate	1 kg	PE bottle	1.06304.1000
		5 kg	PE bottle	1.06304.5000

# Spectromelt® C series Grinding aids

Product	Composition	Content	Packaging	Ord. No.
Spectromelt <sup>®</sup> C 10	Grinding and tabletting aid	5 kg	PE bottle	1.13990.5000
	Tablet weight: 333 mg			
Spectromelt® C 15	Grinding and tabletting aid	5 kg	PE bottle	1.17062.5000
	Tablet weight: 333 mg			
Spectromelt® C 15 MP	Grinding and tabletting aid	1 kg	PE bottle	1.17074.1000
	Micropowder			
Spectromelt <sup>®</sup> C 20	Grinding and tabletting aid	5 kg	PE bottle	1.13934.5000
	Tablet weight: 333 mg			
Hoechst wax C micropowder	Tabletting aid	1 kg	PE bottle	1.09014.1000

# Additives for fluxing and grinding

Product	Content	Packaging	Ord. No.
Boric acid Suprapur®	50 g	PE bottle	1.00765.0050
	500 g	PE bottle	1.00765.0500
Lithium carbonate Suprapur®	50 g	PE bottle	1.05676.0050
	250 g	PE bottle	1.05676.0250
Sodium carbonate anhydrous Suprapur®	50 g	PE bottle	1.06395.0050
	500 g	PE bottle	1.06395.0500
Lithium bromide hydrate Suprapur®	50 g	PE bottle	1.05668.0050
	250 g	PE bottle	1.05668.0250
Lithium bromide solution w=25 %	50 ml	PE bottle	1.00884.0050
Potassium Iodide Suprapur®	50 g	PE bottle	1.05044.0050
	500 g	PE bottle	1.05044.0500
Sodium nitrate Suprapur®	50 g	PE bottle	1.06546.0050
Lithium nitrate Suprapur®	100 g	PE bottle	1.05653.0100
	500 g	PE bottle	1.05653.0500
Lithium Fluoride Suprapur®	50 g	PE bottle	1.05686.0050
	500 g	PE bottle	1.05686.0500
Lanthanum (III) Oxide	25 g	Glass bottle	1.10982.0025
for atomic absorption spectroscopy			

# Application notes for Spectromelt® fluxing and grinding

Spectromelt <sup>®</sup> type	Recommended for digestion of
A10, A 100, A1000	Basic oxides carbonates, ground samples, magnesite, aluminum silicates, bauxite,
	rare earth oxide, cement, sludges, phosphate
A 12, 14	Borax, cement, sludges, sand, chromium
A 20, A 22	Acidic oxides, silicates, sand, clay, spineless
B 10	Iron, iron-, chromium-, tin- and titanium ores, rare earth minerals
C 10, C 20	Ores, slags; in steel and cement analysis
LiBr, Kl	Non wetting agent increases the surface tension of the melt and simplifies removal
	out of the crucible.

# High purity acids and bases Suprapur<sup>®</sup> | Ultrapur

Digestion with acids is often used during sample preparation. The reagents for sample preparation have an important impact on the outcome of the measurement. In modern instrumental trace analysis any impurity can disturb the measurement. Acids with low impurities should be used to minimize the interference on the instrument signal. High purity acids and bases Suprapur<sup>®</sup> and Ultrapur offer the most suitable purity of wet digestion materials and they have low impurities during the entiry minimum shelf life.



# High purity acids and bases

# Contents

Quality standards	222
Reagents for wet digestion	224
• Ordering information	226

Page

# Advantages

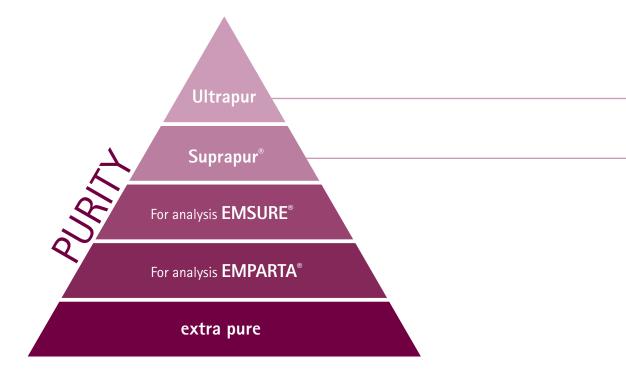
- Choose your purity grade for instrumental trace analysis
- Suprapur<sup>®</sup> acids filled in borosilicate glass or extra pure PE material
- Ultrapur acids filled in PFA bottles
- Certificate of Analysis gives an extensive impurity profile



www.merckmillipore.com/acids

# Different grades to meet your needs

Take a look at our high purity material for wet digestion Suprapur<sup>®</sup> and Ultrapur, especially for instrumental trace analysis whenever you are using an acid, you have to consider your application, your target and of course your budget. Each application is different and therefore we offer a wide range of acids in different grades and many different packaging sizes and materials. No matter if you need to perform a highly critical analysis or preparing basic lab applications, with our brand names you will always find the most suitable reagent.



# Quality grades for your individual needs

# Ultrapur

High purity acids, hydrogen peroxide | Highly sophisticated instrumental trace analysis, e.g. ICP-MS | Specified parameters in ppt range

### Suprapur®

High purity acids, hydrogen peroxide | Highly sophisticated instrumental trace analysis, e.g. AAS and ICP-OES | Specified parameters in ppb range

### Acids EMSURE® and EMPARTA®

- For classical analysis we offer a range of acids for analysis EMSURE® and EMPARTA®
- For further information please have a look in chapter »Acids for analysis«

### Acids extra pure

- For preparative lab applications and cleaning purposes
- For further information please have a look in chapter »Acids for analysis«

# Reagents for wet digestion Suprapur<sup>®</sup> | Ultrapur

The more sensitive the instrumental detection becomes, the more important it is to use highly pure reagents. Merck Millipore provides a whole range of standards and reagents that meet the demands of modern instrumental analysis. During the sample preparation it is important to know the blank values of the reagents. On the certificate you find the specification and actual batch values of the parameters.

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### Suprapur<sup>®</sup> acids and bases are suitable for trace analysis in the ng/g (ppb) range.

Suprapur<sup>®</sup> acids are filled in borosilicate or extra pure PE bottles. The material minimizes any elemental impurity of the acid, so the specification of the unopened bottle is kept during the minimum shelf life. The bottles are cleaned and pre-conditioned before filling. Quality control is done after filling. This gives you the assurance that the certified batch values are those values of the filled acid. Suprapur<sup>®</sup> reagents are packed in a stable outer-box.

Hydrogen peroxide Suprapur<sup>®</sup> is packed in a black bottle to protect it against light. The bottle is made of extra pure PE material to avoid any contamination. To make it more safe for you: hydrogen peroxide bottles are closed with the SafetyCap. The SafetyCap with the PTFE membrane releases the pressure, but avoids any contamination.

### Ultrapur reagents are preferred for ultra-trace analysis in the pg/g (ppt) range.

Ultrapur reagents are produced by sub-boiling distillation. The slowly distilled reagents subsequently have the lowest possible trace-impurities. Ultrapur reagents are exclusively filled in pre-conditioned PFA (fluorpolymer) bottles. This material meets the highest demands of all users for ultra-trace instrumental analysis, e.g. ICP-MS. Ultrapur® reagents are packed in a stable outer-box.

### More information

▶ See chapter »Packaging« on page 18 ▶ See chapter »Acids for analysis« on page 94

# Suprapur®



# Ordering information Suprapur®

High purity acids and bases Suprapur®

	Designation	Content	Packaging	Ord. No.
A	Acetic acid 100 % Suprapur®	250 ml	Glass bottle	1.00066.0250
		1	Glass bottle	1.00066.1000
	Ammonia solution 25 % Suprapur®	250 ml	PE bottle	1.05428.0250
		500 ml	PE bottle	1.05428.0500
		1	PE bottle	1.05428.1000
		2.5 l	PE bottle	1.05428.2500
В	Boric acid Suprapur®	50 g	PE bottle	1.00765.0050
		500 g	PE bottle	1.00765.0500
F	Formic acid 98 – 100 % Suprapur®	250 ml	Glass bottle	1.11670.0250
		1	Glass bottle	1.11670.1000
ł	Hydrobromic acid 47 % Suprapur®	250 ml	Glass bottle	1.00306.0250
		1	Glass bottle	1.00306.1000
	Hydrochloric acid 30 % Suprapur®	250 ml	PE bottle	1.00318.0250
		500 ml	PE bottle	1.00318.0500
		1	PE bottle	1.00318.1000
		2.5 l	PE bottle	1.00318.2500
	Hydrofluoric acid 40 % Suprapur®	250 ml	PE bottle <sup>1</sup>	1.00335.0250
		500 ml	PE bottle <sup>1</sup>	1.00335.0500
		1	PE bottle	1.00335.1000
		2.5 l	PE bottle	1.00335.2500
	Hydrogen peroxide 30 % Suprapur®	250 ml	PE bottle <sup>2</sup>	1.07298.0250
		500 ml	PE bottle <sup>2</sup>	1.07298.0500
		1	PE bottle <sup>2</sup>	1.07298.1000
N	Nitric acid 65 % Suprapur®	250 ml	Glass bottle	1.00441.0250
		1	Glass bottle	1.00441.1000
)	Oxalic acid dihydrate Suprapur®	100 g	PE bottle	1.00489.0100
)	Perchloric acid 70 % Suprapur®	250 ml	Glass bottle	1.00517.0250
		1	Glass bottle	1.00517.1000
	ortho-Phosphoric acid 85 % Suprapur®	250 ml	PE bottle	1.00552.0250
		500 ml	PE bottle	1.00552.0500
		1	PE bottle	1.00552.1000
		2.5 l	PE bottle	1.00552.2500
5	Sodium hydroxide solution 30 % Suprapur®	250 ml	PE bottle	1.05589.0250
		500 ml	PE bottle	1.05589.0500
		1	PE bottle	1.05589.1000
		2.5 l	PE bottle	1.05589.2500
	Sulphuric acid 96 % Suprapur®	250 ml	Glass bottle	1.00714.0250
		1	Glass bottle	1.00714.1000

1 = HDPE dosage bottle >page 108. | 2 = PE bottle with SafetyCap >page 106. | Glass bottles for Suprapur® acids are made of borosilicate glass. | For further information have a look in chapter »Packaging«.

# Ultrapur



# Ordering information Ultrapur

High purity acids and bases Ultrapur

	Designation	Content	Packaging	Ord. No.
Н	Hydrochloric acid 30 % Ultrapur	250 ml	PFA bottle	1.01514.0250
		500 ml	PFA bottle	1.01514.0500
		11	PFA bottle	1.01514.1000
	Hydrofluoric acid 48 % Ultrapur	1	PFA bottle	1.01513.1000
	Hydrogen peroxide 31 % Ultrapur	1	PFA bottle	1.06097.1000
Ν	Nitric acid 60 % Ultrapur	250 ml	PFA bottle	1.01518.0250
		500 ml	PFA bottle	1.01518.0500
		1	PFA bottle	1.01518.1000
S	Sulphuric acid 96 % Ultrapur	250 ml	PFA bottle	1.01516.0250
W	Water Ultrapur	500 ml	PE bottle	1.01262.0500
		11	PE bottle	1.01262.1000

# High purity salts Suprapur<sup>®</sup>

For instrumental analysis we offer high purity of inorganic salts for sample preparation. High purity salts are produced under strict conditions in a clean room area. Sophisticated production and crystallization ensure the purity and batch-to-batch consistency of our products. Quality management is done according to DIN EN ISO 9001.



# High purity salts

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• High purity material	233

# Advantages

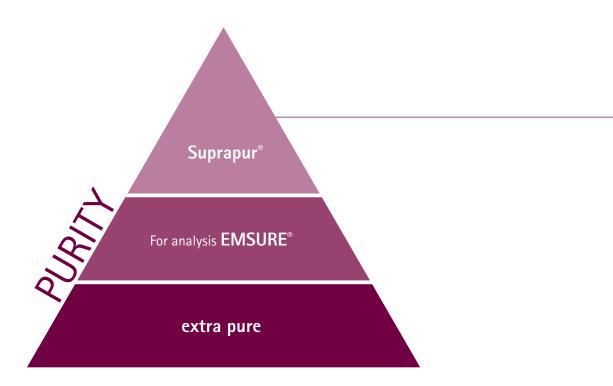
- Highest purity of salts
- Extensive impurity profile in the Certificate of Analysis
- High batch-to-batch consistency



www.merckmillipore.com/salts

# Salts Suprapur®

Production and quality control is done in Darmstadt following modern and most sophisticated methods. This ensures the high purity of the salt. The purity of the salts are 99.99 % or higher up to 99.9999 %. For the products the Certificate of Analysis (CoA) gives an extensive profile of the impurities. The CoA is available on the internet: www.merckmillipore.com



**Recommended storage instructions:** We recommend storing all inorganic salts in airtight containers in a dry place at room temperature when possible. Further storage instructions are included in our Safety Data Sheets.

**Higher water of crystallization content:** In case of temperature fluctuations, products with a higher water of crystallization content have a greater tendency to cake than anhydrous salts. In addition, many products tend to decompose when exposed to heat. These compounds should therefore be stored in a dry place that is as cool as possible or at room temperature (max. 25°C). Please follow the storage instructions on the labels of the products.

# Quality grades for your individual needs

# Suprapur®

High purity salts | Instrumental trace analysis, e.g. AAS | Specified parameters in ppm to ppb range

### Salts EMSURE®

- For classical analysis we offer a range of salts for analysis EMSURE®
- For further information please have a look in chapter »Salts for analysis«

### Salts extra pure

- For preparative lab applications
- For further information please have a look in chapter »Salts for analysis«



# Ordering information Suprapur<sup>®</sup>

# High purity salts Suprapur® A-P

Product	Content	Packaging	Ord. No.
Ammonium chloride Suprapur®	50 g	PE bottle	1.01143.0050
	500 g	PE bottle	1.01143.0500
Ammonium dihydrogen phosphate Suprapur®	50 g	PE bottle	1.01440.0050
Ammonium sulfate Suprapur®	50 g	PE bottle	1.01209.0050
	500 g	PE bottle	1.01209.0050
Barium chloride Suprapur®	50 g	PE bottle	1.01716.0050
	500 g	PE bottle	1.01716.0500
Barium fluoride Suprapur®	50 g	PE bottle	1.01722.0050
Boric acid Suprapur®	50 g	PE bottle	1.00765.0050
	500 g	PE bottle	1.00765.0500
di-Boron trioxide Suprapur®	250 g	PE bottle	1.00169.0250
Calcium carbonate Suprapur®	50 g	PE bottle	1.02059.0050
	500 g	PE bottle	1.02059.0500
Calcium chloride tetrahydrate Suprapur®	100 g	PE bottle	1.02384.0100
Calcium nitrate tetrahydrate Suprapur®		PE bottle	
Calcium intrate tetranyurate Suprapur <sup>®</sup>	100 g		1.02123.0100
Casium ablasida Cumanum	500 g	PE bottle PE bottle	1.02123.0500
Cesium chloride Suprapur®	50 g		1.02039.0050
	250 g	PE bottle	1.02039.0250
Lithium bromide hydrate Suprapur®	50 g	PE bottle	1.05668.0050
	250 g	PE bottle	1.05668.0250
Lithium carbonate Suprapur®	250 g	PE bottle	1.05676.0250
Lithium chloride monohydrate Suprapur®	250 g	PE bottle	1.05677.0250
Lithium fluoride Suprapur®	50 g	PE bottle	1.05686.0050
	500 g	PE bottle	1.05686.0500
Lithium nitrate Suprapur®	100 g	PE bottle	1.05653.0100
	500 g	PE bottle	1.05653.0500
Lithium sulfate monohydrate Suprapur®	50 g	PE bottle	1.05697.0050
Magnesium nitrate hexahydrate Suprapur®	50 g	PE bottle	1.05855.0050
	500 g	PE bottle	1.05855.0500
Potassium bromide Suprapur®	100 g	PE bottle	1.04904.0100
	500 g	PE bottle	1.04904.0500
Potassium carbonate-1.5-hydrate Suprapur®	50 g	PE bottle	1.04926.0050
	250 g	PE bottle	1.04926.0250
Potassium chloride Suprapur®	50 g	PE bottle	1.04938.0050
	500 g	PE bottle	1.04938.0500
Potassium dihydrogen phosphate anhydrous Suprapur®	50 g	PE bottle	1.05108.0050
	250 g	PE bottle	1.05108.0250
	500 g	PE bottle	1.05108.0500
di-Potassium hydrogen phosphate anhydrous Suprapur®	100 g	PE bottle	1.05109.0100
	250 g	PE bottle	1.05109.0250
	500 g	PE bottle	1.05109.0500
Potassium hydroxide hydrate Suprapur®	500 g	PE bottle	1.05002.0500
Potassium iodide Suprapur®	50 g	PE bottle	1.05044.0050
	500 g	PE bottle	1.05044.0500
Potassium nitrate Suprapur®	50 g	PE bottle	1.05065.0050
· · · · · · · · · · · · · · · · · · ·	500 g	PE bottle	1.05065.0500
Potassium sulfate Suprapur®	100 g	PE bottle	1.05152.0100
		. = 00000	

### High purity salts Suprapur® 0-Z

Product	Content	Packaging	Ord. No.
Sodium acetate anhydrous Suprapur®	50 g	PE bottle	1.06264.0050
	500 g	PE bottle	1.06264.0500
Sodium bromide Suprapur®	100 g	PE bottle	1.06363.0100
	250 g	PE bottle	1.06363.0250
Sodium carbonate anhydrous Suprapur®	50 g	PE bottle	1.06395.0050
	500 g	PE bottle	1.06395.0500
Sodium chloride Suprapur®	50 g	PE bottle	1.06406.0050
	250 g	PE bottle	1.06406.0250
	500 g	PE bottle	1.06406.0500
Sodium dihydrogen phosphate anhydrous Suprapur®	50 g	PE bottle	1.06370.0050
	250 g	PE bottle	1.06370.0250
Sodium fluoride Suprapur®	25 g	PE bottle	1.06450.0025
di-Sodium hydrogen phosphate anhydrous Suprapur®	100 g	PE bottle	1.06566.0100
	500 g	PE bottle	1.06566.0500
Sodium hydroxide monohydrate Suprapur®	50 g	PE bottle	1.06466.0050
	500 g	PE bottle	1.06466.0500
Sodium iodide Suprapur®	100 g	PE bottle	1.06519.0100
	500 g	PE bottle	1.06519.0500
Sodium nitrate Suprapur®	50 g	PE bottle	1.06546.0050
Sodium sulfate anhydrous Suprapur®	50 g	PE bottle	1.06647.0050
	500 g	PE bottle	1.06647.0500
di-Sodium tetraborate anhydrous Suprapur®	25 g	PE bottle	1.06309.0025
	250 g	PE bottle	1.06309.0250
Sodium thiosulfate pentahydrate Suprapur®	100 g	PE bottle	1.06509.0100
	500 g	PE bottle	1.06509.0500
Strontium nitrate anhydrous Suprapur®	50 g	PE bottle	1.07871.0050
	250 g	PE bottle	1.07871.0250

# High purity material Suprapur®

Also available as high purity materials for laboratory use are Bromine, lodine and Mercury Suprapur®.

# Ordering information

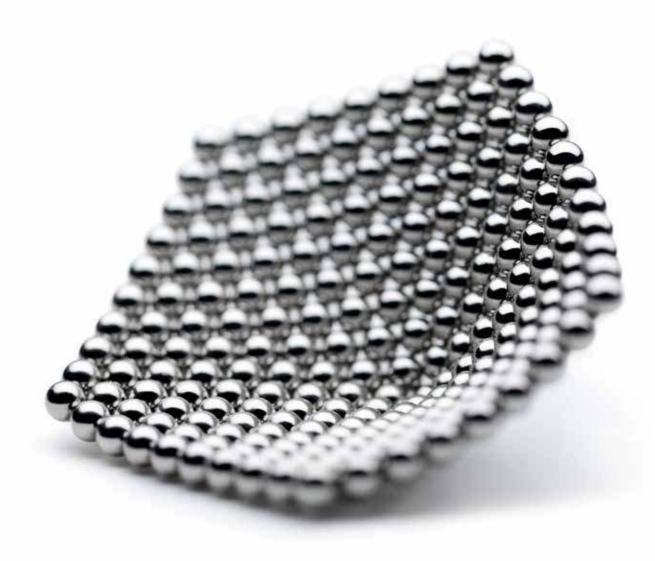
# High purity material Suprapur®

	Product	Content	Packaging	Ord. No.
В	Bromine Suprapur®	250 g	Glass bottle	1.01947.0250
1	Iodine Suprapur®	50 g	Glass bottle	1.04763.0050
		500 g	Glass bottle	1.04763.0500
М	Mercury Suprapur® *	250 g	Glass bottle	1.04404.0250
		1 kg	Glass bottle	1.04404.1000

\* = only available within the EU.

# Safety products and general applications

Safety, simplicity and sustainability. Merck Millipore helps you achieve them all with our large variety of inorganic reagents. Using natural products wherever possible, our reagents conform to the highest quality standards to make your daily work not only safer but also simpler.



# Absorption and filtration

Among the most important applications in laboratory practice are absorption, adsorption and filtration. To help ease your daily work, we offer several reagents as well as filter and clarification materials. Our products can be used for a wide variety of purposes, for example to dry, absorb or bind substances but also to decolorize, clarify or support filtration processes.

# Absorbents for spilled liquids

When accidents happen Chemizorb<sup>®</sup> is there to help. The product range includes specific absorbents for different substances that efficiently absorb spilled aggressive liquids in the laboratory to ensure swift and safe handling.

# Drying agents

Merck Millipore drying agents are the ideal choice whenever drying of gaseous, liquid or solid substances is required. Our diverse range of products helps protect your chemicals against moisture and mold formation during transportation, storage and in the laboratory.

# Auxiliaries for purification and sample preparation Page 276

Routine laboratory work involves purifying, enriching and separating samples for analytical purposes. Merck Millipore products cover various chemical and physical methods as well as an array of other applications, such as removing residues from gases, decalcification, sealing, lubrication, separation, stabilization and as reaction and filling aids.

# Indicators

Indicators are used in laboratories to observe chemical reactions. Whether for monitoring a chemical status, change of reaction or the end-point of a titration, you can rely on Merck Millipore's large portfolio of indicators for accurate results.

# **Cleaning applications**

Thorough, residue-free cleaning is essential for reliable processes in laboratories. No matter what you need to clean or what contaminants you face, Merck Millipore has the ideal solutions for your laboratory and equipment. Discover our established Extran<sup>®</sup> laboratory cleansers.

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# . ....

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# Absorption and filtration

Absorption, adsorption and filtration are among the most important applications in laboratory practice. In the Merck Millipore product range, you will find a wide variety of absorption and adsorption reagents as well as filter and clarification materials, such as activated charcoal, graphite, molecular sieves and sea sand. You can find chemical and physical data on our reagents in the Merck Millipore Chemicals and Reagents catalog and on our website: www.merckmillipore.com



# Absorption and filtration

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# Advantages

- Reliable: Above all thanks to their reliable take-up capacity and purification of a wide range of substances, the premium Merck Millipore reagents for absorption, adsorption or filtration offer great reliability for the implementation of your application
- **Convenient:** The comprehensive Merck Millipore product range allows easy ordering, all from one place
- Economical: Merck Millipore offers packaging sizes of small and larger packages for individual needs, so that the exact quantity required can be easily achieved.



www.merckmillipore.com/absorption-filtration

# Safety and environment



Many of the absorption and adsorption reagents and filter materials Merck Millipore offers for use in laboratories are natural reagents which are not harmful to the environment.

One example is Merck Millipore activated charcoal, which is produced from natural raw materials and, alongside laboratory applications, is also suitable for use in environmental protection.

Calcium oxide, which is used as a  $CO_2$  absorbent among other things, is also one of the Merck Millipore products for the sustainable protection of people and the environment, since it is extracted from selected natural marble.

More information about sustainable protection www.merckmillipore.com/protection



# Definitions

- Absorption [from Latin »absorbere«: to devour, swallow up] describes the take-up of gases or liquids by solid bodies. Solids with cavities of a defined pore size can be used for separation, purification or enrichment. In this way, mixtures usually of gases can be separated into their individual component parts as a result of their varying molecule sizes.
- Adsorption [from Latin »adsorbere«: to suck in] describes the take-up of a substance on the surface of a solid without a chemical reaction. Due to adhesive forces, gaseous, liquid and, in rare cases, solid substances accumulate on the surface or interface of the adsorbent. This releases the bond energy in the form of heat. The larger the surface, the greater the adsorption capacity of a substance. The total surface is therefore often given in »m²/g« for specification. This adsorption strength is reduced by warming and increased by cooling.
- **Filtration** is a mechanical separation process for the separation or purification of substances. The filter and clarification materials used for this are usually reagents with absorbent or adsorbent properties, although materials which work in a similar way to frits or which form capillaries as filter cake can be used to clarify solutions, to decolorize or to remove colloidally dissolved substances.

# Safety and environment – characteristics

► For easy detection, safety and/or sustainable characteristics of our products are highlighted with this symbol.

# Ordering information Absorption

Absorption tubes		CAS No.	Content	Packaging	Ord. No.
Absorption tube for CO <sub>2</sub> (sodium hydroxide on support)		-	3 units	Plastic tube	1.01562.0003
Absorption tube for H <sub>2</sub> O (r	nolecular sieve 0.3 nm with indicator)	-	3 units	Plastic can	1.06107.0003
Application advice	Drying of incoming or outgoing air or genera	Drying of incoming or outgoing air or general gas absorption			
Additional information	Absorption tubes are 15 cm long and have a diameter of 2 cm. Hoses of various dimensions can also be con- nected: hoses with 3 mm external diameter can be pushed into the tube, while hoses with 8 – 12 cm internal diameter can be pushed over the opening. The tube must be opened on both sides before use by removing the two stoppers.			nal	
Absorption tube for CO <sub>2</sub>					
Application advice	$CO_2$ absorption tubes are used to bond carbon dioxide, for example to keep the titer stable during titration with NaOH and to prevent the formation of Na <sub>2</sub> CO <sub>3</sub> through the CO <sub>2</sub> contained in the incoming air. The absorbent, which is placed on an inert carrier, also prevents caking after absorption. The air or gas can still flow through the absorption tube without a change in resistance.				
Additional information	$CO_2$ absorption tubes are filled with sodium absorption tubes has a significantly higher c the degree of depletion. In the case of $CO_2$ si capacity of the $CO_2$ absorption tube is aroun	apacity than caustic alkali pell aturation, the indicator change	ets. A »saturat es from dark g	tion display« show:	
Absorption tube for H <sub>2</sub> O					
Application advice	H <sub>2</sub> O absorption tubes are used to keep reage when working with solvents from withdrawa		e during Karl I	ischer titration or	
Additional information	In $H_2O$ absorption tubes, the filling also cont	ains an indicator displaying th	e degree of de	pletion.	



Calcium oxide		CAS No.	Content	Packaging	Ord. No.
Calcium oxide from marble small lumps ~ 3 - 20 mm		1305-78-8	1 kg	Plastic bottle	1.02109.1000
			25 kg	Fibre carton	1.02109.9025
Application advice	Calcium oxide is used as a drying agent	Calcium oxide is used as a drying agent and absorbent for $CO_2$ and for the creation of calcium bonds, e.g. in			
	the production of glass, paper, rubber an	the production of glass, paper, rubber and soda.			
Information on	Calcium oxide is obtained from selected	natural marble.			
sustainable protection					

# www.merckmillipore.com/absorption-filtration

Charcoal activated		CAS No.	Content	Packaging	Ord. No.	
Charcoal activated for analysis		7440-44-0	250 g	Metal can	1.02514.1000	
			1 kg	Metal can	1.02514.5000	
			20 kg	Fibre carton	1.02514.9025	
Charcoal activated granul	ar about 1.5 mm extra pure food grade	7440-44-0	1 kg	Plastic bag	1.02184.1000	
			5 kg	Fibre carton	1.02184.5000	
			25 kg	Fibre carton	1.02184.9020	
Charcoal activated powde	r extra pure food grade	7440-44-0	5 kg	Fibre carton	1.02183.1000	
			20 kg	Fibre carton	1.02183.9020	
Charcoal activated for an	alysis	7440-44-0	1 kg	Plastic bag	1.02186.9020	
			1 kg	Plastic bag	1.02186.0250	
	20 kg Fibre carton 1.02186.1000					
	1500 m <sup>2</sup> /g), activated charcoal adsorbs mole Since the chemical substances have to diffu	Activated charcoal is used among others for adsorption, decolorization and purification of e.g. gases and liquids and as well as a carrier for catalysts. Thanks to its very large internal surface area (800 to 1500 m <sup>2</sup> /g), activated charcoal adsorbs molecules of different sizes and has a wide variety of uses. Since the chemical substances have to diffuse into the pores during adsorption, powdered carbons are easier to »load« than granulated ones. Granulated carbons are more suitable for flow-through processes with gases as a result of their lower flow resistance.				
Additional information	Activated charcoal is used in the chemicals and food and beverage industries, and in environmental protection. Activated charcoal looks the same as charcoal, but a special »activation process« means that it has a porous, sponge-like structure with a larger internal surface area. The adsorption capacity resulting from this depends among other things on the starting material – fossil charcoal – and the activation process. Depending on the process, small cavities with a pore size of 1 to 25 nm (micro and macro pores) and a larger internal surface area of 800 to 1500 m <sup>2</sup> /g are formed.					
Information on sustainable protection	Activated charcoal is gained from natural raw materials (obtained from natural pine wood or mineral coal) and is used in the food and beverage industry and in environmental protection.					

Charcoal wood	CAS No.	Content	Packaging	Ord. No.		
Charcoal wood powder	7440-44-0	1 kg	Plastic bag	1.02204.1000		
		25 kg	Fibre carton	1.02204.9025		
Application advice	Charcoal wood can be used for adsorption, filtration or clarification for many technical purposes.					
Additional information	The coking or carbonization – heating up to 400°C without air – of beech w	ood produces	charcoal wood with			
	a very large internal surface area.					
Information on	Natural production through the coking of beech wood.					
sustainable protection						

Graphite		CAS No.	Content	Packaging	Ord. No.		
Graphite fine powder extr	a pure	7782-42-5	2.5 kg	Fibre carton	1.04206.2500		
			25 kg	Fibre carton	1.04206.9025		
Application advice	Graphite is used as an absorbent and lub	pricant at temperatures of up to are	ound 350°C.				
Additional information	Graphite is a gray material which shines	like metal, feels greasy and stains	easily as a res	ult of its flaky, eas	ily		
	divisible mass. It is manufactured from high quality coke, which is mixed with pitch and special tars and						
	pressed into molded shapes. These shapes are then heated to up to 3000°C without oxygen. This gives graph-						
	ite a high degree of purity, high capacity	ite a high degree of purity, high capacity for conduction and lubrication and the typical layered hexagonal					
	crystal structure. In order to create grap	stal structure. In order to create graphite in this powdered form, the shapes are broken and ground into					
	this fine powder in a pebble mill in a time-consuming process.						
Information on	Graphite consists mainly of pure carbon.						
sustainable protection							

# Ordering information Absorption

Molecular sieve	CAS No.	Content	Packaging	Ord. No.
Molecular sieve 0.3 nm beads ~ 2 mm	1318-02-1	250 g	Plastic bottle	1.05704.0250
		1 kg	Plastic bottle	1.05704.1000
		10 kg	Bucket, plastic	1.05704.9010
Molecular sieve 0.3 nm beads, with moisture indicator ~ 2 $mm^{1)}$	-	250 g	Plastic bottle	1.05734.0250
	-	1 kg	Plastic bottle	1.05734.1000
Molecular sieve 0.3 nm rods ~ 1.6 mm (1/16")	1318-02-1	250 g	Plastic bottle	1.05741.0250
		1 kg	Plastic bottle	1.05741.1000
		10 kg	Bucket, plastic	1.05741.9010
Molecular sieve 0.4 nm beads ~ 2 mm Reag. Ph Eur	1318-02-1	250 g	Glass bottle	1.05708.0250
		1 kg	Glass bottle	1.05708.1000
		10 kg	Bucket, plastic	1.05708.9010
Molecular sieve 0.4 nm beads, with moisture indicator ~ 2 mm	-	250 g	Glass bottle	1.05739.0250
		1 kg	Glass bottle	1.05739.1000
Molecular sieve 0.4 nm rods ~ 1.6 mm (1/16")	1318-02-1	1 kg	Plastic bottle	1.05743.1000
Molecular sieve 0.5 nm beads ~ 2 mm	1318-02-1	250 g	Glass bottle	1.05705.0250
		1 kg	Glass bottle	1.05705.1000
Molecular sieve 1.0 nm beads ~ 2 mm	1318-02-1	1 kg	Glass bottle	1.05703.1000
Application advice Molecular sieves are suitable for drying alm	ost all gases and liquids			

▶ 1) Molecular sieves with 0.3 nm bead form (105704) and with indicator brown gel (105734) are suitable for use in Karl Fischer titrators.

Sodalime		CAS No.	Content	Packaging	Ord. No.
Sodalime, granules appr. 1	Sodalime, granules appr. 1 – 2.5 mm with indicator for analysis		500 g	Plastic bottle	1.06733.0501
			2.5 kg	Plastic bottle	1.06733.2500
Sodalime pellets with indic	ator for analysis	-	1	Plastic bottle	1.06839.1000
			5 I	Plastic bottle	1.06839.5000
			25 I	Fibre carton	1.06839.9026
Application advice	Soda lime is used for the absorption of $CO_2$ from air and gases. Soda lime has a large internal absorption surface area. The absorption capacity is at least 25 %. In laboratories and technical facilities, soda lime can be used for both small and large absorption systems.				
Additional information	Soda lime contains an indicator which changes from beige to violet, although this may change back from violet to blue in periodic use or after a period of shutdown. This is the result of a back or balance reaction, in which hydroxide groups which have become active again are available on the surface. No $CO_2$ is released during this. Even if the indicator turns beige again, the soda lime does not have more absorption capacity than at the beginning. The $CO_2$ bonding which has already taken place reduces the remaining available capacity.				

### **More information**

▶ »Volumetric solutions« see page 132 ▶ »Karl Fischer reagents« see page 146 ▶ »Drying agents« see page 250

# Ordering information Filtration

Aluminium oxide fibers		CAS No.	Content	Packaging	Ord. No.
Aluminium oxide fibers for gooch crucibles		-	100 g	Plastic bottle	1.15754.0100
			1 kg	Plastic bottle	1.15754.1000
Application advice	Aluminium oxide fibers are used as filtration and insulation agents as a result of their high melting point				
	of 1700 to 1800°C. In contrast to asbestos fibers, they are not carcinogenic and can be easily disposed of.				
Information on	Aluminium oxide fibers from Merck Millipore are non-carcinogenic and can be easily disposed of.				
sustainable protection					

Glass wool	CAS No.	Content	Packaging	Ord. No.	
Glass wool	65997-17-3	250 g	Metal can	1.04086.0250	
		1 kg	Fibre carton	1.04086.1000	
Application advice	Due to its chemical resistance even against strong acids, except hydrofluoric acid, glass wool is used as a filter				
	material in analytical processes, and as a supporting substance for drying agents in drying towers, for example to prevent caking in phosphorus pentoxide. It can also be used for insulation up to around 500°C.				
Additional information	Glass wool is made of soda lime silicic acid glass fibers with a diameter of an	ound 15 µm.			



Sea sand		CAS No.	Content	Packaging	Ord. No.
Sea sand extra pure		7631-86-9	1 kg	Plastic bottle	1.07711.1000
			5 kg	Plastic bottle	1.07711.5000
			25 kg	Fibre carton	1.07711.9025
Sea sand purified by acid a	7631-86-9	1 kg	Plastic bottle	1.07712.1000	
			5 kg	Plastic bottle	1.07712.5000
			10 kg	Plastic bottle	1.07712.9010
			25 kg	Fibre carton	1.07712.9025
Application advice	Sea sand is used e.g in laboratories for filtering and clarifying extremely contaminated water and solutions. In addition, sea sand is used as a grinding aid and heating bath (sand bath). To remove contamination which can be dissolved in acid, sea sand is boiled in hydrochloric acid and ignited for the carbonization of organic components to achieve the quality level necessary for laboratory use.				
Additional information	The grain size of sea sand is 0.1 to 0.3 mm.				
Information on sustainable protection	Sea sand is a natural product which consists predominantly of silicon oxide and small quantities of other metal oxides.				

# Absorbents for spilled liquids Chemizorb<sup>®</sup>

Mishaps and accidents happen. With Chemizorb<sup>®</sup>, however, you can remove spilled aggressive or other unpleasant liquids quickly and safely. Chemizorb<sup>®</sup> consists of porous mineral or synthetic copolymers that are chemically inert and, depending on the type, are capable of taking up 100 to 400 percent of their own weight in liquid material.



# Absorbents for spilled liquids

# Contents

• The »allrounders« Chemizorb <sup>®</sup> powder and granules	246
• The »all-in-one« Chemizorb <sup>®</sup> mercury set	247
• The »specialists« Chemizorb <sup>®</sup> alkalis, acid, hydrofluoric acid	248
Handling and disposal	249

# Advantages

With Chemizorb<sup>®</sup> you can remove spilled liquids:

- Easily
- Quickly
- Safely
- Environment-friendly



www.merckmillipore.com/chemizorb

# The »allrounders«

# Chemizorb® powder

- Is insoluble in water and in all other media that are liquid at room temperature.
- The powder is characterized by an extremely high absorbance capacity.
- Due to its large surface area, it is capable of absorbing two to four times its own weight in aqueous solutions and one to two times its weight in organic solvents and viscous oils.

# Ordering information

Chemizorb <sup>®</sup> powder	Content	Packaging	Ord. No.
Chemizorb® powder absorbent for spilled liquids	500 g	PE bottle	1.02051.0500
	25 kg	Fibre carton	1.02051.9025

### **Chemizorb®** granules

Chemizorb<sup>®</sup> granules are insoluble in water and in all other media that are liquid at room temperature. The granules possess a slightly lower absorbance capacity than the powder: due to its rough granular form, it can absorb only up to 100 percent of its own weight; on the other hand, however, it is much easier to dose. The granule form is used in all those places where it is not suitable to use the powder form, for example in draughty rooms or outside. Just like the powder, the granules are suited for absorbing alkalis, acids, or also heating-bath liquids.



Chemizorb\*

# Ordering information

Chemizorb <sup>®</sup> granules	Content	Packaging	Ord. No.
Chemizorb® granules absorbent for spilled liquids	1 kg	PE bottle	1.01568.1000
	5 kg	Bucket, plastic	1.01568.5000
	20 kg	Paper sack	1.01568.9020
	20 kg	PE drum	1.01568.9021



# The »all-in-one« set for mercury

### **Chemizorb® Mercury**

Chemizorb<sup>®</sup> Mercury is an all-inclusive set of reagents and auxiliaries for the safe and complete removal of drops of mercury and of traces of elementary mercury. The reagents included in the set are sufficient for the safe and simple decontamination of an area of roughly one square meter.



- Any droplets of mercury are first suctioned off using the pipette.
- The contents of the pipette are then emptied into the mercury bottle.
- Any remaining mercury is then strewed with a layer of reagent 1.
- The covered area is then sprayed with reagent 2.
- After a reaction time of 15 to 30 minutes the mercury-containing absorbent is removed from the surface and put into a small tub using a small shovel and a spatula.
- Any remaining material can subsequently be simply cleaned away with a wipe.
- Afterwards all tools and working materials are stored safely in the large can.
- The mercury-containing waste material must be disposed of as special waste in accordance with the official regulations.

# Ordering information

Chemizorb® Hg	Content	Packaging	Ord. No.
Chemizorb® Hg Reagents and accessories	1 set	PE case	1.12576.0001
for absorbent for mercury			
1 set consisting of: 500 g of reagent 1, 100 ml of reagent 2, one small tub, one large disposal can, protective gloves,			
Chemizorb® Hg reagents refill pack	1 set	PE can	1.01569.0001
for Ord. No. 1.12576.0001			
1 set consisting of: 500 g reagent 1 and 100 ml reagent 2			

# The »specialists«

The absorbents for alkalis, acids, and hydrofluoric acid contain, in addition to the carrier material itself, also water-soluble neutralizers and pH indicators. The admixed pH indicators make it possible to keep track of the process of neutralization of the spilled acid or alkalis. It should be borne in mind that the neutralization reaction may involve the generation of heat and gas.

### **Chemizorb®** Alkalis

- Is a powder mixture that consists of a mineral copolymer as the absorbent, an acidic salt as the neutralizer, and a pH indicator.
- Upon contact with the alkalis the absorbent first turns blue in colour.
- After neutralization is complete, the blue colour becomes paler.
- About 100 to 120 g of Chemizorb® OH- is required to absorb and neutralize 100 ml of alkalis.



# Ordering information

Chemizorb® OH-	Content	Packaging	Ord. No.
Chemizorb® $OH^{-}$ absorbent and neutralizer for spilled alkalis, with indicator	1 kg	PE bottle	1.01596.1000

### **Chemizorb®** Acid

- NEW: Improved composition. Now easier to use and with higher absorption capacity.
- Powder mixture that consists of a mineral copolymer as the absorbent, an alkaline neutralizer, and a pH indicator. Suitable for all acid spills (except Hydrofluoric acid)
- The neutralization process results in the release of CO<sub>2</sub>. When the acid has been neutralized, the colour of the indicator changes from red back to yellow.
- About 100 g of Chemizorb® H<sup>+</sup> is required to to absorb and neutralize 100 ml of acid.



# Ordering information

Chemizorb® H*	Content	Packaging	Ord. No.
Chemizorb® H <sup>+</sup> absorbent and neutralizer for spilled acids, with indicator	500 g	PE bottle	1.02491.0500
	2.5 kg	Bucket, plastic	1.02491.2500

# <section-header>

# Chemizorb® Hydrofluoric Acid

- Is a powder mixture consisting of a synthetic copolymer as the absorbent, calcium salt as a neutralizer and precipitant, and a pH indicator.
- The neutralization process results in the release of CO<sub>2</sub>. When the hydrofluoric acid has been neutralized, the colour of the indicator changes from red to orange/yellow.
- About 150 g of Chemizorb<sup>®</sup> HF is required to absorb and neutralize 100 ml of hydrofluoric acid.

# Ordering information

Chemizorb® HF	Content Packaging	Ord. No.
Chemizorb® HF absorbent and neutralizer for spilled hydrofluoric acid, with indicator	1 kg PE bottle	1.01591.1000

# Handling and disposal

Handling

**lling** All Chemizorb<sup>®</sup> products are handled in much the same way:

- Spilled liquids are covered with a sufficient amount of absorbent and, wherever necessary, mixed using a spatula, spoon, or small shovel.
- Wait until the neutralization and absorption processes are complete.
- After the absorbent has been collected, the contaminated surface is cleaned thoroughly with plenty of water.
- A specific set of instructions for use is given on the label of each respective product package.
- **Disposal** The contaminated material Chemizorb<sup>®</sup> and absorbed chemicals is collected in a polyethylene bag and forwarded for disposal in accordance with the company regulations and national guidelines for the hazardous products in question.

# Dry and safe

# Drying agents from Merck Millipore

Merck Millipore drying agents help protect your valuable goods! Products and goods must often be protected against moisture and mould formation, both on long transport routes as well as during their storage. Merck Millipore offers a comprehensive selection of different drying agents for this purpose and many other applications in laboratories.



# Drying agents

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# Advantages

- Reliable: Merck Millipore drying agents help to minimize the effects of moisture on products. This maintains the original condition and prevents follow-up costs caused by any damage.
- Convenient: Merck Millipore drying agents are user-friendly and easy to handle. This prevents time being wasted.
- Economical: Protection using Merck Millipore drying agents increases the longevity of your products. This helps to reduce costs.



www.merckmillipore.com/drying-agents

# Safety and reliability



### Safety and environment

In the drying agents product group, too, Merck Millipore offers products which support the goal of sustainable environmental protection and safety. For example, silica gels with or without orange or brown gel indicators are offered as an alternative to silica gel with blue gel indicator, which is presumably carcinogenic.

### Safety information

When using drying agents, one must be aware of the potential dangers involved. Both acid and basic drying agents can be corrosive and magnesium perchlorate can explode, as can sodium and potassium on contact with certain organic substances resp. with water or chlorinated hydrocarbons. In the case of drying agents that develop hydrogen during the drying process, drying must be carried out in a well-ventilated fume chamber. Blue gel, due to the presence of cobalt chloride, can have a carcinogenic effect (R-phase 49 – may cause cancer by inhalation). Filling and emptying should thus always be carried out in a fume chamber.

### Drying rate

The intensity only indicates the theoretically achievable residual value for water; it may take a long time for equilibrium to be reached. Thus, if a high degree of efficiency is to be achieved, rapid water uptake is important.

The uptake rate is determined by the following steps:

- The H<sub>2</sub>O molecules must be able to leave the material to be dried and must traverse a path to the drying agent.
- The molecules must be able to diffuse into the reactive centers of the drying agent.

Whilst the user can influence the first two points with his experimental setup, the manufacturer of the drying agent must take the following parameters into account if the third point is to be optimized:

- Particle size,
- Pore size and pore distribution,
- Prevention of deactivation of the surface during the drying process.

The ideal drying agents are those where the above parameters do not significantly change during the water adsorption process, e.g. SICAPENT®, magnesium perchlorate, molecular sieves, silica gel, aluminium oxide and calcium hydride. However, many drying agents tend to clump during the water absorption process, disintegrate or form a syrupy layer over unused product. This is a disadvantage when working with gases in drying towers; they tend to become blocked or channels are formed through which the gas flows but in an incompletely dried state.

#### Capacity

The capacity of a drying agent is defined by the mass of water adsorbed per 100 g anhydrous substance. Example: 1 kg drying agent of capacity 20 % can adsorb 200 g of water. The residual water content of heavily loaded drying agent is higher than that of less loaded agent. On the other hand, drying agents are more heavily loaded by gases or liquids with higher water content. Exception: drying agents such as  $CuSO_4$  which form defined hydrates maintain a constant water vapor partial pressure until the next hydrate stage is formed, independent of the mass of water adsorbed.

#### Regeneration

Some drying agents can be regenerated. To do this, the drying agent is heated to restore its equilibrium. Due to the fact that water is absorbed rapidly, regenerated drying agents must be filled and stored well away from moisture.

### **Applications**

The user-friendly drying agents from Merck Millipore are suitable for a wide range of laboratory applications, for example for drying gases, liquids and solids. With such a wide variety of grain and packaging sizes, you are sure to find the suitable drying agent, either for the classic method of static or dynamic drying processes.

Products with no tendency towards clumping are particularly suited for the dynamic drying process including, for example, calcium hydride, magnesium perchlorate, aluminium oxide, silica gel, or molecular sieves.

More information about sustainable protection www.merckmillipore.com/protection

### Safety and environment – characteristics

▶ For easy detection, safety and/or sustainable characteristics of our products are highlighted with this symbol.

## Drying methods

#### **Drying methods**

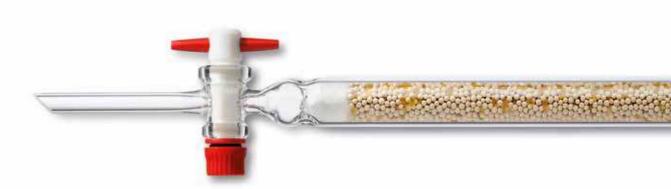
Non-sensitive solids can be dried at higher temperatures in a drying cabinet. However, drying at room temperature in a desiccator or at higher temperatures using a drying pistol is more gentle. Application of a vacuum facilitates the diffusion of the water molecules from the solid to the drying agent; the drying rate is hence somewhat faster.

### Static drying

In the classical drying of liquids, the drying agent is added, the whole allowed to stand, stirred (e.g. with a magnetic stirrer), shaken or boiled under reflux (details can be found in relevant textbooks of organic chemistry). It is important that the liquid is moved in such a way that it comes into contact with the drying agent. The liquid is then filtered or decanted. Should compounds be formed due to reaction with the water, these must be subsequently removed by distillation.

The frequently used drying agents calcium chloride, potassium carbonate, sodium sulfate and calcium sulfate have a medium drying effect only on solvents when used statically. Drying agents such as sodium or the earthalkaline oxides, however, are not as efficient as often thought due to their reactive surfaces being relatively small and in addition covered by a coating that hinders access of water molecules. In addition, as laboratory accidents are relatively frequent with these materials, they should not be used for this purpose.





#### **Dynamic drying**

In order to increase the drying rate and to achieve better utilization of the drying agent, liquids and gases can be passed through drying towers or drying tubes filled with a drying agent. However, if diffusion and flow rate are not to be hindered, the drying agents used should not be susceptible to clumping or deliquescence. For this reason, drying agents such as calcium hydride, magnesium perchlorate, aluminium oxide, silica gel and molecular sieves are particularly suitable. Untreated phosphorus pentoxide tends to clump when in contact with water and is thus normally unsuitable for dynamic drying. SICAPENT<sup>®</sup>, however, is a drying agent where  $P_2O_5$  has been coupled to an inert carrier; it remains flowable also when loaded 100 % and allows gases to flow through without resistance.

The drying process can be optimized by using a drying agent of small particle size. In this way, the surface area can be significantly increased and hence the column length and packing decreased. However, it should be taken into account that the flow rate is reduced due to the greater flow resistance in the column. The diagram shows a drying process for gases using silica gel in a drying column: An orange gel turns colorless when loaded with water. The moist gas enters the column at the left hand side with water content  $C_{4}$  and leaves it on the right in a dry condition  $C_{E}$ ; however, at this point, the gas contains more than the minimum residual water achievable with the drying agent in question. The drying agent in the left hand part of the column is already loaded to the maximum with water and is in equilibrium with the moist gas entering. The actual drying process - the transfer of water from gas to silica gel takes place in the segment known as the »Mass Transfer Zone – MTZ«. Over the drying period, the MTZ migrates towards the right hand side of the column (steps 2, 3, 4) until it reaches the end and the moist gas leaves. In order to avoid the gas leaving, the gas flow is interrupted well in time; this has the effect that a small part of the column remains unutilized. However, such dynamic drying procedures are mostly better than static ones. (This is shown in the general calculation on the next page.)



Drying process for gases using silica gel in a drying column.

## Calculations

**General calculation** of relative humidity of the atmosphere: The absorptivity of the atmosphere for humidity increases with the temperature until saturation. 1 m<sup>3</sup> air at 11°C is saturated with 10.0 g water, at 20°C with 17.3 g, at 30°C with 30.4 g and at 40°C with 51.2 g.

**Calculation** of the amount of drying agent required: 1000 l gas containing 10 mg/l water are to be dried at 25°C to a residual water content of 1 mg  $H_2O/l$ . 1000 l x 10 mg  $H_2O/l - 1000$  l x 1 mg  $H_2O/l = 9$  g  $H_2O$  are to be adsorbed.

**Calculation** of the required amount of drying agent for static drying: At the end of the drying process, the residual water content of the gas is in equilibrium with the drying agent. The loading of the silica gel necessary to achieve the desired residual water content can be taken from the table in the ordering information of silica gel, page 272: 1 mg H<sub>2</sub>O/I residual water  $\triangleq$  loading of 5.2 g H<sub>2</sub>O / 100 g silica gel. To absorb 9 g H<sub>2</sub>O, 9/5.2 x 100 g = about 200 g silica gel are required.

**Calculation** of the required amount of drying agent for dynamic drying: In this case, the greater part of the drying agent is in equilibrium with the water content of 10 mg/l of the gas flowing into the column. Thus, a higher loading – about 20 g  $H_2O$  / 100 g silica gel – is possible than in the case of static drying where the entire drying agent is in equilibrium with the low residual water content. Even if in the case of dynamic drying half of the drying agent remains unutilized, 100 g are sufficient compared with 200 g for static drying.

As the flowing gas has much less contact with the drying agent than with the static method, the much lower values for residual water content as cited in the literature for static drying are not quite achieved. If such low residual water content is to be achieved, it is necessary to connect a further column with a more effective drying agent. If the gas is circulated over a drying column in a closed room, even if dynamic, only the capacity of a static method can of course be achieved.

Calculationof the column diameter: Based on the flow rate and the given volume flow (volume/time unit), the smallest<br/>allowable column cross-section can be calculated.<br/>Example: 3.6 l of 2-propanol per hour are to be dried (= 3600 ml / 60 min).<br/>At a flow rate of 10 cm/min\* the minimum cross-sectional area is 6 cm² corresponding to approx. 30 mm diameter.

## **Definitions**

## Column dimensions

In order to be able to utilize the drying agent to the full, the Mass Transfer Zone [MTZ] and the length of non-utilized column must be kept to a minimum.

Narrow columns have proved to be of advantage in this case:

- For gases, a ratio for length to diameter of greater than 5 is recommended. Columns filled with beads or granular silica gel should be at least 1 m long.
- For liquids, columns of 60 cm in length and 2 3 cm in diameter to 2 m and 6 cm respectively are recommended (for further details, see »drying of solvents«).

To determine the necessary column volume, the required amount of drying agent should be divided by the bulk density. Example: 100 g silica gel of bulk density of 70 g / 100 ml have a volume of 143 ml.

- Flow rate However, the ratio length to cross-section should not be so large that high flow rates result as this would lengthen the MTZ considerably. Recommended flow rates (bases on the free cross-section of the column) for gases: 5 15 m per minute, for liquids: 2.5 30 cm per minute. These values have been established experimentally as being optimal.
- **Drying gases** Gases should be dried using the dynamic method (see »drying methods«). Very moist gases should first be dried using a drying agent of high capacity: CaH<sub>2</sub>, CaSO<sub>4</sub>, Mg(ClO<sub>4</sub>)<sub>2</sub>, molecular sieve, H<sub>2</sub>SO<sub>4</sub>, or silica gel. Fine drying can then be attained using phosphorus pentoxide, SICAPENT<sup>®</sup>, CaH<sub>2</sub>, Mg(ClO<sub>4</sub>)<sub>2</sub> or molecular sieve. Further details are contained in the section describing the relevant drying agents.

## Drying agents

### for solvents with low water absorption capacity

Solvents with a low water-absorbing capacity can generally be dried using static methods; they should be allowed to stand in their reservoirs for up to several days with occasional shaking in contact with a suitable drying agent (e.g. 100 – 200 g molecular sieve (MS) per liter solvent).

The residual water content that can be attained with molecular sieves (MS) is less than  $10^{-4}$  percent by weight corresponding to 1 ppm = 1 mg H<sub>2</sub>O = approx. 0.05 mmol H<sub>2</sub>O per liter solvent. 250 g molecular sieve can dry more than 10 l hydrophobic solvent whilst becoming 14 – 18 % loaded with H<sub>2</sub>O. Of course, dynamic drying as described in textbooks can also be used.

When drying hydrophobic solvents dynamically with aluminium oxide, silica gel or molecular sieve, the flow rate should be up to 30 cm per minute. In this way, using a column of diameter 2.5 cm and 5 cm<sup>2</sup> cross-section, up to 6 l per hour can pass through. Columns of diameter 2.5 cm and a length of 60 cm containing some 200 g of molecular sieve have proven useful for such applications.

	Solvents	CaCl <sub>2</sub> – Calcium chloride	CaH <sub>2</sub> – Calcium hydride	CaO – Calcium oxide	Distillation	$K_2CO_3$ – Potassium carbonate	KOH – Potassium hydroxide	LiAlH <sub>4</sub> – Lithium aluminium hydride	Molecular sieve 0.4 nm	Molecular sieve 0.5 nm	Na – Sodium	Na <sub>2</sub> SO <sub>4</sub> – Sodium sulfate	P <sub>2</sub> O <sub>5</sub> – Phosphorus pentoxide
Α	n-Amyl acetate					•				•			
	n-Amyl alcohol								•				
	Aniline						•			٠			
	Anisole	•	٠		•					•	•		
В	Benzene	•	•		•					٠	•		
	Benzyl alcohol				•					٠			
	Bromobenzene	•			•					٠			
	Bromoform	•								٠			•
	tert-Butyl methyl ether		•					•		٠	•		
С	Carbon disulfide	•											•
	Carbon tetrachloride	•			•				•				•
	Chlorbenzene	•			•					٠			
	Chloroform	•							•				•
	Cyclohexane	•					•		•	•			
	Cyclopentane		•					•		٠	•		
D	n-Decane		•					•	•				
	1,2-Dichlorobenzene	•			•					•			
	Dichloromethane	•							•				
	Dichloroethane	•			•					٠			
	Diethyl ether	•	•					•	•		•		
	Diethyl ketone					•			•				
	Diethylene glycol dibutyl ether	•	•						•		•		
	Diisoamyl ether		•					•	•		•		
	Diisopropyl ether	•	•						•		•		
-	Dipropyl ether		•					•	•		•		
E H	Ethyl methyl ketone n-Heptane		-			٠		-	•				
П	n-Hexane		•					•	•		•		
1	Isoamyl alcohol		•	•		•		•	•	•			
1	Isobutyl methyl ketone	•		•		•				•			
	Isooctane	•	•			•		•		•	•		
N	Nitrobenzene	•	-					-		•	-		
	Nitropropane	•			•					•			
Р	n-Pentane		•					•	•				
1	Petroleum ether, petroleum, petroleum benzene	•					•	•		•			
т	Tetrachloroethylene				•	•				•		•	
	Toluene	•	•		•				•		•		
	1,1,1-Trichlorethane	•			•					•			
	Trichloroethylene				•	•				•		•	
	1,1,2-Trichlorotrifluoroethane		•						•				
x	Xylene	•	•		•				•		•		

## Drying agents

### for solvents with medium to unlimited water adsorption capacity

 $H_2O + drying agent \rightleftharpoons H_2O / drying agent / compound (1)$  $H_2O + solvent \rightleftharpoons H_2O solvated (2)$ Solvent + drying agent  $\rightleftharpoons$  Solvent / drying agent / compound (3)

Due to the competitive reactions (2) and (3), the attainable residual water contents are some 1000 times higher than in air – unless drying agents such as calcium hydride are used where no equilibrium exists due to one of the products (in this case  $H_2$ ) leaving the equation.

In general, residual water values of  $10^{-3}$  % by weight are adequate. Further drying is no longer meaningful, in particular if the dried solvent is refilled under air: even if poured quickly, the H<sub>2</sub>O content increases from  $1 \cdot 10^{-3}$  to  $2 - 4 \cdot 10^{-3}$  %. A further source of contamination with water is e.g. non-greased ground glass, e.g. in desiccators, through which significant amounts of water vapor can diffuse. Suitable drying agents are recommended in the listing below. As conventional drying with chemical agents is adequately described in textbooks of preparative organic chemistry, only dynamic drying with the help of water-miscible solvents and molecular sieves (MS) is described here.

The following values can be attained using this method: **Residual water content:** 0.001 – 0.005 % weight H<sub>2</sub>O in the solvent **Capacity:** at a desired residual water content of max. 0.001 %, the molecular sieve used may not be loaded greater than: Diethyl ether 14 g H<sub>2</sub>O / 100 g molecular sieve Ethyl acetate 6 g H<sub>2</sub>O / 100 g molecular sieve Dioxane 4 g H<sub>2</sub>O / 100 g molecular sieve Pyridine 2 g H<sub>2</sub>O / 100 g molecular sieve

Loading: depends on the reaction equation (2) of solvents

	Solvents A-M	<b>Water adsorption</b> [g H <sub>2</sub> O/100 g solvent]	Drying agent	Ca – Calcium	CaCl <sub>2</sub> – Calcium chloride	CaH <sub>2</sub> – Calcium hydride	CaO – Calcium oxide		Distillation	K <sub>2</sub> CO <sub>3</sub> – Potassium carbonate	KOH – Potassium hydroxide	Mg – Magnesium	MgO – Magnesium oxide	MgSO4 – Magnesium sulfate	Molecular sieve 0.3 nm	Molecular sieve 0.4 nm	Molecular sieve 0.5 nm	Na – Sodium	$Na_2SO_4$ – Sodium sulfate	P <sub>2</sub> O <sub>5</sub> – Phosphorus pentoxide
Α	Acetic acid	∞						•												
	Acetone	∞								•					٠					
	Acetonitrile	∞			•					•					•					•
	Acetylacetone	∞								•						•				
	tert-Amyl alcohol	14					•										•			
В	1-Butanol	20							•	•						•				
	2-Butanol	44							•	•							•			
	tert-Butanol	∞					•										٠			
	n-Butyl acetate	2.9												•		•				
С	Cyclohexanol	11					•										•			
	Cyclohexanone	8.7								•							•			
D	Diethylene glycol	∞							•							•			•	
	Diethylene glycol diethyl ether	∞			•	•											•	•		
	Diethylene glycol dimethyl ether	∞			•	•											•	•		
	Diethylene glycol monobutyl ether	∞			•	•											•	•		
	Diethylene glycol monoethyl ether	∞			•	•											•	•		
	Diethylene glycol monomethyl ether	∞			•	•											•	•		
	N,N-Diethylformamide	∞				•			•								•			
	N,N-Dimethylformamide	∞				•			•							•				
	Dimethyl sulfoxide	∞				•			•						•					
_	1,4-Dioxane	∞			•	•										•		•		
Е	Ethanol	∞					•					•	•		•					
	Ethanolamine	∞									•				•					
	(2-Ethoxyethyl)-acetate	6.5								•						•			•	•
	Ethyl acetate	9.8								•						•			•	•
	Ethylene glycol dimethyl ether	∞				•			•							•				
	Ethylene glycol	∞							•							•			•	
	Ethylene glycol monobutyl ether	∞							•											
	Ethylene glycol monoethyl ether	∞							•											
	Ethylene glycol monomethyl ether	∞							•							•				
-	Ethyl formiate Formamide	∞					-							•		•			•	
F		∞					•		•						•				•	
G	Glycerol	∞							•							•				
н	1,1,1,3,3,3-Hexafluoro-2-propanol Isobutanol	∞ 15		•			•			•						•				
M	Methanol	∞					•						•		•					
IVI	Methanol Methyl acetate	∞ 8					•			•			•		-	•				
	Methyl formiate	8 24					-			•						•			•	
	Methyl propyl ketone	3.6								•						•				
	Methyl pyridine	3.0 ∞									•					•				
	incurr printine																			

	Solvents N-Z	Water adsorption [g H <sub>s</sub> 0/100 g solvent]	Drying agent	Ca – Calcium	CaCl <sub>2</sub> – Calcium chloride	CaH <sub>2</sub> – Calcium hydride	CaO – Calcium oxide	CuSO <sub>4</sub> – Calcium sulfate	Distillation	K <sub>2</sub> CO <sub>3</sub> – Potassium carbonate	KOH – Potassium hydroxide	Mg – Magnesium	MgO – Magnesium oxide	MgSO <sub>4</sub> – Magnesium sulfate	Molecular sieve 0.3 nm	Molecular sieve 0.4 nm	Molecular sieve 0.5 nm	Na – Sodium	$Na_2SO_4$ – Sodium sulfate	P <sub>2</sub> O <sub>5</sub> – Phosphorus pentoxide
Ρ	1,2-Propanediol	∞					٠					٠	٠			٠				
	1,3-Propanediol	∞					•					•	•			•				
	1-Propanol	∞					•					•	•			•				
	2-Propanol	∞					•					•			•					
	Pyridine	∞					•					•	•		•					
Т	Tetraethylene glycol	∞									•					•				
	Tetrahydrofuran	~				٠					٠					٠				
	Triethanolamine	~									٠						•			
	Triethylene glycol	~							•							•			•	
	Triethylene glycol dimethyl ether	∞							•											

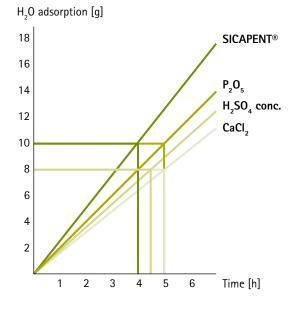
### Water absorption rate of some drying agents

Experimental: 100 g SICAPENT® or 75 g of other drying agents were placed in a vacuum desiccator alongside a dish of water. After 1 h the increase in weight of the drying agents were established using gravimetric analysis. The results obtained are shown in the figure.

### Examples of flow rate

The flow rate for water-miscible solvents should be less than 10 cm/minute. This corresponds to max. flow rates of:

Flow rate	Column diameters
50 ml/min	25 mm
70 ml/min	30 mm
200 ml/min	50 mm



Solvent	Initial water content [% by weight]	Residual water content [ % by weight]	Quantity of solvent dried [I]	Type [nm]
Acetonitrile	0.05 - 0.2	0.003	3 – 4	0.3
Benzene	0.07	0.003	>10	0.4
Chloroform	0.09	0.002	>10	0.4
Cyclohexane	0.009	0.002	>10	0.4
Dichloromethane	0.17	0.002	>10	0.4
Diethyl ether	0.12	0.001	10	0.4
Diisopropyl ether	0.03	0.003	10	0.4
Dimethylformamide	0.06 - 0.3	0.006	4 – 5	0.4
1,4-Dioxane	0.08 - 0.3	0.002	3 – 10	0.5
Ethanol	0.04	0.003	10	0.3
Ethyl acetate	0.015 - 0.2	0.004	8 - 10	0.4
Methanol	0.04	0.005	10	0.3
2-Propanol	0.07	0.006	7	0.3
Pyridine	0.03 - 0.3	0.004	2 – 10	0.4
Carbon tetrachloride	0.01	0.002	>10	0.4
Tetrahydrofuran	0.04 - 0.2	0.002	7 – 10	0.5
Toluene	0.05	0.003	>10	0.4
Xylene	0.045	0.002	>10	0.4

Dynamic drying of solvents with molecular sieves using a column of 25 x 600 mm (250 g molecular sieve) or of 50 x 2,000 mm (2 kg molecular sieve).

#### Amount of solvent dried

The amount of dry solvent obtainable for solvents that are readily miscible with water cannot be accurately given as this is dependent on the initial water content which is mostly unknown. However, if the solvent is dried statically to a low  $H_2O$  content (e.g. with approx. 100 g of molecular sieve, enough for 1 l of solvent), the subsequent dynamic process can be used to dry 10 l of the solvent to 0.001 - 0.002 % weight using 200 g of molecular sieve. For drying the strongly hygroscopic alcohols methanol, ethanol and 2-propanol to 0.002 % weight of residual water, however, some 2 kg of 0.3 nm molecular sieve is necessary. Column dimensions: Ø 50 mm, length 2 m. An overview of the attainable drying effect with a series of water-saturated solvents is given in the table above.

### **Practical procedure**

It should initially be checked whether, in addition to water, the solvent to be dried is adsorbed by the molecular sieve. To do this, place 10–20 beads in a test tube along with several ml of the solvent. Significant increase in temperature – in certain circumstances even boiling – indicates co-adsorption according to (3). If this is the case, either a molecular sieve of smaller pore size, where there is no co-adsorption, should be used or the flow rate should be reduced to max. 2.5 cm per minute. The appropriate pore sizes where no further co-adsorption takes place are given in the table.

Initially the solvent should be applied to the column slowly until the entire column has been wetted within 15–30 minutes. As a rule, the first fraction collected contains an increased water content; this should either be discarded or re-applied to the column. In the case of fresh molecular sieve, the first fraction may contain some particles and be somewhat turbid; this fraction should either be disposed of or filtered.

## Ordering information Drying agents A-C



Calcium [Ca]		CAS No.	Content	Packaging	Ord. No.			
Calcium granular particle s	ize about 2 – 6 mm	7440-70-2	100 g	Glass bottle	1.02053.0100			
			500 g	Glass bottle	1.02053.0500			
For drying	Alcohols							
Application	During the drying process, insoluble metal hydr	oxide is initially formed follo	wed by meta	l alcoholate,				
which is soluble in alcohol. Hence, subsequent to drying, the solution must be distilled.								
Capacity Stoichiometric								

Calcium chloride [CaCl <sub>2</sub> ]		CAS No.	Content	Packaging	Ord. No.
Calcium chloride anhydro	us powder Reag. Ph Eur	10043-52-4	500 g	Plastic bottle	1.02378.0500
			2.5 kg	Plastic bottle	1.02378.2500
Calcium chloride anhydro	us, granular ~ 1 – 2 mm	10043-52-4	1 kg	Plastic bottle	1.02379.1000
			5 kg	Plastic bottle	1.02379.5000
Calcium chloride anhydro	us, granular ~ 2 – 6 mm	10043-52-4	1 kg	Plastic bottle	1.02391.1000
			5 kg	Fibre carton	1.02391.5000
			25 kg	Fibre carton	1.02391.9025
Calcium chloride anhydro	us, granular ~ 6 – 14 mm	10043-52-4	1 kg	Plastic bottle	1.02392.1000
			5 kg	Fibre carton	1.02392.5000
			25 kg	Fibre carton	1.02392.9025
For drying	Acetone, ethers, numerous esters, aliphati	c, olefinic, aromatic and halogen	ated hydrocarl	oons, neutral gases	5.
Unsuitable for drying	Alcohols, ammonia, amines, aldehydes, ph	enols, several esters and ketones:	these compo	unds are bound	
	by CaCl <sub>2</sub> .				
Application	Drying of liquids, filling drying tubes; not	suitable for the drying of fast-flo	wing gases as	pore diffusion is	
	hindered due to deliquescence during wat	er uptake.			
Residual water content	65 % H <sub>2</sub> 0				
in air	content				
Capacity	98 %				
Regeneration	At 250°C in a drying oven				

Calcium hydride [CaH <sub>2</sub> ]		CAS No.	Content	Packaging	Ord. No.		
Calcium hydride for synthe	sis, ~ 1–10 mm	7789-78-8	100 g	Glass bottle	8.02100.0100		
			500 g	Glass bottle	8.02100.0500		
For drying	Gases, organic solvents, including ketones a	nd esters.					
Unsuitable for drying	Compounds with active hydrogen, ammonia	i, alcohols.					
NB							
Application	As calcium hydride is a very effective drying	g agent and reacts vigorously wi	th water, the	substances to be			
	dried should contain only low amounts of w	vater. In reaction with water, hyd	lrogen is relea	ased (always work i	in		
	a fume hood!) according to the equation Ca	$H_2 + H_20 ->2 H_2 + Ca0.$					
	The fine voluminous powder formed may bl	ock drying towers. CaH <sub>2</sub> is super	ior to sodium	as a drying agent	as		
	it possesses a much larger surface area. The	CaO formed does not adhere to	the $CaH_2$ sur	face and itself acts	as		
	a drying agent. CaO + $H_2O$ ->Ca(OH) <sub>2</sub> .						
Disadvantage	Due to the higher activity and reactivity that	an Na, CaH $_2$ is less stable if store	d incorrectly.	Hence, once the			
	package has been opened, it should be store	ed in a desiccator.					
Residual water content <0.00001 mg H <sub>2</sub> 0/I							
in air							
Capacity	Stoichiometric						

Calcium oxide [CaO]		CAS No.	Content	Packaging	Ord. No.
Calcium oxide from marble	small lumps ~ 3 – 20 mm	1305-78-8	1 kg	Plastic bottle	1.02109.1000
			25 kg	Fibre carton	1.02109.9025
For drying	Neutral and basic gases, amines, alcohols, ethers.				
Unsuitable for drying	Acids, acid derivatives, aldehydes, ketones, esters.				
Residual water content	0.003 mg H <sub>2</sub> 0/I				
in air					
Capacity	Limited as the surface is coated with a less perme	eable layer, especially in the	e presence of	CO <sub>2</sub> .	

Copper sulfate [CuSO₄]		CAS No.	Content	Packaging	Ord. No.
Copper(II) sulfate anhydrou:	s for analysis EMSURE®	7758-98-7	250 g	Plastic bottle	1.02791.0250
			1 kg	Plastic bottle	1.02791.1000
For drying	Low fatty acids, alcohols, esters.				
Unsuitable for drying	Amines, nitriles, ammonia.				
Residual water content	1.4 mg H <sub>2</sub> 0/I				
in air					
Regeneration	Above 50°C under vacuum.				
Advantage	Can be used as indicator: Colorless anhydrou	us copper(II)sulfate becomes b	olue as copper(l	l)sulfate 5-hydrate.	

## Ordering information Drying agents D-M

Desiccant sachets [SiO <sub>2</sub> ]		CAS No.	Content	Packaging	Ord. No.						
	ca gel with humidity indicator (orange gel)	-	50 units	Metal can	1.03804.0001						
sachet: 7 x 9 cm	ca ger with humany maleator (orange ger)		50 units	Wietar can	1.03004.0001						
	lica gel with humidity indicator (orange gel)		10 units	Metal can	1.03805.0001						
sachet: 15 x 14 cm	inca ger with humanly malcator (orange ger)	-	TO UNITS	Wictar can	1.03003.0001						
	lica gel with humidity indicator (orange gel)	_	10 units	Metal can	1.03806.0001						
sachet: 15 x 20.5 cm	inca ger with humanly indicator (orange ger)	TO UNITS	wictar can	1.03000.0001							
	a gel with humidity indicator (orange gel)		100 units	Metal can	1.03803.0001						
sachet: 4 x 7 cm	a ger with humany material (orange ger)		1000 units	Fibre carton	1.03803.0002						
	Further desiccant sachets, e.g. 500 g, on request.										
For drying	Humidity										
	,										
Application	Sachets filled with silica gel protect valuable a	•									
	along with sensitive machine components and										
	Sachets maintain the function of sensitive opt		•								
Capacity	Silica gel has a high adsorptive capacity for m	oisture: 20 % of its own weig	ht at 25°C an	d 80 % relative							
	humidity.										
Indicator change	At approx. 7–10 g adsorbed $H_2O$ / 100 g silica	gel, the color change is from	orange to col	orless.							
in orange gel											
Regeneration	Silica gel (orange gel) can be regenerated in a	drying oven at 130 – 140°C.	Desiccant sac	het only up to							
	80°C, because the adhesive of the bag can me	lt.									



Lithium aluminium hydride	[Li(AIH₄)]	CAS No.	Content	Packaging	Ord. No.			
Lithium aluminium hydride	- powder, for synthesis	16853-85-3	25 g	Metal can	8.18875.0025			
Lithium aluminium hydride	- tablets, for synthesis	16853-85-3	25 g	Metal can	8.18877.0025			
For drying	Hydrocarbons, ethers.							
Unsuitable for drying	Acids, acid derivatives (chlorides, anhyd	rides, amides, nitriles), aromatic nitro	o compounds.					
Application	Li(AIH₄) reacts vigorously, on occasion e	xplosively, with water whilst releasir	ig hydrogen.					
Hence, the solvents to be dried should have a very low initial water content.								
Capacity	Capacity Stoichiometric							

Magnesium [Mg]		CAS No.	Content	Packaging	Ord. No.	
Magnesium, turnings acc. t	to Grignard for synthesis	7439-95-4	250 g	Metal can	8.05817.0250	
			1 kg	Metal can	8.05817.1000	
Magnesium powder particle size about 0.06 - 0.3 mm7439-95-4			1 kg	Metal can	1.05815.1000	
For drying	Alcohols					
Application	Magnesium turnings must be activated with	Magnesium turnings must be activated with iodine prior to use. During the drying process insoluble metal				
	hydroxide is initially produced, followed by metal alcoholate, which is soluble in alcohol. Thus after drying,					
	distillation is necessary.					
Capacity	Stoichiometric					

Magnesium oxide [MgO]		CAS No.	Content	Packaging	Ord. No.
Magnesium oxide for analysis		1309-48-4	100 g	Plastic bottle	1.05865.0100
			500 g	Plastic bottle	1.05865.0500
For drying	Alcohols, hydrocarbons, basic liquids.				
Unsuitable for drying	Acid compounds.				
Residual water content	0.008 mg H <sub>2</sub> 0/I				
in air					
Regeneration	At 800°C				

Magnesium perchlorate [M	g(ClO₄)₂]	CAS No.	Content	Packaging	Ord. No.	
Magnesium perchlorate hyd	lrate [about 83 % Mg(ClO₄)₂], desiccant,	64010-42-0	500 g	Metal can	1.05873.0500	
about 1 – 4 mm						
For drying	Inert gases, air; adsorbs ammonia as strongly as water.					
Unsuitable for drying	Numerous solvents in which it is soluble, e.g.	Numerous solvents in which it is soluble, e.g. acetone, dimethyl formamide, dimethyl sulfoxide, ethanol,				
	methanol, pyridine, organic compounds.	methanol, pyridine, organic compounds.				
Application	In drying towers for the drying of rapid flowing gases; with increasing $ m H_2O$ loading the packing becomes					
	looser. $Mg(ClO_4)_2$ can be removed easily as it	does not stick to the walls.				
Residual water content	0.0005 mg H <sub>2</sub> 0/I to 10 % H <sub>2</sub> 0 content   0.002	mg H <sub>2</sub> 0/I to 32 % H <sub>2</sub> 0 content				
in air						
Capacity	48 %, corresponding to 6 moles crystal wate	r.				
Safety information	Explosion risk when in contact with a reducing	ng atmosphere, in particular in	the presence	of acids		
	or compounds that can be hydrolyzed to form acids. Mg(ClO <sub>4</sub> ) <sub>2</sub> may only be heated in vessels made of					
	inorganic materials.					
Regeneration	At 240°C under vacuum.					

Magnesium sulfate [MgSO₄]		CAS No.	Content	Packaging	Ord. No.	
Magnesium sulfate anhydrous for analysis EMSURE®		7487-88-9	1 kg	Glass bottle	1.06067.1000	
			25 kg	Plastic drum	1.06067.9025	
For drying	Almost all compounds including acids, acid de	Almost all compounds including acids, acid derivatives, aldehydes, esters, nitriles and ketones.				
Residual water content	1.0 mg H <sub>2</sub> 0/I					
in air						
Regeneration	At 200°C in a drying oven.					

## Ordering information Drying agents M

Molecular sieves	CAS No.	Content	Packaging	Ord. No.
Molecular sieve 0.3 nm beads ~ 2 mm <sup>1)</sup>	1318-02-1	250 g	Plastic bottle	1.05704.0250
		1 kg	Plastic bottle	1.05704.1000
		10 kg	Bucket, plastic	1.05704.9010
Molecular sieve 0.3 nm beads, with moisture indicator ~ 2 mm <sup>1)</sup>	-	250 g	Plastic bottle	1.05734.0250
		1 kg	Plastic bottle	1.05734.1000
Molecular sieve 0.3 nm rods ~ 1.6 mm (1/16")	1318-02-1	250 g	Plastic bottle	1.05741.0250
		1 kg	Plastic bottle	1.05741.1000
		10 kg	Bucket, plastic	1.05741.9010
Molecular sieve 0.4 nm beads ~ 2 mm Reag. Ph Eur	1318-02-1	250 g	Glass bottle	1.05708.0250
		1 kg	Glass bottle	1.05708.1000
		10 kg	Bucket, plastic	1.05708.9010
Molecular sieve 0.4 nm beads, with moisture indicator ~ 2 mm	-	250 g	Glass bottle	1.05739.0250
		1 kg	Glass bottle	1.05739.1000
Molecular sieve 0.4 nm rods ~ 1.6 mm (1/16")	1318-02-1	1 kg	Plastic bottle	1.05743.1000
Molecular sieve 0.5 nm beads ~ 2 mm	1318-02-1	250 g	Glass bottle	1.05705.0250
		1 kg	Glass bottle	1.05705.1000
Molecular sieve 1.0 nm beads ~ 2 mm	1318-02-1	1 kg	Glass bottle	1.05703.1000

Molecular sieves are suitable for drying practically all gases and liquids. They can be used in desiccators, drying tubes, for keeping absolute solvents dry, filling columns for drying gases or solvents and for selective adsorption. (e.g. phosgene from chloroform).

#### **Advantages**

- Easy-to-use: Practically chemically inert, non-toxic, no disposal problems, dried liquids can be decanted.
- High adsorption capacity even with low water content of the substance to be dried.
- High adsorption capacity even at high temperatures.
- High adsorption affinity for polar and unsaturated organic molecules; however, H<sub>2</sub>O is always preferentially adsorbed.
- Selective adsorption: only molecules that can pass through the pores are adsorbed.







Molecular sieves – contin	ued						
Temperature	Molecular sieves abs	sorb H20 whilst essentially	maintaining their capacity at temperatures w	here both			
	aluminium oxide and silica gel begin to release water. Between 0 and 150°C, the capacity decreases						
	gradually from 23 to	o 7 % with a residual water	content of 10 mg $H_2O/I$ .				
Residual water content	Min. 0.0001 mg H <sub>2</sub> 0	Min. 0.0001 mg $H_20/l$ at 25°C. The less loaded a molecular sieve is the more intensively it dries.					
in air	The supplied origina	l packed molecular sieve co	ntains approx. 1 – 2 % water. This tends not t	to interfere			
	with the drying proc	ess. However, if the require	ments of the drying process are high, the sub	stance			
	must be activated as	s described under »regenera	tion«.				
Typical values for	Loading [g H <sub>2</sub> O/100	) g molecular sieve]	Residual water content [mg H <sub>2</sub> O/l]				
molecular sieve 0.4 nm	1		0.0001				
	3		0.001				
	6		0.01				
	15		0.1				
	20		0.5				
Capacity	15 – 24 % at 25°C. I	If low residual water conter	nt is to be attained, the capacity can only be p	partially utilized			
	(see table above): At	a desired residual water co	ontent of 0.01 mg $H_2O/I$ , the loading may not	exceed			
	6 g H <sub>2</sub> 0 / 100 g mol	ecular sieve.					
Indicator	The indicator (brown	n gel) changes from brown	to yellowish at a $H_2O$ uptake of approximately	,			
	7 – 10 g / 100 g mo	lecular sieve.					
Regeneration	This can be carried o	This can be carried out as often as required; the max. regeneration temperature is 450°C. Molecular sieves					
	can be dried in a drying oven above 250°C to a water content of 2 – 3 g / 100 g. The remaining water can be						
	removed at 300 – 350°C using a vacuum oil pump (10-1-10-3 mbar), whereby, as is usual, a cold trap						
	containing carbon dioxide coolant or liquid air should be connected. Water pumps, due to their high partial						
	water vapor pressure, are completely unsuitable for this purpose. For safety reasons, molecular sieves						
	that have been used to dry solvents should be freed from possible solvent by mixing with water prior to						
	regeneration. Molec	ular sieves with moisture in	dicator should not be heated above 150°C.				
Chemical and physical	Molecular sieves are	crystalline, synthetic zeoli	tes. Their crystal gratings are similar to a cage	e with			
properties	numerous hollow sp	aces. The cavities are acces	sible from all sides by pores of exactly defined	d dimensions:			
	depending on the ty	depending on the type of molecular sieve, these can be 0.3, 0.4, 0.5 or 1.0 nm in diameter. If, due to					
	heating, the water in	n the hollow spaces is remo	ved, the material becomes an extremely activ	ve adsorbent.			
	However, only those	molecules are adsorbed the	at are small enough to pass through the pores	s (sieve effect).			
	Pore diameter	Туре	Composition	Structure			
	0.3 nm	ЗA	Potassium sodium aluminium silicate	Zeolite			
	0.4 nm	4A	Sodium aluminium silicate	Zeolite			
	0.5 nm	5A	Sodium and calcium aluminium silicate	Zeolite			
	1.0 nm	13A/X	Sodium aluminium silicate	Zeolite			
Physical properties	The molecular sieve	crystallites obtained by hyd	rothermal manufacture are formed into rods a	and beads using			
	1 – 2 % clay as a binding agent. Vibration caused by transport may bring about some abrasion which collects in						
	the first fraction dur	ing dynamic drying.					
	Bulk density		0.75 kg/l				
	Surface (BET)		800 m²/g				
	Form supplied		Powder, beads (~ 2 mm), rods (~ 1.6 mm, ~	~ 3.2 mm)			
	Effective pore diam	eter depending on type	0.3, 0.4, 0.5 or 1.0 nm				
	Hollow space volum	10	0.3 cm³/g				
	Specific heat		>0.8 KJ/kg				

## Ordering information Drying agents P-S

Phosphorus pentoxide [P20	<b>〕</b>	CAS No.	Content	Packaging	Ord. No.	
di-Phosphorus pentoxide e	extra pure	1314-56-3	1 kg	Glass bottle	1.00540.1000	
			25 kg	Plastic drum	1.00540.9025	
di–Phosphorus pentoxide f	or analysis ACS, ISO, Reag. Ph Eur	1314-56-3	100 g	Glass bottle	1.00570.0100	
			500 g	Glass bottle	1.00570.0500	
For drying	Neutral and acid gases, saturated alipathic and aromatic hydrocarbons, nitriles, alkyl and aryl halogenides and					
	carbon disulfide.					
Unsuitable for drying	Alcohols, amines, acids, ketones, ethers, chlorinated and fluorinated hydrocarbons.					
Residual water content	0.00002 mg $H_2$ 0/l to 25 % water absorption	on with SICAPENT®, corresponding	g to 2 mole H <sub>2</sub>	O per mole $P_2O_5$ .		
in air						
Capacity	P <sub>2</sub> O <sub>5</sub> : 40 % SICAPENT <sup>®</sup> : 33 %					
Application note	On adsorbing water, phosphorus pentoxide	e becomes covered with a film of	polymetaphos	phoric acid		
	which hinders the diffusion of $H_2O$ molecu	les. This effect can be avoided by	using SICAPEN	NT <sup>®</sup> as the		
	polymetaphosphoric acid formed from $P_2O_s$ and water is immediately adsorbed by the carrier substance.					
	As a result, the drying agent is available as	As a result, the drying agent is available as a fine, flowable granulate.				
Regeneration	Not possible					

Potassium carbonate [K <sub>2</sub> CO	] [	CAS No.	Content	Packaging	Ord. No.
Potassium carbonate for analysis EMSURE® ACS, ISO, Reag. Ph Eur		584-08-7	500 g	Plastic bottle	1.04928.0500
			1 kg	Plastic bottle	1.04928.1000
Potassium carbonate for analysis EMSURE® ACS, ISO, Reag. Ph Eur 584-08-7			50 kg	Fibre carton	1.04928.9050
For drying	Ammonia, amines, acetone, nitriles, chlorinated hydrocarbons.				
Unsuitable for drying	Acids, substances that tend to react when ex	Acids, substances that tend to react when exposed to water-removing basic conditions.			
Application	Drying liquids.				
Regeneration	At 160°C; becomes finely powdered from 100	)°C.			

Potassium hydroxide [KOF	1	CAS No.	Content	Packaging	Ord. No.		
Potassium hydroxide pelle	ts for analysis EMSURE®	1310-58-3	500 g	Plastic bottle	1.05033.0500		
			1 kg	Plastic bottle	1.05033.1000		
			5 kg	Plastic bottle	1.05033.5000		
			25 kg	Fibre carton	1.05033.9025		
			50 kg	Fibre carton	1.05033.9050		
For drying	Basic liquids, e.g. amines and inert and basic	Basic liquids, e.g. amines and inert and basic gases.					
Unsuitable for drying	Acids, acid derivatives (chlorides, anhydrides,	Acids, acid derivatives (chlorides, anhydrides, amides, nitriles).					
Application	Drying liquids. Not suitable for drying fast-fl	owing gases as this hinders di	ffusion due to o	leliquescence.			
	Can be used for drying gases if, apart from m	Can be used for drying gases if, apart from moisture, acid gas should be adsorbed.					
Residual water content	0.002 mg H <sub>2</sub> 0/I						
in air							



SICAPENT®		CAS No.	Content	Packaging	Ord. No.
SICAPENT® with indicator	SICAPENT® with indicator (phosphorus pentoxide drying agent		500 ml	Glass bottle	1.00543.0500
for desiccators) on inert ca	rrier material		2.8 l	Glass bottle	1.00543.2800
Composition	25 % inert inorganic carrier substance and 75 %	phosphorus pentoxide.			
Particle size of carrier	0.1 – 1.6 mm				
Bulk density	approx. 300 g/l				
Flowable up to	100 % water uptake				
Indicator content	0.1 %				
Water content /	H <sub>2</sub> O content [ %]	Indicator color of drying	agent		
Indicator color	0	Colorless			
	20	Green			
	27	Blue-green			
	33	Blue			
Application note	The main advantage of using granulated drying a	gents is the ease of use. Eve	n after signif	icant water uptake	
	(approx. 100 % of its own weight) the substance	remains in particle form. He	nce, subsequ	ent to the drying pro	-
	cess the drying agent can easily be removed from			5	;
	it is some 20 % faster than simple phosphorus pe	ntoxide. In other terms, 20 9	6 more water	is adsorbed	
	in the same time.				
Application	Drying liquids, filling drying tubes. Due to its high	n intensity and granulate for	m, it is partic	ularly suitable for	
	drying fast-flowing gases in drying tubes.				
Safety information	On opening the bottle, fine particles of drying ag	ent may spray out; hence wh	ien opening t	he bottle adhere to	
	the instructions on the label and open carefully w	whilst wearing safety spectad	eles.		

## Ordering information Drying agents S

Silica gel [SiO <sub>2</sub> ]		CAS No.	Content	Packaging	Ord. No.	
Silica gel granules, desiccan	t ~ 0.2 – 1 mm	7631-86-9	1 kg	Plastic bottle	1.01905.1000	
Silica gel granules, desiccan	t ~ 2 – 5 mm	7631-86-9	1 kg	Plastic bottle	1.01907.1000	
			5 kg	Plastic drum	1.01907.5000	
Silica gel with moisture indi	cator (brown gel) desiccant ~ 1 – 4 mm	-	1 kg	Plastic bottle	1.01972.1000	
			5 kg	Plastic bottle	1.01972.5000	
			25 kg	Plastic drum	1.01972.9025	
Silica gel with indicator (ora	nge gel), granulate ~ 1 – 3 mm	-	1 kg	Plastic bottle	1.01969.1000	
			5 kg	Plastic bottle	1.01969.5000	
			25 kg	Plastic drum	1.01969.9025	
Silica gel beads, desiccant ~	2 – 5 mm	7631-86-9	1 kg	Plastic bottle	1.07735.1000	
For drying	Practically all gases and liquids.					
Unsuitable for drying	Alkaline liquids (bases and amines). Orange g	el: strong acid and basic ga	ses, organic solve	nts.		
Application	In a desiccator, for protecting moisture-sensi	tive substances during stora	age and transport	and for		
	maintaining the dryness of anhydrous solven	ts, packing drying towers fo	r gases or solvent	S.		
Application temperature	Up to approx. 65°C the capacity is practically temperature-independent. At higher temperatures					
	the capacity decreases significantly.					
Advantages of white gel	Practically chemically inert, non-toxic, no disposal problems, easy-to-handle. Dried liquids can simply					
	be decanted.					
Residual water content	Min. 0.02 mg $H_2O/I$ , corresponding to a dew p		ed silica gel is wit	h water,		
in air	the more intensive it dries and the lower the residual water content.					
	Loading in g $H_2O$ / 100 g	Residual water cont	ent mg H <sub>2</sub> O/I			
	1	0.003				
	1.5	0.1				
	3.2	0.5				
	5.2	1				
	14	5				
	23	10				
	30	13				
Capacity	20 - 27 % at 25°C. If low residual water cont	ents is to be attained, the c	apacity may only	be partly utilized		
	(see table above): if the desired residual wate	r content of 1 mg/l is to be	attained, the load	ling may not exceed	d	
	5.2 g $H_2O$ / 100 g silica gel.					

Silica gel [SiO <sub>2</sub> ] – contin	ued				
Indicator change	At approx. 7 – 10 g adsorbed $H_2O$ / 100 g silica g	At approx. 7 – 10 g adsorbed $H_2O / 100$ g silica gel, the color change is from orange to colorless.			
in orange gel					
Indicator change	At approx. 7 – 10 g adsorbed $H_2O$ / 100 g silica g	At approx. 7 – 10 g adsorbed $H_2O / 100$ g silica gel, the color change is from brown to yellowish.			
in brown gel					
Regeneration	Regeneration Silica Gel	Temperature / duration in a drying oven			
	White-Gel	Approx. 100 – 180°C / approx. 3 hours			
	Orange-Gel	Approx. 130 – 140°C / approx. 3 hours			
	Brown-Gel	Approx. 120 – 150°C / approx. 3 hours			
	Silica gel is no longer capable of drying	Above 500°C			
Typical chemical and	Analytical data	98 % SiO <sub>2</sub> , remainder $AI_2O_3$ , TiO <sub>2</sub> , Fe <sub>2</sub> O <sub>3</sub>			
physical data	Indicator in orange gel	Iron salt			
	Indicator in brown gel	Iron salt			
	Bulk density	Approx. 0.7 kg/l			
	Surface (BET)	700 m²/g			
	Particle size	0.2 – 1 mm, 1 – 3 mm, 2 – 5 mm			
	Pore size	2.0 – 2.5 nm			
	Specific heat	Approx. 1 KJ/kg°C			
	Heat of adsorption per kg adsorbed water	3200 KJ			



## Ordering information Drying agents S-Z

Sodium [Na]		CAS No.	Content	Packaging	Ord. No.	
Sodium rod diameter 2.5 cm (protective liquid: paraffin oil)		7440-23-5	250 g	Glass bottle	1.06260.0250	
			1 kg	Glass bottle	1.06260.1000	
Sodium rods (protective liquid: paraffin oil) for synthesis 7440-23-5		7440-23-5	250 g	Glass bottle	8.22284.0250	
		1 kg	Glass bottle	8.22284.1000		
For drying	Ethers, saturated aliphatic and aromatic l	Ethers, saturated aliphatic and aromatic hydrocarbons, tertiary amines.				
Unsuitable for drying	Acids, acid derivatives, alcohols, aldehydes, ketones, alkyl and aryl halogenides; these can give rise					
	to extremely vigorous, explosive reaction	S.				
Application	As sodium wire using a sodium press for o	drying liquids. Caution! Sodium	reacts explosively	y with water.		
	Sodium waste should be disposed of using	g a high-boiling alcohol e.g. te	rt-butanol.			
Capacity	Stoichiometric					
NB	Practically all solvents which can be dried	Practically all solvents which can be dried with sodium can also be more intensively dried with				
	calcium hydride.					

Sodium hydroxide [NaOH	]	CAS No.	Content	Packaging	Ord. No.
Sodium hydroxide pellets for analysis EMSURE® ISO		1310-73-2	500 g	Plastic bottle	1.06498.0500
			1 kg	Plastic bottle	1.06498.1000
			5 kg	Plastic bottle	1.06498.5000
			25 kg	Fibre carton	1.06498.9025
			50 kg	Fibre carton	1.06498.9050
For drying	Basic liquids, e.g. amines and inert and	d basic gases.			
Unsuitable for drying	Acids, acid derivatives (chlorides, anhy	ydrides, amides, nitriles).			
Application	Drying liquids. Not suitable for drying	fast-flowing gases as pore diffusior	n is hindered by o	deliquescence.	
	Can be used for drying gases if acid ga	as also has to be adsorbed.			
Residual water content	0.002 mg H <sub>2</sub> 0/I				
in air					

Sodium sulfate [Na <sub>2</sub> SO <sub>4</sub> ]		CAS No.	Content	Packaging	Ord. No.
Sodium sulfate anhydrous g	ranulated for organic trace analysis EMSURE®	7757-82-6	500 g	Glass bottle	1.06639.0500
Sodium sulfate anhydrous, coarse granules for analysis 0.63 – 2.0 mm EMSURE® ACS		7757-82-6	500 g	Plastic bottle	1.06637.0500
			1 kg	Plastic bottle	1.06637.1000
			25 kg	Fibre carton	1.06637.9025
Sodium sulfate anhydrous for analysis EMSURE® ACS, ISO, Reag. Ph Eur 7757-82-		7757-82-6	500 g	Plastic bottle	1.06649.0500
			1 kg	Plastic bottle	1.06649.1000
			5 kg	Plastic bottle	1.06649.5000
			25 kg	Fibre carton	1.06649.9025
For drying	Almost all compounds including fatty acids, alde	hydes, ketones and alkyl and	l aryl halogen	ides.	
Application	Drying liquids; of average effect.				
Regeneration	At 150°C in a drying oven.				

Sulfuric acid [H₂SO₄]		CAS No.	Content	Packaging	Ord. No.
Sulfuric acid 95 – 97 % for analysis EMSURE® ISO		7664-93-9	1	Glass bottle	1.00731.1000
			1	Plastic bottle	1.00731.1011
			2.5 l	Glass bottle	1.00731.2500
			2.5 l	Safebreak btl.	1.00731.2510
			2.5 l	Plastic bottle	1.00731.2511
			25 I	Plastic container	1.00731.9025
For drying	Air, gases such as hydrogen chloride, chlo	rine, carbon monoxide, sulfur dioxi	de, hydrocarb	ons and	
	inert gases.				
Unsuitable for drying	Oxidizing gases such as hydrogen sulfides	Oxidizing gases such as hydrogen sulfides and hydrogen iodides and unsaturated and numerous other organic			
	compounds.				
Application	Sulfuric acid is used in wash bottles for drying gases or in open dishes in desiccators. To increase the surface				

area and to avoid the risk of burns.



# Auxiliaries for purification and sample preparation

Purifying, enriching and separating samples for analytical purposes are routine processes in laboratories. Solid, liquid and gaseous substances must also be purified in production. A wide range of methods can be used for this, including: Absorption, adsorption, distillation, extraction, filtration, crystallization and drying.

In the product range of reagents for sample preparation, Merck Millipore offers many laboratory chemicals for a wide variety of purification methods. Many of these products, however, are not only suitable for purification, but can also be used for example as reaction auxiliaries, filtration aids, fillers, additives or active ingredient carriers.

The final part of the Merck Millipore product range is classic laboratory aids – such as heating bath media, joint greases or boiling chips – which are indispensable for many reactions and distillations. You can find chemical and physical data on our reagents in the Merck Millipore Chemicals and Reagents catalog and on our website: www.merckmillipore.com

## Purification and preparation

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### Advantages

- Reliable: Above all thanks to their reproducible results, the premium Merck Millipore reagents for purification, separation and enrichment offer extremely high batch consistency and thus great reliability for the implementation of your application.
- **Convenient:** The comprehensive Merck Millipore product range is easy to order, all from one place.
- Economical: Merck Millipore offers packaging sizes for the smallest laboratories and larger packages for testing and production facilities, so that the exact quantity required can be easily obtained.



www.merckmillipore.com/purification-preparation

## Safety and environment



### Safety and environment

Merck Millipore pays attention to ensure that chemical reagents for sample preparation are manufactured in an eco-friendly way, without potentially harmful additives. In the manufacture of Merck Millipore decalcification solution, for example, the use of surfactants and any potentially allergenic aromas or colorings is consciously avoided. In contrast to other mineral oil heating baths, the Merck Millipore heating bath fluid is toxicologically harmless and biodegradable.

Our range of selected natural products such as quartz sand and kaolin, a natural, fine and well-crystallized clay material, underlines our strategic goal of protecting people and the environment sustainably. The selected natural product is tested for organic impurities and various anions and cations in the Merck Millipore chemical analysis laboratory and specified in laboratory quality.

#### Safety and environment - characteristics

▶ For easy detection, safety and/or sustainable characteristics of our products are highlighted with this symbol.



Further information about sustainable protection www.merckmillipore.com/protection

## Definitions

Absorption reagents	Absorption [from Latin »absorbere«: to devour, swallow up] describes the take-up of gases or liquids by solid bodies. Solids with cavities of a defined pore size can be used for separation, purification or enrichment. In this way, mixtures – usually of gases – can be separated into their individual component parts as a result of their varying molecule sizes.	
Adsorption reagents	Adsorption [from Latin »adsorbere«: to suck in] describes the take-up of a substance on the surface of a solid without a chemical reaction. Due to adhesive forces, gaseous, liquid and, in rare cases, solid substances accumulate on the surface or interface of the adsorbent. This releases the bond energy in the form of heat. The larger the surface, the greater the adsorption capacity of a substance. The total surface is therefore often given in »m <sup>2</sup> /g« for specification. This adsorption strength is reduced by warming and increased by cooling.	
Filter and clarification materials	The filter and clarification materials are usually reagents with absorbent or adsorbent properties, although materials which work in a similar way to frits or which form capillaries as filter cake can also be used to clarify solutions, to decolorize or to remove colloidally dissolved substances.	
Reaction auxiliaries	Reagents can be involved in a reaction or method directly or indirectly, for example as a filler or additive.	
Joint greases	To lubricate ground glass joints, taps, desiccator lids etc, grease (e.g. silicone or special desiccator grease) is usually applied. A thin, even layer of grease prevents glass components from »caking«. In order for the greases to remain in the joint and close it tightly even at high temperatures or in a vacuum, they must be chemically inert, temperature-resistant and highly viscous.	
Drying agents	Drying agents are not only used for the removal of water or other (not chemically bonded) liquids, but are also effective absorbents. They are used both in sample preparation and in product purification, for the drying of gases, solvents, synthetics or for the storage of sensitive products under consistently dry conditions.	

### More information

▶ »Absorption and filtration« see page 236 ▶ »Drying agents« see page 250

## Ordering information Purification and preparation

Boiling chips [Distillation]		CAS No.	Content	Packaging	Ord. No.
Boiling chips granules ~ 1 -	2 mm	-	250 g	Plastic bottle	1.07912.0250
Boiling chips granules ~ 2 -	8 mm	-	100 g	Plastic bottle	1.07913.0100
			500 g	Plastic bottle	1.07913.0500
- 11	In liquids, for example during distillation, boilin is heated above its boiling point.	g chips prevent »superheatin	g«, which oco	curs when the liquid	
Additional information	Boiling chips made from aluminium oxide or silicate ceramics are available in different grain sizes, depending on the medium and requirements.				
Boiling chips granules ~ 1 -	2 mm				
Application advice	Boiling chips with a grain size of under 2 mm a	re used for heating liquids in	very small v	essels.	
Boiling chips granules ~ 2 - 8 mm					
	Due to their porous structure, boiling chips mad solutions and solvents.	le from silicate ceramics are	best for use i	n watery	

BTS catalyst [Cleaning]		CAS No.	Content	Packaging	Ord. No.
BTS catalyst (about 5 x 3 mm) for gas purification –		1 kg	Glass bottle	1.04182.1000	
Application advice	A BTS catalyst is used for the removal of mixtures thereof, for example for cleanin carbon dioxide, methane, ethane, propan of compositions.	ig noble gases, nitrogen, hydrogen, o	oxygen, carbo	on monoxide,	
Additional information	The BTS catalyst consists of around 30 % further preliminary treatment for the ren black, the reduced form grey, although th with the help of an oxygen-nitrogen mix $2 \text{ Cu+O}_2 \rightarrow 2 \text{ CuO}$	noval of reducing contaminants from ne change in color is relatively weak	m inert gases Return to t	s. The oxidized for he oxide form occ	

Decalcification solution [[	Decalcification]	CAS No.	Content	Packaging	Ord. No.
Decalcification solution base: citric acid about 19 % -		1	Plastic bottle	1.00240.1000	
Application advice	Ivice         Decalcification solution with a citric acid base is used e.g. in laboratories for the decalcification of, e.g.,           faucets, boilers, heating coils, coolers, containers and pipes in which drinking or process water of medium or           high hardness is used. It can also be used as a complexing agent and buffer solution.				
Additional information	This decalcification solution with a citric acid base, 19 % concentration, can easily be made into the neces- sary concentration (1 to 5 %) by dilution and is thus easier to handle than organic acids in powder or crystal form. The use of a citric acid base also makes it less aggressive than other organic or diluted inorganic acids.				
Information on sustainable protection	The use of surfactants, aromas and co	lorings was conciously avoided during	production.		

### www.merckmillipore.com/purification-preparation

Glass beads	CAS No.	Content	Packaging	Ord. No.	
Glass beads 2 mm	-	500 g	Plastic bottle	1.04014.0500	
Glass beads 3 mm	_	500 g	Plastic bottle	1.04015.0500	
Glass beads 4 mm	-	500 g	Plastic bottle	1.04016.0500	
Glass beads 5 mm	-	500 g	Plastic bottle	1.04017.0500	
Glass beads 6 mm	-	500 g	Plastic bottle	1.04018.0500	
Application advice	Glass beads of size 2 to 6 mm can be used in a wide range of ways in laboratories and technical facilities: For filling distillation columns (the size of the glass beads determines the distillation speed and separation); As agitator and mixing beads e.g. when mixing solutions which tend to settle or substances which are difficult to dissolve; For the prevention of superheating during distillation, especially in highly pure solvents or acids; In the concentration or vaporization of solutions (prevents change in weight, even in aggressive substances and at high temperatures. The vaporization or evaporation dish can be reweighed with glass beads.				
Additional information	Glass beads are made from colorless soda lime silica glass. It is the most wid applications and is produced by melting quartz sand, natrite and lime at a te	,	5		



Glass beads 6 mm

Graphite	CAS No.	Content	Packaging	Ord. No.
Graphite fine powder extra pure	7782-42-5	2.5 kg	Fibre carton	1.04206.2500
		25 ka	Fibre carton	1.04206.9025

Heating bath fluid [Auxiliary]		CAS No.	Content	Packaging	Ord. No.
Heating bath fluid for heating baths up to approx. 170°C 31694-55-0			2.5 l	Plastic bottle	1.15265.2500
Application advice Heating bath fluid is used to allow chemical reactions at a higher temperature.					
Additional information	Heating bath fluids can be used up to a temperature of around 170°C. The colorless heating bath fluid				
	consists of a mixture of polyhydric aliphatic alcohols. It can be mixed with water without limitation, and is				
	toxically harmless and biodegradable. In c	ontrast to other mineral oil heatin	g baths, reac	tion vessels can	
	easily be washed out with water. Even if w	vater gets into the hot heating bat	h unintentio	nally, there is usuall	у
	no splashing.				
Information on	Toxically harmless and biodegradable.				
sustainable protection					

Iron (II) sulfide [Analysis] CAS No.			Content	Packaging	Ord. No.
lron(II) sulfide sticks Ø ~ 1 cm		1317-37-9	1 kg	Plastic bottle	1.03956.1000
			25 kg	Fibre carton	1.03956.9025
Application advice	Used in laboratories for the production of h	nydrogen sulfide.			
Additional information	Iron (II) sulfide is dark gray or black, metal-	like pieces, plates or sticks, whi	ch are usually	contaminated with	
	excess Fe-extra pure, crystalline FeS would	be light tombac brown. FeS is i	nsoluble in wa	ter, but dissolves	
	in acids, developing hydrogen sulfide. Since there is non-converted Fe in FeS, the nitrogen sulfide obtained in				
	this way is contaminated with hydrogen.				

## Ordering information Purification and preparation

Kaolin	CAS No.	Content	Packaging	Ord. No.
Kaolin powder	1332-58-7	2.5 kg	Plastic bottle	1.04440.2500
Application advice	Kaolin can be used e.g. as an adsorbent, filler, polisher or carrier. As an adso	rbent, kaolin	can bond cations	
	among other things.			
Additional information	Due to the structure of aluminium silicate, kaolin, also known as china clay,	kaolinite, po	rcelain clay etc.,	
	swells significantly when taking up water, and can absorb up to 80 % water			
Information on	The kaolin offered here is a selected natural product which is tested for orga	anic impuritie	es and various anion	s 💦
sustainable protection	and cations and provided in laboratory quality.			

Magnesia rods [Reaction	detection]	CAS No.	Content	Packaging	Ord. No.
Magnesia rods for the phosphorus salt pearls -		-	100 units	Plastic bottle	1.05809.0100
Application advice	Magnesia rods are used e.g. for the detection of certain elements through characteristic flame color and for reactions with borax and phosphoric salt pearls. The magnesia rods are also used as a carrier or digestion agent for certain substances that can be fused when held into a flame.				
Additional information	Magnesia rods are formed from ignited m 2600°C, does not melt even in the hottest 14 cm long and have a diameter of around	flame and does not show its own	5 5		



Marble granular [Analysis]		CAS No.	Content	Packaging	Ord. No.	
Marble granular for producing $CO_2$		471-34-1	1 kg	Plastic bottle	1.05986.1000	
Marble granular for producing CO <sub>2</sub> 471-34-1			5 kg	Plastic bottle	1.05986.5000	
Application advice	Marble granular is used for the determ	ination of carbon dioxide in water,	, which attacks l	ime. CO2 can be		
	produced from the effect of hydrochlo	produced from the effect of hydrochloric acid on marble.				
Additional information	Marble granular is formed from calcium	Marble granular is formed from calcium carbonate.				

Oil bath filling [Additive]		CAS No.	Content	Packaging	Ord. No.		
Oil bath filling for oil baths up to about 250°C		8002-74-2	1	Alu bottle	1.06900.1000		
			5 l	Alu bottle	1.06900.5000		
			25 I	Steel barrel	1.06900.9026		
Application advice	Oil bath fillings are used up to a temperat	Oil bath fillings are used up to a temperature of around 250°C.					
Additional information	Oil bath filling consists of mineral oils wh	ich boil at high temperatures and a	are free from	resin and acid. W	lhen		
	used for the first time, the oil bath filling	should be heated to the operating	temperature	for a while in a f	ume		
	hood, in order to remove low-molecular c	omponents. If high temperatures a	re maintaine	d for a long time,			
	there is a risk of thermal decomposition. Due to unpleasant odors, the oil bath filling should generally only be			be			
	used in a fume hood.						

Paraffin		CAS No.	Content	Packaging	Ord. No.		
Paraffin 42-44, in block	form	8002-74-2	1 kg	Plastic bottle	1.07150.1000		
			2.5 kg	Plastic bottle	1.07150.2500		
			25 kg	Fibre carton	1.07150.9025		
Paraffin 46-48, in block t	form	8002-74-2	1 kg	Plastic bottle	1.07151.1000		
			25 kg	Fibre carton	1.07151.9025		
Paraffin 51-53, in pastill	e form Ph Eur, BP, NF	8002-74-2	1 kg	Plastic bottle	1.07157.1000		
			2.5 kg	Plastic bottle	1.07157.2500		
			25 kg	Fibre carton	1.07157.9025		
Paraffin 52-54, in pastill	e form Ph Eur, BP, NF	8002-74-2	1 kg	Plastic bottle	1.07300.1000		
			20 kg	Fibre carton	1.07300.9020		
Paraffin 56–58, in pastille form Ph Eur, BP, NF		8002-74-2	1 kg	Plastic bottle	1.07337.1000		
			2.5 kg	Plastic bottle	1.07337.2500		
			20 kg	Fibre carton	1.07337.9020		
Paraffin 57-60, in pastill	e form Ph Eur, BP, NF	8002-74-2	1 kg	Plastic bottle	1.07158.1000		
			25 kg	Fibre carton	1.07158.9025		
Paraffin liquid Reag. Ph E	ur	8012-95-1	1	Plastic bottle	1.07162.1000		
			2.5 I	Plastic bottle	1.07162.2500		
			25 I	Plastic container	1.07162.9025		
Paraffin viscous Ph Eur, B	P, USP	8012-95-1	1	Plastic bottle	1.07160.1000		
			2.5 I	Plastic bottle	1.07160.2500		
			25 I	Plastic container	1.07160.9026		
Application advice	Paraffins have a wide range of uses and	Paraffins have a wide range of uses and applications, e.g. as a heating bath medium, as a waxing agent and					
	lubricant or as an additive. Due to their	high flashpoint and ignition tem	perature, paraffi	ns can be used up to			
	a temperature of around 200°C.						
Additional information	Paraffins (alkanes) consist of saturated		•		n		
	the chain length. Paraffin is not a singl	5		ons of varying chain			
	lengths. This also defines the condition	of aggregation and the defined m	nelting range.				

Polyvidone	CAS N	No.	Content	Packaging	Ord. No.			
Polyvidone 25 Ph Eur, BP	9003-	-39-8	100 g	Plastic bottle	1.07443.0100			
			1 kg	Plastic bottle	1.07443.1000			
Application advice	Polyvidone, also known as polyvinylpyrrolidone (PVP), is u	Polyvidone, also known as polyvinylpyrrolidone (PVP), is used as a protective colloid, stabilizer and binding						
	agent.							
Additional information	Polyvidone is a hygroscopic, amorphous, white-yellow po	owder, a polymer of 1	-vinylpyrrol	idon-(2), and forms a				
	viscous colloidal solution with water. Its amorphous strue	cture means that PV	P has no me	lting point, but rathe	r			
	a glass transition temperature depending on the level of polymerization between around 110 and 180°C.							
	PVP dissolves in water and in a wide range of other organ	nic solvents.						

## Ordering information Purification and preparation

Polyvinyl alcohol [Aux	liary]	CAS No.	Content	Packaging	Ord. No.
Polyvinyl alcohol prote	tive colloid for argentometric titration	9002-89-5	100 g	Plastic bottle	1.14266.0100
Application advice	Polyvinyl alcohol or PVA or PVOH is a man	-made thermoplastic plastic			
	as an adhesive and thickening agent.				
			Station in the	1 M 1 M	
			Carton Sales	in project -	
				- total	
			COMPANY AND		
		1. 19	1.54	Part of the	(4)
		A. C.			
	Quartz fine gro	anular	1 States	Bardi -	
			er an Deserve	the second	
		1 C			

Quartz [Filler and additive]		CAS No.	Content	Packaging	Ord. No.		
Quartz fine granular, washed and calcined for analysis		14808-60-7	250 g	Plastic bottle	1.07536.0250		
			1 kg	Plastic bottle	1.07536.1000		
			5 kg	Plastic bottle	1.07536.5000		
Application advice	As a result of its chemical indifference, qua	artz sand is used as a filler, filtrati	ion aid and a	s a catalyst carrier			
	or pulverization agent in laboratory analys	pulverization agent in laboratory analysis.					
Additional information	The grain size of quartz is 0.2 to 0.8 mm.						
Information on	Quartz sand is a selected natural product v	Duartz sand is a selected natural product which is treated in a similar way to sea sand.					
sustainable protection							

Silicic acid [Filler and ad	ditive]	CAS No.	Content	Packaging	Ord. No.	
Silicic acid precipitated extra pure heavy		10279-57-9	1 kg	Plastic bottle	1.00656.1000	
			20 kg	Fibre carton	1.00656.9020	
Silicic acid precipitated extra pure light DAB 10279-57-9			1 kg	Fibre carton	1.00657.1000	
			10 kg	Fibre carton	1.00657.9010	
Application advice	Synthetic, highly chemically pure silicic	Synthetic, highly chemically pure silicic acid may be used in laboratories as a filler, additive, flow aid and				
	adsorbent.					

Silicone anti-foaming agent	s [Anti-foaming agent]	CAS No.	Content	Packaging	Ord. No.
Silicon anti-foaming agent -			100 g	Plastic bottle	1.07743.0100
			500 ml	Plastic bottle	1.07743.0500
Application advice	Silicone anti-foaming agents prevent the unwanted formation of foam through substances which are active on the surface of watery solutions, such as e.g. emulsifiers. In defoaming processes in laboratories and technical facilities, the quantity used depends on the composition of the substance to be defoamed and the foaming agent itself. Very low dosages are usually necessary – in normal cases around 2 to 10 ppm and in water solutions containing wetting agents, 200 to 1000 ppm. The best amount to use in individual cases can only be found by trial.				
Additional information	lengths which are extremely effective	of a watery emulsion of substituted poly as a »foam suppressor«: their limited al ulate at the phase interface, thus reduc	oility to mix	with water	

Silicone grease [Sealing]		CAS No.	Content	Packaging	Ord. No.	
Silicone grease		-	100 g	Fibre case	1.07746.0100	
Silicone high vacuum grease heavy -			100 g	Fibre case	1.07921.0100	
Application advice	Silicone grease is used in a similar war parts and for sealing a vacuum in join	, , , , , , , , , , , , , , , , , , , ,	the lubricatio	on of moving		
Additional information	between the various grease types is th	Silicone greases are highly viscous polysiloxanes characterized by great chemical resistance. The difference between the various grease types is their viscosity, as can be derived from the additional designation. The higher the viscosity, the more reliable the seal in a vacuum, even at temperatures over 200°C.				

Silicone oil [Auxiliary]		CAS No.	Content	Packaging	Ord. No.
Silicone oil for oil baths up to 250°C		68083-14-7	100 ml	Plastic bottle	1.07742.0100
			1	Plastic bottle	1.07742.1000
Application advice	Silicone oil is used as a heat transfer medium. As a result of the wide operating temperature range				
	of -45 to +230°C, it is also used as thermostat liquid.				
Additional information	Silicone oil is a methylphenylpolysiloxane and an almost colorless liquid, and cannot be mixed with water.				

Stopcock grease [Sealing]		CAS No.	Content	Packaging	Ord. No.
Stopcock grease melting point 45 – 53°C –		-	250 g	Plastic can	1.04318.0250
Application advice	Desiccator grease is used for standard applications in the laboratory, e.g. for sealing joints in laboratory				
	equipment made of glass.				
Additional information	It consists of beeswax and vaseline.				

Triton®	CAS No.	Content	Packaging	Ord. No.
Triton <sup>®</sup> X-100 for analysis	9036-19-5	1	Glass bottle	1.08603.1000
		2.5 I	Glass bottle	1.08603.2500

Water detection paste	CAS No.	Content	Packaging	Ord. No.
Water detection paste N	-	150 g	Alu tube	1.08641.0001
Application advice	This paste can be used to detect water in containers that are filled with fuel, heating oil or other water-immiscible fluids. The level water will be reflect by a color change. A well-defined color change does not only enable to show the presence of water, but also allows to measure the level of the water.			
Additional information	If stored in a cool and dry place with the tube firmly closed, the water detection paste can be kept indefinitely. When using in cool weather, the paste's spreading ability can be improved by warming the tube slightly.			

## Indicators

The word »indicators« is derived from the Latin »indicare« meaning »show« – which corresponds to the function of these products. They help, for example, to display certain information when used in laboratories, and are thus most often used for titration purposes. Chemical reactions and the status of a reaction can be monitored by changes in the color of the indicators. Merck Millipore offers a large portfolio of highly accurate indicators depending on the type of chemical or physical conditions and reactions.



## Indicators

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### Advantages

- Reliable: Our exceptional high-quality analytics and technical expertise allow us to offer you a consistently high level of quality and reliability for your analyses.
- **Convenient:** Our wide range of indicators in various packaging sizes enables you to obtain all the indicators you need for your application from a single source. This saves you the effort of searching around and therefore saves valuable time, which you can use for important applications.
- Economical: Merck Millipore indicators allow you to precisely define the end-point for a wide range of titration types, thereby supporting your financial and scientific activities.
- High quality: The high standard of quality of the Merck Millipore indicators tested by means of extensive analyses ensure reproducible results.



www.merckmillipore.com/indicators

## Acid-alkali indicators or pH indicators

An acid-alkali indicator or pH indicator is a coloring agent which changes color to display whether an aqueous solution reacts as an acid or an alkali. In contrast to pH-indicator strips or indicator papers, which are mainly used to find the current pH value, the main use of pH indicators in laboratory applications is for the quantitative definition of the acid or alkali content of a solution (end-point indication / titration). When used as a rapid end-point indicator of an acid-base titration, the indicator shows a different color at the equivalence point compared to before it. This allows the user to visually recognize the end-point of the titration.

#### How does the change in color occur?

pH indicators are organic coloring agents with an acidic or alkaline character, which exhibit a different color and constitution in their undissociated condition than in the ionic form. When an acid or alkali is added, the color does not change immediately, but continuously within a pH range. In every titration, it is important to determine the equivalence point. As a result of hydrolytic processes, this does not always coincide with the neutral point (pH 7), but can lie in either the acidic or alkaline range. It is therefore necessary to select an indicator for the titration whose transition point has the same or similar pH value to the equivalence point of the system being titrated.

#### The pH-Indicator-Selector. An intelligent tool

On our website, we offer you a clever and convenient tool that helps you quickly find the right pH indicator for your particular application. You simply need to enter the pH value required and you will immediately receive an overview of the suitable indicators, along with color information, transition intervals and other useful product information.

To explore our practical application assistant, please visit: www.merckmillipore.com/ph-indicator-selector

#### More information

www.merckmillipore.com/ph-indicator-selector

# The pH-Indicator-Selector

Malachite green oxalate	green			green-blu	Je								blue	colorles	s
Brilliant green	yellow			gree	n										
Eosin Y	yellow			g	green flour	escence									
Erythrosine B	orange				red										
Methyl green	yellow			blue											
Cresol red		pink		ye	llow-oran	ge		yellow		purpl	e				
Crystal violet		yellow		blue-	violet										
m-Cresol purple		р	ink		yellow			yellow-ora	nge	F	ourple				
Thymol blue			red		yellow-ora	ange		yello	w-green		blue				
Eosin B		c	olorless	t.	oink floures	scence									
Quinaldine red		c	colorless		pink										
Bromocresol green sodium salt			yellow	-green			blue	:							
Dimethyl yellow				red	ł	yellow-o	orange								
Bromophenol blue				yellow	,	blue	-violet								
Congo red				blue	2		yellow-o	range							
Methyl orange				red	1	yellow-c	orange								
Methyl orange solution				red	1	yellow-o	orange								
Bromocresol green					yellow		blue								
Mixed indicator 4.5 acc. to Mortimer						red	blue								
Methyl red						red		yello	w-orange						
Methyl red sodium salt						red		yello	w-orange						
Mixed indicator 5					re	d-violet		green							
Bromocresol purple						ye	llow		purple						
Bromophenol red						yellow-or	range		purple						
4-Nitrophenol						colorle	ss		yel	llow					
Bromothymol blue sodium salt							yellow		blu	Je					
Bromothymol blue							yellow			blue					
Phenol red							ye	llow		violet-re	ed				
Phenol red sodium salt							ye	llow		violet-re	ed				
3-Nitrophenol							c	olorless		yello	ow-oran	ige			
Neutral red								blue-red		yellow-oi	range				
1-Naphtholphthalein								browr		blue-g	reen				
Phenolphthalein									colorless		r	ed-violet			
Phenolphthalein solution (1 % in ethanol)									colorless		r	ed-violet			
Phenolphthalein solution (0.375 % in methanol)									colorless		r	ed-violet			
Thymolphthalein										colorless		bli	Je		
Alkali blue												blue-\	violet	pink	
Indigo carmine													blue	yellow	
Titan yellow													yellow	red	

The pH ranges and color shades shown are approximations

# Ordering information

### Acid-alkali indicators | pH indicators A-D

Product	duct CAS No. End-point (pH / color change) from to		Content	Packaging	Ord. No.	
Alizarin red s mono sodium salt	130-22-3	4.3	6.3	25 g	Glass bottle	1.06278.0025
[C.I. 58005] for analysis and indicator		yellow	pink	100 g	Glass bottle	1.06278.0100
1st transition range (pH 4.3 – pH 6.3)		9.4	12.0			
2nd transition range (pH 9.4 – pH 12.0)		brown-orange	violet			
Alkali blue [C.I. 42765] indicator	1324-76-1	11.5	13.0	25 g	Glass bottle	1.09196.0025
		blue-violet	pink			
Alkali blue solution indicator	-	11.5	13.0	11	Glass bottle	1.09198.1000
		blue-violet	pink	2.5 l	Glass bottle	1.09198.2500
Brilliant green (hydrogen sulfate)			2.6	50 g	Glass bottle	1.01310.0050
[C.I. 42040) for microbiology		yellow	green			
Bromocresol green indicator	76-60-8	3.8	5.4	1 g	Glass bottle	1.08121.0001
ACS, Reag. Ph Eur		yellow	blue	5 g	Glass bottle	1.08121.0005
				25 g	Glass bottle	1.08121.0025
Bromocresol green sodium salt	62625-32-5	3.8	5.4	10 g	Glass bottle	1.01541.0010
indicator water-soluble ACS		yellow-green	blue			
Bromocresol purple indicator	115-40-2	5.2	6.8	5 g	Glass bottle	1.03025.0005
ACS, Reag. Ph Eur		yellow	purple	25 g	Glass bottle	1.03025.0025
Bromophenol blue indicator	115-39-9	3.0	4.6	5 g	Glass bottle	1.08122.0005
ACS, Reag. Ph Eur		yellow	blue-violet	25 g	Glass bottle	1.08122.0025
Bromophenol red indicator	ophenol red indicator 2800-80-8 5.2		6.8	5 g	Glass bottle	1.03023.0005
		orange-yellow	purpur			
Bromothymol blue indicator			7.6	5 g	Glass bottle	1.03026.0005
ACS, Reag. Ph Eur		yellow	blue	25 g	Glass bottle	1.03026.0025
Bromothymol blue sodium salt	34722-90-2	5.8	7.6	10 g	Glass bottle	1.01895.0010
indicator water-soluble ACS		yellow	blue			
Chlorophenol red indicator	4430-20-0	4.8	6.4	10 g	Glass bottle	1.03024.0010
		yellow	purple			
Congo red [C.I. 22120] indicator	573-58-0	3.0	5.2	25 g	Glass bottle	1.01340.0025
Reag. Ph Eur		blue	yellow-orange			
m-Cresol purple indicator	2303-01-7	1.2	2.8	5 g	Glass bottle	1.05228.0005
1st transition range (pH 1.2 – pH 2.8)		pink	yellow			
		7.4	9.0			
2nd transition range (pH 7.4 – pH 9.0)		yellow-orange	purple			
Cresol red indicator Reag. Ph Eur	1733-12-6	0.5	2.5	5 g	Glass bottle	1.05225.0005
1st transition range (pH 0.5 – pH 2.5)		pink	yellow-orange	25 g	Glass bottle	1.05225.0025
	st transition range (pH 0.5 – pH 2.5) p 6		8.5			
2nd transition range (pH 6.5 – pH 8.5)		yellow	purple			
Crystal violet [C.I. 42555] indicator	iolet [C.l. 42555] indicator 548-62-9 0.8		2.6	25 g	Glass bottle	1.01408.0025
ACS, Reag. Ph Eur		yellow	blue-violet	100 g	Glass bottle	1.01408.0100
				1 kg	Glass bottle	1.01408.1000
Dimethylyellow [C.I. 11020] indicator	60-11-7	2.9	4.0	10 g	Glass bottle	1.03055.0010
		red	yellow-orange	5		

### Acid-alkali indicators | pH indicators E-N

Product	CAS No. End-point (color change) from to		Content	Packaging	Ord. No.	
		from	to			
Eosin B (bluish) [C.I. 45400]	548-24-3	1.4	2.4	25 g	Glass bottle	1.15934.0025
for microscopy Certistain®		colorless	rose fluorescence	100 g	Glass bottle	1.15934.0100
Eosin Y (yellowish) [C.I. 45380]	17372-87-1	0.0	3.0	25 g	Glass bottle	1.15935.0025
for microscopy Certistain®		yellow	green fluorescence	100 g	Glass bottle	1.15935.0100
Erythrosine B [C.I. 45430]	16423-68-0	0.0	3.6	10 g	Glass bottle	1.15936.0010
for microscopy Certistain®		orange	red	25 g	Glass bottle	1.15936.0025
Indigo carmine [C.I. 73015]	860-22-0	11.5	13.0	25 g	Glass bottle	1.04724.0025
for analysis		blue	yellow			
Malachite green oxalate [C.I. 42000]	2437-29-8			25 g	Glass bottle	1.01398.0025
for microscopy and for microbiology			green-blue	100 g	Glass bottle	1.01398.0100
				1 kg	Glass bottle	1.01398.1000
Metanil yellow Reag. Ph Eur	587-98-4	2.3	1.2	10 g	Glass bottle	1.59267.0010
		orange-yellow	violet-red			
Methyl green zinc chloride double salt	7114-03-6	0.1	2.3	25 g	Glass bottle	1.15944.0025
[C.I. 42590] for microscopy Certistain®		yellow	blue			
Methyl orange [C.I. 13025] indicator	547-58-0	3.1	4.4	25 g	Glass bottle	1.01322.0025
ACS, Reag. Ph Eur		red	yellow-orange	100 g	Glass bottle	1.01322.0100
				1 kg	Metal can	1.01322.1000
Methyl orange solution 0.1 %	-	3.1	4.4	250 ml	PE bottle	1.01323.0250
indicator pH 3.1-4.4		red	yellow-orange	11	PE bottle	1.01323.1000
(red yellow-orange)						
Methyl red [C.I. 13020] indicator	493-52-7	4.4	6.2	25 g	Glass bottle	1.06076.0025
ACS, Reag. Ph Eur		red	yellow-orange	100 g	Glass bottle	1.06076.0100
				1 kg	Metal can	1.06076.1000
Methyl red sodium salt [C.I. 13020]	845-10-3	4.4	6.2	25 g	Glass bottle	1.06078.0025
water-soluble ACS		red	yellow-orange	100 g	Glass bottle	1.06078.0100
Mixed indicator 4.5 acc. to Mortimer	_	4.3	5.2	250 ml	PE bottle	1.01359.0250
		red	blue			
Mixed indicator 5 for ammonia	_	4.4	5.8	250 ml	Glass bottle	1.06130.0250
titrations <sup>1)</sup>		red-violet	green	1	Glass bottle	1.06130.1000
1-Naphtholphthalein indicator	596-01-0	7.1	8.3	1 q	Glass bottle	1.06246.0001
		brownish	blue-green	5 q	Glass bottle	1.06246.0005
1-Naphtholbenzein indicator	145-50-6	8.0	9.6	5 g	Glass bottle	1.06202.0005
Reag. Ph Eur		orange-brown	blue	5		
Neutral red [C.I. 50040] indicator	553-24-2	6.8	8.0	25 g	Glass bottle	1.01369.0025
and for microbiology <sup>1)</sup>		blue-red	yellow-orange	100 g	Glass bottle	1.01369.0100
3-Nitrophenol indicator	554-84-7	6.6	8.6	25 g	Glass bottle	1.06794.0025
		colorless	yellow-orange			
4-Nitrophenol indicator	100-02-7	5.4	7.5	25 g	Glass bottle	1.06798.0025
		colorless	yellow	100 g	Glass bottle	1.06798.0100
► 1) pH and redox indicator			1000	100 y		1.00730.0100

# Ordering information

### Acid-alkali indicators | pH indicators 0-Z

	Product	from to		Content	Packaging	Ord. No.	
		from to e 2R [C.I. 61110] indicator 4395-65-7 pink blue n in non-aqueous solvents ur		to			
	Oracet blue 2R [C.I. 61110] indicator	4395-65-7	pink	blue	5 g	Glass bottle	1.01487.0005
	Reag. Ph Eur						
Ī	Phenol red Reag. Ph Eur	143-74-8	6.8	8.4	1 g	Glass bottle	1.59375.0001
			yellow	red-violet	100 g	Glass bottle	1.59375.0100
Ī	Phenol red indicator ACS	143-74-8	6.4	8.2	5 g	Glass bottle	1.07241.0005
			yellow	red-violet	25 g	Glass bottle	1.07241.0025
					100 g	Glass bottle	1.07241.0100
					1 kg	Metal can	1.07241.1000
Ī	Phenol red sodium salt indicator ACS	34487-61-1	6.4	8.2	5 g	Glass bottle	1.11748.0005
			yellow	red-violet			
Ī	Phenol red solution indicator	-	6.4	8.2	100 ml	Glass bottle	1.07242.0100
			yellow	red-violet			
	Phenolphthalein indicator	77-09-8	8.2	9.8	25 g	Glass bottle	1.07233.0025
1	ACS, Reag. Ph Eur		colorless	red-violet	100 g	Glass bottle	1.07233.0100
					500 g	Glass bottle	1.07233.0500
Ī	Phenolphthalein solution 0.375 %	-	8.2	9.8	250 ml	Glass bottle	1.07238.0250
i	in methanol indicator		colorless	red-violet	1	Glass bottle	1.07238.1000
Ī	Phenolphthalein solution 1 %	-	8.2	9.8	250 ml	Glass bottle	1.07227.0250
i	in ethanol indicator pH 8.2 – 9.8		colorless	red-violet	1	Glass bottle	1.07227.1000
I	pH-Indicator solution pH 0.0 – 5.0	-	0.0	5.0	100 ml	PE bottle	1.09177.0100
١	with color card						
Ī	pH-Indicator solution pH 4.0 – 10.0	-	4.0	10.0	100 ml	PE bottle	1.09175.0100
I	Universal Indicator with color card				1	PE bottle	1.09175.1000
Ī	PH-Indicator solution pH 9.0 – 13.0	-	9.0	13.0	100 ml	Fibre case	1.09176.0100
١	with color card						
(	Quinaldine red indicator	117-92-0	1.4	3.2	5 g	Glass bottle	1.02282.0005
I	Reag. Ph Eur		colorless	rose			
	SUDAN III, [C.I. 26100]		blue	red	25 g	Glass bottle	1.11747.0025
	Thymol blue indicator	76-61-9	1.2	2.8	5 g	Glass bottle	1.08176.0005
1	ACS, Reag. Ph Eur		red	yellow-orange	25 g	Glass bottle	1.08176.0025
	1st transition range (pH 1.2 – pH 2.8)		7.8	9.5	2.5 kg	Metal can	1.08176.2500
2	2nd transition range (pH 7.8 – pH 9.5)		yellow-green	blue			
	Thymolphthalein indicator	125-20-2	9.3	10.5	5 g	Glass bottle	1.08175.0005
1	ACS, Reag. Ph Eur		colorless	blue	25 g	Glass bottle	1.08175.0025
					100 g	Glass bottle	1.08175.0100
	Titan yellow [C.I. 19540) indicator	1829-00-1	12.0	13.0	25 g	Glass bottle	1.01307.0025
I	Reag. Ph Eur		yellow	red			



#### Application advice

The following rules must be observed when using pH indicators for titration purposes:

- When titrating strong acids with strong bases, a broad range of indicators can be used whose end-points lie in the weakly acidic, in the neutral, or in the weakly alkaline range.
- Weak acids can be titrated with strong bases by using indicators whose end-points lie in the weakly alkaline range.
- Weak bases can be titrated with strong acids by using indicators whose end-points lie in the weakly acidic range.
- The titration of weak bases with weak acids and vice versa produces inaccurate results.
   Only very few indicators which have to be specifically determined according to the case are suitable for this particular type of titration, and then only when a reference solution is titrated at the same time.

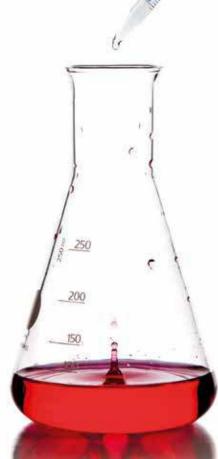
# Indicator solution

### Acid-alkali indicators | pH indicators A-M

	Actu-alkali indicators   pri indicators A-	
	Product	Indicator solution
Α	Alizarin red s mono sodium salt [C.I. 58005]	0.1 – 0.2 g in 100 ml water
	for analysis and indicator	
	Alkali blue [C.I. 42765] indicator	0.1 g in 100 ml ethanol (96 %)
В	Brilliant green (hydrogen sulfate) [C.I. 42040] for microbiology	0.1 g in 100 ml water
	Bromocresol green indicator ACS, Reag. Ph Eur	0.1 g in 100 ml Ethanol (20 %) or 0.04 g in 0.58 ml sodium hydroxide solution dilute 0.1 mol/l and fill up to100 ml with water
	Bromocresol green sodium salt indicator water-soluble ACS	0.1 g in 100 ml water
	Bromocresol purple indicator ACS, Reag. Ph Eur	0.1 g in 100 ml Ethanol (20 %) or 0.04 g in 0.74 ml sodium hydroxide solution dilute 0.1 mol/l and fill up to 100 ml with water
	Bromophenol blue indicator ACS, Reag. Ph Eur	0.04 g in 100 ml Ethanol (20 %) or 0.04 g in 0.69 ml sodium hydroxide solution dilute 0.1 mol/l and fill up to 100 ml with water
	Bromophenol red indicator	0.1 g in 100 ml Ethanol (20 %) or 0.04 g in 0.94 ml sodium hydroxide solution dilute 0.1 mol/l and fill up to 100 ml with water
	Bromothymol blue indicator ACS, Reag. Ph Eur	0.1 g in 100 ml Ethanol (20 %) or 0.04 g in 0.64 ml sodium hydroxide solution dilute 0.1 mol/l and fill up to 100 ml with water
	Bromothymol blue sodium salt indicator water-soluble ACS	0.1 g in 100 ml water
С	Chlorophenol red indicator	0.1 g in 100 ml Ethanol (20 %) or 0.04 g in 0.94 ml sodium hydroxide solution dilute 0.1 mol/l and fill up to 100 ml with water
	Congo red [C.I. 22120] indicator Reag. Ph Eur	0.2 g in 100 ml water
	m-Cresol purple indicator	0.04 g in 100 ml Ethanol (20 %) or 0.04 g in 1.05 ml sodium hydroxide solution dilute 0.1 mol/l and fill up to 100 ml with water
	Cresol red indicator Reag. Ph Eur	0.1 g in 100 ml Ethanol (20 %) or 0.04 g in 1.05 ml sodim hydroxide solution dilute 0.1 mol/l and fill up to 100 ml with water
	Crystal violet [C.I. 42555] indicator ACS, Reag. Ph Eur	0.1 g in 100 ml Ethanol (70 %)
D	Dimethylyellow [C.I. 11020] indicator	0.1 – 0.5 g in 100 ml Ethanol (90 %)
E	Eosin B (bluish) [C.I. 45400] for microscopy Certistain®	0.1 g in 100 ml water
	Eosin Y (yellowish) [C.I. 45380] for microscopy Certistain®	0.1 g in 100 ml water
	Erythrosine B [C.I. 45430] for microscopy Certistain®	0.1 g in 100 ml water
I.	Indigo carmine [C.I. 73015] for analysis	0.25 g in 100 ml Ethanol (50 %) or 1 g in 100 ml water
Μ	Malachite green oxalate [C.I. 42000] for microscopy and for microbiology	0.1 g in 100 ml water
	Metanil yellow Reag. Ph Eur	0.1 g in 100 ml water
	Methyl green zinc chloride double salt [C.I. 42590] for microscopy Certistain®	0.1 g in 100 ml water
	Methyl orange [C.I. 13025] indicator ACS, Reag. Ph Eur	0.4 g in 100 ml Ethanol (20 %) or 0.04 g in 100 ml water
	Methyl orange solution 0.1 % indicator pH 3.1-4.4 (red yellow-orange)	Ready to use indicator solution
	Methyl red [C.I. 13020] indicator ACS, Reag. Ph Eur	0.1 g in 100 ml Ethanol (96 %)
	Methyl red sodium salt [C.I. 13020] water-soluble ACS	0.1 g in 100 ml water
	Mixed indicator 4.5 acc. to Mortimer	Ready to use indicator solution
	Mixed indicator 5 for ammonia titrations	Ready to use indicator solution

### Acid-alkali indicators | pH indicators N-Z

	Product	Indicator solution
Ν	1-Naphtholphthalein indicator	0.1 g in 100 ml Ethanol (96 %)
	1-Naphtholbenzein indicator Reag. Ph Eur	0.1 – 1.0 g in 100 ml 2-propanol
	Neutral red [C.I. 50040] indicator and	0.3 g in 100 ml Ethanol (70 %)
	for microbiology	
	3-Nitrophenol indicator	0.3 g in 100 ml Ethanol (96 %) or 0.08 g in 100 ml water
	4-Nitrophenol indicator	0.2 g in 100 ml Ethanol (96 %) or 0.08 g in 100 ml water
0	Oracet blue 2R [C.I. 61110] indicator for titration in	0.1 – 0.5 g in 100 ml glacial acetic acid
	non-aqueous solvents Reag. Ph Eur	
Р	Phenol red Reag. Ph Eur	0.1 g in 100 ml ethanol
	Phenol red indicator ACS	0.1 g in 100 ml Ethanol (20 %) or 0.04 g in 1.3 ml sodium hydroxide solution
		dilute 0.1 mol/l and fill up to 100 ml with water
	Phenol red sodium salt indicator ACS	0.04 g in 100 ml water
	Phenolphthalein indicator ACS, Reag. Ph Eur	0.1 g in 100 ml Ethanol (96 %)
	Phenolphthalein solution 0.375 % in methanol	Ready to use indicator solution
	indicator	
	Phenolphthalein solution 1 % in ethanol indicator	Ready to use indicator solution
	pH 8.2 – 9.8	
Q	Quinaldine red indicator Reag. Ph Eur	0.1 g in 100 ml Ethanol (60 %)
S	SUDAN III, [C.I. 26100]	0.1 – 0.5 g in 100 ml glacial acetic acid
Т	Thymol blue indicator ACS, Reag. Ph Eur	0.04 g in 100 ml Ethanol (20 %) or 0.04 g in 0.86 ml sodium hydroxide
		solution to be filled up to 100 ml
	Thymolphthalein indicator ACS, Reag. Ph Eur	0.1 g in 100 ml Ethanol (50 %)
	Titan yellow [C.I. 19540] indicator Reag. Ph Eur	0.1 g in 100 ml Ethanol (20 %)



# **Redox indicators**

Redox indicators are reducible or oxidizable materials in which the reduced and oxidized forms of the material have different colors. Redox indicators are used in laboratories to track redox reactions, to determine approximate redox potentials, and to indicate the end-point of a redox titration (in oximetry). In this, the color does not change at a particular pH value, but instead at a particular redox potential, which should be close to the redox potential of the equivalence point.

If an oxidizing volumetric solution is used, the redox potential of the indicator must be higher than the potential of the solution. In the case of a reducing volumetric solution, the redox potential of the indicator must be lower than that of the solution.



# Ordering information

### **Redox indicators** A-Z

	Product	CAS No.	Redoxpot	ential	Color change [form]		Content	Packaging	Ord. No.
	Troduct	CAS NO.		E <sub>m</sub> [pH 7]	[oxidized] [reduced]		content	Tackaying	010.100.
A	Amido black 10 B [C.I. 20470]	1064-48-8	+ 0.57	+ 0.84	yellow-	blue	25 g	Glass bottle	1.01167.0025
	for electrophoresis	1004-40-0	+ 0.37	+ 0.04	brown	oluc	25 y		1.01107.0023
В	2,2'-Bipyridine for analysis	366-18-7	+ 1.03	-	light blue	red	5 g	Glass bottle	1.03098.0005
	(reagent for iron(II) and molybdenum) ACS						25 g	Glass bottle	1.03098.0025
	Brilliant cresyl blue zinc chloride double salt for microscopy Certistain®	51716-96-2	+ 0.58	+ 0.05	blue	colorless	25 g	Glass bottle	1.01368.0025
D	2,6-Dichlorophenol-indophenol	620-45-1	+ 0.67	+ 0.23	blue	colorless	5 g	Glass bottle	1.03028.0005
	sodium salt dihydrate						25 g	Glass bottle	1.03028.0025
	for analysis for the determination								
	of ascorbic acid								
	N,N-Dimethyl-1,4-phenylene-	536-46-9	+ 0.751	-	dark blue	colorless	25 g	Glass bottle	1.03067.0025
	diammonium dichloride for analysis								
	3,3'-Dimethylnaphthidine metal indicator	13138-48-2	+ 0.78	-	purple red	colorless	1 g	Glass bottle	1.03122.0001
	Diphenylamine-4-sulfonic acid	6152-67-6	+ 0.84	-	red-violet	colorless	10 g	Glass bottle	1.03590.0010
	sodium salt redox indicator						-		
-	Diphenylamine-4-sulfonic acid	6211-24-1	+ 0.841)	-	red-violet	colorless	5 g	Glass bottle	1.00255.0005
	barium salt redox indicator								
F	Ferroin solution (1,10-phenanth-	-	+ 1.06	-	blue	orange-	100 ml	Glass bottle	1.09193.0100
	roline iron(ll) sulfate) 1/40 mol/l					red	500 ml	Glass bottle	1.09193.0500
	redox indicator								
	Ferroin indicator solution for waste	-	+ 1.06	-	blue	red	100 ml	Glass bottle	1.09161.0100
	water analysis						500 ml	Glass bottle	1.09161.0500
Ľ	Indigo carmine [C.I. 73015]	860-22-0	+ 0.29	- 0.11	blue	yellowish	25 g	Glass bottle	1.04724.0025
	for analysis <sup>2)</sup>								
м	Methylene blue [C.I. 52015]	-	+ 0.53	+ 0.01	blue	colorless	25 g	Glass bottle	1.15943.0025
	for microscopy Certistain®						100 g	Glass bottle	1.15943.0100
	Methylene blue [C.I. 52015] Reag.	-	+ 0.53	+ 0.01	blue	colorless	10 g	Glass bottle	1.59270.0010
	Ph Eur						100 g	Glass bottle	1.59270.0100
N	Neutral red [C.I. 50040] indicator	553-24-2	+ 0.24	- 0.29	violet-red	colorless	25 g	Glass bottle	1.01369.0025
	and for microbiology <sup>2)</sup>						100 g	Glass bottle	1.01369.0100
	Nile blue (hydrogen sulfate)	3625-57-8	+ 0.4	- 0.12	blue-red	colorless	25 g	Glass bottle	1.15946.0025
	[C.I. 51180] for microscopy								
	Certistain®								
P	1,10-Phenanthroline monohydrate	5144-89-8	+ 1.14	+ 1.06	light blue	red	5 g	Glass bottle	1.07225.0005
	for analysis and redox indicator						10 g	PE bottle	1.07225.0010
							100 g	PE bottle	1.07225.0100
	Safranine O [C.I. 50240] for microscopy Certistain®	477-73-6	+ 0.24	- 0.29	blue-violet	colorless	25 g	Glass bottle	1.15948.0025
	Thionine (acetate) [C.I. 52000]	78338-22-4	+ 0.56	+ 0.06	violet	colorless	25 g	Glass bottle	1.15929.0025
	for microscopy Certistain®	,0000 22 4	1 0.00	. 0.00		201011235	20 9		
	Zinc iodide starch solution for	-	-	-	colorless	blue	500 ml	PE bottle	1.05445.0500
	analysis <sup>3)</sup>								
	<ul> <li>analysis<sup>3</sup></li> <li>▶ 1) in sulfuric acid 1 mol/l ▶ 2) pH a</li> </ul>	nd redox indica	ator 🕨 3) I	ntensive blue	inclusion co	mpound of io	dide starch so	olution	

# Complex or metal indicators

Complex or metal indicators are organic materials that are capable of forming colored complexes with metal ions. In this, the free indicator has a different color to that of the complexed form. For the detection of the end-point of a complexometric titration, it is essential to use one of the metal indicators that form complexes with metal ions which differ in color from the indicators themselves. In addition, the reagent properties must be suited to the sensitivity and the acid-base characteristics of the indicator. Complex or metal indicators are used, for example, in the volumetric analytical determination of the concentration of metal ions, e.g. in complexometric titrations. Determination of water hardness is a common application.

# Ordering information

#### Complex indicators | Metal indicators A-E

Product	CAS No.	For determination of	Content	Packaging	Ord. No.
Alizarin-3-methylamine-N,N-diacetic acid dihydrate for the spectrophotometric determination of fluoride	-	-	5 g	Glass bottle	1.01010.0005
Arsenazo III metal indicator	1668-00-4	Lanthanide, Th, Y	5 g	Glass bottle	1.10107.0005
Aurin tricarboxylic acid ammonium salt for analysis (reagent for aluminium) ACS	569-58-4	Al, Ca, Fe, Mg	10 g	Glass bottle	1.00128.0010
Calcein indicator for metal determination	1461-15-0	Ba, Ca, Cr, Cu, Mg, Sr	5 g	Glass bottle	1.02315.0005
Calcon [C.I. 15705] metal indicator	2538-85-4	Ca, Cd, Mg, Zn	50 g	Glass bottle	1.04594.0050
Calconcarboxylic acid metal indicator	-	Са	5 g	Glass bottle	1.04595.0005
			25 g	Glass bottle	1.04595.0025
1,8-Dihydroxy-2-(4-sulfophenylazo)naph- thalene-3,6- disulfonic acid trisodium salt for analysis (reagent for fluoride, zirconium, thorium)	23647-14-5	Th, Zr	25 g	Glass bottle	1.07998.0025
3,3'-Dimethylnaphthidine metal indicator	13138-48-2	Cd, Zn	1 g	Glass bottle	1.03122.0001
Dimethylyellow [C.I. 11020] indicator	60-11-7	For production of mixed indicators	10 g	Glass bottle	1.03055.0010
1,5-Diphenylcarbazide for analysis	140-22-7	Hg, Ni, Pb, V	25 g	Glass bottle	1.03091.0025
and redox indicator ACS, Reag. Ph Eur			100 g	Glass bottle	1.03091.0100
1,5-Diphenylcarbazone (cont. 50 %	-	-	5 g	PE bottle	1.03087.0005
Diphenylcarbazid) ACS, Reag. Ph Eur			25 g	PE bottle	1.03087.0025
Dithizone for analysis	60-10-6	Ag, Au, Bi, Cd, Co, Cu, Fe, Hg, In, Mn,	5 g	Glass bottle	1.03092.0005
(1,5-diphenylthiocarbazone) Reag. Ph Eur		Ni, Pb, Sn, Zn	25 g	Glass bottle	1.03092.0025
Eriochrome black T [C.I. 14645] indicator	1787-61-7	Ba, Ca, Cd, Hg, In, Lanthanide, Mg,	25 g	Glass bottle	1.03170.0025
for complexometry ACS, Reag. Ph Eur		Mn, Pb, Zn, Zr	100 g	Glass bottle	1.03170.0100
Eriochrome blue-black B [C.I. 14640] metal indicator	3564-14-5	Ca, Cd, Mg, U, Zn, Zr <sub>2</sub>	25 g	Glass bottle	1.03168.0025
Eriochrome cyanine R [C.I. 43820] for analysis (reagent for aluminium)	3564-18-9	Al, Ca, Cu, Fe, Mg, Th, Zr	25 g	Glass bottle	1.03164.0025

# Ordering information

### **Complex indicators | Metal indicators H-Z**

Product	CAS No.	For determination of	Content	Packaging	Ord. No.
H Hematoxylin cryst. [C.I. 75290]	517-28-2	Al, Bi, Cu, Th, Zr		Glass bottle	1.04302.0025
for microscopy	017 20 2		25 g 100 g	Glass bottle	1.04302.0100
Hydroxynaphthol blue metal (pM) indicator	63451-35-4		25 g	Glass bottle	1.04593.0025
Indicator buffer tablets	03431-33-4	Ca, Zn	500 g	PE can	1.08430.0500
for the determination of water hardness	-	Ca, 211	1 kg	PE can	1.08430.1000
with Titriplex <sup>®</sup> solutions			тку	r L Call	1.06430.1000
M Magnesium reagent acc. to Mann and Yoe	14936-97-1	Mg	5 g	Glass bottle	1.08712.0005
(reagent for magnesium)					
Methylthymol blue sodium salt	1945-77-3	Ba, Bi, Ca, Cd, Cu, Fe, Hg, In, Lantha-	1 g	Glass bottle	1.06084.0001
metal indicator		nide, Mg, Mn, Pb, Sc, Sn, Th, Ti, Zn, Zr	5 g	Glass bottle	1.06084.0005
Murexide (ammonium purpurate)	3051-09-0	Ag, Ca, Co, Cu, Mn, Ni, Sc, Th, Zn	5 g	Glass bottle	1.06161.0005
metal indicator ACS, Reag. Ph Eur			25 g	Glass bottle	1.06161.0025
N Naphthol green B [C.I. 10020] indicator	19381-50-1	-	25 g	Glass bottle	1.01306.0025
P Phthalein purple metal indicator	2411-89-4	Ba, Ca, Cd, Mg, Sr	1 g	Glass bottle	1.07297.0001
Reag. Ph Eur			10 g	Glass bottle	1.07297.0010
1-(2-Pyridylazo)-2-naphthol (PAN) metal	85-85-8	Al, Bi, Cd, Co, Cu, Fe, Ga, Hg, In, Mn,	5 g	Glass bottle	1.07531.0005
indicator Reag. Ph Eur <sup>1)</sup>		Ni, Pb, Th, Ti, U, V, Zn			
4-(2-Pyridylazo)resorcinol monosodium salt	16593-81-0	Al, Bi, Cd, Cu, Fe, Ga, Hg, In, Lantha-	5 g	Glass bottle	1.07533.0005
monohydrate metal indicator Reag. Ph Eur		nide, Mn, Ni, Pb, Sr, Th, Ti, Zn			
3,5-Pyrocatecholdisulfonic acid disodium	149-45-1	Fe: blue-green to yellow	100 g	Glass bottle	1.01922.0100
salt monohydrate (tiron) metal indicator					
Pyrogallolred metal indicator	32638-88-3	Bi, Pb, Ni, Co	1 g	Glass bottle	1.07534.0001
R Rhodizonic acid disodium salt indicator	523-21-7	-	5 g	Cardboard box	1.06595.0005
for sulfate titration					
T Thorin indicator for sulfate titration	3688-92-4	Bi, Li, Sc, Th, U, Y	5 g	Glass bottle	1.08294.0005
X Xylenol orange tetrasodium salt	3618-43-7	Bi, Ca, Cd, Co, Cu, Fe, Hg, In,	1 g	Glass bottle	1.08677.0001
metal indicator ACS, Reag. Ph Eur		Lanthanide, Mg, Mn, Pb, Sc, Th, Ti, U,	5 g	Glass bottle	1.08677.0005
		V, Y, Zn, Zr			
Z Zincon for the photometric determination	62625-22-3	2–3 Cd, Hg, Pb, Zn		Glass bottle	1.08739.0001
of copper and zinc			5 g	Glass bottle	1.08739.0005
▶ 1) Application as metal indicator: 0.01-0.1 %	Ethanol (96 %)				



# Fluorescence indicators

Substances, whose fluorescent properties in solution are influenced by a change in the hydrogen-ion concentration, oxidation potential, or metal-ion concentration, are used as fluorescence indicators. Fluorescence indicators are essential for end-point determination of precipitation titration, in which a well-defined change in one of these properties occurs at the equivalence point. When the reagent is added, the analyte forms a precipitate that is hardly soluble.

Fluorescent dyes are also used to stain and visualize cells, cell components, chromosomes and bacteria, as well as to detect antigen-antibody reactions.

### Ordering information

#### Fluorescence indicators A-U

Product	CAS No.	End-point (color change)		Content	Packaging	Ord. No.
Acridine orange zinc chloride double salt	10127-02-3	8.4	10.4	25 g	Glass bottle	1.15931.0025
[C.I. 46005] for microscopy Certistain®	no fluoresce		yellow-green			
Calcein indicator for metal determination	1461-15-0	8.0	10.0	5 g	Glass bottle	1.02315.0005
		no fluorescence	yellow-green			
Chromotropic acid disodium salt dihydrate	5808-22-0	3.1	4.4	25 g	Glass bottle	1.02498.0025
for analysis ACS, Reag. Ph Eur		no fluorescence	light blue			
2',7'-Dichlorofluorescein indicator ACS,	76-54-0	4.0	6.6	5 g	Glass bottle	1.09676.0005
Reag. Ph Eur		blue-green	no fluorescence			
Eosin Y (yellowish) [C.I. 45380]	17372-87-1	0.0	3.0	25 g	Glass bottle	1.15935.0025
for microscopy Certistain®	no fluc		green	100 g	Glass bottle	1.15935.0100
Erythrosine B [C.I. 45430]	16423-68-0	2.5	4.0	10 g	Glass bottle	1.15936.0010
for microscopy Certistain®		no fluorescence	light green	25 g	Glass bottle	1.15936.0025
Fluorescein sodium [C.I. 45350]	518-47-8	4.0	4.5	50 g	Glass bottle	1.03887.0050
indicator Reag. Ph Eur		rose	green	250 g	Glass bottle	1.03887.0250
Fluorescent indicator $F_{254}$	68611-47-2	fluor. green	colorless	50 g	PE bottle	1.09182.0050
2-Naphthol for analysis	135-19-3	8.5	9.5	100 g	PE bottle	1.06234.0100
		no fluorescence	blue	500 g	PE bottle	1.06234.0500
1,2-Phenylenediamine for analysis	95-54-5	3.1	4.4	50 g	Glass bottle	1.07243.0050
		green	no fluorescence			
Phloxin B [C.I. 45410] for microscopy	587-98-4	2.5	4.0	25 g	Glass bottle	1.15926.0025
Certistain®		no fluorescence	light green			
Uranine AP [C.I. 45350] concentrated for	518-47-8	-	-	100 g	Glass bottle	1.08462.0100
the examination of subterranean waters				1 kg	Metal can	1.08462.1000

#### www.merckmillipore.com/indicators



Uranine fluorescent dye: the best and most commonly used tracer dye for determining water flow.

# Indicators for special application

In addition to the previously mentioned indicators, Merck Millipore also offers special indicators tailored to your exact requirements. For further information and consultation, please contact your regional Merck Millipore office.

### Ordering information

#### Indicators for special application

Product	CAS No.	Indicator group	Color cl	Color change		Packaging	Ord. No.
				from to			
Dimidium Bromide for surfactant tests <sup>1)</sup>	518-67-2	Tenside	pink blue		1 g	Glass bottle	1.12130.0001
		indicator			5 g	Glass bottle	1.12130.0005
Disulfine blue vn 150 [C.I. 42045]	129-17-9	Tenside	pink	blue	25 g	Glass bottle	1.12144.0025
for surfactant tests <sup>2)</sup>		indicator					
lodine indicator	-	Indicator for	blue	colorless	100 g	PE bottle	1.04764.0100
		iodometry					
Application advice for a two-phase titration	n 1) mixed wit	h Disulfine blue 2)	mixed wit	h Dimidium Bromi			

# Cleaning applications Extran<sup>®</sup>

Extran<sup>®</sup> laboratory cleansers are the perfect solution for cleaning your laboratory utensils to Merck Millipore's well-known quality standards. Reliable processes in laboratories and product facilities are only possible with thorough, residue-free cleaning. Only in this way is it possible to ensure proper scientific working procedures. Everything that comes into contact with chemicals or biological substances must be free of impurities, both before and after use. Merck Millipore's high-quality Extran<sup>®</sup> cleaning agents have fulfilled these requirements for over 25 years.



# Cleaning applications

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### Advantages

- Reliable and residue-free cleaning with Extran<sup>®</sup>.
- Extran<sup>®</sup> is free from NTA protecting the health of the laboratory staff.
- Extran<sup>®</sup> is free from scents and dyestuffs and does not contain chlorine or other toxic ingredients.
- The active ingredients in Extran<sup>®</sup> are biodegradable.
- Extran<sup>®</sup> is the ideal all-purpose cleaner.



www.merckmillipore.com/extran

# Safety and environment



#### Efficient, safe and environment-friendly

Extran® cleans reliably, leaving no residues. This prevents residues from being transferred into the next analysis or test. Merck Millipore provides a practical and easy-to-use application aid to prove the freedom from residues of nonionic surfactants after the cleaning process by means of a photometric test. This helps you in preparing your own individual cleaning validation, saving you time and money.

Extran® is free from scents and dyestuffs and does not contain chlorine or other toxic ingredients. It thus prevents odors, protecting the health of the laboratory staff in the process. Our laboratory cleansers are also free of silicones and oxidants.

#### **Biodegradable**

The active ingredients in Extran<sup>®</sup> are biodegradable and Extran<sup>®</sup> is manufactured under stringently controlled production conditions and fulfils the requirements of environmental protection. In almost all cases, Extran<sup>®</sup> makes the use of chromosulfuric acid, which is still common on the market, unnecessary. It is thus gentle on the environment and on the health of staff.

More information about sustainable protection www.merckmillipore.com/protection

#### The ideal all-purpose cleaner

Depending on the type of contamination and the material to be cleaned, the Extran® range of products offers the ideal solution for the cleaning of your laboratory utensils and production locations. Put your trust in over 25 years of Extran® experience from Merck Millipore and use our detergents for manual cleaning or machine cleaning in laboratory washing machines. Both processes generally require different detergents. You can find more details on this later on in this chapter.

For further information, new additions to the range, Safety Data Sheets and of course our certificates of analysis, please see our website.

#### www.merckmillipore.com/extran



# Cleaning application examples

	Cleaning applications A-0	Decalcification solution	Extran <sup>®</sup> MA 01	Extran <sup>®</sup> MA 02	Extran® MA 05	Sodium hydroxide solution	Machine washing	Extran® AP 11	Extran® AP 12	Extran® AP 13	Extran® AP 16	Extran® AP 17	Extran® AP 21	Extran® AP 22	Extran® AP 33	Extran® AP 41
	an	calcit	tran®	tran®	tran®	dium	lach	tran®								
	g approximent of	ă		ŭ		S	2	ŭ		ă	ă	ŭ	ш	ŭ	ň	ŭ
Α	Alkyd resins		•		•				•							
	Aluminum			•												
	Amines	•											•	•		
	Analytical laboratories Balsam resin		•	•	•			•			•					
В	Baisam resin Bitumen		•		•				•			•				
	Blood		•		•				•							
	Brass		-	•	•				•							
	Breweries		•	•		•		•	•	•		•				•
	Bronze			•												-
с	Calcareous deposits on equipment	•											•	•		
-	Carbonates	•											•			
	Cells			•												
	Chemical glassware		•	•	•			•			•					
	Culture media								•			•				
D	Dairies		٠	•	٠	•		•	•	•		•				
	Distillation residues		٠		•					•		•				
Е	Enzyme test receptacles		٠	•	٠			•	•	•	٠	•	٠	•	٠	•
F	Fat residues		٠		•					•						
	Felt-tip pen		•	•	•			•	•	•	٠	•				
	Foam														•	
	Food industry		٠	•	٠	•		•	٠	•		•				
	Food waste		•	•				٠	٠	•	٠	٠				
G	Glass and porcelain equipment		٠	•					٠			•				
	Grease for joints		•						٠	•		٠				
Н	Heavy oils		٠						٠			•				
	Hydroxides	•											•	•		
L	Laboratory floors		•			•										
	Lenses for glasses			•												
Μ	Metal equipment			•				•			•					
	Mucus															•
Ν	Neutralization	•											•	•		
0	Nickel		•	•	•			•	•	•	•	•	•	•	•	•
0	Oil		•						•			•				

	Cleaning applications P-Z	Manual washing	Decalcification solution	Extran® MA 01	Extran® MA 02	Extran® MA 05	Sodium hydroxide solution	Machine washing	Extran® AP 11	Extran® AP 12	Extran® AP 13	Extran® AP 16	Extran® AP 17	Extran® AP 21	Extran® AP 22	Extran® AP 33	Extran® AP 41
Р	Petri dishes		•	٠	•	•			•	•	•	٠	•	٠	•	٠	•
	Phosphate analysis equipment					•											
	Pipettes				•				•			٠					
	Plaster residues		•	٠	•				•	•							
	Plastic equipment								•			٠					
	Precision equipment				٠												
	Protein residues			•		٠				٠			•				
	Proteins			•		•				•			•				
٥	Quartz equipment				٠												
R	Rubber				٠				•			•					
S	Saliva			٠		٠											•
	Silicones (oils, greases, resins)			٠						•	•		•				
	Stainless steel		•	٠	•	٠	•		•	٠	•	٠	•	٠	•	٠	•
Т	Thin film plates			•		•			•			•					
	Tiles in laboratory			٠			•										
	Tough residues						٠			•			•				
U	Ultrasound			٠	•	•											
W	Wax			٠		•											
Z	Zinc				•												

# Manual washing

The Extran<sup>®</sup> MA types for manual washing are universally applicable concentrates for the production of water baths which work reliably and residue-free.

#### General application advice

- Water is used to prepare the cleaning solution. If slight sedimentation of the hardener occurs, more Extran<sup>®</sup> must be added. De-mineralized water boosts the cleaning effect.
- For cleaning, the items to be cleaned are simply immersed completely in the solution.
- Once cleaning is finished, they are rinsed first with tap water and then with demineralized water.
- The baths can be used for a longer time without a noticeable decrease in the cleaning effect.
- If necessary, the rinsing liquid can be supplemented with fresh Extran<sup>®</sup>.
- The length of application is less than 2 hours.
- For »difficult cases« such as plaster, blood or heavy oil, the items to be cleaned are simply left in the bath a little longer.
- Heat speeds up the cleaning process.
- Extran<sup>®</sup> is also ideally suited to ultrasound cleaning.





Extran <sup>®</sup> MA 01 liquid, alk	aline	Content	Packaging	Ord. No.
Extran <sup>®</sup> MA 01 alkaline		1	Plastic bottle	1.07555.1000
		2.5 l	Plastic bottle	1.07555.2500
		5 I	Plastic bottle	1.07555.5000
		10 I	Plastic container	1.07555.9010
		25 l	Plastic container	1.07555.9025
Ingredients	Ionic and non-ionic surfactants, phosphates, excipient in low quantities			
Application advice	Universal cleaner for the removal of heavy contamination. In wiping tables,	iles, floors. I	n soaking for the	
	automated cleaning of laboratory equipment. Do not use on alkali-sensitive	materials su	ch as aluminium.	

	automated cleaning of laboratory equipment. Do not use on alkali-sensitive materials such as autominium.
Properties	Liquid   alkaline   chlorine-free   free from odorants / dyestuffs
Dosing	The ideal dosage depends on the hardness of the water and the level of contamination of the item
	to be cleaned.
	Recommended application concentrations
	For normal contamination: 2 %
	For heavier contamination: 5 %
	For very tough stains up to 20 %
pH value	pH = 11.6 [in a 2 % solution]
	pH = 12.0 [in a 5 % solution]
Accessories	Dosing feeder made from PP, 20–28 ml for 1   Extran® bottle 9.57571.1020

### The advantages of the 1 I bottle Extran® MA 01 for you

More convenient:	The 1 I bottle, onto which the separately available dosing unit can be easily mounted,
	is easy to handle and thus allows increased convenience when cleaning.
Cost effectiveness:	Modern dosing systems are a prerequisite for top cleaning results and optimum profitability.
	Precise and accurate dosing is the basis of cost effectiveness, made possible using the
	measuring unit tailored to the 1 l bottle.
Safety:	The dosing unit makes repeatable dosing and better control over the concentration possible.
	This ensures workplace safety and the health of staff.

Extran <sup>®</sup> MA 02 liquid,	neutral	Content	Packaging	Ord. No.
Extran <sup>®</sup> MA 02 neutra	l	2.5 l	Plastic bottle	1.07553.2500
		5 I	Plastic bottle	1.07553.5000
		10 I	Plastic container	1.07553.9010
		25 I	Plastic container	1.07553.9025
Ingredients	Ionic and non-ionic surfactants, phosphates, excipient in low quantities			
Application advice	Universal cleaner for the gentle cleaning of appliances made from alkali-ser	sitive metals	such as	
	aluminum, zinc and alloys with similar behavior. Suitable for metal applianc	es and precis	ion measuring	
	devices made from glass and quartz such as burettes, pipettes, cells, blood g	as analyzers	and other medical	
	equipment which is sensitive to aggressive detergents and also rarely has pr	oblematic co	ntamination.	
Properties	Liquid   neutral   chlorine-free   free from odorants / dyestuffs			
Dosing	The ideal dosing depends on the hardness of the water and the level of contain	mination of t	he item	
	to be cleaned.			
	Recommended application concentrations			
	For normal contamination: 2 %			
	For heavier contamination: 5 %			
pH value	pH = 7.5 [in a 5 % solution]			

Extran <sup>®</sup> MA 05 liquid,	, alkaline, phosphate-free	Content	Packaging	Ord. No.
Extran <sup>®</sup> MA 05 liquid,	, alkaline, phosphate-free concentrate	2.5 l	Plastic bottle	1.40000.2500
		5 I	Plastic bottle	1.40000.5000
		10 I	Plastic container	1.40000.9010
		25 I	Plastic container	1.40000.9025
Ingredients	Anionic and non-ionic surfactants, alkalescent additives, free of NTA (nitrilot	ri acetic acid	)	
Application advice	Universal cleaner for the removal of tough stains. Unlimited use also possible	with very ha	ard water.	
	Do not use on alkali-sensitive materials such as aluminum. Use is especially r	ecommended	d everywhere where	
	micro-phosphate tests are carried out.			
Properties	Liquid   alkaline   phosphate-free   chlorine-free   free from odorants / dyest	uffs   NTA-fr	ree	
Dosing	The ideal dosage depends on the hardness of the water and the level of conta	mination of	the item	
	to be cleaned.			
	Recommended application concentrations			
	For normal contamination: 2 %			
	For heavier contamination: 5 %			
	For very tough stains up to 20 %			
pH value	pH = 11.6 [in a 2 % solution]			
	pH = 12.0 [in a 5 % solution]			

# Automated cleaning

The various types of Extran<sup>®</sup> AP were created in cooperation with leading appliance manufacturers especially for use in laboratory washing machines and tested in these machines for suitability. As well as a distinctive cleaning power with extensive universal effects, the very low formation of foam is also an important property. The good solubility in water of all components minimizes residues on appliances which have been cleaned.

To neutralize displaces alkali residues and remove remaining traces of alkali, an acid rinser should be used after every main wash cycle.

#### All neutralizing agents are suitable.

- Extran<sup>®</sup> AP 21 acidic with phosphoric acid
- Extran<sup>®</sup> AP 22 acidic with citric acid





Extran <sup>®</sup> AP 11 powder,	mild alkaline	Content	Packaging	Ord. No.
Extran <sup>®</sup> AP 11 mildly a	Ikaline	2 kg	Plastic bottle	1.07558.2000
		10 kg	Bucket, plastic	1.07558.9010
		25 kg	Fibre carton	1.07558.9025
Ingredients	Phosphates, alkali salts			
Application advice	Universal cleaning agent for the gentle cleaning of alkali-sensitive items. Cle be tainted with allergenic detergents, such as jewelry, glasses. A corrosion inl intensive prevention of corrosion of glass and ceramics. Extran® AP 11 mild a heavy agitation of the solution in a washing machine.	nibitor is inclu	uded for the	ng
Properties	In powder form   mild alkaline   surfactant-free   chlorine-free   contains a co odorants / dyestuffs	prrosion inhib	itor   free from	
Dosing	The ideal dosage depends on the hardness of the water and the level of conta to be cleaned.	mination of t	he item	
	Recommended application concentrations			
	For normal conditions: 0.2 – 0.4 %, i.e. 20 – 40 g of Extran® AP 11 are used for	or around 10	l of water	
pH value	pH = 11.3 [in a 0.3 % solution when ready for use]			

Extran® AP 12 powder, a	alkaline	Content	Packaging	Ord. No.
Extran <sup>®</sup> AP 12 alkaline		2 kg	Plastic bottle	1.07563.2000
		10 kg	Bucket, plastic	1.07563.9010
		25 kg	Plastic drum	1.07563.9025
Ingredients	Phosphates, sodium hydroxide, alkali salts			
Application advice	Active universal cleaning agent for the main wash cycle, which cleans even h	eavily soiled i	items and removes	
	dried or burned-on residues. Particularly suitable for the removal of starch ar	d protein res	idues.	
	Extran® AP 12 alkaline does not foam even during heavy agitation of the solu	tion in a was	hing machine.	
Properties	In powder form   alkaline   surfactant-free   chlorine-free   free from odorant	s / dyestuffs		
	Extran® AP 12 is free from organic surfactants and emulsifiers, but contains of	complexing a	gents and can there	-
	fore be used in both soft and hard water.			
Dosing	The ideal dosage depends on the hardness of the water and the level of conta	mination of t	he item	
	to be cleaned.			
	Recommended application concentrations			
	For normal conditions: 0.2 – 0.4 %, i.e. 20 – 40 g of Extran® AP 12 are used for	or around 10	l of water	
pH value	pH = 12.3 [in a 0.3 % solution when ready for use]			

Extran <sup>®</sup> AP 13 powder,	alkaline with detergents	Content	Packaging	Ord. No.
Extran <sup>®</sup> AP 13 alkaline	with detergents	2 kg	Plastic bottle	1.07565.2000
		10 kg	Fibre carton	1.07565.9010
		25 kg	Plastic drum	1.07565.9025
Ingredients	Non-ionic surfactants, phosphates, sodium hydroxide, alkali salts			
Application advice	Intensive cleaning agent for the main wash cycle. Particularly effective again	st grease and	l oil deposits.	
	Other organic and inorganic residues are also removed.			
Properties	In powder form   alkaline   chlorine-free   contains a corrosion inhibitor   free	from odoran	ts / dyestuffs	
	Extran® AP 13 contains organic surfactants and emulsifiers and foams little.	The product	contains	
	complexing agents and can therefore be used even in hard water without fur	ther addition	s.	
Dosing	The ideal dosage depends on the hardness of the water and the level of conta	mination of t	the item	
	to be cleaned.			
	Recommended application concentrations			
	For normal conditions: 0.2 – 0.4 %, i.e. 20 – 40 g of Extran® AP 13 are used for	or around 10	l of water	
pH value	pH = 12.3 [in a ready-to-use solution]			

Extran <sup>®</sup> AP 16 liquid, m	ild alkaline	Content	Packaging	Ord. No.
Extran <sup>®</sup> AP 16 liquid, m	ild alkaline concentrate	2.5 l	Plastic bottle	1.40001.2500
		5 I	Plastic bottle	1.40001.5000
		10 l	Plastic container	1.40001.9010
		25 I	Plastic container	1.40001.9025
Ingredients	Complexing agents, alkali salts, free from NTA			
Application advice	Universal cleaning agent for the gentle cleaning of alkali-sensitive items. Cle	aning of item	ns which cannot be	
	tainted with allergenic detergents, e.g. jewelry, glasses.			
	Extran® AP 16 mild alkaline does not foam even during heavy agitation in a v	vashing mach	nine.	
Properties	Liquid   mild alkaline   phosphate-free   surfactant-free   chlorine-free   free	from odorant	s / dyestuffs   NTA-fr	ee
	Extran® AP 16 mild alkaline is a liquid main cleaning agent with mild alkaline	e properties f	or automatic dosage.	
Dosing	The ideal dosage depends on the hardness of the water and the level of conta	mination of	the item	
	to be cleaned.			
	Recommended application concentrations			
	For normal conditions: 0.3 – 0.5 %, i.e. 30 – 50 ml of Extran® AP 16 are used	for around 10	0 l of water	
pH value	pH = 11.2 [in a 0.3 – 0.5 % solution]			
Accessories	Adapter made from PP, for 10 I and 25 I Extran® cans			9.67212.0001

Extran <sup>®</sup> AP 17 liquid, a	alkaline	Content	Packaging	Ord. No.
Extran® AP 17 liquid, a	alkaline concentrate	2.5 l	Plastic bottle	1.40006.2500
		5 I	Plastic bottle	1.40006.5000
		10 I	Plastic container	1.40006.9010
Ingredients	Complexing agent, sodium hydroxide solution, free from NTA			
Application advice	Active universal cleaning agent for the main wash cycle which cleans and rer	noves even h	eavily soiled items.	
	Particularly suitable for the removal of starch and protein residues.			
	Extran® AP 17 alkaline does not foam even during heavy agitation in a washi	ng machine.		
Properties	Liquid   alkaline   phosphate-free   surfactant-free   chlorine-free   free from	odorants / dy	estuffs   NTA-free	
	Extran® AP 17 is free from organic surfactants and emulsifiers, but contains	complexing a	igents and can theref	ore
	be used in both soft and hard water.			
Dosing	The ideal dosage depends on the hardness of the water and the level of conta	mination of	the item	
	to be cleaned.			
	Recommended application concentrations			
	For normal conditions: 0.3 – 0.5 %, i.e. 30 – 50 ml of Extran® AP 17 are used	in around 10	l of water	
pH value	pH = 12.2 [in a 0.3 solution]			
Accessories	Adapter made from PP, for 10 I and 25 I Extran <sup>®</sup> cans			9.67212.0001

Extran® AP 21 acidic with phosphoric acid     2.5 I     Plastic bottle       10 I     Plastic container	Ord. No.	Content Packaging	ran® AP 21 liquid, acidic with phosphoric acid
10 I Plastic container	1.07559.2500	2.5 I Plastic bottle	ran® AP 21 acidic with phosphoric acid
	1.07559.9010	10 I Plastic container	
25   Plastic container	1.07559.9025	25 I Plastic container	

Ingredients	Phosphoric acid				
Application advice	The acid special cleaner can be used both as a pre-wash agent and a rinsing agent with				
	a neutralizing effect.				
	Pre-wash agent: When used as a pre-wash agent, it primarily dissolves carbonates and hydroxides from				
	the residues. Protein substances and organic bases, such as amines, are often removed better in an acidic				
	pre-wash as in an alkaline main wash cycle.				
	Rinsing agent: As a rinsing agent, i.e. after the alkaline main wash cycle, it is especially suitable				
	for removing remaining traces of alkali on the cleaned material or, in the case of solution carry-over,				
	for neutralization. This acidic cleaning agent is also well suited to the removal of calcareous deposits in				
	the washing machine.				
Properties	Liquid   acidic   surfactant-free   chlorine-free   free from odorants / dyestuffs				
	Extran® AP 21 is an acidic pre-wash and neutralization agent with a phosphoric acid base.				
Dosing	Added automatically using a dosing device or manually.				
	Recommended application concentrations				
	Around 0.1 – 0.3 %, i.e. 10 – 30 ml of Extran® AP 21 are added to around 10 l of water				
pH value	pH = 2.0 [in a ready-to-use solution]				
Accessories	Adapter made from PP, for 10 I and 25 I Extran <sup>®</sup> cans	9.67212.0001			

Extran <sup>®</sup> AP 22 liquid,	acidic with citric acid	Content	Packaging	Ord. No.	
Extran® AP 22 acidic with citric acid		2.5 l	Plastic bottle	1.07561.2500	
		10 I	Plastic container	1.07561.9010	
		25 I	Plastic container	1.07561.9025	
Ingredients	Citric acid, non-ionic surfactants, low levels of excipient, phosphate-free				
Application advice	The acidic special cleaner can be used both as a pre-wash agent and a ri with a neutralizing effect.	nsing agent			
	<b>Pre-wash agent:</b> When used as a pre-wash agent, it primarily dissolves hydrophysical protein substances and organic bases, such as amines, are often removed be alkaline main wash cycle.				
	<b>Rinsing agent:</b> As a rinsing agent, i.e. after the alkaline main wash cycle, it is for removing remaining traces of alkali on the cleaned material or, in the cas for neutralization. This acidic cleaning agent is also well suited to the remov washing machine. The product is recommended for cases in which gentle con for particular reasons. Particularly suitable for the gentle removal of calcare sensitive metal and glass surfaces.	e of solution al of calcareo nditions must	carry-over, us deposits in the be maintained		
Properties	Extran® AP 22 is an acidic pre-wash and neutralization agent with a citric ad	cid base.			
Dosing	Added automatically using a dosing device or manually.				
	Recommended application concentrations				
	Around 0.1 – 0.3 %, i.e. 10 – 30 ml Extran® AP 22 are used for around 10 l of water				
pH value	pH = 3.0 [in a ready-to-use solution]				
Accessories	Adapter made from PP, for 10 I and 25 I Extran <sup>®</sup> cans			9.67212.0001	

Extran® AP 33 Defoamer		Content	Packaging	Ord. No.
Extran <sup>®</sup> AP 33 liquid, anti	-foaming agent	2.5 l	Plastic bottle	1.40007.2500
Ingredients	ents Inorganic polymers, low levels of excipient, contains silicon, produced without added formaldehyde			
Application advice	If the residues to be removed foam significantly themselves, the development of foam is prevented by			
	adding this special defoamer. Strong foamers include all kinds of emulsifiers, e.g. soaps, which sometimes only			
	develop during the wash cycle due to the saponification of fats, and numerous protein stains.			
Dosing	0.5 – 3 ml per 10 l wash cycle			

Extran® AP 41 powder, enzymatic		Content	Packaging	Ord. No.	
Extran® AP 41 enzymatic		2 kg	Plastic bottle	1.07570.2000	
		25 kg	Plastic drum	1.07570.9025	
Ingredients	Enzymes, phosphates, alkali salts				
Application advice	plication advice Alkaline cleaning agent for use in washing machines. Especially for the removal of dried tissue and saliva				
	residues, of mucus, protein and blood, in catheters, breathing tubes, breathing bags etc. Ideal conditions for				
	cleaning are between 55 and 65°C, since the enzymes do not work above 70°C.				
	We recommend Extran® AP 22 acidic with citric acid as an acidic rinsing agent.				
Dosing	Recommended application concentrations				
	0.3 %, i.e. 30 g of Extran <sup>®</sup> AP 41 are used for each 10 l wash cycle				
pH value	pH = 11.4 [in a ready-to-use solution]				



# Ordering information General cleaning applications

Chromosulfuric acid		Content	Packaging	Ord. No.
Chromosulfuric acid for cleaning glass vessels		1	Glass bottle	1.02499.1000
		2.5 l	Glass bottle	1.02499.2500
General information	Chromosulfuric acid is an excellent cleaning agent for tough cases, for exam with carcinogenic substances. Carcinogenic residues can be oxidative destro chromosulfuric acid. The effect is based on the chromium(VI) oxide CrO <sub>3</sub> , a v agent. During the oxidation process, the red-brown chromium(VI) oxide is re of chromium. The depletion level can thus be assessed from the change in co testing: fresh chromosulfuric acid is red-brown, used is green in color.	ative destroyed by treatment with de CrO <sub>3</sub> , a very strong oxidation ) oxide is reduced to the green trivalent state change in color without further		
Safety advice	Extreme care must be taken when working with chromosulfuric acid due to oxidizing properties and the possibility of the formation of poisonous chrom amount of heat generated when mixed with water, chromosulfuric acid must by adding water (strongly corrosive splashes!). If dilution is necessary, this ca acid to water while stirring. The equally very poisonous chromium(VI) oxide is formed when chlorides are present in the residues to be removed. For all t procedures using chromosulfuric acid should only be undertaken in a well-w protective clothing, impermeable gloves and protective goggles are to be wo are printed on the label of every pack.	ium(VI) vapo t never be di an only be do chloride (chr hese reasons entilated are	or. Due to the large luted one by adding the omylchloride) 5, cleaning a. Furthermore,	
Removal of residues	Chromium solutions must be treated as special waste and their disposal left t Neutralize any spilled acid with sodium hydrogen carbonate or lime sand. New textiles or sawdust.		·	

Decalcification solution	citric acid base around 19 %	Content	Packaging	Ord. No.	
Decalcification solution	base: citric acid about 19 %	1	Plastic bottle	1.00240.1000	
Ingredients	Citric acid, other organic acids in small quantities				
Application advice	The product is recommended for cleaning instances in which particularly gentle conditions have				
	to be maintained. Particularly suitable for the gentle removal of calcareous deposits, for example on taps				
	or sensitive metal and glass surfaces. The decalcification solution is made from pharmacopoeic raw				
	materials and thus satisfies the highest quality standards.				
Properties	Liquid   acidic   phosphate-free				
Dosage	The concentration for application is around 1 – 5 %, i.e. 100 – 500 ml of decalcification solution is added to				
	around 10 l of water. The decalcification process can be accelerated by applying a little heat. Do not use on				
	corrosive materials.				

Sodium hydroxide solution		Content	Packaging	Ord. No.
Sodium hydroxide solution about 32 % extra pure		2.5 l	Plastic bottle	1.05587.2500
		5 I	Plastic bottle	1.05587.5000
		25 I	Plastic container	1.05587.9025
		200 l	Plastic barrel	1.05587.9200
Ingredients	Sodium hydroxide			
Application advice	Basic cleaning agent for cleaning applications in which residues from surfact	Basic cleaning agent for cleaning applications in which residues from surfactants or complexation agents are to be avoided. Through the use of these prepared solutions, the time-consuming and dangerous		
	agents are to be avoided. Through the use of these prepared solutions, the tin			
	breakdown of solid sodium hydroxide can be avoided.			
Properties	Liquid   strongly alkaline   chlorine-free			

### Ordering information Accessories

Adapter for canister	Content	Packaging	Ord. No.
Adapter composed of PP for 10 I and 25 I Extran <sup>®</sup> canister		Plastic bag	9.67212.0001

#### One for all: The universal adapter

Larger packing units, such as the 10 l and 25 l cans, are used for cleaning equipment. In this, it is important for workplace safety that the cans are connected tightly to the machines so that no spraying can occur. Around the world, various types of cleaning apparatus are used with very individual connection systems. Merck Millipore has developed a universal adapter especially for this. With its help, different types of machine can be safely connected to the detergent containers. This avoids detergent being lost, while simultaneously enabling reliable cleaning.

#### The advantages of the universal adapter for 10 I and 25 I cans for you

Safety:	If the can is connected tightly to the machine, spillage is avoided and the health and personal safety
	of the staff is thus protected.
Reliability:	External contamination (from the air) can be avoided, thus preventing analytical results from being
	corrupted or influenced and ensuring reliable and exact results.

Dosing aid	Content	Packaging	Ord. No.
Dosing aid made of PP, 20 ml, 28 mm, natural for Extran®, AWH	1 piece	Plastic bag	9.57571.1020

#### It all comes down to the dosage

For cleaning to be effective and reliable, the detergent must be dosed precisely. This also ensures that the agent is used economically: too little cleans insufficiently, too much leaves residues. That is why Merck Millipore offers 1 l bottles with dosing aids, which ensure efficient dosing and are safe to handle when cleaning manually. The dosing aid can also be ordered separately if required and can be reused again and again.



# Service

New possibilities and information tools for further dimensions.

Discover the multimedia world of Merck Millipore Chemicals and enjoy the benefits of a service that is focused on you - our valued customer! In addition to the information provided in this catalog, we also publish a wealth of online and print media regarding our inorganic reagents. Our specialized product brochures give details of our individual product groups, their applications and your advantages in using them.

Please visit our www.merckmillipore.com/inorganic-reagents website, where you will find detailed information about our products as well as useful tools to simplify your routine work.









# Print media

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# Periodic system

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# Online media

### www.merckmillipore.com/inorganic-reagents

You can explore our broad product range by visiting our website. Our individual inorganic reagents are classified under the following three principal subject categories:

- Classic inorganic analysis
- Instrumental inorganic analysis
- Safety products and general applications



## Quick links Try out our Quick links! These take you directly to individual product groups of interest and to videos and useful tools in the Service center.

Videos Featuring various products for inorganic analysis, our fascinating videos cover potential applications, filling and production processes, and much more! Discover more about Merck Millipore salts, acids, reference materials and products for titration by watching our quick videos!

#### Service center Our Service center offers you a selection of extremely practical tools.

Website guide will help you navigate through our web pages quickly and easily.

Analytical application finder presents you a list of applications from which you can choose the most appropriate substance or substance group covering your particular needs.

pH-Indicator-Selector displays a table of pH indicators and their ranges so you can make a quick, informed decision about which one to use for your specific analysis.

Concentration density finder tells you the precise density and molarity at 20°C of any solution selected from the dropdown menu.

LabTools booklet and periodic table of the elements are tools that support you in your laboratory work. The LabTools booklet provides useful advice on how to handle and work with chemicals in the laboratory and is a useful source of information on preventing application errors.

Access to detailed information, as our Merck Millipore Safety Data Sheets as well as Regulations & Requirements information.

Online form to contact us directly, if you have any questions!

# Online media

## www.merckmillipore.com/inorganic-reagents

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Labmail	Labmail contains information about new developments and trends, and features analytical tests, chemicals and reagents that are important for quality control, in-process control and R&D. www.merckmillipore.com/labmail
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Apps Have you already seen our »Periodic table of the elements« app? Why not download it to your iPhone, iPad or iPod touch\*? Merck Millipore's state-of-the-art »Periodic table of elements« app tells you everything you need to know about the chemical elements – in a user-friendly, pocket-sized format. www.merckmillipore.com/apps

#### More information

www.merckmillipore.com/apps www.merckmillipore.com/periodic-table

## Print media

To help you use our products to your advantage, we offer a selection of brochures that cover specific product groups, their benefits, special usage advice and details of our services. Precise instructions on using the individual products are given to simplify your laboratory work, help avoid application errors and save you time and money – all leading to greater efficiency. On the following page is a selection of special product brochures on our inorganic reagents together with their catalog numbers.



### Classic inorganic analysis

- Acids for analysis EMSURE® | W.281145
- Inorganic salts | W.281127
- Just in Case | W.281115
- Made by Merck Millipore Caustic alkalis and alkaline solutions of high purityand defined quality | W.281122
- New Quality in Pharmacopoeia Analysis High audit rates require reliable solutions | W.281130

#### Instrumental inorganic analysis

- Apura® Water determination according to Karl Fischer | W.283125
- Certipur® Not all reference materials are the same | W.283120
- The Magic Box Titripac® for volumetric solutions and buffer solutions | W.283143
- Titripur® The measure of all things Titration Guide for titer determination | W.281136
- Titripur® The measure of all things Volumetric Solutions from Merck Millipore | W.281144

#### Product safety and general applications

- Drying Agents | W.283123
- Extran® detergents The perfect solution for cleaning your laboratory utensils | W.283119
- Chemizorb<sup>®</sup> Absorbents for spilled liquids | W.283137

These and other brochures can be ordered from your regional Merck Millipore office and are also available online as PDF files, which you can download from the relevant product page. Please contact your local Merck Millipore representative or visit www.merckmillipore.com/inorganic-reagents

#### Magazines

Our »lab mail« and »phar mail« magazines keep you up to date with the latest developments and trends in the industry. You can either subscribe to them free of charge by contacting the Merck Millipore Group in your area, or enjoy the electronic version online.

More information

www.merckmillipore.com/inorganic-reagents

# Glossary

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#### Inorganic Reagents for laboratory use

- Classic inorganic analysis
- Instrumental inorganic analysis
- Safety products and general applications



# Periodic table of the elements

The periodic table of the elements is a tool that support you in your laboratory work. You can have a look at the pericodic table on the next page or visit our website to discover the features of the interactive periodic table of the elements. Merck Millipore's innovative information tools are tailored to meet the interests of the new chemist generation. Try it out yourself!

## Legend of the periodic table



- 1 Atomic number
- 2 Element symbol
- 3 Relative atomic mass [g/mol]
- \* most stable isotope
- 4 Melting point [°C]
- **5** Boiling point [°C]
- 6 Electronegativity [Allred, Rochow]
- 7 Oxidation states
- 8 Electron configuration

- Nonmetals
- Halogens
- Inert gases
- Alkaline-earth metals
- 📕 Alkali metals
- Transition metals
- Lanthanides
- Actinides
- Other metals
- Semi metals

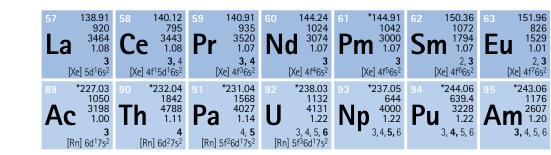


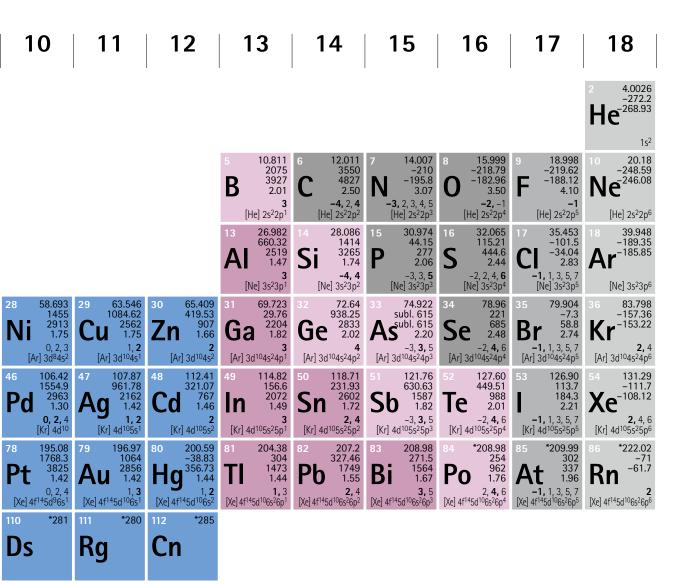
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1	1 1.0079 -259.14 -252.87 2.20 -1, 1 1s1								
2	3 6.941 180.54 1342 0.97 1 [He] 2s <sup>1</sup>	4 9.0122 1287 2469 1.47 2 [He] 2s <sup>2</sup>							
3	11 22.990 97.72 Na 101 [Ne] 3s1	12 24.305 650 Mg 1090 1.23 2 [Ne] 35 <sup>2</sup>							
4	19 39.098 63.38 759 0.91 [Ar] 4s <sup>1</sup>	20 40.078 842 1484 1.04 2 [Ar] 4s <sup>2</sup>	21 44.956 1541 2836 1.20 3 [Ar] 3d <sup>1</sup> 4s <sup>2</sup>	22 47.867 1668 3287 1.32 3,4 [Ar] 3d <sup>2</sup> 4s <sup>2</sup>	23 50.942 1910 3407 1.45 0, 2, 3, 4, 5 [Ar] 3d <sup>34</sup> 5 <sup>2</sup>	24 51.996 1907 <b>Cr</b> 2671 1.56 0, 2, <b>3</b> , 6 [Ar] 3d <sup>5</sup> 4s <sup>1</sup>	25 54.938 1246 Mn 2061 1.60 -1, 0, 2, 3, 4, 6, 7 [Ar] 3d <sup>5</sup> 4s <sup>2</sup>	26 55.845 1538 <b>Fe</b> 2861 1.64 -2, 0, 2, <b>3</b> , 6 [Ar] 3d <sup>6</sup> 4s <sup>2</sup>	<b>27</b> 58.933 1495 <b>2927</b> 1.70 -1, 0, 2, 3 [At] 3d <sup>7</sup> 4s <sup>2</sup>
5	37 85.468 39.31 <b>Rb</b> 688 0.89 1 [Kr] 5s <sup>1</sup>	38 87.62 777 1382 0.99 2 [Kr] 5s <sup>2</sup>	<b>39</b> 88.906 <b>Y</b> 3336 1.11 <b>3</b> [Kr] 4d <sup>1</sup> 5s <sup>2</sup>	40 91.224 1855 4409 1.22 4 [Kr] 4d <sup>2</sup> 5s <sup>2</sup>	41 92.906 2477 <b>Nb</b> 4744 1.23 3, <b>5</b> [Kr] 4d <sup>4</sup> 5s <sup>1</sup>	42 95.94 2623 <b>Mo</b> 1.30 0, 2, 3, 4, 5, <b>6</b> [Kr] 4d <sup>5</sup> 5s <sup>1</sup>	43 *97.907 2157 4265 1.36 7 [Kr] 4d <sup>5</sup> 5s <sup>2</sup>	44 101.07 2334 4150 1.42 -2, 0, 2, 3, 4, 6, 8 [Kr] 4d <sup>7</sup> 5s <sup>1</sup>	45 102.91 1964 3695 1.45 0, 1, 2, 3, 4, 5 [Kr] 4d <sup>8</sup> 5s <sup>1</sup>
6	55 132.91 28.44 671 0.86 1 [Xe] 6s <sup>1</sup>	56 137.33 727 Ba 1897 0.97 2 [Xe] 6s <sup>2</sup>	57–71 Lanthanides	72 178.49 2233 4603 1.23 4 [Xe] 4f <sup>14</sup> 5d <sup>2</sup> 6s <sup>2</sup>	73 180.95 3017 <b>Ta</b> 5458 1.33 5 [Xe] 4f <sup>14</sup> 5d <sup>3</sup> 6s <sup>2</sup>	74 183.84 3422 5555 1.40 0, 2, 3, 4, 5, <b>6</b> [Xe] 4f <sup>14</sup> 5d <sup>4</sup> 6s <sup>2</sup>	$\substack{\textbf{75} \\ \textbf{86.21} \\ \textbf{3186} \\ \textbf{5596} \\ \textbf{1.46} \\ \textbf{0, 2, 4, 6, 7} \\ \textbf{[Xe] 4f^{14}5d^56s^2} }$	$\begin{matrix} \textbf{76} & 190.23 \\ 3033 \\ \textbf{Os} & 5012 \\ 1.52 \\ -2, 0, 2, 3, \textbf{4}, 6, 8 \\ [Xe] 4f^{14} 5d^6 6s^2 \end{matrix}$	77 192.22 2446 4428 1.55 -1, 0, 1, 2, 3, 4, 6 [Xe] 4f <sup>14</sup> 5d <sup>7</sup> 6s <sup>2</sup>
7	87 *223.02 27 677 0.86 1 [Rn] 7s <sup>1</sup>	88 *226.03 700 Ra <sup>707</sup> 0.97 2 [Rn] 7s <sup>2</sup>	89–103 Actinides	104 *261.11 <b>Rf</b>	105 *262.11 <b>Db</b>	106 *266.12 Sg	107 *264.12 Bh	108 *277 Hs	109 *268.14 Mt

Lanthanides

## Actinides





64 157.25 1312 Gd 3273 1.11 3 [Xe] 4f <sup>7</sup> 5d <sup>1</sup> 6s <sup>2</sup>	65 158.93 1356 3230 1.10 3, 4 [Xe] 4f <sup>9</sup> 6s <sup>2</sup>		67 164.93 1461 2720 1.10 3 [Xe] 4f <sup>11</sup> 6s <sup>2</sup>	68 167.26 1529 <b>Er</b> 2868 1.11 3 [Xe] 4f <sup>12</sup> 6s <sup>2</sup>	69 168.93 1545 1950 1.11 2, 3 [Xe] 4f <sup>13</sup> 6s <sup>2</sup>	70 173.04 824 1196 1.06 2, 3 [Xe] 4f <sup>14</sup> 6s <sup>2</sup>	71 174.97 1652 1402 1.14 [Xe] 4f <sup>14</sup> 5d <sup>1</sup> 6s <sup>2</sup>
96 *247.07 1340 3110 1.20 3, 4	97 *247.07 1050 <b>Bk</b> 1.20 3, 4	98 *251.08 900 Cf 1.20 3, 4	es *252.08 860 1.20 3	100 *257.1 1527 Fm 1.20 3	101 *258.10 827 Md 1.20 3	102 *259.10 827 No 1.20 2, 3	103 *262.11 1627 Lr 3

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